NEW STABILITY-INDICATING ULTRA PERFORMANCE LIQUID CHROMATOGRAPHY METHOD DEVELOPMENT AND VALIDATION OF LENVATINIB MESYLATE IN BULK DRUG AND PHARMACEUTICAL DOSAGE FORMS

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Received: 17 May 2018, Revised and Accepted: 9 June 2018

ABSTRACT

Objective: The objective of the present study was to develop and validate a new stability-indicating method for the quantification of lenvatinib mesylate in bulk drug and pharmaceutical dosage form using ultra performance liquid chromatography (UPLC).

Methods: The optimized chromatographic conditions for elution of drug included UPLC HSS C18 (100 mm × 2.1 mm, 1.8 m) column, mixture of 0.1% orthophosphoric acid and acetonitrile (50:50 v/v%) mobile phase run on an isocratic mode at a flow rate of 0.3 mL/min, 240 nm detection wavelength, and column oven temperature maintained at 30°C.

Results: The retention time for lenvatinib was found to be 1.24 min. The developed method was validated for various validation parameters in accordance with the International Conference on Harmonization guidelines. The method obeyed Beer’s law in the concentration range of 2.5–15 μg/mL with a correlation coefficient of 0.9996. The percentage relative standard deviation and percentage recovery were determined to be 0.4 and 99.66–100.30%, respectively. The developed method was found to be accurate, precise, specific, linear, rugged, and robust. Forced degradation studies were conducted by exposing the drug to diverse stress conditions such as acidic, basic, peroxide, neutral, photolytic, and thermal conditions. The net degradation was obtained within the limits.

Conclusion: The developed method for the estimation of lenvatinib can be employed to routine analysis of pharmaceutical dosage form.

Keywords: Lenvatinib mesylate, Ultra performance liquid chromatography, Stability indicating, Method development, Validation.

INTRODUCTION

Lenvatinib mesylate (Fig. 1) [1,2] chemically known as 4-[3-chloro-4-(N’-cyclopropylureido)phenoxyl]-7-methoxyquinoline-6-carboxamide methanesulfonate. It is a white to pale reddish-yellow powder, slightly soluble in water and practically insoluble in ethanol. It is a pKa value of 5.05. It belongs to anticancer category and utilized for the treatment of various kinds of thyroid cancer [3,4]. It acts as receptor tyrosine kinase inhibitor that inhibits the kinase activities of vascular endothelial growth factor receptors [5]. It also inhibits other RKTs that have been implicated in pathogenic angiogenesis, tumor growth, and cancer progression in addition to their normal cellular functions, including fibroblast growth factor receptors.

Ultra performance liquid chromatography (UPLC) [6-8] is a relatively new technique giving new possibilities in liquid chromatography, especially concerning the decrease of time and solvent consumption. UPLC chromatographic system is designed in a special way to withstand extreme system back pressures.

In accordance to literature review, it was known that there were only few methods such as reverse-phase high-performance liquid chromatography (RP-HPLC) [9-11], ultraviolet (UV) spectroscopy [11], and liquid chromatography coupled with tandem mass spectrometry method [12-14] developed for the estimation of lenvatinib.

As there was no UPLC method developed for the estimation of lenvatinib, the present study was intended to develop and validate a stability-indicating UPLC method for the quantitative determination of lenvatinib in bulk drug and pharmaceutical dosage form.

METHODS

Chemicals and reagents
Lenvatinib mesylate working standard was supplied as a gift sample from Spectrum Labs, Hydnebad. Lenvima capsules were purchased from a local pharmacy. All the chemicals used for the development of the method were of AR grade purchased from Merck, Mumbai. All the solvents used were of HPLC grade purchased from Sigma-Aldrich, Mumbai.

Analytical conditions and instruments
The ACQUITY UPLC system equipped with binary solvent manager, sample manager, UV detector, and UPLC HSS C18 (100 mm × 2.1 mm, 1.8 µ) column was used for the determination of lenvatinib. The analytical conditions included 0.1% orthophosphoric acid and acetonitrile (50:50 v/v%) as mobile phase run on an isocratic mode at a flow rate of 0.3 mL/min. The column was kept at 30°C and detection was done at 240 nm wavelength. Additional equipment included pH meter, ultrasonic bath sonicator, and weighing balance.

Preparation of mobile phase
Mixture of 0.1% aqueous orthophosphoric acid buffer and acetonitrile in the ratio of 50:50 v/v% was used as mobile phase.

Preparation of standard and sample solution
10 mg of lenvatinib working standard was dissolved in 100 mL of diluent. 1 mL of the above standard stock solution was diluted to 10 mL diluent.

Average weight of 20 lenvima capsules was calculated and an amount equivalent to 10 mg of lenvatinib was dissolved in 100 mL of diluent.

Preparation of forced degradation samples

1. Base degradation samples were prepared by dissolving 10 mg of lenvatinib standard working solution in 100 mL of 0.1N NaOH solution.
2. Acid degradation samples were prepared by dissolving 10 mg of lenvatinib standard working solution in 100 mL of 0.1 N HCl solution.
3. Oxidative degradation samples were prepared by dissolving 10 mg of lenvatinib standard working solution in 100 mL of 3% w/v hydrogen peroxide.
4. Stress heating samples were prepared by dissolving 10 mg of lenvatinib standard working solution in 100 mL of distilled water and heating it at 80°C for 24 hours.
5. Photolytic degradation samples were prepared by dissolving 10 mg of lenvatinib standard working solution in 100 mL of distilled water and exposing it to sunlight for 24 hours.

Validation

The validation was performed in accordance with the International Conference on Harmonization guidelines. The developed method obeyed Beer’s law in the concentration range of 2.5–15 μg/mL with a correlation coefficient of 0.9996. The percentage relative standard deviation and percentage recovery were determined to be 0.4 and 99.66–100.30%, respectively.
The above sample stock solution was filtered and diluted 1–10 mL with diluent.

Method validation
The developed method was validated in compliance with the International Conference on Harmonization (ICH) guidelines [15] the following parameters are as follows:

Accuracy
Accuracy of the method was established by preparing three concentration levels of 50%, 100%, and 150%. Then, percentage recovery was calculated.

Precision
Precision of the method was determined by injecting six solutions of sample and then percentage relative standard deviation (RSD) was calculated.

Specificity
For the determination of specificity of the method, a placebo solution was prepared and compared with the drug solution for any interference of placebo peak with the drug peak.

Linearity
For the linearity study, serial dilutions of the drug solution were prepared in the concentration range of 2.50–15 µg/mL. A graph was plotted by taking concentration on the x-axis and peak area on y-axis.

Limit of detection (LOD) and limit of quantitation (LOQ)
LOD and LOQ are calculated using standard deviation and slope of the calibration curve by placing in the formula mentioned in the ICH guidelines.

Ruggedness
Ruggedness was evaluated by performing precision on different days and the percentage RSD was calculated.

Robustness
Robustness was determined by varying the optimized conditions such as variation of the organic phase in the mobile phase by ±10%, variation of flow rate by ±0.1 mL/min, and variation of column oven temperature by ±5°C.

Solution stability
Solution stability was assessed by performing the assay of drug solutions after storing for 24 h.

Forced degradation studies
Forced degradation studies [16] were conducted by exposing the drug solution to various degradation conditions such as acidic (2 N hydrochloric acid, 60°C for 30 min), basic (2 N sodium hydroxide [NaOH], 60°C for 30 min), oxidative (20% hydrogen peroxide [H₂O₂], 60°C for 30 min), neutral (refluxing the drug in water for 6 h, 60°C), thermal (105°C for 6 h), and photolytic (UV chamber for 7 d) conditions. The net degradation for unusual conditions was noted.

RESULTS

DISCUSSION
For the development of a method for the estimation of lenvatinib, various mobile phases and stationary phases were tried to elute the drug with good peak parameters. The optimized conditions used were mobile phase consisting of composition 0.1% o-phosphoric acid and acetonitrile (50:50 v/v%) run on an isocratic mode with flow rate of 0.3 mL/min, UPLC HSS C18 (100 mm × 2.1 mm, 1.8 µm) column maintained at 30°C temperature. Lenvatinib was detected at a wavelength of 240 nm as the drug shows maximum absorbance at that wavelength as shown in Fig. 2.

The standard solution, sample solution, and the blank solution were prepared as per the procedure mentioned above. The standard solution was injected into the UPLC system for evaluating the system suitability parameters. The results were presented in Table 1 and chromatograms were shown in Fig. 3.

The method was considered to be accurate as the percentage recovery for lenvatinib was found to be 99.66–100.30%. The percentage RSD for lenvatinib was determined to be 0.4, indicating that the method was precise. The developed method was found to be specific to the drug when compared with a placebo solution, as there was no interference.
The method was found to perform Beer's law in the concentration range of 2.5–15 µg/mL with a correlation coefficient of 0.9996. The linearity plot was presented in Fig. 5. The LOD and LOQ for lenvatinib were found to be 0.16 µg/mL and 0.50 µg/mL, respectively. The method was found to be rugged as the percentage RSD when compared with day difference was found to be within the limits, robust when did by varying the optimized conditions and stable in solution form also after storing for 24 h.

Forced degradation studies indicated that the samples remained stable under different stress conditions such as acidic, basic, oxidative, photolytic, and thermal conditions. The net degradation was considered to be within the limits. The results were outlined in Table 2 and chromatograms were shown in Fig. 6.

**CONCLUSION**

The new stability-indicating method was established for quantitative determination of lenvatinib in pharmaceutical dosage and bulk drug using UPLC. The developed method was validated in accordance with the ICH guidelines. The method was found to be accurate, precise, specific, linear, rugged, robust, and stable in solution. Forced degradation studies concluded that the drug was stable in various stress conditions.

This method can be applied to the routine analysis of lenvatinib in pharmaceutical dosage form.

**ACKNOWLEDGMENTS**

The authors are thankful to the Spectrum Labs, Hyderabad, for providing the lenvatinib as the gift samples and also for providing the required facilities to carry out this work.

**AUTHOR’S CONTRIBUTIONS**

All authors contributed equally to this manuscript.

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### Table 2: Results of forced degradation studies

<table>
<thead>
<tr>
<th>Stress condition</th>
<th>% Assay</th>
<th>% Area of degradation peak</th>
<th>% Degradation</th>
</tr>
</thead>
<tbody>
<tr>
<td>2 N HCl for 30 min at 60°C</td>
<td>95.75</td>
<td></td>
<td>4.25</td>
</tr>
<tr>
<td>2 N NaOH for 30 min at 60°C</td>
<td>95.71</td>
<td></td>
<td>4.29</td>
</tr>
<tr>
<td>20% H₂O₂ for 30 min at 60°C</td>
<td>96.61</td>
<td></td>
<td>3.39</td>
</tr>
<tr>
<td>Water for 6 h at 60°C</td>
<td>99.04</td>
<td></td>
<td>0.96</td>
</tr>
<tr>
<td>UV light 200 wts/h or 7 d</td>
<td>97.27</td>
<td></td>
<td>2.73</td>
</tr>
<tr>
<td>105°C for 6 h</td>
<td>97.13</td>
<td></td>
<td>2.87</td>
</tr>
</tbody>
</table>
CONFLICTS OF INTEREST

The authors state that they have no conflicts of interest. It has not meant published elsewhere. Moreover, it has not been simultaneously presented for publication elsewhere. All authors have agreed to the submission to the journal.

REFERENCES