



SIMULTANEOUS AND ACCURATE DETERMINATION OF VITAMINS B₁, B₆, B₁₂ AND ALPHA-LIPOIC ACID IN MULTIVITAMIN CAPSULE BY REVERSE-PHASE HIGH PERFORMANCE LIQUID CHROMATOGRAPHIC METHOD

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ABSTRACT

A simple, precise, rapid and selective reverse phase high performance liquid chromatographic (RP-HPLC) method has been developed and validated for the simultaneous determination of Benfotiamine (B₁) 100 mg, Pyridoxine hydrochloride (B₆) 100 mg, Mecobalamin (B₁₂) 1000 mcg and Alpha-lipoic acid 100 mg in multivitamin capsules. The method uses X-Terra reverse phase (RP-18, 250 x 4.6 mm, 5 µm) column and gradient elution. The aqueous mobile phase contained 0.05 M phosphate buffer adjusted to pH 2.5 and acetonitrile. Separation and quantification was achieved by changing the proportion of the system linearly with a time-schedule programme. Detection was carried out in the range of 200 to 600 nm using photodiode array detector and set at 320 nm and further analysis was carried out using a UV detector. This method has been validated and found to be applicable in routine analysis for multivitamin capsules. The precision is exemplified by relative standard deviations of 1.1% for benfotiamine, 0.9% for pyridoxine hydrochloride, 0.7% for mecobalamin and 1.2% for Alpha-lipoic acid. Good linearity was observed between the concentration of the analytes and peak area with correlation coefficients (R²) of 0.9997, 0.9990, 0.9995, and 0.9998 respectively. Mean recoveries obtained during spiking experiments were in the range of 99.8-100.2.

Keywords: RP-HPLC, Vitamin B₁, B₆, B₁₂ and Alpha-lipoic acid, UV detection, Validation.

INTRODUCTION

Vitamins are an essential group of food ingredients which have to be supplied in sufficient amounts with diet, vitamins are broad group of organic compounds that are minor, but essential constituents of food required for normal growth, self-maintenance and functioning of human and animal bodies. Mainly B complex vitamins play different specific and vital functions in metabolism and their lack or excess produces specific diseases. The B abitaminosis are usually attributed to nutritional deficiency, competition between bacteria, disturbances of the intestinal flora and excessive exposure to sunrays. Multivitamin capsules regularly used in therapeutic management, in improper dietary feeding and for strengthening purpose. This has stimulated research on accurate and efficient analytical methods for the determination of vitamins which is problematic because of their instability and the complexity of the matrices in which they are usually analysed¹⁻³.

The standard⁴ and official analytical methods⁵, which are tedious, sometimes non-specific and time consuming, involve pre-treatment of the sample through complex chemical, physical, or biological reactions to eliminate the interferences commonly found, followed by individual methods for each different vitamin. These methods include spectrophotometric, polarographic, fluorimetric, enzymatic, and microbiological procedures.

For the determination of the compounds in single and multi-component mixtures the number of papers is more restricted. They include, among others, the determination of B₁ and B₆ in pharmaceutical preparation by ratio spectra derivative spectrophotometry and HPLC⁶, electrothermal atomic absorption spectrometric methods⁷, and capillary electrophoresis⁸⁻¹⁰. Among them the techniques most widely used are chromatographic methods and in particular, HPLC which provides rapid, sensitive and accurate methods for vitamin determination and have the advantages of solvent economy, easy coupling with other techniques, and that small amounts of sample required for analysis.

Individual vitamins can be chromatographed isocratically as well as certain combinations of two or three vitamins; the simultaneous chromatography of more complicated mixtures may require a gradient elution program. Determinations can be carried out, among others by normal-phase¹¹, ion pairing¹², chromatography, liquid-phase chromatography with narrow bore column¹³, HPLC after

solid-phase extraction¹⁴⁻¹⁵, counter current¹⁶, reverse phase¹⁷⁻²⁰ being the most common method. Most methods are successful in single component preparations. Other chromatographic systems separate B₁, B₆, and B₁₂ when they exist in the same range²¹ but hindered when the amount of B₁ and B₆ exceeds by a hundred or even thousand times the amount of B₁₂ present in the complex. Most of the published methods involve the use of several detection methods that includes electrochemical²² or UV-Visible absorbance with a single or variable wavelength or photodiode array²³. The proposed method is very simple, rapid, and validated according to ICH guidelines.

In the present work, a simultaneous determination of B-complex vitamins (B₁, B₆, and B₁₂) and Alpha-lipoic acid in multivitamin capsule is proposed using a simple gradient program by setting the wavelength at 320 nm. Also the chromatographic system developed clearly separates the four analytes from their degradation products and this makes the presented method advantageous with respect to the existing literature.

MATERIALS AND METHODS

Instrument

A Waters Alliance series 2690 HPLC separation module equipped with two reciprocating pumps, a Waters series auto-sampler with an effective volume 100 µl and a Waters series 996 photo diode array detector with a dual beam of variable wavelength system was used for scanning the sample. A Waters Alliance series 2690 HPLC equipped with two reciprocating pumps, a Waters series auto-sampler with an effective volume 100 µl and a Waters series 2487 spectrophotometric detector with a dual beam variable wavelength system was used for regular analysis of samples. The chromatographic peaks were recorded and elaborated automatically by employing computerized Waters program 'Millennium 32'.

Reagents and materials

All the chemicals and solvents were HPLC grade and were employed as supplied by the manufacturers without further purification was used in all procedures. Highly purified water obtained through a Milli-Q water purification system (Millipore-Q Elix 3 model, USA), acetonitrile was 'gradient grade for liquid chromatography' purchased from Ranbaxy Chemical Company and monobasic potassium phosphate was HPLC grade purchased from Merck

Company. Highly purified working standards Benfotiamine (B₁ with purity 97.8%)-Sami Labs, Pyridoxine Hydrochloride (B₆ with purity 98.8%)-Madras Pharmaceuticals, Mecobalamin (B₁₂ with purity 92.6%)-Biocon Ltd, and Alpha-lipoic acid (purity 99.6%)-Biocon Ltd, The analyzed samples (Benfage Fort Capsules) were placebo capsules containing 100, 100, 1 and 100 mg of B₁, B₆, B₁₂ and Alpha-lipoic acid per capsule respectively.

Preparation of mobile phase

The gradient elution system started with a aqueous 0.05 M monobasic potassium phosphate buffer adjusted to pH 2.5 with dilute ortho phosphoric acid solution (Solvent A), for 5 minutes. Then it was changed linearly for another 15 minutes to a binary

mobile phase of Solvent A: Solvent B acetonitrile (30:70 v/v), afterwards the initial system was returned linearly for five minutes. The mobile phase was degassed by filtering through a Milli-Q 0.45 µm pore membrane filter.

These conditions were chosen in terms of peak shape, column efficiency, chromatographic analysis time, selectivity and resolution. Several gradient programmes were tried to separate all the four analytes. The gradient started with high % of buffer content to elute out the very polar pyridoxine hydrochloride, followed by a gradual increase of organic modifier to elute other low polar components. After several attempts the following gradient elution program for proposed method is illustrated in Table 1.

Table 1: Gradient elution programme for the simultaneous determination of vitamins B₁, B₆, B₁₂ and Alpha-lipoic acid

Time (min)	Solvent A%	Solvent B%	Gradient type
0	100	0	Linear
5	100	0	
20	30	70	
25	100	0	
Post time 5 min			
		Total run time 25 min	

Chromatographic conditions

A Reverse phase X-Terra column RP-18 was a 250X4.6 mm id., 5 µm particle size and end capped to minimize unreacted silanol effects. After sample preparation immediately transferred in an amber vial and kept at stable temperature, 20°C. The wavelength was set at 320 nm to separate all the four analytes in a single injection. The flow rate was 1.5 ml/min and the volume injected was 20 µl.

PROCEDURE

Preparation of standard solutions

(Diluents: Mixture of pH 2.5 phosphate buffer and acetonitrile in the ratio of 50:50 (v/v) was prepared and used as diluent).

The same proportion of all the vitamins as those in pharmaceutical formulation investigated, stock and standard solutions of all the compounds were prepared every day freshly in diluent respectively, by accurately weighing and dissolving all vitamins. A separate stock solution of vitamin B₁₂ was prepared because of the difference between the level of this vitamin and the others in the pharmaceutical formulation.

Vitamin B₁₂ standard stock solution

About 10 mg of Mecobalamin (B₁₂) working standard was weighed accurately and transferred into a 50 ml amber volumetric flask, dissolved and diluted up to the mark with diluent.

Mixed standard solution

About 100 mg each of Benfotiamine (B₁), Pyridoxine hydrochloride (B₆) and Alpha-lipoic acid working standard was weighed accurately and transferred into 100 ml amber colored volumetric flask. A portion of diluent and 5.0 ml of Mecobalamin (B₁₂) stock solution was added and sonicated for 5 minutes. The volume was made upto 100 ml with diluent and the solution was filtered through 0.45 µm membrane filter.

Sample preparation

Not less than 20 capsules were weighed accurately to determine the average fill weight. The capsule contents were emptied from the shell and mixed well. The shell weight was taken to deduce the average weight of each capsule. A portion of powder equivalent to average fill weight of one capsule was quantitatively transferred into 100 ml amber colored volumetric flask. About 80 ml of diluent was added and sonicated for 15 minutes and the solution was diluted to volume with diluent, mixed well and left to precipitate. The sample solution was then filtered through 0.45 µm membrane filter.

RESULTS AND DISCUSSION

Method development

During method development it became obvious that different work up procedures have to be used for determination of four components due to differences in solubility and stability. So diluent 0.05 M monobasic potassium phosphate buffer adjusted to pH 2.5 and acetonitrile in the ratio 50:50 was selected for dissolving all the four components. A solution of each single vitamin was injected separately into the chromatograph in order to determine the degree of resolution of its components.

The selectivity of a chromatographic separation is described by the combined affinities that the mobile phase and stationary phase exert on the sample components. A reliable chromatographic assay also requires acceptable resolution, reasonable retention times and good peak symmetry. As a consequence in the preliminary studies the optimum chromatographic conditions were investigated in isocratic and gradient elution systems. Similarly different proportions of acetonitrile in the mobile phase exerted a striking change in the K' values of mecobalamin whereas the peak behaviour of Pyridoxine HCl, Benfotiamine and Alpha-lipoic acid remain unaffected.

The profile of a gradient elution system affects the retention of solutes in a similar way to the proportions of the solvent with greater elution strength in a binary mobile phase under isocratic conditions. Under isocratic conditions, the bandwidths regularly increase as the retention time increases for all sample compounds. In contrast an advantage of gradient elution is that the bandwidths can be approximately constant both for early and later elution analytes. Therefore a binary gradient elution system was developed with a beneficial effect on the sensitivity of mecobalamin, which exists 1000-fold less in the B-complex and a single wavelength was fixed to detect all the four components.

A standard mixture of vitamins was then analyzed as a mixture and chromatographic conditions were optimized to maximize peak resolution. Elution of the vitamins from the column was in order of decreasing polarity of the mobile phase used: as the proportion of acetonitrile increased the retention times for the studied vitamins decreased and some of them eluted with solvent front. The greater the proportion of proportions of phosphate buffer, the better peak shapes and resolution of the studied vitamins.

The separation pattern of the vitamins B₁ showed at t_R-10.8 min, B₆ showed at-2.3 min, B₁₂ showed at-12.2 min and Alpha-lipoic acid at-17.0 min. The retention time of the B-complex vitamin was reproducible over a period of month, which makes the method most suitable for routine analysis. The peak asymmetry (T) and theoretical plate counts of all the four components were very close to the ideal value 1.2 and not less than 2000 respectively. The

resolution factors R_s , between the chromatographic peaks, were calculated from the equation $R_s = 2(t_2 - t_1)/(W_1 + W_2)$, where t_2 , t_1 are the retention time of the two components and W_1 , W_2 are the peak width at the base of the two respective peaks, obtained by

extrapolation of the relatively straight sides to the baseline. A representative chromatogram together with other observed system suitability chromatographic parameters are illustrated in Fig.1 and Table 2.

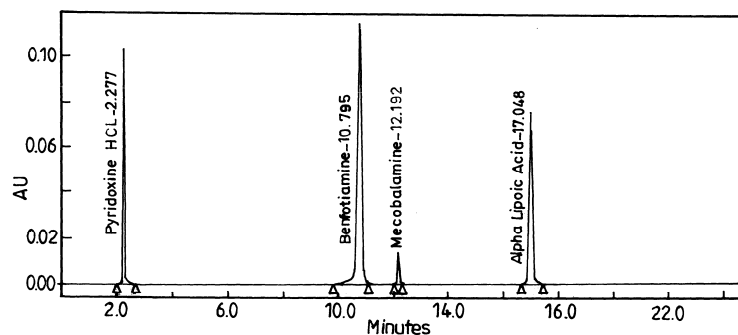


Fig. 1: Typical standard chromatogram for all the four components

Table 2: High performance liquid chromatographic system suitability parameters of the drug separation using the gradient elution system

Active ingredients	System suitability parameters			
	Retention time t_R (min)	Tailing factor	Theoretical plate count	Resolution factor R_f
Benfotiamine	10.79	0.8	35270	47.8
Pyridoxine HCl	2.27	1.1	9998	1.0
Mecobalamin	12.19	1.2	105347	6.7
Alpha-lipoic acid	17.04	1.1	221827	32.3

Mecobalamin exists in small concentration compared to the other vitamins and absorbs poorly above 230 nm; it was not visualized. This particular problem was overcome by working at 320 nm where all the four components showed their maximum absorbance.

In order to test the applicability of this procedure to a commercial vitamin formulation, 'Benfage Forte' was chromatographed and it was shown in Fig. 2. Integration of separated peak area was done

with the integrator or the computer programme and each vitamin was determined by using the peak area-concentration relationship obtained in the standardization step. The sample peaks were identified by comparing both the relative retention times and the UV-Vis spectrum of each one with those of the standard reference vitamins. The protocol affords reproducible quantification of the different vitamins with error less than 10%, which is the standard level in any pharmaceutical quality control.

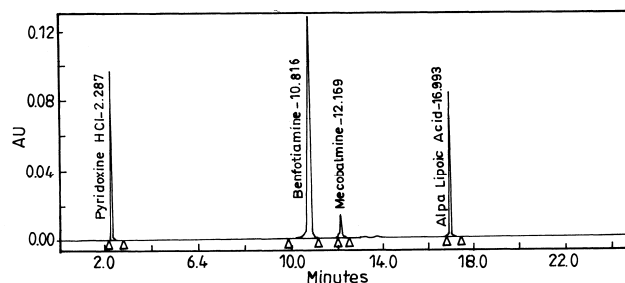


Fig. 2: Typical sample chromatogram for the separation of four components from capsule

Method validation

Validation of the analytical method is the process that establishes by laboratory studies in which the performance characteristics of the method meet the requirements for the intended analytical application. Methods need to be validated or revalidated before their introduction into routine life. The validity of a specific method should be demonstrated in laboratory experiments using samples or standards that are similar to the unknown samples analyzed in the routine. The preparation and execution should follow a validation protocol. This method has been validated with respect to precision, linear range, accuracy, specificity and robustness.

Precision

System precision

Six replicate standard solutions of Benfotiamine, Pyridoxine Hydrochloride, Mecobalamin and Alpha-lipoic acid were injected

into the HPLC system. The area response along with % relative standard deviation (RSD) is tabulated in Table 3a and the % RSD observed on the replicate injection indicates the acceptable reproducibility and thereby the precision of the system.

Method precision

Method precision was determined by measuring repeatability and ruggedness (intermediate precision (between-day precision or time-different intermediate precision)) for each vitamin during 3 days at the same concentration levels. The variation coefficients varied from 0.7 to 1.2%.

Repeatability

Complete analysis was performed in replicate on all samples to allow the calculation of average deviations as a measurement of chromatographic reproducibility. There was one good agreement with the fact that vitamin amounts added by the manufacturers

were usually slightly higher than the label claim, e.g., accounting for a decomposition of vitamins (e.g., Mecobalamin) during tablet storage. Considering an inhomogeneity of the vitamin content of unknown magnitude due to capsule formulation, the precision of the methods is even greater. A good agreement between the concentration values of the reference samples and the amount of vitamins found after analysis has been obtained with the proposed method and also the relative standard

deviations obtained in the analysis of vitamins was summarized in Table 3.

Ruggedness

The ruggedness of the method was demonstrated by the analysis of the samples by two different analysts using two different sets of instruments. RSD of the two sets of data indicated the ruggedness of the method which was shown in Table 4.

Table 3 a.: System precision

Injection number	Area response			
	Benfotiamine	Pyridoxine Hydrochloride	Mecobalamin	Alpha-lipoic acid
1	1025012	326720	73088	443896
2	1019322	319928	72780	444035
3	1012631	318287	72735	445047
4	1017663	315502	72208	442568
5	1018752	317597	72248	444071
6	1008158	317135	72375	443888
Average	1016923	319195	72572	443918
S.D.	5837	3960	350	792
%R.S.D.	0.6	1.2	0.5	0.2

Table 3 b.: Quantification of active components in multivitamin capsule

Vitamin	Vitamin content (mg)	SD	RSD (%)	Recovery (%)*
Benfotiamine	100	1.2	1.3	99.1
Pyridoxine HCl	100	0.9	0.9	98.0
Mecobalamin	1	1.5	1.1	136.1
Alpha-lipoic acid	100	0.8	0.8	103.9

*Complete analysis was performed in replicate.

Table 4: Ruggedness of the method

Vitamin	Vitamin content (mg)	SD	RSD (%)	Recovery (%)*
Benfotiamine	100	0.9	0.9	98.2
Pyridoxine HCl	100	1.1	1.1	98.4
Mecobalamin	1	0.9	0.7	137.1
Alpha-lipoic acid	100	1.2	1.2	103.7

* Complete analysis was performed in replicate

Robustness

The robustness of an analytical method is a measure of its capacity to remain unaffected by small but deliberate variations in method parameters and provides an indication of its reliability during normal usage. The variations may be variation in flow rates by ± 2 nm, variation in pH by ± 0.2 units. The observed results were lie within the limit.

Linear range

Linearity was obtained in the range of standard concentration of each vitamin was shown in Table 5. A series of eleven solutions at

low and high concentration levels (from 50 to 150% of the nominal concentration levels) were prepared, each solution was injected three times and the regression was calculated by the method of least-squares. Peak areas were calculated and the results interpolated on the calibration graph for each vitamin. It could be shown that the correlation is linear over a wide range covering all common capsule weights and vitamin contents.

Thus, possible volume errors resulting from different amounts of capsule material added to the homogenization vessel are negligible.

Table 5: Linearity of the chromatographic system

Vitamin	Linear range ($\mu\text{g ml}^{-1}$)	R ²	Intercept	Slope
Benfotiamine	489.9-1469.6	0.9997	-9114	1039
Pyridoxine HCl	499.4-1498.1	0.9990	13919	274
Mecobalamin	4.6-13.9	0.9995	957	7679
Alpha-lipoic acid	497.7-1469.6	0.9998	-5431	432

Accuracy

The accuracy of the described procedure was assessed by adding known amounts of standard vitamins equal to one dosage form, to a powdered placebo capsules prior to analysis in the presence of maize starch, PVP, talc, magnesium stearate, eudragit E, aluminium hydroxide and sucrose. The recovery data was determined by comparing the areas obtained with that of standard solution

analyzed according to the method. The results of the quantitation of Benfotiamine (B₁), Pyridoxine hydrochloride (B₆), Mecobalamin (B₁₂) and Alpha-lipoic acid, 100, 100, 1, 100 mg/capsule respectively, in multivitamin placebo capsules were shown in the Table 6. The accuracy was carried out at 80%, 100%, and 120% of specification limit. These were in good agreement with the added amount and the recovery of the compounds from each dosage form was quantitative.

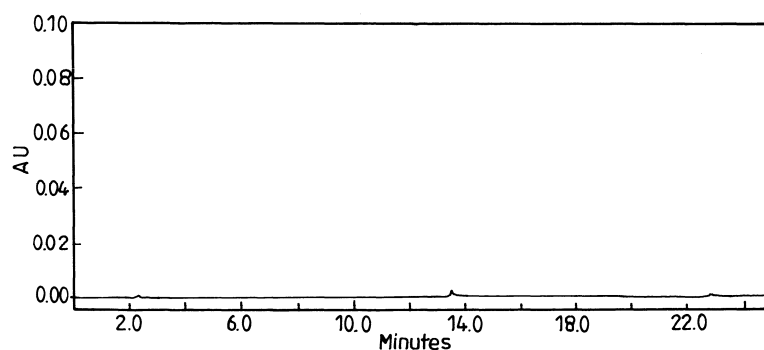
Table 6: Recovery study for the determination of placebo multivitamin capsules by adding 100 mg/capsule for vitamins B₁, B₆, Alpha-lipoic acid and 0.1 mg/capsule for B₁₂ - 80%, 100% and 120% accuracy level.

S.No	Benfotiamine		Pyridoxine HCl		Mecobalamin		Alpha-lipoic Acid	
	Found mg/cap	Recovery %	Found mg/cap	Recovery %	Found mg/cap	Recovery %	Found mg/cap	Recovery %
Accuracy Level - 80%								
1	81.09	100.6	79.67	99.5	0.84	100.0	80.64	99.7
2	80.45	100.8	79.20	98.4	0.83	101.2	84.04	101.3
3	78.72	99.9	80.49	98.7	0.81	98.8	79.50	99.0
Mean		100.4		98.9		100.0		100.0
S.D.		0.5		0.6		1.2		1.2
% RSD		0.5		0.6		1.2		1.2
Accuracy Level - 100%								
1	96.22	99.1	97.07	98.8	0.97	99.0	102.23	100.2
2	96.21	98.8	101.76	101.4	0.98	101.0	104.31	100.0
3	97.38	99.4	100.37	100.9	0.99	99.0	100.38	99.2
Mean		99.1		100.4		99.7		99.8
S.D.		0.3		1.4		1.2		0.5
% RSD		0.3		1.4		1.2		0.5
Accuracy Level - 120%								
1	114.44	99.5	119.55	100.1	1.19	100.8	123.42	101.1
2	117.67	99.8	120.98	101.3	1.20	98.4	121.67	100.6
3	119.36	100.7	120.34	100.6	1.21	101.7	120.67	100.8
Mean		100.0		100.7		100.3		100.8
S.D.		0.6		0.6		1.7		0.3
% RSD		0.6		0.6		1.7		0.3

Specificity

The specificity of the present method was described by peak purity testing in which it compares the spectra within a peak to the peak apex spectrum to determine if the peak consists of one or more spectrally distinct component. It calculates a spectral contrast angle (purity angle) and a threshold angle (purity threshold) for each spectral comparison between the reference spectrum and the other

peak spectra. Purity angle should be less than the purity threshold for each component and there was no purity flag indicates that the excipients present in the dosage forms did not interfere with the analysis (no noticeable interference from the excipients was observed in the chromatograms) and a representative chromatogram together with other observed purity parameters are illustrated in Figs. 3 and 4 and Table 7.

**Fig. 3: Typical placebo chromatogram for recovery study****Table 7: Observed peak purity parameters of each component in multivitamin capsules**

Vitamin	Purity angle	Purity threshold	Purity flag *
Benfotiamine	0.079	1.037	No
Pyridoxine HCl	0.285	2.332	No
Mecobalamin	0.241	1.184	No
Alpha-lipoic acid	0.078	1.056	No

* No purity flag indicates that there is no interference with the main peak by the excipients and hence peak purity passes for all the four components (Limit: Purity angle should be less than the purity threshold)

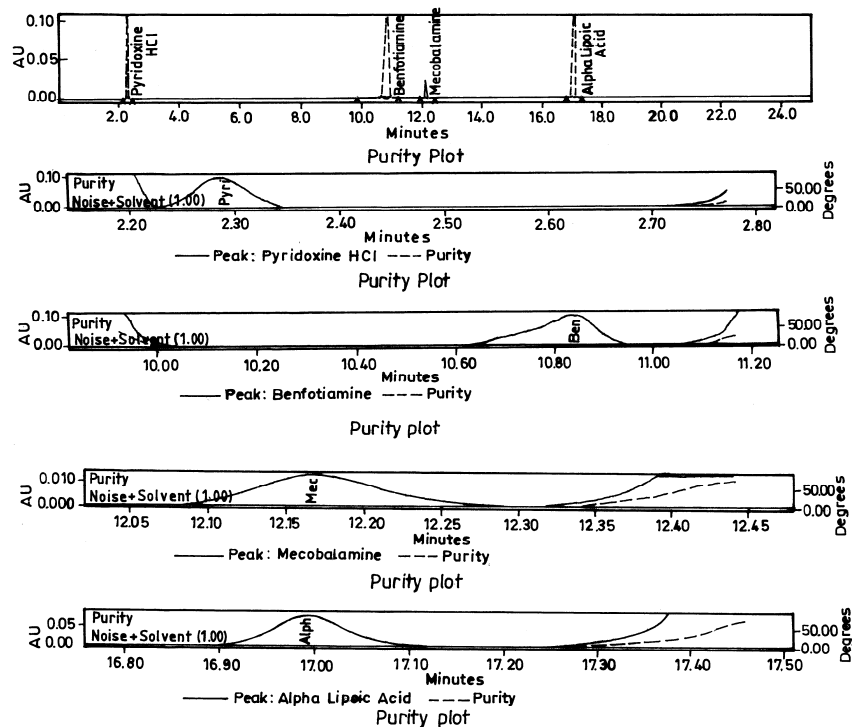


Fig. 4: Typical chromatogram for the purity determination of all the four components

CONCLUSION

In the preceding method for determination of vitamin B₁, B₆, B₁₂ and Alpha-lipoic acid levels in capsule, use of an XTerra column RP-18 and a gradient elution mode was applied. This method is successful for vitamin B₁, B₆, B₁₂ and Alpha-lipoic acid formulations even in the ratio between B₁₂ and the others 1:100 or 1:1000. Isocratic conditions could not be used mainly because the chromatographic peaks are not well resolved. The method is simple, specific and rapid with a high degree of accuracy and precision and applied successfully to the routine analysis of B₁, B₆, B₁₂ and Alpha-lipoic acid in B-complex capsules. The simplicity of the procedure should make it highly desirable for quality control of multi-vitamin products in the pharmaceutical and health food industries.

REFERENCES

1. William Martindale. James E.F. Reynolds, editor. Martindale: The Extra Pharmacopoeia. 31st ed. London: Royal Pharmaceutical Society; 1997.
2. Eitenmiller RR, Lin Ye, Landen WO. Vitamin Analysis for the Health and Food Sciences. Boca Raton: CRC Press; 1999.
3. American Medical Association (AMA) Drug Evaluations. 5th ed. AMA Division of Drugs in cooperation with the American Society for Clinical Pharmacology and Therapeutics; Chicago: 1983.
4. Strohecker R, Henning HM. Vitamin Assay Tested Methods. Verlag Chemie. Limburg. 1965.
5. The United States Pharmacopoeia, 24th Revision, U.S.Pharmacopoeial Convention, Inc., Rockville, MD: 2004.
6. Markopoulou CK, Kagkadis KA, Koundourellis JE. An optimized method for the simultaneous determination of vitamins B₁, B₆, and B₁₂ in multivitamin tablets by high performance liquid chromatography. J. Pharm. Biomed. Anal 2002; 30: 1403-1410.
7. Official Methods of Analysis (OMA). 13th ed. Association of Official Analytical Chemists; Washington DC: 1980.
8. Dinelli G, Bonetti A. Micellar electro kinetic capillary chromatography analysis of water-soluble vitamins and multi-vitamin integrators. Electroph 1994; 15: 1147-1150.
9. Lambert D, Adjalla C, Felden F, Benhayoun S, Nicholas JP, Gueant JL. Identification of vitamin B₁₂ and analogues by high-performance capillary electrophoresis and comparison with high-performance liquid chromatography. J. Chromatogr 1992; 608: 311-315.
10. Fotsing L, Fillet M, Bechet I, Ph Hubert, Crommen JJ. Determination of six water-soluble vitamins in a pharmaceutical formulation by capillary electrophoresis. Pharm. Biomed. Anal 1997; 15: 1113-1123.
11. De Leenheer AP, Lambert WE, Nelis HJ. Modern chromatographic analysis of vitamins-chromatographic science series. New York: Marcel Dekker; 1992.
12. Woollard DC. New ion-pair reagent for the high-performance liquid chromatographic separation of B-group vitamins in pharmaceuticals. J. Chromatogr 1984; 301: 470-476.
13. Blanco D, Escotet VJ, Fidalgo LE, Gutierrez MD. Simultaneous determination of vitamins D₃, E and K₁ and retinyl palmitate in cattle plasma by liquid chromatography with a narrow-bore column. J. Pharm. Biomed. Anal 1994; 660: 49-55.
14. Chatzimichalakis PF, Samanidou VF, Verpoorte R, Papadopyannis IN. Development of a validated HPLC method for the determination of B-complex vitamins in pharmaceuticals and biological fluids after solid phase extraction. J. Sep. Sci 2004; 27: 1181-1188.
15. Iwase H. Ultramicrodetermination of cyanocobalamin in elemental diet by solid-phase extraction and high-performance liquid chromatography with visible detection. J. Chromatogr 1992; 590: 359-363.
16. Shinomiya K, Yoshida K, Kabasawa Y. Countercurrent chromatographic separation of water-soluble vitamins by cross-axis coil planet centrifuge using an ion-pair reagent with polar two-phase solvent system. J. Liq. Chrom. Rel. Technol 2001; 24: 2615-2623.

17. Moreno P, Salvado V. Determination of eight water- and fat-soluble vitamins in multi-vitamin pharmaceutical formulations by high-performance liquid chromatography. *J. Chromatogr. A* 2000; 870: 207-215.
18. Hoeller U, Brodhag C, Knoebel A, Hofmann P, Spitzer V. Automated determination of selected water-soluble vitamins in tablets using a bench-top robotic system coupled to reversed-phase (RP-18) HPLC with UV detection. *J. Pharm. Biomed. Anal* 2003; 31: 151-158.
19. Wimalasiri P, Wills RBH. Simultaneous analysis of thiamin and riboflavin in foods by high-performance liquid chromatography. *J. Chromatogr* 1985; 318: 412-416.
20. Dinc E, Kokdil G, Onur FA. Comparison of matrix resolution method, ratio spectra derivative spectrophotometry and HPLC method for the determination of thiamine HCl and pyridoxine HCl in pharmaceuticals. *J. Pharm. Biomed. Anal* 2000; 22: 915-923.
21. Orr M.L. Pantothenic acid, Vitamin B₆ and Vitamin B₁₂ in Foods. Home Economics-US Department of Agricultural Bulletin; Research Report No. 36, Washington DC: 1969.
22. Wang E, Hou. Determination of water-soluble vitamins using high-performance liquid chromatography and electrochemical or absorbance detection. *J. Chromatogr* 1988; 447: 256-262.
23. Chu K, Tin K. Analysis of commercial multi-vitamin preparation by HPLC with diode array detector. *Anal. Lett* 1998; 31: 2707-2715.