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THERAPEUTIC INDICES OF SOME ANTIFUNGAL ORGANIC COMPOUNDS

EUGENE SEBASTIAN J. NIDIRY*

Division of Flower and Medicinal Crops, ICAR-Indian Institute of Horticultural Research, Hesaraghatta Lake P.O., Bengaluru, Karnataka, India. Email: nidiry@yahoo.co.in

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ABSTRACT

Objective: The objective of the study was to evaluate the therapeutic indices (TIs) some antifungal organic compounds and to compare their safety.

Methods: TIs of 55 organic compounds were evaluated by dividing their mammalian toxicity values given as median lethal doses ([mg/kg] rat, oral) collected from literature by experimentally determined fungitoxicity values evaluated as median effective concentration (EC₅₀ [mg/L]) for the mycelial growth inhibition of *Colletotrichum gloeosporioides* on potato dextrose agar medium.

Results: Large variations in TIs are observed in different classes of compounds, namely, alcohols, carboxylic acids, phenols, esters, and aldehydes. The relationship between carbon chain length of saturated monohydric alcohols and mammalian toxicity shows that toxicity increases from chain length 1 to 5, declines from 5 to 6, increases from 6 to 7, and steadily declines from 7 to 10. Thus, the relationship is by and large parabolic. The relationship between carbon chain length and molar antifungal activity is also parabolic. In spite of this, large differences are found in TIs making 1-decanol (TI=42.1) a very safe antifungal compound and 1-pentanol (TI=0.17) a hazardous antifungal compound. This is because mammalian toxicity reaches a maximum around carbon chain length of 5, while in the case of antifungal activity, it reaches a maximum around carbon chain length of 10.

Conclusions: Among the 55 compounds, whose TIs were evaluated, some compounds with high TIs (>10) which can be considered safe as antifungal compounds are citronellol, geraniol, 1-decanol, 1-octanol, phenyl acetaldehyde, methyl anthranilate, 1-naphthol and 2-naphthol. Some compounds with low TIs (<0.50) which can be considered hazardous as antifungal compounds are resorcinol, hydroquinone, oxalic acid, and acetyl acetone.

Key words: Therapeutic indices, Safe antifungal compounds, Hazardous antifungal compounds, Median lethal doses, Median effective concentrations.

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INTRODUCTION

Fungi are the causal organisms of many diseases of man, animals, and plants. They also cause post-harvest spoilage of agricultural products. Natural and synthetic antifungal compounds have been extensively used to control fungi [1-4].

Development of new antifungal compounds is based on extensive structure-activity relationship (SAR) studies on existing antifungal compounds. In earlier publications from this institute, SARs of several such compounds were reported [5-9]. However, high activity of any compound does not make it a good candidate to be recommended for human use because of the possibility of that compound having high toxicity to human beings also. Thus, therapeutic index (TI) [10,11] also known as selective toxicity coefficient [12,13] which is the ability of a compound to injure one kind of living matter without harming some other kind even though both are in intimate contact is considered as an important criterion for recommending any compound for its safe use as antifungal compound. Since this awareness is important, TI values of 55 common organic compounds whose antifungal activity values had been reported earlier from this institute [5-9] were evaluated and are reported in this paper. These TI values were calculated by taking quotient of mammalian toxicity expressed as median lethal doses (LD₅₀ mg/kg [rat, oral]) collected from literature to antifungal activity expressed as median effective concentration (EC50 [mg/L]) for the mycelial growth inhibition of Colletotrichum gloeosporioides on potato dextrose agar (PDA) medium. Large variation in TI values is observed in different classes of compounds. The variation in the case of monohydric saturated straight chain alcohols is interpreted in terms of SAR studies.

METHODS

The percent mycelial growth inhibitions of *C. gloeosporioides* on PDA medium at five different concentrations the compounds were

determined by poisoned food technique [14]. The median ECs for mycelial growth inhibition (EC_{50}) were calculated by probit analysis [15]. The LD₅₀ values were collected from literature [16,18-27]. Therapeutic indices (TIs) of each compound were evaluated by taking quotient of LD₅₀ (mg/kg) to EC₅₀ (mg/L).

RESULTS

The data on mammalian toxicity, fungitoxicity, and TIs of 55 compounds are presented in Table 1. The listed compounds consist of 23 alcohols, 8 carboxylic acids, 8 phenols, 6 esters, 4 aldehydes, 4 ketones, 1 indole, and 1 amine. For SAR studies of saturated monohydric straight chain alcohols, which are the first nine compounds listed in Table 1, partition coefficient, molar antifungal activity, and molar mammalian toxicity values were calculated and these are presented in Table 2.

DISCUSSION

Large variations are seen in the TIs within any class of compounds.

The first nine compounds in Table 1 are saturated monohydric straight chain alcohols. Even within these nine compounds, large variations in terms of TI are seen making 1-decanol (TI=42.1) a very safe antifungal compound and 1-pentanol (TI=0.17) hazardous as an antifungal compound. The molar antifungal activity values (pEC_{50} [moles/L]) and molar mammalian toxicity values (pLD_{50} [moles/Kg]) of these compounds are presented in Table 2. It is obvious that in the case of monohydric straight chain alcohols, hydrophobicity increases with carbon chain length [7]. The mammalian toxicity increases up to carbon chain length of 5, then it shows a slight decline for carbon chain length of 6, then it increases for carbon chain length of 7 and then decreases steadily up to chain length of 10. Thus, the relationship between carbon chain length and mammalian toxicity is by and large parabolic. This is invariably because of the parabolic relationship between

S. No.	Compound	Chemical class	LD ₅₀ (mg/Kg)	Reference for LD ₅₀	EC ₅₀ (mg/L)	TI
1	Methanol	Alcohol	5628	[18]	24,687	0.23
2	Ethanol	Alcohol	7060	[20]	17,683	0.40
3	1-Propanol	Alcohol	1870	[18]	8679	0.22
4	1-Butanol	Alcohol	790	[20]	3694	0.21
5	1-Pentanol	Alcohol	370	[20]	2129	0.17
6	1-Hexanol	Alcohol	720	[21]	1271	0.57
7	1-Hepanol	Alcohol	500	[20]	546	0.91
8	1-Octanol	Alcohol	3200	[20]	298	10.74
9	1-Decanol	Alcohol	4720	[20]	112	42.14
10	Citronellol	Alcohol	3450	[20]	84	41.07
11	Geraniol	Alcohol	3600	[22]	100	36.00
12	Benzyl alcohol	Alcohol	1250	[18]	1181	1.06
13	Phenyl ethyl alcohol	Alcohol	1609	[23]	1000	1.61
14	2-Propanol	Alcohol	5045	[20]	18,175	0.28
15	2-Methyl 1-propanol	Alcohol	2460	[22]	8205	0.30
16	3-Methyl 1-butanol	Alcohol	1300	[20]	3713	0.35
17	1-Octen-3-ol	Alcohol	340	[18]	1286	0.26
18	Menthol	Alcohol	2046	[19]	452	4.53
19	Neomenthol	Alcohol	3180	[21]	488	6.52
20	Tertiary butyl alcohol	Alcohol	3500	[20]	17106	0.20
21	Linalool	Alcohol	2790	[20]	1090	2.56
22	Ethylene glycol	Alcohol	4700	[20]	34,050	0.14
23	Glycerol	Alcohol	4090	[20]	59,665	0.07
24	Acetic acid	Carboxylic acid	3310	[20]	838	3.95
25	Propionic acid	Carboxylic acid	2600	[20]	343	7.58
26	Lactic acid	Carboxylic acid	3543	[22]	4767	0.74
27	Benzoic acid	Carboxylic acid	1700	[20]	843	2.02
28	4-Hydroxybenzoic acid	Carboxylic acid	6000	[19]	2485	2.41
29	Oxalic acid	Carboxylic acid	375	[25]	1153	0.33
30	Tartaric acid	Carboxylic acid	7500	[20]	5230	1.43
31	Citric acid	Carboxylic acid	3000	[22]	6703	0.45
32	Phenol	Phenol	317	[20]	233	1.36
33	Catechol	Phenol	300	[25]	239	1.26
34	Resorcinol	Phenol	301	[20]	1540	0.20
35	Hydroquinone	Phenol	320	[21]	1716	0.19
36	Pyrogallol	Phenol	789	[26]	458	1.72
37	Phloroglucinol	Phenol	4000	[25]	9959	0.40
38	1-Naphthol	Phenol	1870	[20]	60	31.17
39	2-Naphthol	Phenol	1960	[20]	46	42.61
40	Ethyl acetate	Ester	11300	[16]	8414	1.34
41	Ethyl acetoacetate	Ester	3980	[20]	2423	1.64
42	Menthyl acetate	Ester	5001	[27]	1263	3.96
43	Benzyl acetate	Ester	2490	[22]	999	2.49
44	Methyl anthranilate	Ester	2910	[21]	149	19.53
45	Benzyl benzoate	Ester	1700	[26]	19,480	0.09
46	1-Hexanal	Aldehyde	4890	[21]	2363	2.07
47	Phenyl acetaldehyde	Aldehyde	1550	[20]	278	5.58
48	Benzaldehyde	Aldehyde	1300	201	927	1.40
49	<i>p</i> -Anisaldehyde	Aldehyde	1510	[20]	712	2.12
50	Acetone	Ketone	5800	[20]	118,584	0.05
51	Methyl isobutyl ketone	Ketone	2080	[24]	12,609	0.16
52	Acetyl acetone	Ketone	55	[20]	1587	0.03
53	Menthone	Ketone	1600	[23]	1761	0.91
54	Indole	Indole	1000	[22]	187	5.35
55	Aniline	Amine	250	[22]	457	0.55
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Table 1: Therapeutic indices of some common antifungal organic compounds

LD₅₀: Median lethal dose (mg/Kg, rat, oral) collected from the respective references in the next column, EC₅₀: Median effective concentration (mg/L) for the mycelial growth inhibition of *Colletotrichum gloeosporioides* on PDA medium, TI: Therapeutic index

hydrophobicity (partition coefficient) and mammalian toxicity. The relationship between carbon chain length (hydrophobicity) and antifungal activity is also parabolic. However, in this case, the activity increases up to carbon chain length of 10. Earlier studies [7,17] have shown that antifungal activity of straight chain alcohols decreases after chain length of 10. This explains the low therapeutic index (hazardous nature) of 1-pentanol and high therapeutic index (safety) of compounds such as 1-octanol and 1-decanol. These high TIs of unsaturated alcohols such as citronellol and geraniol may also be because they are C-10 alcohols. Alcohols with low TIs include methanol, ethanol, propanol and butanol, 1-pentanol, and 1-octen-3-ol. In the case of carboxylic acids also, large variation is seen in TIs making propionic acid a very safe antifungal compound and oxalic acid hazardous as an antifungal compound. In the case of phenols, 1-naphthol and 2-naphthol can be considered safe antifungal compounds while resorcinol and hydroquinone can be considered hazardous as antifungal compounds. Among esters, benzyl benzoate has low therapeutic index. Methyl anthranilate has very high therapeutic index and can be considered safe. Among aldehydes, phenyl acetaldehyde with high therapeutic index can be considered as a safe antifungal compound. Since all the ketones given in Table 1 have low TI values, they cannot be considered safe

S. No.	Compound	No. of carbons	CLogP	pEC ₅₀ (moles/L)	pLD ₅₀ (moles/Kg)
1	Methanol	1	-0.77	0.11	0.75
2	Ethanol	2	-0.11	0.42	0.81
3	1-Propanol	3	0.43	0.83	1.51
4	1-Butanol	4	0.97	1.30	1.97
5	1-Pentanol	5	1.51	1.62	2.38
6	1-Hexanol	6	2.05	1.92	2.15
7	1-Hepanol	7	2.59	2.33	2.37
8	1-Octanol	8	3.13	2.64	1.61
9	1-Decanol	10	4.21	3.15	1.52

Table 2: Calculated partition coefficients, molar fungitoxicity, and molar mammalian toxicity values of monohydric straight chain alcohols

CLogP: Calculated partition coefficient [7]. pEC₅₀ (moles/L): Negative logarithm of median effective molar concentration for mycelial growth inhibition of *Colletotrichum* gloeosporioides on PDA medium pLD₅₀ (moles/kg): Negative logarithm of median molar LD₅₀ (rat, oral)



Fig. 1: Variation in fungitoxicity [pEC₅₀(moles/L)] and mammalian toxicity [pLD₅₀(moles/Kg)] of monohydric saturated straight chain alcohols

as an antifungal compound, but aniline with low TI is unsafe as an antifungal compound.

CONCLUSIONS

Among the 55 antifungal organic compounds whose therapeutic indices (TI) were evaluated, some compounds with high therapeutic indices (TI>10) which can be considered safe as antifungal compounds are citronellol, geraniol, 1-decanol, 1-octanol, phenyl acetadehyde, methyl anthranilate, 1-naphthol and 2-naphthol . Some compounds with low therapeutic indices (TI<0.50), which can be considered hazardous as antifungal compounds are resorcinol, hydroquinone, oxalic acid and acetyl acetone.

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AUTHORS' CONTRIBUTIONS

Eugene Sebastian J. Nidiry was involved in the evaluation of fungitoxicity data (EC_{50}), collection of mammalian toxicity data (LD_{50}) from literature, and calculation of TIs and interpretation of the data.

CONFLICTS OF INTEREST

The author declares that there are no conflicts of interest.

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