Academic Sciences

International Journal of Pharmacy and Pharmaceutical Sciences

ISSN- 0975-1491

Vol 5, Suppl 4, 2013

Research Article

DEVELOPMENT AND VALIDATION OF A NEW AND STABILITY INDICATING RP-HPLC METHOD FOR THE DETERMINATION OF IVACAFTOR IN PRESENCE OF DEGRADANT PRODUCTS

PAWANJEET. J. CHHABDA¹, M. BALAJI², SRINIVASARAO .V²

Department of Biochemistry, Ahmednagar College, Ahmednagar, University of Pune, Department of Chemistry, Gitam Institute of Science, GITAM University, Visakhapatnam, India. Email: pawanjeetvps@rediffmail.com

Received: 29 Sep 2013, Revised and Accepted: 21 Oct 2013

ABSTRACT

The objective of this work is to develop a novel stability indicating high performance liquid chromatographic method for determination of Ivacaftor in bulk drug and pharmaceutical dosage form. The separation was achieved by using C8 column with mobile phase consisting of buffer-acetonitrile (30:70 % v/v) with a flow rate of 1.0ml/min. Detection was carried out at 225nm. Linearity was observed over the concentration range of 15-300 μ g/ml with r²= 0.9999. The percentage relative standard deviation in accuracy and precision studies was found to be less than 2%. Ivacaftor was subjected to stress conditions including acidic, alkaline, oxidation, photolysis and thermal degradation. Ivacaftor is more sensitive towards acidic and alkaline degradation. The method can be used to analyze commercial solid dosage contain Ivacaftor with good recoveries for routine analysis.

Keywords: Ivacaftor, Validation, HPLC, Stability indicating.

INTRODUCTION

Ivacaftor is a CFTR potentiator approved for patients with the G551D mutation of cystic fibrosis, which account for 4-5% cases of cystic fibrosis. Ivacaftor was developed by Vertex Pharmaceuticals in conjunction with the Cystic Fibrosis Foundation. Ivacaftor is available as tablets at the dose of 150 mg in the market under the brand name of Kalydeco. Ivacaftor is chemically N-(2, 4-di-tert-butyl-5-hydroxyphenyl)-1, 4-dihydro-4oxoquinoline-3-carboxamide with empirical formula is $C_{24}H_{28}N_2O_3$ and molecular weight 392.49(9-10).

Various methods in the literatures involve determination of Ivacaftor by HPLC (5), pharmacokinetics, pharmacodynamics (1-4). However no method is available for stability indicating HPLC method of Ivacaftor in bulk drug and pharmaceutical dosage form. In the present work we have developed a new, simple precise and stability indicating method for determination of Ivacaftor in bulk drug and pharmaceutical dosage form.



Fig. 1: Structure of Ivacaftor

MATERALS AND METHODS

Chemicals & Reagents

Ivacaftor is available as tablets with brand name KALYDECO was purchased from local market, containing Ivacaftor 150mg. HPLC grade acetonitrile, AR grade ortho phosphoric acid were purchased from Merck, Mumbai. High pure water was prepared by using Millipore Milli-Q plus purification system.

Chromatographic Conditions

A Alliance e2695 separation module (Waters corporation, Milford, MA) equipped with 2998 PDA detector with empower 2 software used for analysis. Buffer consisted of 0.1% orthophosporic acid in water (1ml of phosphoric acid in 1000 ml of water). A Zorbax eclipsed XDB C8 (4.6x150) mm,3.5 μ m column and isocratic mixture

of solution A (Buffer) solution B (Acetonitrile) used as stationary and mobile phase respectively. The isocratic program was fixed as A: B (30:70%v/v). Water and acetonitrile (20:80% v/v) used as diluent. The column oven maintained at 30°c with 1.0ml flow rate. An injection volume 5µl was used. The elution compounds were monitored at 225 nm.

Preparation of Stock and standard solutions

Accurately 150mg of Ivacaftor standard dissolved in 100ml diluent to get a concentration of $1500\mu g/ml$. Further 10ml of stock solution was taken in 100ml flask and diluted up to the mark with diluent to get concentration of $150\mu g/ml$.

Preparation of sample (Tablets)

20 tablets of Ivacaftor were powdered and an amount of powder equivalent to 150mg of drug was weighed and transferred to the 100ml flask added 10ml diluent and placed in an ultrasonicator for 10minites made up to the volume with diluent, and filtered through a $0.45\mu m$ nylon syringe filter. 10ml of this solution was taken into 100 ml flask and diluted volume with diluent to get concentration $150\mu g/ml$.

Forced Degradation studies

Acid Degradation studies

Acid decomposition was carried out in 0.1N HCL at concentration of 1500μ g/ml lvacaftor and after refluxation for 24hrs at 80°c, the stressed sample was cooled, neutralized and diluted as per requirement with diluents filtered and injected. The resulting chromatogram is shown in fig.3 (g). The results are tabulated in table 5.

Alkali Degradation studies

Base decomposition was carried out in 0.1N NaOH at concentration of 1500μ g/ml lvacaftor and after refluxation for 24hrs at 80°c, the stressed sample was cooled, neutralized and diluted as per requirement with diluents filtered and injected. The resulting chromatogram is shown in fig.3 (i). The results are tabulated in table 5.

Oxidation

Oxidation was conducted by using 5%H2O2 solution at room temperature. After 24hrs, 10ml of solution was taken in 100ml flask and diluted up to the mark with diluent to get concentration of 150 μ g/ml filtered and injected. The resulting chromatogram is shown in fig.3 (k). The results are tabulated in table 5.

Temperature Stress studies

1g of Ivacaftor sample was taken into a petridish and kept in oven at 80°c for 7 days. 150mg of sample was taken into 100 ml flask diluted volume with diluent, further 10ml to 100ml made up with diluent. The results are tabulated in table 5.

Photo stability

1g of Ivacaftor was taken in to a petridish and kept in photo stability chamber 200 W.hr/m² in UV Fluorescent light and 1.2M LUX Fluorescent light. 150mg of sample was taken in 100ml flask, dissolved in diluent, further 10ml in 100ml flask diluted volume with diluent. The results are tabulated in table 5.

RESULTS AND DISCUSSION

HPLC Method Development and Optimization

The analytical method conditions were selected after testing the different parameters such as column, wavelength, aqueous and organic phase, buffer concentration, mobile phase ratio, diluent, concentration of

analyte, flow and other parameters. Zorbax eclipsed XDB C8 (4.6x150) mm, $3.5 \ \mu m$ column was used because of its high resolution capacity and low degree of tailing. For mobile phase selection, the preliminary trials using different compositions of mobile phases containing water and acetonitrile gave poor peak shape. For improving peak shape instead of water ortho phosphoric acid and acetonitrile (30:70) and thus, better peak shape was obtained. Water and acetonitrile ($20:80 \ v/v$) used as diluent because lvacaftor freely soluble. The detection wavelength was chosen as 225nm for lvacaftor because they have better absorption and sensitivity at this wavelength (fig-2). Hence selected method was best among the all trails by many aspects.

Method Validation

Specificity

A study to establish the interference, blank detection was conducted. Diluent was injected as per the test method. Solution of standard and sample were prepared as per test method and injected into the chromatographic system. The chromatograms of blank, standard and sample were shown in the fig a, b, c.





Precision

Precision study was established by evaluating method precision and intermediate precision study. Method precision of the analytical method was determined by analyzing six sets of sample preparation. Intermediate precision of the analytical method was determined by performing method precision on another day and another analyst under same experiment condition. The % RSD was calculated. The %RSD range was obtained as 0.33 and 0.47 for method precision and intermediate precision respectively (Table 4) which is less than 2% indicating that the method is more precise.

Accuracy

The accuracy of the method was assessed by determination of recovery for three concentrations (corresponding to 50,100 and 150% of test solution concentration) covering the range of the method. For each concentration three sets were prepared and injected. The drug concentrations of Ivacaftor were calculated, the percentage recovery was found to be 99.46-99.93% with %RSD 0.05 - 0.31(<2.0%) indicating that the method is more accurate (table 2)

Linearity

The linearity plot was prepared with six concentration levels (30, 60, 120,150,180 and 300 μ g/ml of Ivacaftor). These concentration levels were respectively corresponding to 20, 40, 80,100,120 and 200 % of test solution concentration. The results obtained are shown in table 1. The peak areas were plotted against the corresponding concentrations to obtain the calibration curve (figure 4).

Robustness

Robustness of method was checked by making slight deliberate changes in chromatographic conditions like flow rate ($\pm 0.1 \text{ ml/min}$) mobile phase composition and column temperature ($\pm 5^{\circ}$ c). The results are tabulated in table 3.Under all the deliberately varied

chromatographic conditions, the reproducibility of results was observed to be reasonably good. Hence the proposed method has good robustness for the assay of Ivacaftor in bulk and dosage forms.

LOD and LOQ

The LOD and LOQ were determined at a signal to noise ratio of 3:1 and 10:1 respectively by injecting a series of test solutions of known concentrations within the linearity range. Precision study was also carried out at the LOQ level by injecting six pharmaceutical preparations. The LOD and LOQ were to be 0.13μ g/ml and 0.40μ g/ml respectively. The %RSD value was noticed to be less than 2.0% at LOQ concentration level.

Solution stability and Mobile phase stability

Solution stability checked for stability of standard and sample solutions. Solution stability checked at each interval initial 2,4,6,8,12,16,20 and 24 hours. For standard solution stability and sample solution stability %assay value calculated at each interval. %RSD (NMT 2.0%) between initial assay value and assay value obtained at predetermined time interval calculated.

Forced Degradation Studies

Stress studies on Ivacaftor were carried out under oxidation, thermal stress, photolysis, acid and alkali hydrolysis conditions. Significant degradation was observed in acid (fig 3g) and base (fig 3i) of Ivacaftor . There was slight significant degradation of Ivacaftor upon exposure to dry heat at 80°c for 7days and photolysis total impurity increased to 1.15% and 2.72%. In peroxide oxidation (fig 3k) no significant change was observed, which indicated that the drug was stable against these stress conditions. Purity plots of acid, base and peroxide (3h, 3j, 3l) revealed that there was no interference from the impurities, degradation products and excipients to determine the assay of drug substance in pure and pharmaceutical formulation.















0.50





Fig. 3: Typical chromatograms of (a) Blank (b) Standard (c) Sample (d) precision injections (e) Linearity injections (f) Accuracy injections (g) Acid sample (h) Purity plot of Acid (i) Base sample (j) Purity plot of Base(k) Peroxide sample(l) Purity plot of Peroxide

Table 1: Results for linearity of Ivacaftor

Linearity level	%Level	Area	
1	20	707017	
2	40	1434034	
3	80	2874072	
4	100	3605902	
5	120	4307043	
6	200	7271085	
Correlation co-efficient		0.99997	
	intercept	-32681.5	
	slope	36420.08	



Fig. 4: Linearity of Ivacaftor

Table 2: Recoveries study for Ivacaftor

Accuracy Level	Set No	Amount Added (µg/ml)	Amount Found (µg/ml)	Recovery (%)	Average recovery	Std Dev.	% RSD
	1	75.14	74.73	99.1			
50%	2	75.25	74.98	99.64	99.46	0.31	0.31
	3	75.09	74.82	99.64			
	1	150.35	149.89	99.69			
100%	2	150.23	149.98	99.83	99.79	0.09	0.09
	3	150.28	150.06	99.85			
	1	225.13	225.02	99.95			
150%	2	225.25	224.98	99.88	99.93	0.05	0.05
	3	225.18	225.12	99.97			

Table 3: Robustness results for Ivacaftor

Robust conditions	Variation	Retention time(min)	USP Tailing	USP Plate count
	0.9ml	4.02	1.38	5124
Flow	1.0ml	3.69	1.24	6137
	1.1ml	3.37	1.22	6235
	25°c	3.91	1.29	6100
Temperature	30°c	3.69	1.24	6137
-	35°c	3.52	1.17	6239
	75	3.12	1.13	6325
%Acetonitrile	70	3.69	1.24	6137
	65	3.97	1.32	5973

Table 4: Precision results for Ivacaftor

Study	Set no	Assay (%)	Mean assay (%)	Stdev	RSD%
	1	99.45			
	2	99.58			
Method precision	3	100.1			
	4	100.32	99.84	0.33	0.33
	5	99.69			
	6	99.89			
	1	99.45			
	2	100.26			
Intermediate precision	3	99.9			
	4	99.26	99.78	0.47	0.47
	5	99.42			
	6	100.38			

Table 5: Forced degradation results for Ivacaftor

Stress condition	Drug recovered (%)	Drug decomposed (%)
Standard drug	100	
Acid degradation	70.56	29.44
Alkali degradation	83.75	16.25
Oxidation degradation	100	0.00
Thermal degradation	98.85	1.15
Photolytic degradation	97.28	2.72

CONCLUSIONS

The isocratic new reverse phase HPLC method developed for the analysis of Ivacaftor in bulk and pharmaceutical dosage forms is selective, precise and accurate. The method is useful for routine analysis due to short run time. This method is free from interference of the other active ingredients and additives used in the formulation. Degradation impurities did not interfere with the retention time of Ivacaftor, and method is thus stability indicating.

ACKNOWLEDGEMENTS

The authors are grateful of M/S GITAM Institute of Science, GITAM University, Visakhapatnam, India for providing research facilities.

REFERENCES

1. Eckford PD, Li C, Ramjeesingh M, Bear CE. Cystic fibrosis transmembrane conductance regulator (CFTR) potentiator VX-

770 (ivacaftor) opens the defective channel gate of mutant CFTR in a phosphorylation-dependent but ATP-independent manne, J Biol Chem. 2012 Oct 26;287(44):36639-49

- Bob Lubamba | Barbara Dhooghe | Sabrina Noel | Teresinha Leal, Cystic fibrosis: Insight into CFTR pathophysiology and pharmacotherapy, Clinical Biochemistry, oct2012
- Tsukasa Okiyoneda, Guido Veit, Johanna F Dekkers, Miklos Bagdany, Naoto Soya, Haijin Xu,Ariel Roldan,Alan S Verkman, Mark Kurth,Agnes Simon,Tamas Hegedus, Jeffrey M Beekman & Gergely L Lukacs Mechanism-based corrector combination restores ΔF508-CFTR folding and function, *Nature Chemical Biology* 9,444–454(2013)
- 4. Mouawia h, Saker a, Jais jp, Benachi a, Bussieresl, Lacour b, Bonnefont jp, Frydman r, Simpson jl, Paterlini-Brechot, Circulating trophoblastic cells provide genetic diagnosis in 63 fetuses at risk for cystic fibrosis or spinal muscular atrophy, 2012 - Reprod Biomed Online 25(5):508-20

- B.Lakshmi, T.V.Reddy, A Novel RP-HPLC Method for the 5. Quantification of Ivacaftor in formulations, fist,021-033.
- 6. ICH, Q1(B), Harmonized Tripartite Guideline, Stability testing: Photostability Testing of New Drug Substances and Products, in: Proceeding of the International Conference on Harmonization, Geneva. Nov (1996).
- ICH Q2 (R1): Validation of analytical procedures Text and Methodology, Fed. Reg (19 May 1997) 62:27463 Snyder LR, Kirkland JJ, Glajch JI. PracticalHPLC Method Development.2nd ed.; 1997.p. 2-21 www.wikipedia.org/wiki/ Ivacaftor 7.
- 8.
- 9.
- 10. www.chemblink.com/products/ lvacaftor