

COMPUTATIONAL PROCEDURE FOR DETERMINING PHYSICOCHEMICAL PROPERTIES OF DOXORUBICIN DIALDEHYDE STARCH (DOX-DAS) AND DOXORUBICIN-DIALDEHYDE STARCH NANOPARTICLES (DOX-DASNP)

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ABSTRACT

Doxorubicin (or Adriamycin) is well known anti-cancer agent. It is an anthracycline antibiotic. The use of doxorubicin against neoplasms is limited due to its severe cardiotoxicity. The cytotoxicity of doxorubicin can be minimized by linking it to an affinity tag. In this report, the molecular structure, Binding Energy(BE), Dipole Moment (DM), Gibbs free energy of solvation ($\Delta G_{\text{(solvation)}}$) and some physicochemical properties of Doxorubicin Dialdehyde starch (DOX-DAS) and Doxorubicin-Dialdehyde starch nanoparticles (DOX-DASNP) conjugated complexes were investigated using Density functional Theory (DFT) calculations. Our results indicate that this complex mentioned above can be used to improve anti cancer activity and water-solubility of doxorubicin.

Keywords: Anti-cancer drug, DFT calculations, DOX-DASNP, DOX-DAS, Doxorubicin.

INTRODUCTION

Doxorubicin is an anthracycline ring antibiotic that is widely used as a cancer therapeutic. Common treatment side effects include nausea and vomiting, loss of appetite, diarrhea and swelling, but development of a system capable of selectively delivering the drug to the intended target cells may be able to reduce these complications. There are many approaches to drug delivery via drug/drug carrier combinations, such as encapsulation, hydrogel formation, nanoaggregation, and micellar delivery. For doxorubicin delivery, encapsulation and micellar delivery have received increased attention because this system can protect and carry the drug directed to its intended target¹. In experimental studies carried out by some other researchers, it has been illustrated that As a natural biomaterial, starch is a traditional filling agent which has been used in curatorial field for a long time²⁻⁴, and in recent years it is found to have more properties through physical and chemical denaturalization, thus it could be used diffusely in medicine field as drug carrier material^{5,6}. As one kind of starch ramification, dialdehyde starch is prepared by oxidation of starch with periodic acid or periodate, which is a selective oxidizing agent, and can cleave the C2, C3 linkage of anhydroglucose units with the formation of

dialdehyde. Dialdehyde starch not only is biocompatible and biodegradable as starch, but also has more chemical activities. For example, it can react with hydrazine, acid, amine, imine and so on, so it is much more widely used in medicine field^{7,10}. Doxorubicin is a traditional anticancer and antimetabolization drug, which is widely applied to cancer treatment¹¹, but it also has its limitation because of its heavy poisonous side-effect. So it is considered to be combined with natural or synthetical macromolecule polymer and prepared to polymer drug-loaded particles, which resulted in target delivery and sustaining release^{12,13}. Doxorubicin Dialdehyde starch (DOX-DAS) and Doxorubicin-Dialdehyde starch nanoparticles (DOX-DASNP) conjugated complexes were synthesized by LIU XuanMing and colleagues¹⁴. The conjugation scheme is illustrated in Fig. also optimized structures of DOX-DAS and DOX-DASNP have been showed in Fig.2. The geometry structure of DOX-DAS and DOX-DASNP were optimized at B3LYP/6-31g* level of theory and then the Gibbs free energy of solvation ($\Delta G_{\text{(solvation)}}$) were calculated at B3LYP/6-31g* level of theory using Gaussian 03¹⁵. In this study unit number of DAS is 4. Based on these previous experiments, we concluded that doxorubicin can be conjugated to a biopolymer that selectively targets cancer cells. Quantum mechanical molecular simulation can be used to study drug delivery¹.

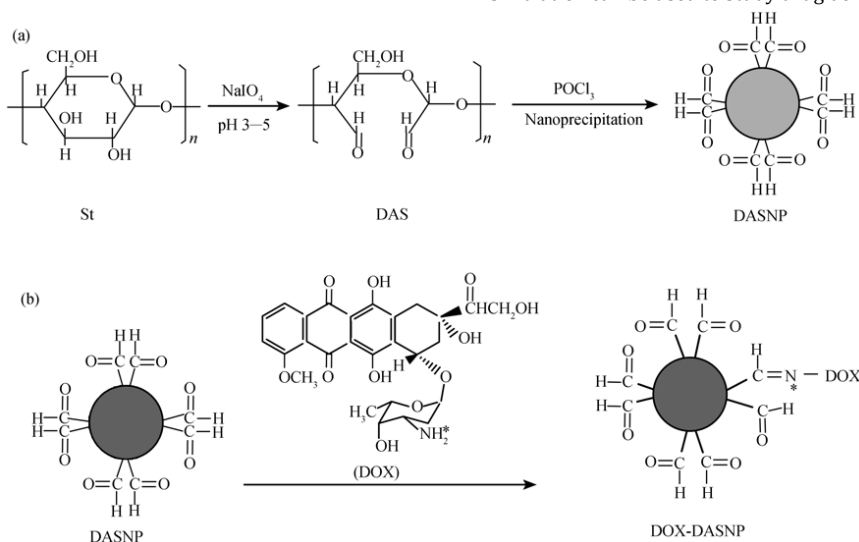


Fig. 1: Synthesis of DASNP and DOX-DASNP. (a) Synthesis of DASNP; (b) synthesis of DOX-DASNP¹⁴

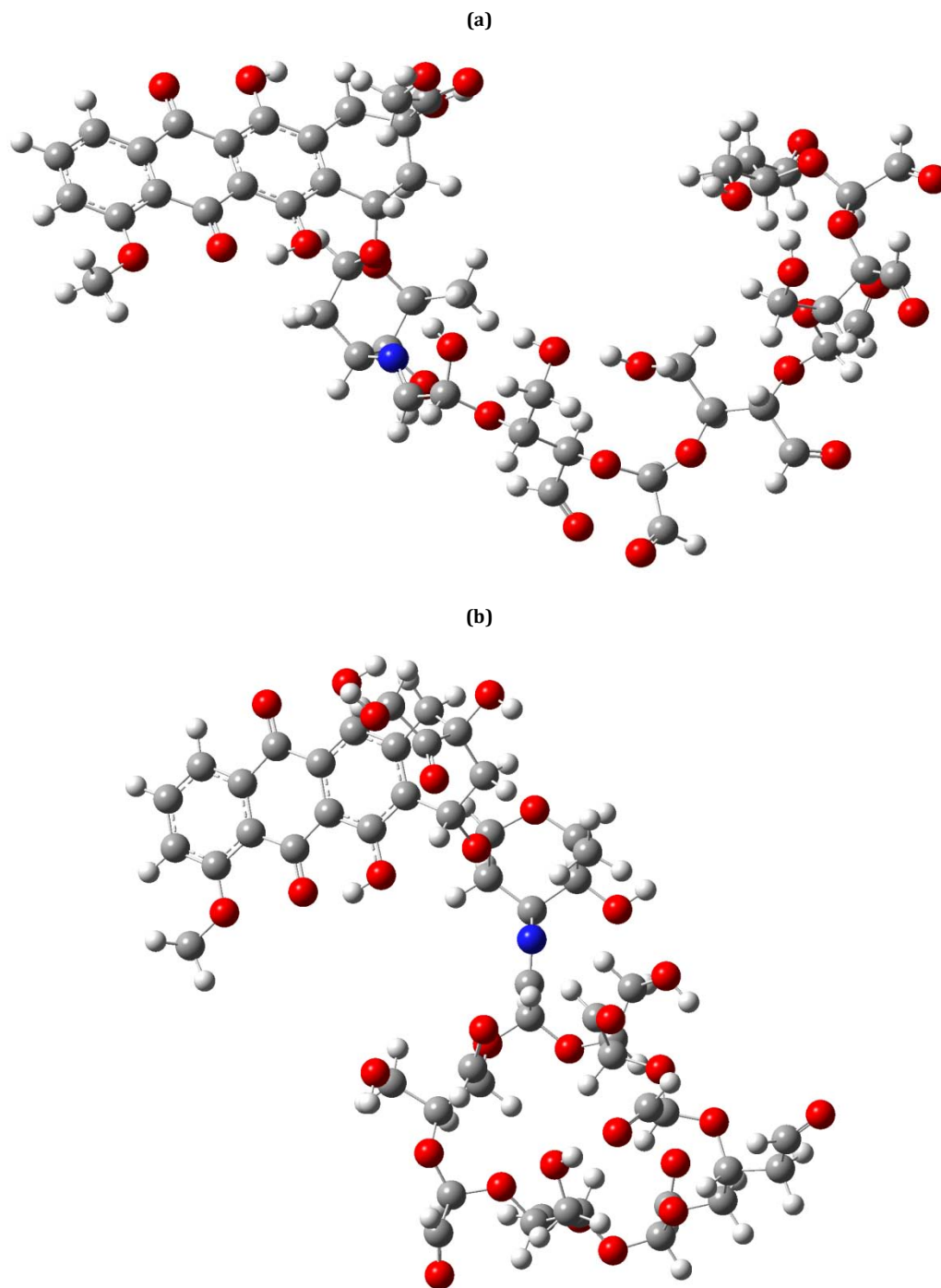


Fig. 2: Structures of DOX-DAS (a) and DOX-DASNP (b) n=4

In this study, we intend to show some the characteristics of doxorubicin, DOX-DAS and DOX-DASNP which have been mentioned above and have been obtained by other researchers experimentally through predictable computational calculations including molecular energy, binding energy, dipole moment, $\Delta G_{(solvation)}$, distance bound and angle bound¹⁶.

RESULTS AND DISCUSSION

The geometry structure of DOX-DAS and DOX-DASNP were optimized at B3LYP/6-31g* level of theory and then the Gibbs free energy of solvation ($\Delta G_{(solvation)}$) were calculated at B3LY/6-31g*

level of theory using Gaussian 03¹⁵. Table 1 presents the geometrical parameters of two different complexes mentioned above around linking position (imine group), see also Fig 3.

Some physicochemical properties of DOX-DAS , DOX-DASNP and Doxorubicin conjugates such as Refractivity, polarizability, Hydration energy, binding energies (BE), Gibbs free energy of solvation ($\Delta G_{(solvation)}$) and Dipole moment (DM) are obtained from optimal structure¹⁷ which have been shown in Table 2. The binding energy per molecule was computed using the formula (1):

$$\Delta E = E_{\text{complex}} - E_{\text{drug}} - E_{\text{carrier}} \quad (1)$$

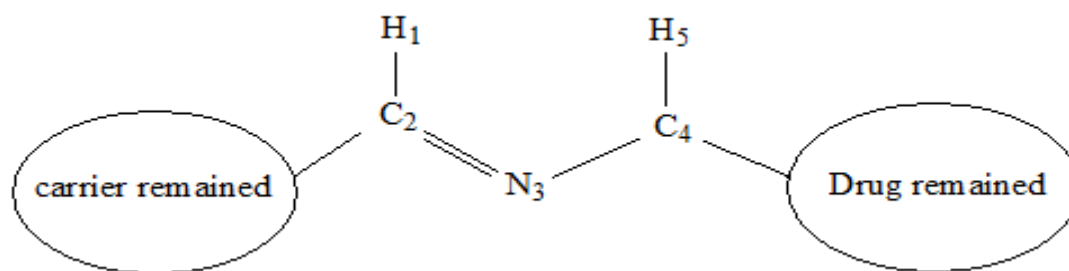


Fig. 3: Structure of linking position in DOX-DAS and DOX-DASNP complexes.

Table 1: Geometrical parameter of complexes around linking position

Complex	R(H ₁ -C ₂) (Å)	R(C ₂ =N ₃) (Å)	R(N ₃ -C ₄) (Å)	C ₂ -N ₃ -C ₄ (°)	N ₃ -C ₄ -H ₅ (°)
DOX-DAS	1.097	1.262	1.449	121.889	110.155
DOX-DASNP	1.101	1.265	1.455	117.753	110.315

Table 2: Some calculated physicochemical properties of DOX-DAS, DOX-DASNP and Doxorubicin

physicochemical properties	DOX-DAS	DOX-DASNP	Doxorubicin
Refractivity ^a	265.15	261.64	135.50
Polarizability	103.78	102.37	52.00
Hydration energy ^a	8.17	3.73	-24.03
Surface area ^a (Å ²)	1218.90	1113.46	729.45
ΔG (solvation) (kcal/mol)	-51.43	-49.4	-23.21
Dipole moment(Debye)	21.019	9.643	6.848
BE (ev/mol)	-1071.845	-1071.845	

^aData were calculated using HyperChem 8 software

CONCLUSION

Density functional Theory (DFT) calculations were applied to study some physicochemical properties of DOX-DAS, DOX-DASNP and Doxorubicin. Regarding the calculation results, hydrophilicity of DOX-DAS and DOX-DASNP are higher than that of Doxorubicin; this fact can be verified through the Gibbs free energy of solvation (ΔG solvation) obtained for these complexes using Gaussian 03. Therefore, DOX-DAS and DOX-DASNP are more soluble than of Doxorubicin. These calculations also show that doxorubicin release from DOX-DAS and DOX-DASNP complexes takes place in a long period of time. On the other hand it means that we have a gradual release which is predictable according to Binding Energy. Our results indicate that these complexes mentioned above can be used to improve anti cancer activity and water-solubility of Doxorubicin.

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