

**IN VITRO FREE RADICAL SCAVENGING ACTIVITY AND IN VIVO ANTICATARACT POTENTIAL OF *HEMIDESMUS INDICUS* ROOTS IN STREPTOZOTOCIN INDUCED DIABETIC RATS**

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**ABSTRACT**

**Objective:** *Hemidesmus indicus* R. Br. (HI) (Indian sarsaparilla) is traditionally used in Indian medicine for human therapy. The present study was aimed to evaluate the antioxidant activity of HI root extracts, and its effects on delaying (or) management of cataract progression in streptozotocin (STZ) induced diabetic rats.

**Methods:** The plant roots were extracted in different solvents, estimated total phenolic content, total flavonoid content and conducted free radical scavenging potential through 1,1-Diphenyl-2-picrylhydrazyl (DPPH), 2,2 azinobis (3-ethyl benzothiazoline-6-sulphonic acid diammonium salt (ABTS) NO, OH and reducing power assays. Further, the protective effect was evaluated in STZ induced diabetic rats through administered of ethyl acetate extract of HI (200 and 400 mg/kg body weight) orally for 8 weeks to study its effect on oxidative stress and diabetic cataract progression.

**Results:** In primary results demonstrated that the ethyl acetate extract has been showed considerable free radical scavenging activity and significantly gain body weight, gradual decreasing blood glucose levels and delay cataract progression has been observed in HI treated rats. Furthermore, we observed significantly decreased oxidative stress markers, and restoration of the antioxidant system in HI treated rats. Lens sections from experimental animal again evidenced to conform the ameliorative effect of HI on diabetic cataract.

**Conclusion:** This obtained result's confirmed that the protective effect of HI is attributed to the antioxidant activities of plant roots and its active principles.

**Keywords:** *Hemidesmus indicus*, Free radicals, Antioxidant activity, Streptozotocin, Diabetic cataract.

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**INTRODUCTION**

Diabetic cataract, characterized by progressive opacification of the transparent eye lens, which is a part of the earliest secondary complication of diabetes [1]; leading to visual impairment and blindness worldwide predominantly in developing countries like India [2,3]. Due to the high prevalence of these countries, a diabetic cataract may pose a major problem in the management of blindness [4]. Evidence demonstrates that both the hyperglycemia and oxidative stress are well known for increase the burden of cataract formation by various risk factors [5]. Long-term hyperglycemia achieves greater oxidative stress by producing reduced sugars, and these reducing sugars react with lipids and proteins causing the cumulative production of reactive oxygen species (ROS) [6,7]. ROS are oxygen-derived free radicals such as hydroxyl radicals (OH•), superoxide anion (O<sub>2</sub>•), hydroperoxyl (OOH•), peroxy (ROO•), and alkoxy (RO•) and as well as singlet oxygen (<sup>1</sup>O<sub>2</sub>) [8]. Moreover, ROS could cause oxidative damages to biological macromolecules (proteins, lipids, DNA, etc.) which lead to degenerative processes and chronic diseases such as diabetic cataract [9,10]. Antioxidant compounds can prevent (or) repair the body's cells, especially through free radical scavenging, reducing agents, complexes of pro-oxidant metals and quenching of singlet oxygen. Consequently, increase the antioxidant capacity of the plasma and reduce the risk of diseases [11,12].

In recent times attention toward finding natural antioxidants have been significantly increased to replace synthetic antioxidants as medicinal products [13], which are said to be toxic to animals including human beings and the exploitation of the various secondary metabolites of the plants was highlighted. Plants constitute the main source of natural antioxidant molecules (phenols and flavonoids) which have the

capacity to neutralize the ROS [14] and protect the human body from free radicals and retard the progress of many chronic diseases [15,16]. Scientific examination and validation of the traditional therapeutic use of the plant medicines may be translated of new and effective drugs as occurred in the past. The continued search for natural antioxidants has gained importance in recent years because of the improving awareness of herbal remedies for various chronic diseases [17].

*Hemidesmus indicus* (L.) R. Br. (HI), belonging to family Asclepiadaceae, locally known as anantamul, twining shrub that has been used as folk medicine, Unani preparations, Indian Ayurveda system and its roots were used extensively to treat the different types of diseases in humans [18]. The root of HI considered to be demulcent, diaphoretic, diuretic, tonic and also utilized in loss of appetite, fever, skin diseases, leucorrhea, syphilis, and rheumatism. Earlier studies reported that an alcoholic extract of its roots possesses antithrombotic, antihepatotoxic, antinociceptive, anti-inflammatory, renoprotective, antidiarrheal, and antienterobacterial activities [19-23] and is found to be effective in the control of blood glucose levels through restoration of insulin production [24,25]. The root has been used for treating blood diseases, diarrhea, respiratory disorders, skin diseases, syphilis, fever, bronchitis, asthma, eye diseases, epileptic fits in children, kidney and urinary disorders, loss of appetite, burning sensation, rheumatism, and in gastric ailments [26,27]. Hence compounds, which can scavenge the excess of free radicals formed or inhibit their production or protect membranes from peroxidation [28]. However, the pharmacological effect of HI against secondary diabetic complications such as cataract has still not been studied. Therefore, in the present study evaluates that the antioxidant property of HI and its role in delaying cataract progression in streptozotocin (STZ) induced diabetic rats.

## MATERIALS AND METHODS

### Materials

STZ, quercetin, 1,1-diphenyl-2-picrylhydrazyl (DPPH), and 2,2-azino-bis(3-ethyl benzothiazoline-6-sulphonic acid diammonium salt (ABTS) obtained from Sigma-Aldrich (MO, USA) and sodium carbonate, sodium phosphate, potassium ferrocyanide, ascorbic acid, gallic acid, potassium persulfate, and Folin-Ciocalteu reagent were purchased from Merck Pvt Ltd (Mumbai, India). Riboflavin, nitro blue tetrazolium, pyrogallol and other reagents, analytical grade chemicals and solvents were obtained from SD Fine Chemicals, India, and indigenous companies in India.

### Collection and preparation of plant material

The roots of the plant HI were collected in July 2014 from the Nallamala forest, Mahaboob Nagar district, Telangana state, India. The plant was identified and authenticated by the chief taxonomist, Prof. P. Ramachandra Reddy, Department of Botany, Osmania University, Hyderabad, and a voucher specimen (Bat/OU/0119/HYD) was deposited in the department of botany for further reference. The roots of HI have been dried under shade for 1 month and powdered in a mechanical grinder into a coarse powder. About 300 g of powder was macerated in 1.5 l of four different solvents (methanol, ethyl acetate petroleum ether, and distilled water) in a dark room at room temperature for 5-8 days with intermittent shaking. The solvent extracts were filtered and then concentrated using a rotary evaporator and kept in a vacuum desiccator for complete removal of the solvent to yield a semi solid mass. The crude extracts were weighed and stored in the refrigerator until further use.

### Determination of total phenolic content (TPC)

The TPC of each plant extract was determined using the Folin-Ciocalteu reagent according to the method described by Ainsworth and Gillespie [29]. In this method, a volume of 1 ml of the plant extract (1 mg/1 ml) was added to 2 ml of Folin-Ciocalteu reagent (diluted 1:10 with distilled water) and was neutralized with 2 ml of 20% sodium carbonate solution. The mixture was permitted to incubate at room temperature for 30 minutes with intermittent shaking for color development. Then, the absorbance of the resulting blue color was measured at 765 nm using a double beam UV-VIS spectrophotometer (Hitachi U-2910). The TPC of all the extracts was determined by the linear equation of a standard curve prepared with gallic acid and calculated as mean±standard deviation (SD) (n=3) and expressed as mg GAE/g dry extract.

### Estimation of total flavonoid content (TFC)

TFC was determined using the form of a flavonoid-aluminum complex by a spectrophotometric assay as described earlier, with some modifications [30]. A volume of 0.5 ml of the plant extract (1 mg) was mixed with 0.1 ml of 10% aluminum chloride, 0.1 ml of potassium acetate (1 M), and 4.3 ml of 80% methanol were mixed. The reaction mixture was incubated at room temperature for 30 minutes for color development. The absorbance was measured at 415 nm using UV-VIS spectrophotometer (Hitachi U-2910). The calculation of total flavonoids in the extracts has been determined from the linear equation of a standard curve prepared with quercetin and calculated as mean±SD (n=3) and expressed mg QE/g of dry extract.

### Scavenging of DPPH radical

The effect of HI extracts on DPPH radical has been carried out, employing the method described earlier by Huang *et al.*, 2011 [31]. DPPH is a molecule containing a stable free radical, and in the presence of an antioxidant, which can donate an electron to DPPH, the purple color typical for free DPPH radical decays and the absorbance change is measured at  $\lambda=517$  nm. Briefly, the assay mixture contained 2 ml of 0.004% DPPH solution prepared in methanol and 0.2 ml of standard (or) sample solution of various concentrations (5-300  $\mu\text{g/ml}$ ) in methanol. The mixture was shaken vigorously and allowed to stand at room temperature for 30 minutes, and the absorbance of the resulting solution was measured at 517 nm using a spectrophotometer (Hitachi

U-2910). The radical scavenging activity was measured as a decrease in the absorbance of DPPH and was calculated using the following formula:

$$\text{DPPH scavenging activity (\%)} = (A_{\text{Control}} - A_{\text{Sample}}) / A_{\text{Control}} \times 100$$

### Nitric oxide (NO) scavenging activity assay

Griess Ilosvay reaction is used to determine the nitrite ions which are generated by aqueous sodium nitroprusside solution interacts with oxygen and NO at physiological pH [32]. The pink chromophore formed of the reaction mixture was read at 546 nm against the blank. The percentage inhibition of activity was calculated using the following formula:

$$\text{NO scavenging activity (\%)} = (A_{\text{Control}} - A_{\text{Sample}}) / A_{\text{Control}} \times 100$$

The extract concentration providing 50% inhibition ( $\text{IC}_{50}$ ) was calculated and obtained by interpolation from a linear regression analysis.

### Hydroxyl radical scavenging activity

The reaction mixture, containing HI extracts (0-300  $\mu\text{g/ml}$ ), was incubated with EDTA (0.1 mmol),  $\text{FeCl}_3$  (0.1 mmol),  $\text{H}_2\text{O}_2$  (1 mmol), deoxyribose (3.75 mmol), and ascorbic acid (0.1 mmol) in potassium phosphate buffer (20 mmol, pH 7.4) for 60 minutes at 37°C [33]. The reaction was terminated by adding 1.0 ml of trichloroacetic acid (2% w/v), 1.0 ml of thiobarbituric acid (TBA) (1% w/v) and then heating the tubes in boiling water 15 minutes. The contents have been cooled to develop the pink color measured at 535 nm against the blank. Decreased absorbance of the reaction mixture indicates increased antioxidant activity.

### Determination of ABTS radical scavenging activity

The free radical scavenging activity of HI was measured by discoloration of  $\text{ABTS}^+$  radical cation using the modified method [34].  $\text{ABTS}^+$  radical cation ( $\text{ABTS}^+$ ) was produced by oxidation of ABTS (7 mM) with potassium persulfate (2.4 mM). Plant extracts (1 ml) have been allowed to react with 2.5 ml of ABTS solution ( $0.703 \pm 0.005$  at 734 nm), and ascorbic acid was used as a standard. Percentage inhibition was calculated using this formula:

$$\text{ABTS radical scavenging activity} = (A_{\text{Control}} - A_{\text{Sample}}) / A_{\text{Control}} \times 100$$

Where,  $A_{\text{Control}}$  is the absorbance of ABTS radical with methanol and  $A_{\text{Sample}}$  is the absorbance of ABTS radical with extract/standard and data expressed mean±SD (n=3).

### Determination of reducing power

The reducing power of the plant extracts was determined using the earlier described method [35]. Briefly, different amount of extract (0-300  $\mu\text{g}$ ) in 1 ml of methanol was mixed with 2.5 ml of phosphate buffer (0.2 M, pH 6.6), 2.5 ml of potassium ferricyanide (10%). The reaction mixture was incubated at 50°C for 20 minutes after centrifugation for 10 minutes at 3000 rpm. Further 2.5 ml of the supernatant solution, 2.5 ml of distilled water and 0.5 ml of  $\text{FeCl}_3$  (0.1% freshly prepared) were mixed properly, and absorbance was read at 700 nm. Increased absorbance indicating stronger reducing power and the results have been compared with that of the rutin as standard.

### Experimental animals and treatment

2-month-old male Sprague-Dawley rats with an average body weight of  $180 \pm 8$  g (obtained from the National Center for Laboratory Animal Sciences, National Institute of Nutrition, Hyderabad, India.) have been used for this study under standard laboratory conditions. All the animals were fed with a semi synthetic AIN-93 diet [14] and water *ad libitum* throughout the experimental period. Animal care and experimental protocols were approved by the Institutional Animal Ethical Committee. Diabetes mellitus has been induced using an intraperitoneal injection of STZ, at a single dose of 35 mg/kg body weight dissolved in 0.1 M citrate buffer (pH 4.5). The

fasting blood glucose levels were monitored, and these levels over than 200 mg/dl have been considered as diabetic model rats.

The experiment was carried out in four groups of eight rats each:

- Group I: Control rats (received sham, 0.1 M citrate buffer, pH 4.5)
- Group II: Diabetic control (STZ induced, single i.p. 35 mg/kg bw)
- Group III: Diabetic rats, administered orally with ethyl acetate extract of HI (EAHI) at a low dose (200 mg/kg bw/day)
- Group IV: Diabetic rats, administered orally with EAHI at a high dose (400 mg/kg bw/day).

Blood glucose level was recorded using glucose monitoring system (Dr Morepen GlucoOne kit) once a week after an overnight fasting. The body weights have been taken using a digital balance and cataract progression were monitored on a weekly basis using a slit lamp microscope (VISIL 02 Excel) [5].

#### Preparation of lens homogenate

At the end of the experimental period (8<sup>th</sup> week), the rats were sacrificed by CO<sub>2</sub> asphyxiation, and the lenses were dissected by posterior approach then placed into pre-weighed Eppendorf tubes and frozen at -80°C until further analysis. A 10% homogenate was prepared from the lens in 50 mM phosphate buffer (pH 7.4) and the activity of the lens enzymes and soluble protein was measured in the soluble fraction of the lens homogenate (15,000×g at 4°C/20 minutes) while the lens TBA reactive substances (TBARS) levels and total protein have been determined in the total homogenate.

#### Estimation protein content and *in vivo* antioxidant levels

The reduced glutathione (GSH) was estimated by the method of Ellman, 1959 [36]. The total reaction was measured at 412 nm against blank and values were compared with a standard curve of GSH. Lipid peroxidation is based on the reaction of malondialdehyde with TBA to form TBARS, which have a pink color with absorption maxima at 540 nm and TBARS levels in the lens homogenate has been analyzed using the method of Matsunami *et al.* [37]. Protein carbonyl groups were estimated by the earlier described method [38], and the concentration of protein carbonyls was calculated using a molar extinction coefficient ( $\epsilon$ 365 nm=21 mM<sup>-1</sup> cm<sup>-1</sup>). The total and soluble proteins have been estimated in the lens by the using method of Lowry *et al.*, 1951 [39]. The reaction has been measured at 540 nm against the blank and values are calculated from a standard curve of bovine serum albumin. The specific activity of the antioxidant enzyme superoxide dismutase (SOD) was estimated spectrophotometrically [40]. The total reaction was carried out at 25°C for 3 minutes, and change in absorbance at 420 nm with a suitable blank was also recorded. One unit of SOD activity is defined as the amount of the enzyme required to inhibit 50% of autoxidation of pyrogallol. Catalase (CAT) activity was assayed spectrophotometrically described by earlier described method [41]. The reaction mixture was measured at 570 nm against control, and the activity was expressed as units per milligram protein (one unit is the amount of enzyme that utilizes 1 mmol of H<sub>2</sub>O<sub>2</sub>/min). GSH peroxidase (GPx) activity was measured using method [42]. Moreover, this assay result depends on the rate of GSH oxidation by H<sub>2</sub>O<sub>2</sub>, as catalyzed by the GPx present in the supernatant and it was read at 430 nm.

#### Histopathological examination

For histopathological examinations, eyes have been collected from rats by posterior approach in each experimental group and preserved in 10% formaldehyde. Formalin-fixed eyes were collected and dehydrated in a graded series of ethanol (70-100%), cleared in xylene, and finally embedded in paraffin. Thereafter, 5 mm thin sections were prepared and stained with hematoxylin and eosin (H and E) for histopathological examination [43]. The photomicrographs of the respective tissue sections were taken using Olympus BX41 research optical microscope fitted with Olympus DP 25 digital camera.

#### Statistical analysis

All data have been expressed as the mean±SD. Statistical significance was performed by ANOVA followed by Tukey's test and p≤0.05, was considered to be statistically significant using SPSS 20.

## RESULTS

#### Extraction yield, TPC and TFC

Table 1 summarizes the percentage of yield of crude successive extracts (ethyl acetate, methanol, petroleum ether, and aqueous) of HI and results revealed that methanol extract found to be higher yield than those of the other extracts. Results of the TPC of various extracts of HI have been significant and shown in Table 1, and it was calculated using the following linear equation based on the calibration curve of gallic acid;  $y=0.0009x-0.013$ ,  $R^2=0.9981$ . TFC was calculated as quercetin equivalents as showed in Table 1. This study showed the highest flavonoid content in the EAHI, and the results have been calculated using the following linear equation based on the calibration curve of quercetin;  $y=0.0025x+0.0681$ ,  $R^2=0.9905$ .

#### Free radical scavenging potential of HI extracts

The effects of extracts on different free radicals are represented in Fig. 1. All the extracts demonstrated significant scavenging abilities which varied depending on the solvent/dose used and their results (IC<sub>50</sub> values) have been shown in Table 2. For the DPPH radical, EAHI showed a higher inhibitory potential along with the standard (L-ascorbic acid) as described in Fig. 1a; while for the ABTS assay, the methanol extract of HI (MEHI) showed a higher radical scavenging activity with a low IC<sub>50</sub> value (19.66±1.01µg/ml) than those of other extracts Table 2 and the results have been expressed as % inhibition (Fig. 1b). However, all the extracts' has been found to be significant (p≤0.05) in relation to that of ascorbic acids. The results NO scavenging potential of the different extracts and the standard ascorbic acid represented in Fig. 1c. Among them, EAHI showed the highest NO inhibitory potential with a low IC<sub>50</sub> value (167.36±1.66 µg/ml) as compared with standard ascorbic acid (60.91±0.79 µg/ml) as represented in Table 2. Hydroxyl radical scavenging activity of various extracts has been presented in Fig. 1d, using rutin as standard.

The EAHI extract showed a good activity in depleting H<sub>2</sub>O<sub>2</sub>, with an IC<sub>50</sub> value of 198.69±0.42 µg/ml as compared with well-known flavonoid, rutin (80.25±3.15 µg/ml) and represented in Table 2. The reductive activities of the various plant extracts are shown in Fig. 2. We noticed that EAHI extract had the higher reducing ability than among other extracts. According to their ability, they can be arranged in order as BHT > EAHI > MEHI > AQHI > petroleum ether HI extracts.

#### Pharmacological evaluation of HI in STZ induced diabetic rats

Since EAHI roots showed greater antioxidant potential than among other extracts, we tested its potential further study on diabetes and its complications in STZ induced rodent model. All rats (Group I-IV) have been fed *ad libitum*, and it was noticed that there was an increase in food intake in all diabetic rats (Group II-IV) compared to the control rats. Despite increased food intake, the body weight of diabetic control rats was found to be low (Group-II 168.03±3.36 g), in relation to the control rats (284.55±0.39 g), and HI fed rats (Group III 202.35±2.53 g and Group IV 264±0.82 g), respectively.

Fig. 3a demonstrates the levels of blood glucose in diabetic control rats have been significantly higher than in relation to control rats and oral administration of EAHI suppressed the elevated of blood glucose levels, which is significantly lower as compared to diabetic control rats (p<0.005). Control group lenses had no signs and symptoms of cataract development throughout the experimental period; however, cataract progression has been seen in diabetic control rats over a period of 8 weeks, which were significantly delayed cataract progression in rats supplemented with the EAHI (Group III and IV). Fig. 3b has been shown the representative stages of cataract progression and photographs have been showed the opacification of the lens (Fig. 4) at the end of the experiment (8<sup>th</sup> week).

#### Effect of EAHI on oxidative stress and enzymatic alterations in lens

Further, we assessed oxidative stress by measuring TBARS (a lipid oxidation product), protein carbonyls (oxidative products of protein),



**Table 1: Quantitative analysis of yield, TPC and TFCs in roots of HI, using gallic acid and quercetin as common reference compounds, respectively**

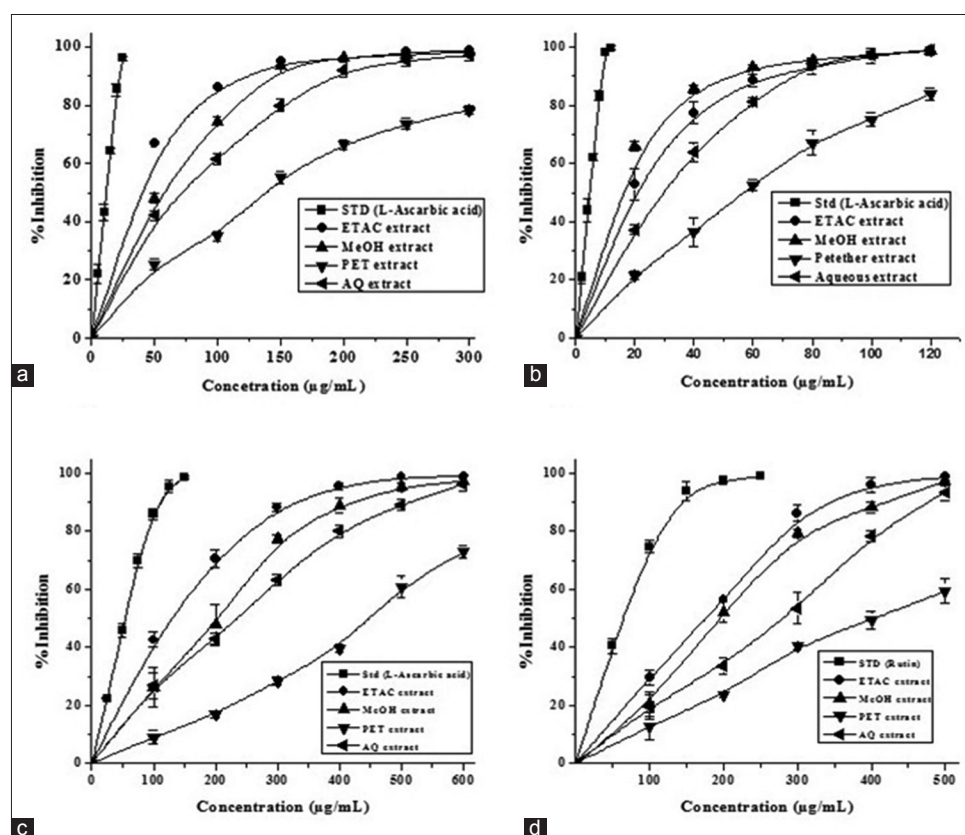
Extract	TPC (mg/g GAE of dry extract)	TFC (mg/g QE of dry extract)	Yield of extraction (% w/v)
Methanol	114.23±7.31*	44.35±1.43*	12.8
Ethyl acetate	94.00±6.38*	70.24±4.51*	3.9
Petroleum ether	10.11±1.12*	7.65±0.2*	8.2
Aqueous	27.34±4.54*	27.41±2.07*	5.1

Values were represented as mean±SD (n=3). \*Denotes in column are statistically significant (p<0.05). Values were represented as mean±SD (n=3). Superscript \* denotes in column are statistically significant (p<0.05). GAE: Gallic acid equivalent. QE: Quercetin equivalent, TPC: Total phenolic content, TFC: Total flavonoid content, HI: *Hemidesmus indicus*

**Table 2: Antioxidant activity of HI roots in different solvents systems**

No	Extraction	IC <sub>50</sub> values (µg/mL)			
		DPPH	ABTS	NO	Hydroxyl radical
1	Ethyl acetate	46.42±1.15*	28.77±3.28*	167.36±1.66*	198.69±0.42*
2	Methanol	73.87±2.26*	19.66±1.01*	236.78±13.54*	219.83±7.02*
3	Pet ether	158.69±2.02*	62.7±1.12*	425.31±27.49*	408.49±21.6*
4	Aqueous	94.78±3.78*	37.94±1.05*	257.51±6.93*	266.43±9.88*
5	L-ascorbic acid	12.02±0.22	5.06±0.09	60.91±0.79	NA
6	Rutin	NA	NA	NA	80.25±3.15

Values were represented as mean±SD (n=3). \*Denotes statistically significant (p<0.05) compared with standard. NA: Not analyzed, NO: Nitric oxide, DPPH: 1,1-diphenyl-2-picrylhydrazyl, ABTS: 2,2 azinobis (3-ethyl benzothiazoline-6-sulphonic acid diammonium salt, IC<sub>50</sub>: Inhibitory concentration 50%, SD: Standard deviation, HI: *Hemidesmus indicus*



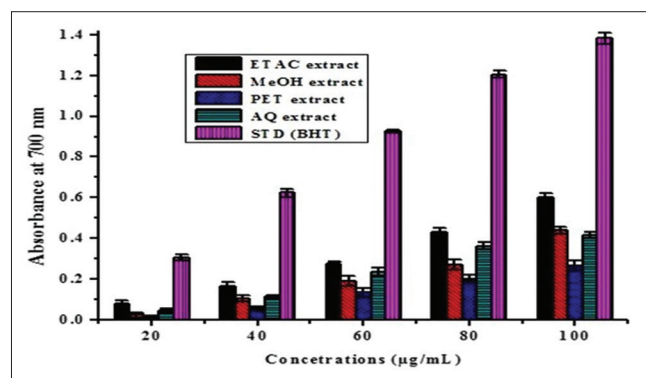
**Fig. 1: Percentage scavenging activities of roots of *Hemidesmus indicus* and its fractions through 1,1-diphenyl-2-picrylhydrazyl (a), 2,2 azinobis (3-ethyl benzothiazoline-6-sulphonic acid diammonium salt) (b), nitric oxide (c), and hydroxyl radical (d) assays compared with referenced standards. Data were expressed as mean±standard deviation, (n=3) (p<0.05). ETAC: Ethyl acetate, MeOH: Methanol, PET: Petroleum ether, and AQ: Aqueous**

GSH (a cellular antioxidant), and antioxidant enzymes such as SOD, CAT, and GSH peroxidase (GPx) in the rodent lens. The levels of TBARS, protein carbonyls have been increased and GSH levels have been reduced in the lenses of diabetic control rats. Interestingly, rat lenses of

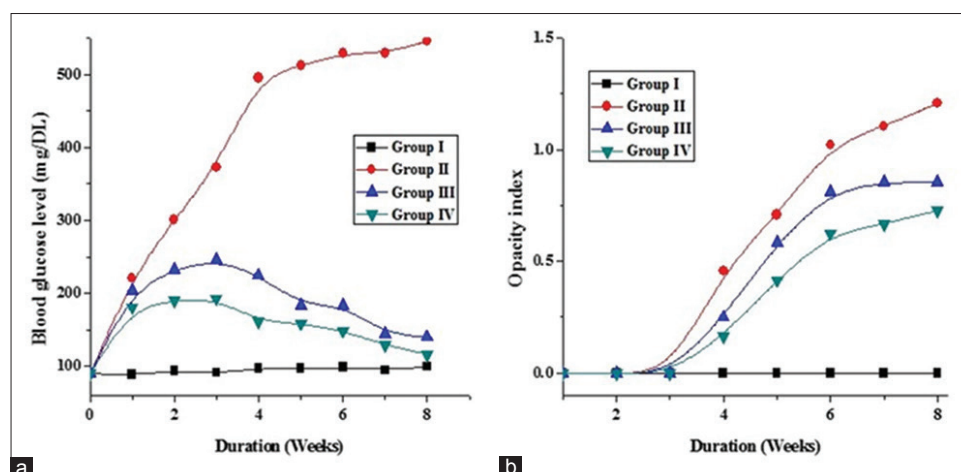
treated with EAHI prevented lipid and protein oxidation as evidenced by significantly decreased TBARS, protein carbonyls and the loss of GSH (Table 3). Antioxidant enzymes such as SOD, CAT and GPx have been significantly decreased in the lens of diabetic control rats, and their

enzymatic activities were restored EAAHI treated rat lenses, which were significantly higher ( $p < 0.05$ ) as compared to diabetic groups (Table 3).

The underlying mechanism, insolubilization of proteins have been considered to be the ultimate change that results in lens opacification; therefore, we analyzed the total and soluble protein content in all group rat lenses. There was a significant decrease in both total and soluble protein in diabetic control rats compared with the control rats, and supplementation of EAAHI rats improved the percentage of soluble protein as showed in Fig. 5.



**Fig. 2: Antioxidant activity of roots of *Hemidesmus indicus* through the reducing power assay in different solvent systems and compared with known standard (BHT). Data were expressed as mean±standard deviation, (n=3) ( $p \leq 0.05$ ). ETAC: Ethyl acetate, MeOH: Methanol, PET: Petroleum ether, AQ: Aqueous, and BHT: Butylated hydroxytoluene**



**Fig. 3: Ethyl acetate extract of *Hemidesmus indicus* (EAAHI) prevented diabetic cataract progression in streptozotocin induced diabetic in rats. The effect of EAAHI on blood glucose levels (a) and grading of cataract progression was performed by slit-lamp examination (b) a weekly basis. The data are the mean±standard deviation (n=6)**

**Table 3: Ethyl acetate extract of HI prevented oxidative stress and restoration of antioxidant enzymes in STZ induced diabetic in rat lens**

Parameters	Group-I	Group-II	Group-III	Group-IV
GSH	14.44±0.41	6.49±0.14*	8.27±0.12**	11.65±0.34**
TBARS	8.92±0.41	19.42±0.89*	14.74±0.33**	11.47±0.73**
Protein carbonyls	13.46±0.29	32.03±0.63*	23.96±0.47**	17.66±0.37**
SOD	4.53±0.74	1.33±0.58*	1.86±0.20*	3.61±0.31#
CAT	8.76±0.57	3.71±0.24*	5.58±0.52**	7.88±0.25#
GPx	24.20±2.41	16.51±0.85*	18.18±0.82*	21.26±1.29#

After 8 weeks duration of experiment, lenses were homogenized and GSH ( $\mu\text{mol/g lens}$ ), TBARS ( $\text{nmol/g lens}$ ), protein carbonyls ( $\text{nmol/mg protein}$ ), SOD ( $\text{units}/100 \text{ mg of protein}/\text{min}$ ), CAT ( $\text{units}/100 \text{ mg of protein}/\text{min}$ ), and GPx ( $\mu\text{mol of NADPH oxidized}/\text{h}/100\text{mg of protein}$ ) have been estimated. The data are the mean±SD (n=3). \*Indicates statistically significant from Group I (analyzed by ANOVA;  $p \leq 0.05$ ). #Indicates statistically significant from Group II (analyzed by ANOVA;  $p \leq 0.05$ ). HI: *Hemidesmus indicus*, GSH: Glutathione

### Histopathological evaluation

Histopathological examinations in lens tissues of all rats have been observed using H and E stain. The control group rats showed the normal morphology of the lens fibers (Fig. 6a) and were in diabetic control rat lens gain more staining in the area of lens fibers, which are indicating protein aggregation and cataract progression (Fig. 6b). The supplementation of EAAHI rat lens (Group III IV) showed lower protein aggregations in lens fibers as control lens. These observations are suggesting that delaying cataract progression and ameliorative effect of EAAHI confirmed with dose-dependent (Fig. 6c and d).

### DISCUSSION

Diabetes and hyperglycemia are primary causal factors, play a key role in the onset of diabetic complications such as cataract which contribute to blindness worldwide [44,45]. Recent studies reported that diabetes could become a major threat to public health and the management of cataract blindness [46]. Cataract is a long-term complication of diabetes, the only remedy for the treatment of such kind disease is surgery with undesirable defects. Several studies reported that oxidative stress and hyperglycemia have been suggested as common underlying mechanisms of cataractogenesis due to diabetes and augmentation of the antioxidant defenses of the lens have been shown to prevent or delay cataract [47,48]. Therefore, any strategy that prevents (or) slow down the progression of cataract can have a significant effect on human health. Since ancient time, medicinal plants are a major source of secondary metabolites, which play a key role in the highly effective free radical scavengers and antioxidant and biological activities [49]. The antioxidant activity of alcoholic extract of HI has been demonstrated earlier in numerous methods [50-52]. However, the potential of free radical scavenging ability of HI roots in chemical and biological systems has not been comprehensively investigated. In this study, we primarily

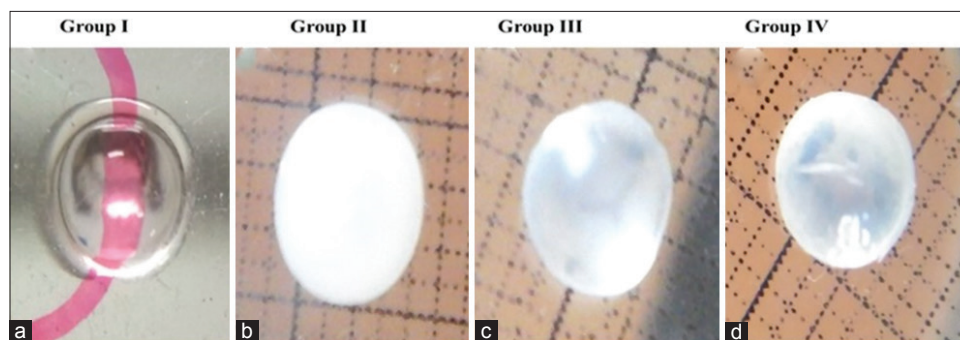


Fig. 4: Photographs show the effect of ethyl acetate extract of *Hemidesmus indicus* on the delaying of cataract progression in streptozotocin induced rat lenses after 8 weeks post-treatment. (a) Group I, (b) Group II, (c) Group III, and (d) Group IV

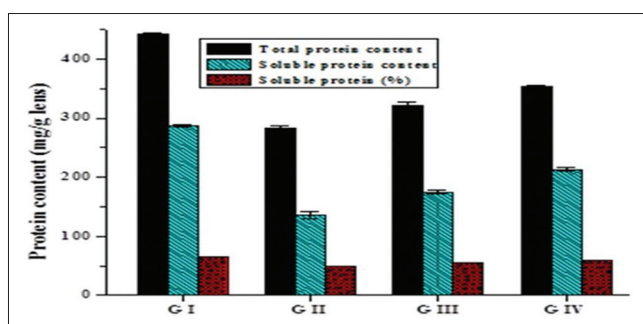


Fig. 5: Effect of ethyl acetate extract of *Hemidesmus indicus* treatment on the protein content of streptozotocin induced diabetic rat lens. Values are given as mean  $\pm$  standard deviation (n=3)

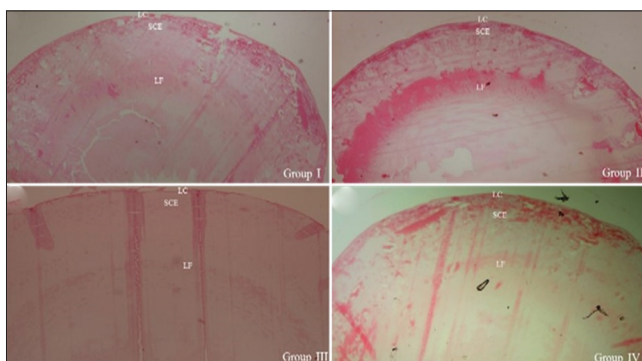


Fig. 6: The end of the experiment eyes were collect and dissected from different groups of rats (Groups I-IV), histopathological studies were performed by Hematoxylin and Eosin stain and examine under a microscope (Olympus CX21) for histological changes

investigated the amount of phytoconstituents (TPC and TFC) and free radical scavenging activity of different extracts of HI by the methods of DPPH, ABTS, NO, OH, and reducing power in *in vitro*. The significance of this observation was further investigated the EAHl extract on hyperglycemia and cataract progression in STZ induced diabetic rat model.

In recent times, the search for phytochemicals (phenols and flavonoids) has been rising due to their potential use in the therapy of various chronic and infectious diseases [53]. Looking back to our results it was seen that ethyl acetate most effective solvent for extracting phenolics from HI followed by methanol. The total phenolic and flavonoid contents of the ethyl acetate extraction have been higher than those of methanol, petroleum ether, and aqueous extracts. This result could

be explained the high content of total phenolics in EAHl indicated the potent antioxidant properties of this part. In this study, it appeared that the higher TPC and TFC of the plant extracts resulted in higher antioxidant activity which could be attributed to the redox properties such as reducing agents, hydrogen donors, free radical scavenger, singlet oxygen quenchers, and metal chelators [54].

DPPH is a stable free radical and can accept an electron or hydrogen radical to become a durable diamagnetic molecule. All the plant extracts tested inhibited the DPPH radical and result proved that the extracts are in a position to donating an electron or hydrogen which could react with DPPH radical in different levels. This difference could be attributed to an unequal contest of antioxidant molecules such as polyphenol, flavonoids dissolve based on the polarity of solvents. The EAHl had the lowest  $IC_{50}$  ( $46.42 \pm 1.15 \mu\text{g/mL}$ ) indicates greatest scavenging activity compared with other extracts. Moreover, antiradical activity of the extracts to trap DPPH radical depends on the availability and the ability of these extracts act as proton donors to the presence of polyphenols content [55]. ABTS is oxidized by potassium persulfate to its cation radical  $ABTS^+$ , which is intensely colored. Our results showed that all the tested samples demonstrated a scavenging potential of the ABTS with varying inhibitory potential and interestingly, the MEHI demonstrating the highest ABTS inhibitory power and this result confirmed that the presence of higher phenolic level in the extract justified its highest scavenging property on ABTS radical [56]. Eventually, a good relationship was also observed between phenolics and antioxidant activity which further confirms this view [57]. NO is an important chemical mediator that is involved in several biological processes, and an elevated production of NO could lead to several diseases. According to this study, EAHl showed higher NO radical scavenging power while remaining extracts depending on their phenolic composition. Earlier studies reported that the antioxidant activity of phenolic acids is related to the quantity, number, and position of hydroxyl groups in the molecule [58]. Hydroxyl radicals are highly reactive in nature, and it's capable to damage biological molecules such as DNA, lipids, and proteins to increase oxidative damage in the human body [59]. Our results demonstrated that EAHl exhibited a highest OH scavenging power compared with other extracts. It can, therefore, be assumed that these extracts have good antioxidant properties. Studies recommended several testing methods for the investigation of antioxidant activity of extracts from natural resources [60]. Regarding our results, EAHl exhibited highest reductive activity at different concentrations, and this extract has good electron donors for power reducing activity. Overall results conclude that, among four different extracts employed in this present study, EAHl showed significant antioxidant activity. Moreover, studies have also been extended to know the efficacy of EAHl root in STZ induced diabetic rats.

STZ is a chemical agent used to induce hyperglycemia in experimental animals mainly due to the rampant generation of ROS, selective pancreatic islet b-cell cytotoxicity and the cytotoxic events subsequently cause b-cell necrosis [61,62]. In this study, STZ



significantly induced hyperglycemia, and oral administration of the EAHI (200 and 400 mg/kg body weight) for 8 weeks affected a significant decrease in blood glucose levels and gain body weight. STZ induced is characterized by a severe loss of body weight which was observed in diabetic control rats and EAHI treated rats (Group III and IV) showed significant improvement in body weight ( $p \leq 0.05$ ) when compared with diabetic control rats. The loss of body weight in diabetic control rats might be the results of protein wasting due to unavailability of carbohydrate for utilization as an energy source [63]. The present study confirms the antihyperglycemic property of HI roots in STZ induced diabetic rats. Administration of crude of EAHI at a dosage of (200 and 400 mg/kg body weight) tended to bring blood glucose levels toward near normal levels. However, the extract did not produce any hypoglycemic effect in normal rats. Hence, the EAHI may be considered to have a good antihyperglycemic active principle(s) without causing any hypoglycemic effect unlike insulin, and other synthetic drugs and the safe dose for HI administration has been reported previously [25]. However, the serum glucose level of diabetic rats treated with HI is still high and is more than the threshold (150 mg/dl) that is necessary for the induction of cataract [64]. Our results indicate that cataract was established in the diabetic group after the 5<sup>th</sup> week of STZ injection and progressed up to the end of the study. However, the significant decrease in the cataract score is seen in Group III and IV due to the EAHI administration. Therefore, the observed delay of the onset cataract progression of after the administration of HI in diabetic rats is possibly due to other factors in addition to its glucose lowering property. To investigate this hypothesis, numerous biochemical pathways related to the formation of diabetic cataract have been investigated.

During the progression of the cataract, the total and soluble protein content of the diabetic lenses was reduced significantly. The reduction may be caused by the formation of high molecular weight aggregates of proteins, oxidative stress, cross-linking of proteins, and protein leakage from osmotic stress [65,66]. As our results, the treatment of rats with HI considerably prevented the total and soluble protein loss in diabetic lenses. The present study showed that increase in protein carbonyls, TBARS and decreased GSH levels may be assigned to oxidative stress in diabetic control rat lenses. Moreover, then treated with EAHI have been proven to prevent accumulation of TBARS and protein carbonyls and preventing the loss of GSH. Further, altered activities of the antioxidant enzymes, including SOD, CAT, and GPx are due to the increase of oxidative stress in diabetic conditions as reported previously [67]. Our results demonstrate that the EAHI, with its antioxidant activities, could significantly increase the activity of SOD, CAT, and GPx enzymes in diabetic lenses. The decrease of enzymes activities in the diabetic control group suggests that the molecules responsible for the antioxidant enzyme activities in that medium were exhausted by the higher concentration of the oxidants [68] and these results agreement with earlier reports [69]. Furthermore, the histopathological examination showed that the treatment of diabetic rats with lower and higher doses of EAHI extracts protected the lens structures.

## CONCLUSION

Extracts of HI have a good amount of polyphenolics, and significant free radical scavenging ability. After completion of the primary study, we selected EAHI to further investigation on antioxidant, antihyperglycemic, and anticataract potential in the STZ induced diabetic rats. Overall results confirmed the antioxidant activities of HI root extract led to inhibition of physiological, molecular and histopathological alterations in diabetic rats. Nevertheless, further studies need to be investigated to isolate and characterize the bioactive compounds responsible for these activities and its constituents as promising therapeutic agents for DM and its complications.

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