FORMULATION AND STABILITY EVALUATION OF ATENOLOL GEL IN TWO DIFFERENT BASES

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INTRODUCTION

Atenolol is one of the β-selective adrenergic blocking agents in the treatment of angina pectoris. It is reported to have extensive hepatic first-pass effect after oral administration [1,2]. It is also very well known that conventional therapy such as orally, may result higher fluctuations in plasma concentration of the drug and unwanted side effect [3]. Hence, the transdermal drug delivery system that provides a predetermined constant drug delivery would be beneficial for effective and safe therapy [4]. Most of the transdermal delivery is formulated as topical preparations. The main advantage of this system is to overcome the first-pass metabolism. Other advantages of topical preparations are avoidance of the risks and inconveniences of intravenous therapy and reducing the effect of varied conditions on drug absorption, like pH changes, presence of enzymes, gastric emptying time are [5-7].

Gels, as semisolid base for transdermal preparation, are a class of dosage form which have a higher aqueous component that permits greater dissolution of drugs and also permits easy migration of the drug through a vehicle that is essentially a liquid, compared with the ointment or cream base [8-11]. Kim and Shin developed and studied the release of atenolol from the ethylene-vinyl acetate matrix containing various plasticizers. Many other study also developed increased release from topical formulation [12-15]. Hence, the objectives of this study were to formulate and investigate the physicochemical stability of atenolol in two different gel bases. The influence of gelling agent, drug loading effect, and skin irritation studies were also evaluated.

MATERIALS AND METHODS

Materials
Atenolol (PT Kalbe Farma), Ammonia 21%, Aqupec HV-505 (Sumitomo Seika), Aquadest, DMDM hydantoin, Ethyl acetate, (hydroxypropyl methylcellulose [HPMC], Bratachem), Methanol (Bratachem), Nutrient Agar (Oxoid), Propylene glycol (Bratachem), and Triethanolamine (TEA, Bratachem) were used.

Atenolol content was determined by Spectrophotometry UV-visible (Specord 200). The pH was measured by pH-meter 744 Methrohm while gel viscosity was measured by Viscotester Rion (VT-04 F).

Methods

Preparation of gels
HPMC gels
The composition of gel formulation is shown in Table 1. The HPMC was dispersed with continuous stirring until uniformly dispersed and the solution was allowed to cool to 50°C. The other ingredients were added with continuous stirring. Atenolol dissolved in propylene glycol was added slowly with stirring. Other ingredients were added slowly while stirring until homogenous gel was obtained.

Aqupec HV-505 gels
Aqupec HV-505 powder was dispersed in hot water and stirred with a stirrer at 2500 rpm, and the gels then were left overnight at ambient temperature. Triethanol amine was added until the clear gel was obtained. Atenolol was dissolved in propylene glycol and was added slowly with stirring into solution; the other ingredients were added with continuous stirring.

Physical stability investigation
Parameters investigated including organoleptic and bleeding investigation, pH, and viscosity measurement.

Microbial investigation
This study was conducted to investigate the efficacy of DMDM hydantoin as the preservative after the gels were stored for 14 days. The study was conducted by counting method.
Qualitative and quantitative determination of atenolol in gel

Qualitative analysis was performed by thin-layer chromatography (TLC) into gel. Samples were dissolved in methanol and the eluent was ammonia:methanol (1:4). The spot was observed at UV light 254 nm.

Quantitative analysis was conducted by measuring the atenolol content in the sample. 1 g of gel was accurately weighed and placed in a 100 ml volumetric flask containing 30 ml of methanol, stirred for 30 min, and made up to volume. The amount of atenolol was calculated using standard obtained curve. The determination was conducted at 1st day and 56th day of preparation.

Safety test

The safety test was conducted into 10 volunteers using patch test method.

RESULTS AND DISCUSSION

Physical stability investigation of atenolol in Aqupec HV-505 and HPMC bases

Organoleptic investigation showed that all formulae have smooth texture, odorless, and spreadable. Gel with HPMC base was white opaque, while the one with Aqupec HV-505 was clear and transparent. The result was stable during 56 days of storage. Based on aesthetic considerations, Aqupec HV-505 1% was best one for gel base. So it is used for next experiment to which variating the Atenolol concentration in gels.

pH of gels in various bases

Ideal pH for topical preparation is 5.5–10. The result showed that pH of F1 was 7.58–7.80, and it means that all gels with Aqupec HV-505 and HPMC base fulfilled the pH requirement. The pH was decreased during 56 days of storage, especially FB formulas, but they all still in required pH for topical preparation. pH of gels during storage is shown in Fig. 1.

Viscosity of gels

Investigation on gels’ viscosity gave the results that viscosity was decreased during the time of storage (Fig. 2). A decrease in viscosity may due to indirectly monitor chemical degradation because changes at the molecular level may cause changes in viscosity [16-18].

Further investigation was conducted by varying the atenolol concentration on 1% Aqupec HV-505. Based on organoleptic investigation, almost all gels were also clear, transparent, and well spreadable. Aqupec HV-505 1.5% became turbid, especially, after 0.5% atenolol was added. No bledding happened in all gels during 56 days of storage. Formula used in gels with various concentration of atenolol is shown in Table 2.

Physical stability investigation of 1% Aqupec HV-505 gels with various concentration of atenolol

Organoleptic investigation

Based on the result of organoleptic investigation including color, smell, and consistency during 56 days of storage, it was concluded that F1, F3, and F5 had homogenous and stable during the time of investigation at ambient temperature. The results show that there were no visible changes in the gel formulation.

pH of gel

The pH of F1 was 7.74–8.30. Addition of atenolol causes an increase in pH. It may be caused by the physicochemical of Atenolol itself as weak base. The increase in the concentration of atenolol resulted in bigger pH value of gels. The pH of gels with various concentration of atenolol is shown in Fig. 3.

Viscosity of gels

The viscosity of gels containing atenolol was lower compared with those without atenolol, which is due to internal and molecular change of atenolol addition. Furthermore, the results revealed that the increase of atenolol concentration in gel formulation did not affect the viscosity. Effect of time of storage was investigated for 56 days. The results...
showed that gel formulation was stable after 56 days of storage. Viscosity of gels with various concentration of atenolol during the time of storage is shown in Fig. 4.

Microbial Investigation
The test was conducted to prove that preservative used in formulation was effective. It was found that there was no bacterial growth after 14 days of investigation which can be concluded that the preservative is effectively worked in the gels.

Qualitative analysis by TLC
The purpose of the study was to investigate atenolol content in gels during the time of storage. First, the investigation was conducted by qualitative analysis using TLC method. Furthermore, the quantitative analysis was conducted to investigate the homogeneity content of atenolol in gel preparation. The study was performed into all gels at the day of preparation and after 56 days of storage. Qualitative analysis by TLC method showed that Rf of atenolol in gels and atenolol powder as standard were unchanged after formulation, as well as after 56 days of storage as shown in Table 3.

Quantitative analysis
Atenolol content in gels is shown in Fig. 5. From the results, it can be concluded that drug content in gels did not significantly changed during the time of storage. Stability of atenolol in gel formulation can be explained by inert matrix using as gel basis. HPMC and Carbopol as matrix used in gel basis revealed the stability for atenolol after 56 days of storage.

Safety test
Safety test was conducted by Patch test method to find out whether the formulation may be caused or aggravated by a contact allergy to the skin. Positive reactions were observed; skin reaction nearly becomes red and/or itchy. The safety test gave the result that all gels were save to be used because it did not give the allergic reaction to the volunteers.

CONCLUSIONS
Organoleptic investigation showed that all formulae have smooth texture, odorless, and spreadable. Gel with HPMC base was white opaque, while the one with Aqupec HV-505 were clear and transparent which was stable during 56 days of storage. Gels with Aqupec HV-505 and HPMC base were fulfilled the pH requirement for topical preparation. Viscosity of FA and FB was decreased during the time of storage. The addition of atenolol concentration did not affect the viscosity significantly.

AUTHORS’ CONTRIBUTIONS
The entire author equally contributed to this work.

CONFLICTS OF INTEREST
The authors have no conflicts of interest.

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