Academic Sciences

International Journal of Pharmacy and Pharmaceutical Sciences

ISSN- 0975-1491

Vol 5, Issue 2, 2013

Research Article

SYNTHESIS AND CHARACTERIZATION OF NOVEL MANNICH BASES OF BENZIMIDAZOLE DERIVATIVES FOR ANTIBACTERIAL AND ANTIFUNGAL ACTIVITY

M.VIJEY AANANDHI^{1*}, ABHAY KUMAR VERMA¹, R.SUJATHA², R.KAMAL RAJ³

¹Department of Pharmaceutical Chemistry, School of Pharmaceutical sciences, Vels University(VISTAS), Chennai 600117, India, ²N.K.R Government Arts College for Women, Namakkal 637001 India, ³Department of Pharmaceutical Chemistry, Saastra College of Pharmaceutical Education and research, Nellore 524311 India. Email: mvaanandhi@gmail.com

Received: 17 Jan 2013, Revised and Accepted: 02 Mar 2013

ABSTRACT

Objectives: A series of mannich bases of benzimidazole derivatives were synthesized from o-phenylenediaamine in two steps via benzimidazole intermediates

Methods: The synthesized Compounds 3a-c were characterized by IR, 1H NMR, mass and elemental analysis and were evaluated for their anti fungal activity against a panel of two pathogenic fungal strains namely, *Aspergillus niger, and Candida albicans* by two fold serial dilution method, antibacterial activity against *B.subtilis*, *S.aureus*, *E. Coli*, *S. Typhi*

Results: All the compounds showed significant inhibitory activity against the microbes with the 100μ g/ml which produces 100% inhibition against the microorganism. Anti bacterial activity was determined using Ciprofloxacin as a standard and antifungal activity was determined using standard Ketoconazole,

Conclusions: Out of the synthesized compounds 3a, 3b and 3c shows excellent antibacterial activity and compound 3a showed good antifungal activity than others.

Keywords: Benzimidazole derivatives, Anti fungal activity, Antibacterial activity

INTRODUCTION

The benzimidazole is an important pharmacophore in modern drug discovery. Literature review reveals that Manich bases of benzimidazole derivatives exhibits diverse pharmacological activities like, antimicrobial[1], analgesic[2], anti-inflammatory[3], anthelmintic[4], antiviral as well as antitumor[5], etc. In addition benzimidazoles are very important intermediates in organic reactions. Research in this area is still unexplored and is directed towards the synthesis of compounds with enhanced biological activity. Based on the above observation it is worthwhile to prepare newer novel Mannich bases of benzimidazoles with enhanced antimicrobial activity.

MATERIALS AND METHODS

The melting points were determined in open glass capillaries and are uncorrected. IR spectra were recorded on Shimadzu FT-IR 8400-S spectrophotometer by KBr pellet technique. ¹H-NMR and ¹³C-NMR spectra were recorded on AMX-400 NMR spectrophotometer at 400 MHz using DMSO-d₆ as the solvent and tetra methyl silane (TMS) as internal standard. The chemical shifts are expressed in δ ppm. The splitting patterns were designated as follows; s: singlet; d: doublet; q: quartet; m: multiplet. LCMS were recorded by using Shimadzu LCMS-2010A instrument by ESI. Molecular ion (M⁺) value in m/z units is provided along with percent relative abundance in parenthesis. Synthesis of the intermediate and target compounds was accomplished according to the steps depicted in **Scheme 1**.

Reagents and conditions: i) formic acid and 10%NaoH ii) paminobenzoic acid, P-Dimethyl amino benzaldehyde and ethanol iii) p-aminobenzoic acid, p-hydroxy benzaldehyde and ethanol iv) paminobenzoic acid, P-methoxy benzaldehyde and ethanol

General method for the preparation of Synthesis of benzimidazole(2)

O-Phenylenediaamine**(1)** (1.08 g, 10.0 mmol) was placed in a 100 ml flask and it was added 0.7 g (0.52ml, 13.6 mmol) of 85% formic acid. The mixture was heated on a water bath for 2 hr at 100 $^{\circ}$ C, cooled and 10%NaOH solution was added slowly with constant rotation of the flask, until the mixture was just alkaline to litmus. Crude benzimidazole was filtered off at the pump, washed with ice cooled

water. The crude product was dissolved in boiling water, discoloring carbon was added to it, and the mixture was digested for 15 minutes. Filtered rapidly at the pump though a preheated bucker funnel, the filtered was cooled to about 10 $^{\circ}$ C, the product was filtered off, washed with cold water and dried to give pure benzimidazole.

General procedure for the synthesis of Mannich bases of benzimidazole derivatives(3a-c)

To a solution containing benzimidazole(2) (1.18g,10 mmol), in 20 ml of ethanol, 1.40g(10 mmol) of aromatic aldehyde and 1.37g(10 mmol) of 4-aminobenzoic acid were added with constant stirring for 1 hr. the reaction mixture was reflux for 10 hr. On ice cooling the product formed was filtered, dried in vacuum and recrystallized with ethanol

4-[1H-Benzimidazole-yl(p-dimethylaminobenzal)methylamino] benzoic acid(3a)

UV λ_{max} (DMSO): 279 nm; IR (KBr) v_{max} : 1591 (C =C, Ar), 3252 (C-H, Ar), 3259 (C-H,s), 1410 (C-H, b), 945 (C-C, Ring), 1352 (C-N, s), 1681 (C=N, s), 3429 (C-N, 3°amine), 2360 (COOH,s), 1681 (C=O, s); ¹H NMR(DMSO- *d*₆) δ : 7.26 (Ar-H, m), 0.86 (-CH₃m) 2.36, 2.00 (-NH, d), 11(OH, COOH, s); LC-MS: *m/z* 446.58. (M⁺); Mol. formula C₂₇H₃₄N₄O₂. m.p.190 -191 °C, Yield 73%.

4-[1H-Benzimidazole-yl-(p-hydroxybenzal) methylamino]benzoic acid(3b)

UV λ_{max} (DMSO): 278 nm; IR (KBr) v_{max} : 1609 (C =C, Ar), 3217 (C-H, Ar), 3196 (C-H,s), 1384 (C-H, b), 941 (C-C, Ring), 1348 (C-N, s), 1665 (C=N, s), 3390 (C-N, 3°amine), 3639(OH, s), 2804 (COOH,s), 1635 (C=O, s); ¹H NMR(DMSO- *d*₆) &: 7.26 (Ar-H, m), 6.67(-CH₂, m), 4.00 (-NH, d), 11(OH, COOH, s), 5.00 (OH, Ar, s); LC-MS: *m/z* 434.58 (M⁺); Mol. formula C₂₆H₃₂N₃O₃ m.p.195 -196 °C, Yield 75%.

4-[1H-Benzimidazole-yl-(p-methoxybenzal) methyl-amino] benzoic acid(3c)

UV λ_{max} (DMSO): 279 nm; IR (KBr) v_{max} : 1512 (C =C, Ar), 3394 (C-H, Ar), 3394 (C-H,s), 1458 (C-H, b), 996 (C-C, Ring), 1350 (C-N, s), 1600 (C=N, s), 3454 (C-N, 3°amine), 2350 (COOH,s), 1688 (C=O, s), 1253(OCH₃, s); ¹H NMR(DMSO- d_6) δ : 7.26 (Ar-H, m), 6.67(-CH₂, m),

4.00 (-NH, d), 11(OH, COOH, s), 6.65 (-OCH₃, s); LC-MS: *m/z* 448.68 (M⁺); Mol. formula C₂₇H₃₄N₃O₃ m.p.158 -160 °C, Yield 83%.

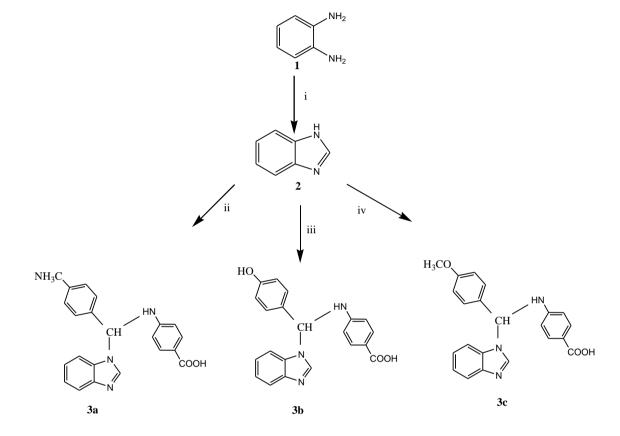
In vitro antifungal activity[6-7]

In vitro anti fungal activity of the synthesized compounds was evaluated by two fold serial dilution method. Media used was Potato Dextrose Broth (PDB). Initially, the stock culture of *Aspergillus niger*, and *Candida albicans* were revived by inoculating in broth media and grown at 37° C for 48 hrs. The tubes of the above media PDB (5 ml) were prepared and each tube was added with compounds (100 µg) and inoculated with 100 µl of 48 hr old cultures. The control tubes with Ketoconazole and DMSO were also prepared. All the tubes were incubated at

37°C for 48 h with constant shaking and the absorbance of biomass were measured 660 nm against autoclaved, uninoculated media as blank. The result of *in vitro* anti fungal activity is expressed as Minimum Inhibitory concentration (MIC) and is given in **Table 1**.

Antibacterial activity[8-10]

All the compounds were screened in- vitro for their antibacterial activity against *Staphylococcus aureus*, *Bacilllus subtilis*, *Solmonella typhi*, *Escherichia coli* by cup- plate method at 100μ g/ml concentration using Ciprofloxacin as standard and DMSO as solvent control. After 24 hr of incubation at 37° C, the MIC was measured. The results are tabulated in **Table-2**



Scheme 1: Synthesis of mannich bases of benzimidazole derivatives

Table 1: In vitro antifungal activity of mannich bases of benzimidazole derivatives

Compound	Concentration	Zone of inhibition (mm)	
	(µg/ml)	Candida albicans	Aspergillus niger
3a	100	11	10
3b	100	9	9
3c	100	8	8
Standard (Ketoconazole)	100	23	22
Standard (DMSO)	-	-	-

Table 2: In vitro Antibacterial activity of mannich bases of benzimidazole derivatives

Compound	Concentration (µg/ml)	Zone of inhibition (mm)			
		Gram positive		Gram negative	
		B.subtilis	S.aureus	E. coli	S. typhi
3a	100	13	12	8	13
3b	100	12	12	13	14
3c	100	13	10	10	12
Standard (Ciprofloxacin)	100	24	22	21	22
Control (DMSO)	-	-	-	-	-

RESULTS AND DISCUSSION

The melting points of all synthesized compounds were found in open capillary tubes and readings were uncorrected. The results of IR spectra were given in spectral detail heading which showed absorption bands for aromatic C-H,N-H,C-N,C=N,C=O,OH,C-Cl, and O-CH₃ groups. The results of the ¹H NMR spectra given under spectral detail heading showed that the numbers of hydrogen atoms present in all the synthesized compounds were exact when compared to the number of hydrogen atoms in the expected compounds. the molecular mass of the synthesized compounds was nearer to the molecular mass of the expected compounds

For all three compounds antibacterial activity was determined using standard Ciprofloxacin, results were tabulated in **Table-2**. All the compounds showed significant inhibitory activity against the microbes with the 100μ g/ml which produces 100% inhibition against the microorganism. Depending on the functional group present in the aromatic ring, different MIC values were obtained. Out of the synthesized compounds 3a, 3b and 3c shows excellent antibacterial activity. The compounds substituted with Methoxy and hydroxyl group (**3b & 3c**) showed higher antibacterial activity was determined using standard Ketoconazole, results were tabulated in **Table-1**. out of three compounds, compound and **3a** showed good antifungal activity than others.

ACKNOWLEDGEMENT

The authors are thankful to Vels University (VISTAS) and its management for providing research facilities and encouragement and to our friends those who helped us to complete this research

REFERENCE

- 1. Khadar et al, microwave assisted synthesis of 1,3,4-oxadiazoles carrying benzimidazole moiety and their antimicrobial properties, Ind J of Chem 2010;49:1130-1134.
- Khan S et al, 2-Substituted Benzimidazoles as Antiinflammatory and Analgesic agents. Indian J. Heterocycl. Chem 1997;7:55-58.
- 3. Gunasekaran et al, Synthesis, Antiinflammatory and Antibacterial activities of substituted Phenyl Benzimidazole. Asian J. Chem 2007;19(1):116-120.
- 4. Murthi et al, Synthesis, characterization and in vivo anthelmintic activity of some novel N-Mannich bases of benzimidazoles, J. Indian Chem. Soc 2010;87:627-631.
- 5. Pandey et al, 1, 2-Disubstituted Benzimidazole as potential antiviral agents. Indian J. Heterocycl. Chem 2005;14:217-220.
- 6. Jarrahpour et al, Synthesis, Antibacterial, Antifungal and Antiviral Activity Evaluation of Some New bis-Schiff Bases of isatin and Their Derivatives. Molecules 2007; 12: 1720-1730.
- Bharati et al, Synthesis of Mannich base with benzimidazoles and its antifungal activity, Ind. J.Hetrocycle.Chem 2003;12:249-252.
- 8. Podunavac-kuzmanovi et al, Antibacterial evaluation of some benzimidazole derivatives and their zinc (II) complexes, J. Serb. Chem. Soc. 2007;72(5):459–466.
- 9. David et al, Synthesis of bisbenzimidazole and antimicrobial activity. J. Med. Chem 2004;9:158-163.
- 10. Sundari et al, Synthesis of some thiazine substituted benzimidazoles, antibacterial and antifungal activity, Ind. J. Hetrocycle. Chem 2004;14:47-50.