

ISOLATION AND IDENTIFICATION OF NATURALLY OCCURRING LARVICIDAL COMPOUND ISOLATED FROM *ZINGIBER ZERUMBET* (L).J.E. SMITH

TRI MURINI¹, MAE SRI HARTATI WAHYUNINGSIH^{1*}, TRI BASKORO T SATOTO², ACHMAD FUDHOLI³, MUHAMMAD HANAFI⁴

¹Department of Pharmacology and Therapy, Faculty of Medicine, Universitas Gadjah Mada, Indonesia, ²Department of Parasitology, Faculty of Medicine, Universitas Gadjah Mada, Indonesia, ³Department of Technology Pharmacy, Faculty of Pharmacy, Universitas Gadjah Mada, Indonesia, ⁴Research Centre for Chemistry, Indonesian Institute of Sciences (LIPI), PUSPIPTEK Serpong, Banten 13510, Indonesia.
Email: maeshw@ugm.ac.id

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ABSTRACT

Objective: A naturally occurring larvicidal compound is an alternative to eradication of *Aedes aegypti* larvae. This compound is toxic to larvae but relatively safe for human. Rhizome of *Lempuyang gajah* (*Zingiber zerumbet* (L.) J. E. Smith) has been used traditionally to prevent mosquito bites. An initial study indicated that petroleum ether (PE) soluble fraction of the methanol (MeOH) extract of *Z. zerumbet* was toxic against larvae lethal concentration (LC₅₀), 67.01±2.35 versus 153.57±4.01 ppm (MeOH ext.). Therefore, this study aimed to isolate and identify the compounds with larvicidal activity from *Z. zerumbet* rhizome.

Methods: The PE soluble (PE-soluble) fraction was subjected to a bioassay-guided fractionation and isolation method to obtain four (4) fractions (I-IV). Two compounds (1 and 2) were isolated from Fraction I that was the most active compared to other fractions (II-IV). Compound 1 turned out to be more active than compound 2; therefore, compound 1 was then identified by means of spectroscopic data.

Results: Based on the LC₅₀ values, compound 1 was more active than compound 2 (LC₅₀ 41.75±0.05 and LC₉₀ 57.66±3.37 [1] versus 1122.27±1.80 and 1875.69±1.35 ppm [2]), and compound 1 was identified as Zerumbone.

Conclusion: Zerumbone was the main active compound; in the future, this compound can be formulated as a standardized preparation based on the content.

Keywords: *Zingiber zerumbet*, Larvicidal, Lethal concentration 50, Identification, Zerumbone.

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INTRODUCTION

Dengue fever remains a major health problem in third world countries. Approximately 100 million cases of dengue fever are estimated to occur annually in the world [1]. No ideal treatment is currently available for eliminating the viruses that cause dengue fever. Some efforts to discover dengue vaccines have recently been made, but there are no clear signs of success. In the past few decades, chemical insecticides have been used to control mosquitoes as a dengue vector. However, this method is not recommended due to the risk of physiological and biochemical resistance [2,3].

Safe and environmentally-friendly insecticides are, therefore, more encouraged, including the use of naturally-occurring compounds [4], such as *Lempuyang gajah* (*Zingiber zerumbet*) that is traditionally used to protect the skin from mosquito bites. A preliminary study of this species found that the rhizome of *Z. zerumbet* contained a variety of naturally-occurring compounds, such as terpenoids and phenolic compounds, including flavonoids, alkaloids, and saponins [5-7]. In addition, the dichloromethanolic extract of *Z. zerumbet* rhizome tested against *Artemia salina* Leach (Brine shrimp lethality test) resulted in 40.4 µg/mL lethal concentration (LC₅₀), while on *Anopheles nuneztovari*, the LC₅₀ reached 62.8 µg/mL [8]. These values indicate that the rhizome of *Z. zerumbet* has the potential to be developed into a larvicide.

Previous research on the effects of methanolic extract of *Z. zerumbet* rhizome on *Aedes aegypti* larvae found the value of LC₅₀ reaching

153.57±4.01 ppm. In addition, the petroleum ether (PE)-soluble fraction of the methanolic extract performed even higher activities (LC₅₀ = 67.01±2.35 ppm) compared to the methanolic extract [9]. Determining the structures of the active compound in the PE-soluble fraction is, therefore, deemed crucial to discover a safe and environmentally-friendly larvicide. The objective of this study was to explore deeper the results of the previous research to find out isolat of *Z. zerumbet* (L) J.E. Smith, which possess larvicidal activity, and what compound responsible for action.

METHODS

Z. zerumbet rhizomes were obtained from Jatimulyo, Kulon Progo District of Yogyakarta, in May 2014. The rhizome was authenticated at the Department of Pharmaceutical Biology (BF/284/Ident/Det/VI/2014), the Faculty of Pharmacy of Universitas Gadjah Mada, Yogyakarta – Indonesia. All the organic solvents were of pro-analytical grade, and the instar larvae III-IV of *A. aegypti* were obtained from the Parasitology Laboratory of the Faculty of Medicine, Universitas Gadjah Mada according to standard [1].

The instruments used were rotary vacuum evaporator (Heidolfv 2000, Germany), oven (Mettler, Germany), and silica gel GF₂₅₄ (E Merck). The ultraviolet (UV) spectrum was recorded on UV spectrometer (Shimadzu UV-365), while IR spectrum was on Perkin Elmer Spectrum 1000, nuclear magnetic resonance (NMR) spectra were observed using Bruker HX 500, and mass spectrum was recorded using Shimadzu GC-17, QP-5000.

Larvicidal bioassay

Preparation of *A. aegypti* colony

Colonization of *A. aegypti* was done in the Parasitology Laboratory, Faculty of Medicine using [1] procedure. Adult *A. aegypti* was put in a cage completed with an aspirator and a cup containing water for the mosquito to lay eggs. The *A. aegypti* eggs were then placed on a tray containing 1500 mL of water, and after 1–2 days the eggs hatched into larvae. During the process of colony formation, a chicken liver was added as the food. Identification of instar larvae III–IV of *A. aegypti* was performed 7 days after the eggs hatched, observed macroscopically and using a loupe whenever necessary.

Larvicidal activity test

The test used 7 concentrations for Fraction I and the isolated compounds (40, 44.16, 48.75, 53.82, 59.42, 65.60, and 75 ppm), while for Fractions II, III, and IV, concentrations at 1000, 1120, 1254.40, 1404.93, 1573.52, 1762.34, and 2000 ppm were applied [10]. Each concentration was suspended with a couple drops of tween-80, and then distilled water was added until 100 mL in a 200 mL flask. Each flask contained 25 instar larvae III–IV of *A. aegypti*. The number of dead larvae was counted after 24 h on adding tested samples. Larvae were claimed dead when they drowned or did not move even when they were touched with a pipette on siphon or thorax. The test was replicated 3 times. The data were presented in LC_{50} and LC_{90} values that were determined through the probit regression analysis using A. Woods U.N.S.W. 1.1 version program.

Statistical analysis

Data were analyzed using one-way ANOVA. Significant differences between the treatments were determined using Tukey's multiple-range test ($p < 0.05$).

Isolation and structure identification

Isolation of larvicidal active compounds was conducted based on the bioassay-guided fractionation and isolation method (Fig. 1) [11] with slight modification. Each extract, fraction, and isolated compound obtained in this process was monitored through a larvicidal assay.

Fractionation, isolation and identification of larvicidal compound

The vacuum dried MeOH extract was triturated with PE to obtain PE-soluble (3.5 g) and PE-insoluble fractions (precipitate) [9]. The PE-soluble indicating larvicidal activity was fractionated through vacuum liquid chromatography (VLC, SiO_2) eluted by PE–EtOAc with increasing amount of EtOAc to produce four combined Fractions (I–IV) (Fig. 2). Larvicidal active compound was isolated from Fraction I (1.0 g) that displayed the best larvicidal activity among the other fractions obtained by preparative thin layer chromatography (TLC) (SiO_2 , GF₂₅₄ prep. grade, PE: EtOAc 1:3 v/v) to make two compounds (1 and 2). Both compounds were tested for their larvicidal activity, and it turned out that compound 1 was more active than compound 2. The larvicidal active compound identity was figured out according to its spectroscopic (UV, IR, ^{13}C - and 1H -NMR) data and comparison with reported literature.

RESULTS AND DISCUSSION

The fractionation using VLC chromatography with PE solvent resulted in 10 fractions combined into four fractions (Fraction I [1000 mg], Fraction II [100 mg], Fraction III [90 mg], and Fraction IV [70 mg]) (Table 1). Each fraction was tested for their larvicidal activity against *A. aegypti* larvae.

The values of LC_{50} and LC_{90} of the four fractions indicate that Fraction I had the highest larvicidal activity with 51.70 ± 0.46 ppm of LC_{50} followed by Fraction III, IV, and II with 1323.71 ± 1.04 ppm, 1484.52 ± 1.82 ppm, and 1491.00 ± 0.80 ppm of LC_{50} values, respectively (Table 1). The table shows that Fraction I performed the highest activity as a larvicide against *A. aegypti* larvae as opposed to Fractions II, III, and IV. Danga et al. [12] conducted a study of larvicide against *A. aegypti* larvae obtained from the fractionation of *Plectranthus glandulosus* leaf extract

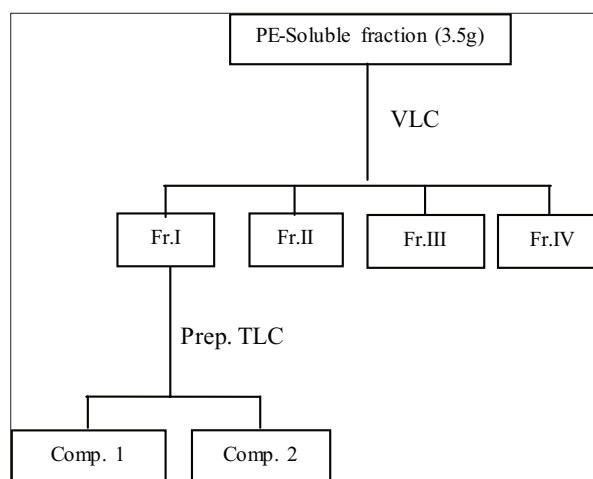


Fig. 1: Scheme of bioassay-guided fractionation for larvicidal compound isolated from the rhizome of *Zingiber zerumbet*

Table 1: LC_{50} and LC_{90} values of *Z. zerumbet* Fractions (I–IV) against *A. aegypti*

Fraction	Weight (mg)	Mean±SD (ppm)	
		LC_{50}	LC_{90}
I	1000	51.70 ± 0.46^a	72.38 ± 1.67^b
II	100	1491.00 ± 0.80^a	2099.75 ± 1.90^b
III	90	1323.71 ± 1.04^a	2220.43 ± 1.57^b
IV	70	1484.52 ± 1.82^a	2166.58 ± 0.95^b

Z. zerumbet: *Zingiber zerumbet*, *A. aegypti*: *Aedes aegypti*. The value expressed as mean±SEM. n=3 fraction. Statistical analysis to compare Fraction I (active) with other Fraction (II–IV=non active) was performed by ANOVA followed by Tukey test. Values^{a,b} were statistically different compared to fraction active (I) $p < 0.05$. SD: Standard deviation, LC_{50} : Lethal concentration

using several solvents with varied polarizations. Fractions from four solvents (hexane, chloroform, ethyl acetate, and methanol) resulted in 89.08, 209.38, 1987.10, and 2533.66 ppm of LC_{50} , respectively. A similar study was performed by Lame et al. [13], examining the methanolic extracts of *Annona senegalensis* and *Boswellia dalzielii*. The LC_{50} values of hexane, chloroform, and ethyl acetate fractions from the methanolic extract of *A. senegalensis* leaves were 379.3, 595.2, and 1240.3 ppm, respectively. Meanwhile, the *B. dalzielii* methanolic extract produced 537.1 and 785.5 ppm of LC_{50} from its hexane and chloroform fractions. Other research on the larvicidal activity of three extracts using organic solvents indicated that the hexane extract was highly effective against *A. aegypti* larvae. The LC_{50} for 24 h were found to be 240.9 ppm, 142.2 ppm, and 87.2 ppm for chloroform, PE, and hexane extracts, respectively, each against the instar III of *A. aegypti* larvae [14]. A similar research using crude extracts obtained from PE (A), ethyl acetate (B), and methanol (C) of *C. infundibuliformis* against *Anopheles stephensi*, *A. aegypti* and *Culex quinque fuscatus* indicated that the PE extract (A) has a good mortality against *A. stephensi* and the methanolic extract (C), can inhibit *A. aegypti* to a higher level when compared to the other extracts [15]. Another study on larvicidal activities of six varying extracts of *Acalypha indica* leaves against *A. albopictus* indicated that *A. indica* leaf extract could be one of the new potential bio larvicidal agent against the vector *A. albopictus* [16]. Fractions have higher larvicidal activity than extracts because a fraction contains an active compound that is not as complex as an extract. Differences in the solvent polarity for extraction and fractionation also affect the larvicidal activity. The crude extract of bioactive compounds or phytoconstituents isolated from medicinal plants can be an alternative to synthetic insecticides currently used today. The bioassay-guided fractionation showed that the fraction active from *Sphaeranthus indicus* extract has mosquito larvicidal activity against all the three species [17] the biological activities of plant

extracts are likely caused by various compounds, including phenolates, terpenoids, and alkaloids found in the plants [18].

Fraction I as the most active larvicidal fraction was isolated using the preparative TLC. Two isolated compounds, compound 1 weighing 170 mg and 50 mg of compound 2, were obtained (Table 2). After a larvicidal activity test, the activity of compound isolate 1 was compared to that of compound 2 for LC_{50} (41.75 ± 0.05 vs. 1122.12 ± 1.80) as well as for LC_{90} (57.66 ± 3.37 vs. 1875.69 ± 1.35). Considering the values of LC_{50} and LC_{90} of the four fractions (Table 1) and compared with isolated compounds 1 and 2 (Table 2), compound 1 had much lower values of LC_{50} and LC_{90} than the others. The LC_{50} of compound 1 was only 10 ppm lower than Fraction I because compound isolate 1 became the most dominant component in Fraction I (LC_{50} : 51.70 ± 0.46 vs. 41.75 ± 0.05 ppm). It means isolated compound 1 served as the main active compound of *Z. zerumbet* as a larvicide. The bioassay-guided isolation is a relatively new method applied globally for discovering the active compounds of natural resources, and a wide range of active compounds have been successfully isolated using this method, including the isolation of a cytotoxic compound tagitinin C against melanoma cell lines (M19) [19]. Therefore, the molecular structure of isolated compound 1 needed further identification.

Structure identification of compound 1

Compound 1 appeared as white amorphous powder. The infrared spectra (KBr, cm^{-1}) of compound 1 indicated the presence of aliphatic

hydrocarbon as shown by the bands at 2947, 2916, and 2885 cm^{-1} . A strong band at 1643 cm^{-1} attributed to α, β unsaturated $C=O$ functional group, while a characteristic absorption band of $C=C$ group and $C-O$ bond appeared as the bands at 1444 and 1267 cm^{-1} , respectively (Fig. 3).

The 1H -NMR ($CDCl_3$, 500 MHz) spectra of compound 1 showed the characteristics of four peaks of methyl ($-CH_3$) singlets (s) at δ 1.08, 1.24, 1.57, and δ 1.77 (s). Four protons of 2 CH_2 groups appeared at a different chemical shift and multiplicity as well due to the differences in CH_2 proton environments. Therefore, one CH_2 proton

Table 2: LC_{50} and LC_{90} values of compounds isolated from the rhizome of *Z. zerumbet* against *A. aegypti*

Compound	Weight (mg)	Mean \pm SD (ppm)	
		LC_{50}	LC_{90}
1	170	41.75 ± 0.05^a	57.66 ± 3.37^b
2	50	1122.27 ± 1.80^a	1875.69 ± 1.35^b

The value expressed as mean \pm SEM. n=3 isolated. Statistic analysis to compare isolate 1 and isolate 2 was performed by Tukey's test. Values^{a, b} were statistically different compare to isolate active (1) were significantly different $p < 0.05$. *Z. zerumbet*: *Zingiber zerumbet*, SD: Standard deviation, LC_{50} : Lethal concentration

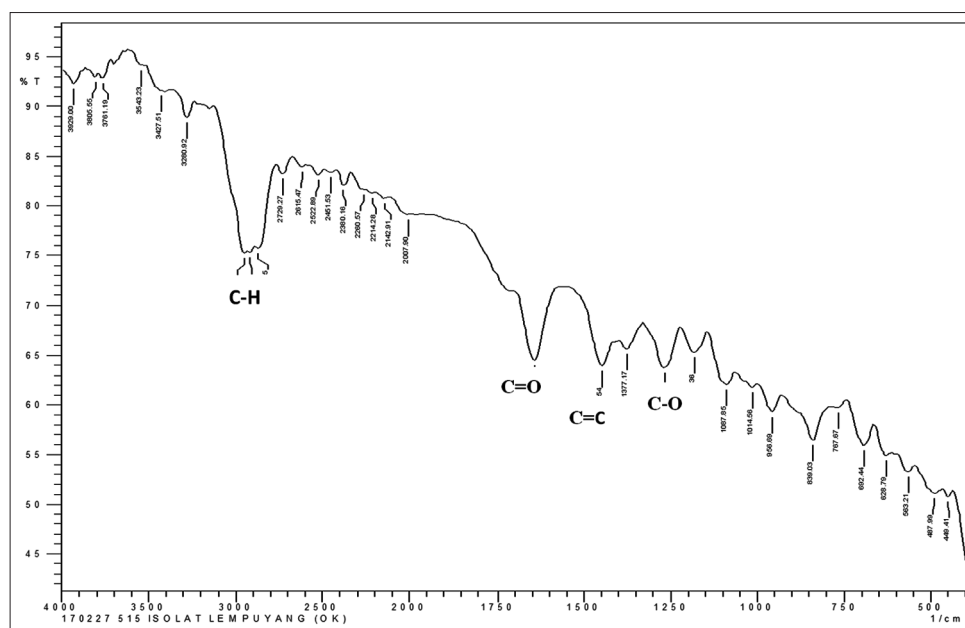


Fig. 2: Infrared spectrum (KBr, cm^{-1}) of compound 1 of *Zingiber zerumbet*

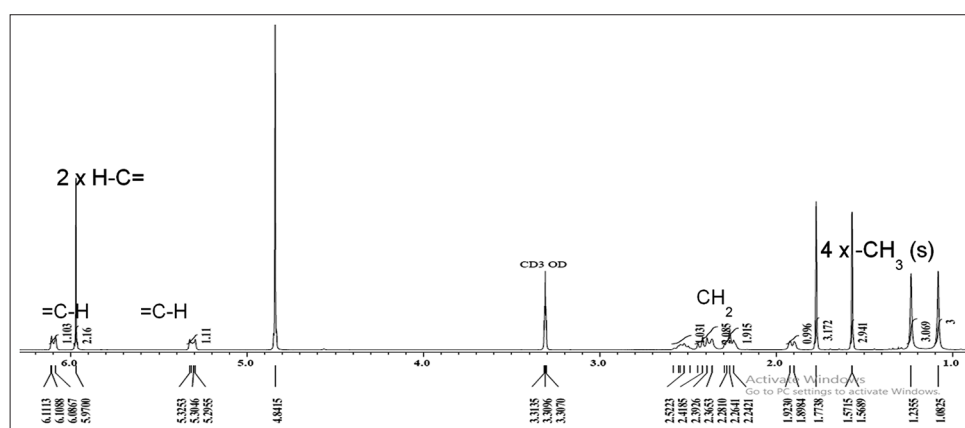


Fig. 3: 1H -nuclear magnetic resonance ($CDCl_3$, 500 MHz) spectra of compound 1 of *Zingiber zerumbet*

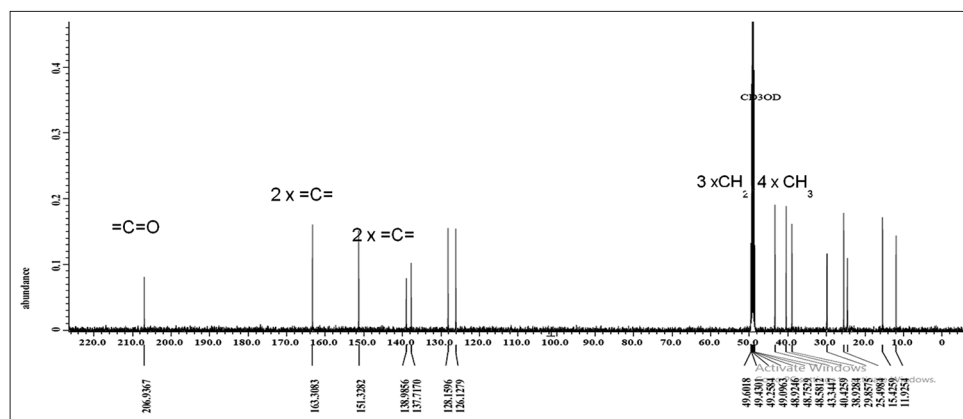


Fig. 4: ^{13}C -nuclear magnetic resonance (CDCl_3 , TMS, 500 MHz) spectra of compound 1 of *Zingiber zerumbet*

appeared at δ 1.91 (1H, d, 13.2 Hz) and 2.28 (1H, d, 13.2 Hz), while other 2 CH_2 protons appeared at δ 2.26 (2H, m) and 2.36 (2H, m), in which the protons split between each other causing multiplet appearances. In addition, there were four signals appeared at the aromatic proton region, and they were identified as protons of unsaturated carbons (δ 5.29 [dd, 14.9 and 4.5 Hz], δ 6.11 [d], and δ 5.97 [s, 2H]) (Fig. 4). Based on the data, compound 1 was suspected as zerumbone, and further spectroscopic data were, therefore, pursued.

The ^{13}C -NMR spectrum exhibited 15 signals indicating that there were 15 carbons present in the molecule. The characteristic signal at δ 206.94 ppm was a carbonyl carbon, specifically as a ketone carbon. Two quaternary carbons at δ 138.98 and 137.72 ppm were identified as unsaturated carbons (C-3 and C-7). The three methylene groups appeared at δ 42.23, 40.34, and 39.45 ppm. The two upfield signals at δ 11.56 and 14.71 ppm were identified as a gem dimethyl (C-14, C-15) (Fig. 4).

Based on the above data and comparison with reported data [20], it could be confirmed that the compound was zerumbone. In addition, a mass spectrum exhibited a molecular ion peak at m/z 218.16 and had molecular elements identified as $\text{C}_{15}\text{H}_{22}\text{O}$ that matched the reported data of zerumbone (Fig. 5).

Studies of *Z. zerumbet* rhizome have been widely published. Several aromatic and flavonoid compounds have also been isolated from the chloroform extract of *Z. zerumbet* rhizome, while some terpenoid compounds were isolated from *Z. zerumbet* roots [21,22]. The existence of sesquiterpenoids (zerumbone, humulene, caryophyllene, and zingiberene) as well as monoterpenoids (borneol, linalool, camphene, eucalyptol, and myrcene) has also been reported [23,24]. Salvador-Neto *et al.* [25] found that sesquiterpene (+)-obtusal isolated from *Alga Laurencia dendroidea*. Agardh had a potent larvicidal activity against *A. aegypti* larvae. Zerumbone has been reported to have antinociceptive, anti-inflammatory, and antitumor activities *in vivo*. Meanwhile, in several *in vitro* studies, it was found that zerumbone showed antiplatelet and anti-proliferative aggregations [26]. Dash *et al.* [27] suggested that zerumbone could be utilized as a lead molecule in the world of herbal therapeutics functioning as an immunomodulator drug for the treatment of chronic infection and various autoimmune disorders.

CONCLUSION

This study showed that compound 1, the major larvicidal compound isolated from the rhizome of *Z. zerumbet*, was identified as 2,6,9-humulatrien-8-one (Zerumbone) on the basis of spectroscopic data and comparison with literature data. Zerumbone demonstrated the best larvicidal activities with LC_{50} 41.75 ± 0.05 ppm and LC_{90} value of 57.66 ± 3.37 ppm.

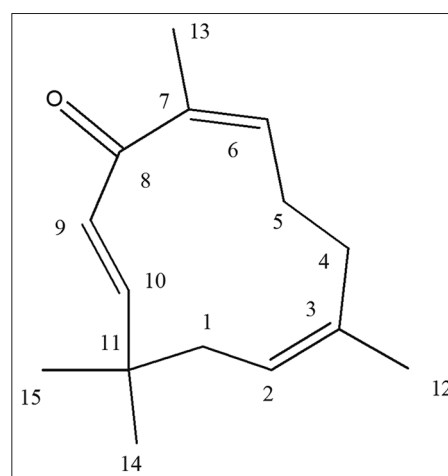


Fig. 5: Structure of zerumbone

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AUTHORS CONTRIBUTION

TM and MSH contributed to the preparation of manuscript and research design, data collection, analysis and interpretation. TBTS and AF contributed to revision of the manuscript especially on the Larvicidal activity. MH contributed to revision of the manuscript especially on the Structure identification of compounds Isolated from *Zingiber zerumbet* (L). J.E. Smith. All the authors read and approved the final manuscript.

CONFLICTS OF INTERESTS

Authors declare that there are no conflicts of interest.

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