DESIGN AND IN VITRO CHARACTERIZATION OF FLOATING TABLETS OF METRONIDAZOLE

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ABSTRACT

Objective: The objective of the present study was to formulate the effervescent floating matrix tablet of metronidazole and to evaluate the effect of varying concentrations of hydrophilic polymers on drug release.

Methods: Drug excipients interaction was studied by Fourier transform infrared spectrophotometer. The effervescent floating matrix tablets were prepared by direct compression technique using hydroxypropyl methycellulose (HPMCK4) and xanthan gum alone and in combination as release retardants. Microcrystalline cellulose was used as diluent. Sodium bicarbonate was used as effervescent agent. The prepared matrix tablets were evaluated for their physicochemical parameters such as weight variation, hardness, friability, content uniformity, buoyancy time, and in vitro dissolution.

Results: Micromeritic properties and post-compression parameters were evaluated and all the parameters were found within the acceptable limit. The drug release data were subjected to different models to evaluate release kinetics and mechanism of drug release. The matrix tablets prepared with xanthan gum and a mixture of xanthan gum and HPMCK4 were retarded the drug release up to 12 h. The release mechanism of metronidazole was evaluated on the basis of release exponent n value in Peppas model. The n value of the formulations ranged from 0.46 to 0.89 which indicated Case II transport and zero-order release.

Conclusion: Floating matrix tablet is the simple, efficient, and economic method to sustain the release of metronidazole to eradicate Helicobacter pylori in peptic ulcer disease.

Keywords: Floating matrix tablets, Metronidazole, Xanthan gum, Helicobacter pylori.

INTRODUCTION

Oral route is the most convenient route for the drug delivery due to low cost of the therapy, ease of administration leads to high level of patient compliance and flexibility in the designing of dosage form. Novel oral drug delivery systems classified into oral controlled release dosage forms (OCRDFs) and targeting dosage forms [1]. Variable and short gastric residence time results in decreased efficacy of the administered dose when administered in the form of OCRDF. To improve the performance of OCRDF, a new concept was discovered by scientists in drug delivery that is gastroretentive drug delivery systems (GRDDSs). A GRDDS can be defined as a system which remains in the stomach for a sufficient time period against all the physiological barriers, releasing the active moiety in a controlled manner and finally becomes metabolized in the body. Gastroretentive drug delivery system can be a useful tool in delivery of drugs that are primarily absorbed in the duodenum which is upper part of the gastrointestinal tract [2]. Floating drug delivery is the class of targeted oral drug delivery system. A floating drug delivery system floats in the gastric juice without affecting the gastric emptying rate. This technique can provide therapeutically effective drug concentration for longer durations, thereby reducing the dosing frequency and minimizing fluctuations and also helps to increase a drug's gastric residence time and reduces the variability in bioavailability [3].

Helicobacter pylori (Campylobacter pylori) is a Gram-negative, microaerophilic bacterium, usually found in the stomach or the upper part of the small intestine. H. pylori infection is a common worldwide infection that is an important cause of peptic ulcer disease and gastric cancer. Peptic ulcers are sores that develop in the lining of the stomach (gastric ulcers), lower esophagus (esophageal ulcers), and/or first part of the small intestine (duodenal ulcers), as a result of inflammation caused by the bacteria H. pylori, erosions, and stomach acids. American College of Gastroenterology guidelines, 2017 for the treatment of H. pylori infection includes the following therapeutic strategies for the first-line treatment, bismuth, proton-pump inhibitors, tetracyclines, and a nitroimidazole (strong recommendation), particularly in those with previous macrolide exposure or rare penicillin allergy [4]. Nitroimidazole (metronidazole) antibiotics are most active antibiotics for the treatment and prevention of infections caused by anaerobic bacteria. Metronidazole is used as an antibiotic, amebicide, and antiprotozoal agent. Metronidazole used for preparing floating tablets, to retain in the stomach, because it is locally active in the gastric mucosa. The prolonged effect of locally active metronidazole increases its clinical efficacy for better eradicating H. pylori in peptic ulcer disease [5-9].

Drug release from the matrix tablets was controlled by the polymer properties and its structure. Hydrophilic polymers are one of the most commonly used drug release retardants in matrix tablets. Hydroxypropyl methylcellulose (HPMCK4) is a semi-synthetic water-soluble cellulose derivative. It has been widely used in the development of sustained release dosage forms. Natural gums are inert and non-toxic. Xanthan gum is a natural, biosynthetic, edible, and high-molecular-weight gum. Xanthan gum is an extracellular polysaccharide produced by fermentation of carbohydrate with Xanthomonas campestris bacteria, is hydrophilic swellable polymer. The primary structure of xanthan gum consists of 1, 4 linked β-D-glucose residues, having a trisaccharide side chain of β-D-mannose-β-D-glucuronic acid-α-D-mannose attached to alternate D-glucose units of the main chain. The anionic character of this polymer is due to the presence of both glucuronic acid and pyruvic acid groups in the side chain. It develops a weak structure in water which creates high viscosity solutions at low concentration [10,11].

The objective of the present study was to prepare the effervescent floating matrix tablets of metronidazole using hydrophilic polymers.
such as hydroxypropyl methylcellulose, xanthan gum alone, and in combination. Sodium bicarbonate was used as gas generating agent to float and to achieve sustained release of metronidazole to eradicate *H. pylori* in the treatment of peptic ulcer.

**MATERIALS AND METHODS**

**Materials**

Metronidazole was purchased from Yarrow Chemical Ltd., Mumbai, India. Xanthan gum, hydroxypropyl methylcellulose K4M, microcrystalline cellulose, and sodium bicarbonate were purchased from Vijaya Enterprises Ltd., Hyderabad, India; magnesium stearate and talc were purchased from Aman Scientific Traders, Vijayawada, India. All other reagents, chemicals, and solvents used were of analytical grade.

**Methods**

**Drug-excipient compatibility study**

Infrared spectra of the pure drug and drug with polymers were recorded on a Fourier transform infrared spectrophotometer. The disc method was employed to study possible interactions between drug and selected polymers. Infrared spectrum was taken by scanning the sample in KBr (IR grade) discs and analyzed over a wavenumber range of 4000–400 cm\(^{-1}\). Transmittance spectra were recorded.

**Preparation of floating matrix tablets**

Floating matrix tablets of metronidazole were prepared by direct compression technique. All the ingredients were individually screened through sieve no. 60 except glidant and lubricant. Required quantities of metronidazole, polymer, diluents, and sodium bicarbonate were accurately weighed according to the composition shown in Table 1 and mixed in a polybag for about 30–45 min. The obtained blend was lubricated with talc and magnesium stearate for another 5 min, and then, the resultant mixture was directly compressed into tablets with 12 mm flat faced circular punches using 12-station rotary tablet machine (CEMACH, Ahmedabad, India) [12].

**Micromeritic properties of blended powder**

**Angle of repose**

The frictional forces in a loose powder can be measured by the angle of repose. Angle of repose is defined as maximum angle possible between the surface of a pile of the powder and the horizontal plane. The angle of repose was determined by the fixed funnel method. The accurately weighed powder was taken in a funnel. Angle of repose was calculated using the following equation [13]:

\[
\tan \theta = \frac{h}{r}
\]

Where, \(\theta\) = Angle of repose,

\(h\) = Height of the pile, cm

\(r\) = Radius of the base of the pile, cm.

Carr’s consolidation index

The Carr’s consolidation index of the powder blend is defined as:

\[
\text{Carr's consolidation index} = \frac{\text{Tapped density} - \text{bulk density}}{\text{Tapped density}} \times 100
\]

Bulk density is the ratio of mass of powder to the bulk volume. Tapped density is the ratio of mass of powder to the tapped volume.

**Evaluation of matrix tablets**

The tablets were evaluated for their physical properties such as weight variation, hardness, and friability.

**Weight variation**

A total of 20 tablets of each formulation were selected randomly and weighed individually using an electronic weighing balance. The average weight was calculated and individual tablet weight was compared with average weight [14].

**Hardness and friability**

Hardness studies were carried out by a Monsanto hardness tester. Six determinations were carried out for each sample to calculate the hardness of prepared tablets. Hardness of tablets was measured in terms of kg/cm\(^2\). The friability of the tablets was measured using Roche friabilator. 10 tablets were weighed accurately, placed in the chamber, and rotated for 4 min (100 rotations). At the end, the dust on the tablet was cleaned, weighed accurately again and the percent friability was computed from the weight of tablets. Friability was calculated using the following equation [15]:

\[
\text{Friability} = \frac{\text{Initial weight} - \text{Final weight}}{\text{Initial weight}} \times 100
\]

**Uniformity of drug content**

A total of 10 tablets were weighed from each formulation and triturated in mortar to a fine powder. Powder equivalent to 200 mg of metronidazole was extracted in 100 ml of 0.1 N hydrochloric acid and filtered. The drug content was determined by measuring the absorbance of sample using ultraviolet (UV)-visible spectrophotometer (LAB INDIA) at 277 nm after appropriate dilution with 0.1 N hydrochloric acid. The drug content was determined using calibration curve [16].

**Buoyancy lag time determination**

The buoyancy of the tablets was studied at 37±0.5°C, in 100 ml 0.1 N hydrochloric acid. The time required for the tablet to rise to the surface and total floating time was recorded [17].

**In vitro dissolution study**

Dissolution test of floating matrix tablets of metronidazole was performed using a USP type II (paddle) dissolution testing apparatus. The rotational speed was set at 50 rpm and the temperature remained constant at 37±0.5°C. 900 ml of 0.1 N hydrochloric acid was used as dissolution medium. 5 ml aliquots were withdrawn at predetermined

<table>
<thead>
<tr>
<th>Composition (mg/tab)</th>
<th>Formulation</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>F1</td>
</tr>
<tr>
<td>Metronidazole</td>
<td>200</td>
</tr>
<tr>
<td>HPMC K4M</td>
<td></td>
</tr>
<tr>
<td>Xanthan gum</td>
<td></td>
</tr>
<tr>
<td>Sodium bicarbonate</td>
<td>65</td>
</tr>
<tr>
<td>MCC</td>
<td>100</td>
</tr>
<tr>
<td>Magnesium stearate</td>
<td>5</td>
</tr>
<tr>
<td>Talc</td>
<td></td>
</tr>
<tr>
<td>Talc</td>
<td>5</td>
</tr>
<tr>
<td>Total weight</td>
<td>500</td>
</tr>
</tbody>
</table>

HPCMK4M: Hydroxypropyl methylcellulose; MCC: Microcrystalline cellulose
time intervals and the same amount of fresh medium was added to maintain the sink conditions. The collected samples were filtered through a 0.45 µm membrane filter and the concentration of dissolved drug was measured at 277 nm using a UV-visible spectrophotometer (LABINDIA). Cumulative percentage of drug release was calculated using equation obtained from a calibration curve [18].

Kinetic analysis of drug release data

To study the release mechanism of metronidazole from the sustained release floating matrix tablets, the in vitro drug release data were fitted into the following the mathematical models:

- Zero order
  \[ Q_t = Q_0 + k_t t \]  
- First order
  \[ \log C = \log C_0 - k_t t/20303 \]  
- Higuchi
  \[ Q_t = k_t t^{1/2} \]  
- Korsmeyer–Peppas
  \[ Q_t/Q_{ts} = k_t t^n \]  
- Hixon-Crowell cube root
  \[ (W_{t}^{1/3} - W_{1/3}) = k_t t \]

Where, \( Q_0, Q_t \) and \( Q_{ts} \) are the amounts of drug dissolved initially, at time \( t \) and at time infinite. \( C_t \) and \( C \) are the concentrations of drug initially and at time \( t \), \( W_t \) and \( W \) are the amounts of drug in the pharmaceutical dosage form initially and at time \( t \), and \( k_t, k, k_t, k_p \) and \( k_h \) refer to the rate constants obtained from the linear curves of the respective models. \( n \) is the diffusional exponent that characterizes the mechanism of drug release. The values of the coefficients were calculated using regression analysis between log \( Q/Q_0 \) and log time. The diffusional exponent \( n \) value was obtained from the slope of the regression equation and \( k_t \) was calculated from antilog of the intercept value.

If the value of \( n \) for a cylinder is <0.45, it suggests the Fickian release (diffusion controlled), if \( n \) >0.45 and <0.89, it is non-Fickian release (diffusion and polymer relaxation) and >0.89 it suggests super Case II release (relaxation and erosion) for swelleable systems.

For cylindrical systems like tablets, the \( n \) values of 0.45 and 0.89 represent pure diffusion or erosion release, respectively [19].

RESULTS AND DISCUSSION

Micromeritic properties of blended powder

The powder mixture of all the formulations was evaluated for angle of repose, bulk density, tapped density, and compressibility index. The compressibility index for all the formulations was found to be between 13.31±0.3° and 18.77±0.5°.

Evaluation of physical parameters

Sustained release floating matrix tablets of metronidazole were prepared by direct compression technique employing varying concentrations of HPMCK4, xanthan gum alone, and in combination. The physical properties of metronidazole floating matrix tablets were given in Table 2. Sustained release floating matrix tablets of metronidazole were prepared by direct compression technique. A total of nine formulations were prepared.

Effect of polymer concentration on matrix integrity of the tablets

Maintenance of matrix integrity of the tablets is an important parameter that needs to be studied because if the tablet does not maintain its physical integrity, it could be broken down into smaller fragments and escape from the gastrointestinal tract. In the present study, metronidazole has been formulated into sustained release floating tablet to release drug in the gastric environment to show its local action in the treatment of peptic ulcer. Buoyancy studies indicated that HPMCK4 alone could not maintain matrix integrity up to 12 h and disintegrated completely after 6 h as well as rapid erosion of the tablet was observed when placed in the dissolution medium due to its low viscosity. However, when combined with xanthan gum, it can control the release of drug and also matrix integrity was maintained up to 12 h.

In vitro dissolution study

Metronidazole release from the floating matrix tablets was studied for 12 h in 0.1 N hydrochloric acid. Total nine formulations were made with hydroxypropyl methylcellulose K4 and xanthan gum alone and in combination. Polymers employed in different ratios, i.e., 1:0.625 (25%), 1:0.75 (30%), and 1:0.875 (35%). The results of dissolution studies indicated that the burst release was observed with hydroxypropyl methylcellulose. Sustained release was observed with xanthan gum alone and its combination with HPMCK4. Formulation (F4) containing 25% of xanthan gum as release retardant was selected as best. The comparative drug release profiles are given in Fig. 1. Metronidazole release from all the matrix tablets was dependent on the concentration of release retardant contained in the tablet. The drug release from the matrix tablets was decrease as the concentration of release retardant was increased. HPMCK4 alone could not control the release

Table 2: Physical properties of floating matrix tablets of metronidazole

<table>
<thead>
<tr>
<th>Formulation code</th>
<th>Weight variation*</th>
<th>Hardness*</th>
<th>Friability</th>
<th>Content uniformity</th>
</tr>
</thead>
<tbody>
<tr>
<td>F1</td>
<td>497.30±1.36</td>
<td>3.32±0.14</td>
<td>0.72±0.24</td>
<td>98.23±1.32</td>
</tr>
<tr>
<td>F2</td>
<td>494.12±1.13</td>
<td>3.54±0.16</td>
<td>0.69±0.16</td>
<td>99.20±1.28</td>
</tr>
<tr>
<td>F3</td>
<td>498.76±1.26</td>
<td>3.50±0.12</td>
<td>0.70±0.18</td>
<td>99.64±1.34</td>
</tr>
<tr>
<td>F4</td>
<td>499.90±1.06</td>
<td>4.59±0.15</td>
<td>0.56±0.12</td>
<td>98.26±1.35</td>
</tr>
<tr>
<td>F5</td>
<td>498.64±1.32</td>
<td>4.78±0.10</td>
<td>0.66±0.11</td>
<td>99.05±1.34</td>
</tr>
<tr>
<td>F6</td>
<td>497.45±1.16</td>
<td>3.46±0.16</td>
<td>0.64±0.13</td>
<td>98.19±1.26</td>
</tr>
<tr>
<td>F7</td>
<td>498.82±1.09</td>
<td>4.52±0.14</td>
<td>0.72±0.13</td>
<td>99.56±1.32</td>
</tr>
<tr>
<td>F8</td>
<td>499.85±1.28</td>
<td>4.38±0.12</td>
<td>0.69±0.09</td>
<td>98.94±1.23</td>
</tr>
<tr>
<td>F9</td>
<td>498.64±1.09</td>
<td>4.22±0.13</td>
<td>0.71±0.10</td>
<td>99.56±1.20</td>
</tr>
</tbody>
</table>

**All values are expressed as mean±SD, *n=20 and *n=6. SD: Standard deviation

Table 3: Buoyancy study of floating matrix tablets of metronidazole

<table>
<thead>
<tr>
<th>Formulation code</th>
<th>Floating lag time (min)</th>
<th>Floating time (h)</th>
<th>Matrix integrity</th>
</tr>
</thead>
<tbody>
<tr>
<td>F1</td>
<td>&lt;1</td>
<td>&lt;12</td>
<td>-</td>
</tr>
<tr>
<td>F2</td>
<td>&lt;1</td>
<td>&lt;12</td>
<td>-</td>
</tr>
<tr>
<td>F3</td>
<td>&lt;1</td>
<td>&lt;12</td>
<td>-</td>
</tr>
<tr>
<td>F4</td>
<td>&lt;2</td>
<td>&gt;12</td>
<td>+</td>
</tr>
<tr>
<td>F5</td>
<td>&lt;1</td>
<td>&gt;12</td>
<td>+</td>
</tr>
<tr>
<td>F6</td>
<td>&lt;1</td>
<td>&gt;12</td>
<td>+</td>
</tr>
<tr>
<td>F7</td>
<td>&lt;2</td>
<td>&gt;12</td>
<td>+</td>
</tr>
<tr>
<td>F8</td>
<td>&lt;2</td>
<td>&gt;12</td>
<td>+</td>
</tr>
<tr>
<td>F9</td>
<td>&lt;2</td>
<td>&gt;12</td>
<td>+</td>
</tr>
</tbody>
</table>
of metronidazole from the tablets because the amount of polymer was not sufficient to maintain the matrix integrity due to its low molecular weight and viscosity. Xanthan gum could control the release of metronidazole up to 12 h due to its high molecular weight. These results indicate that xanthan gum has higher drug retarding ability than HPMCK4. The presence of xanthan gum alone and in combination with HPMCK4 retarded the initial release of drug from the tablets due to high swelling property.

**Drug release kinetics**

The release data of matrix tablets were fitted into various mathematical models (zero order, first order, Higuchi, and Peppas) to evaluate the kinetics and mechanism of drug release from the tablets. The model that best fits the release data is selected based on the correlation coefficient value in various models. The model that gives high R² value considered as the best fit of the release data. The R² values in zero, first order, Higuchi–Peppas, and Hixson–Crowell models are given in Table 4. The results indicated that the drug release from the matrix tablets followed zero-order kinetics. To evaluate the drug release mechanism from the matrix tablets plots of percent drug released versus square root of time as per Higuchi’s equation, log time versus log percentage of drug release as per Peppas and time versus cube root of drug remained as per Hisson-Crowell were constructed. When the release data were analyzed as per Peppas equation, the release exponent n value...
was ≤0.89 with all the tablets indicating Case II release, i.e., polymer swelling and relaxation might be the mechanism for the drug release from the matrix tablets.

### Compatibility studies

From the Fourier-transform infrared spectroscopy studies shown in Fig. 2, it is very clear that there are no interactions between drug and excipients. All the peaks responsible for the active functional groups were present in the best formulation (F4). The characteristic and observed peaks are mentioned in Table 5.

### CONCLUSION

The effervescent floating matrix tablets of metronidazole were prepared by direct compression method employing HPMCK4, xanthan gum as drug release retardants, and sodium bicarbonate used as gas generating agent. The concentration of gas generating agent and release retardants considerably affected the buoyancy time, matrix integrity, and profile of drug release. Thus, by changing the concentration of gas generating agent and polymer in formulations, the characterization of the matrix tablets could be controlled. All the formulations showed floating lag time <2 min. Xanthan gum alone and in combination with HPMCK4 has successfully sustained the release of metronidazole. The drug release from the floating matrix tablets was found to be zero-order kinetics with Case II transport, involves polymer swelling and chain relaxation as mechanism of drug release. Floating tablets of metronidazole sustained the drug release for 12 h can be expected to eradicate *H. pylori* in peptic ulcer disease.

### ACKNOWLEDGMENT

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### CONFLICTS OF INTEREST

Authors have no conflicts of interest in publishing this research work.

### AUTHORS’ CONTRIBUTION

All the authors have contributed equally.

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### Table 4: Correlation coefficients (R) values of drug release kinetics and mechanism of floating matrix tablets of metronidazole

<table>
<thead>
<tr>
<th>Formulation</th>
<th>Zero order</th>
<th>First order</th>
<th>Higuchi</th>
<th>Peppas</th>
<th>Hixson-Crowell</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>R²</td>
<td>n</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>F1</td>
<td>0.826</td>
<td>0.971</td>
<td>0.975</td>
<td>0.957</td>
<td>0.46</td>
</tr>
<tr>
<td>F2</td>
<td>0.881</td>
<td>0.960</td>
<td>0.995</td>
<td>0.991</td>
<td>0.47</td>
</tr>
<tr>
<td>F3</td>
<td>0.953</td>
<td>0.949</td>
<td>0.973</td>
<td>0.979</td>
<td>0.75</td>
</tr>
<tr>
<td>F4</td>
<td>0.976</td>
<td>0.890</td>
<td>0.962</td>
<td>0.983</td>
<td>0.89</td>
</tr>
<tr>
<td>F5</td>
<td>0.979</td>
<td>0.781</td>
<td>0.936</td>
<td>0.988</td>
<td>0.90</td>
</tr>
<tr>
<td>F6</td>
<td>0.989</td>
<td>0.959</td>
<td>0.944</td>
<td>0.994</td>
<td>0.85</td>
</tr>
<tr>
<td>F7</td>
<td>0.975</td>
<td>0.761</td>
<td>0.971</td>
<td>0.994</td>
<td>0.80</td>
</tr>
<tr>
<td>F8</td>
<td>0.955</td>
<td>0.896</td>
<td>0.970</td>
<td>0.990</td>
<td>0.78</td>
</tr>
<tr>
<td>F9</td>
<td>0.992</td>
<td>0.938</td>
<td>0.955</td>
<td>0.996</td>
<td>0.86</td>
</tr>
</tbody>
</table>

### Table 5: Data of Fourier transform infrared spectra of pure and best formulation

<table>
<thead>
<tr>
<th>Functional group</th>
<th>Characteristic peak/cm⁻¹</th>
<th>Observed peaks-pure drug/cm⁻¹</th>
<th>Observed peaks-best formulation/cm⁻¹</th>
</tr>
</thead>
<tbody>
<tr>
<td>NO₃ (stretching)</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Symmetric</td>
<td>1375–1385</td>
<td>1369.50</td>
<td>1369.50</td>
</tr>
<tr>
<td>Asymmetric</td>
<td>1475–1550</td>
<td>1535.39</td>
<td>1535.39</td>
</tr>
<tr>
<td>C=C (stret) aromatic</td>
<td>1450–1600</td>
<td>1479.45</td>
<td>1479.45</td>
</tr>
<tr>
<td>C=N (stret) imidazole</td>
<td>1100–1230</td>
<td>1155.40</td>
<td>1155.33</td>
</tr>
<tr>
<td>O-H (stret) alcholic</td>
<td>3200–3550</td>
<td>3219.30</td>
<td>3219.30</td>
</tr>
<tr>
<td>C=N (stret) imidazole</td>
<td>1640–1690</td>
<td>1654.98</td>
<td>1653.05</td>
</tr>
</tbody>
</table>

**REFERENCES**


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