

## SYNTHESIS, REACTIONS, AND PHARMACOLOGICAL APPLICATIONS OF 2-AMINOBENZIMIDAZOLES: AN UPDATE

APEKSHA MOTGHARE\*, PARIMAL KATOLKAR

Department of Pharmaceutical Chemistry, Kamla Nehru College of Pharmacy, Butibori, Nagpur (MS), India.

Email: parimal.katolkar@gmail.com, apekshamotghare6@gmail.com

Received: 09 April 2022, Revised and Accepted: 25 May 2022

### ABSTRACT

Many drug molecules comprises of benzo fused heterocycles with two nitrogen atoms, that is, benzimidazole and its derivatives. Many biological active molecules contain that 2-aminobenzimidazole cores are among the foremost common structural components in medicinal chemistry. 2-aminobenzimidazole and its derivative have wide range of biological and pharmaceutical activities. In this review, the authors summarize synthesis, various chemical reactions, and biological activities of 2-aminobenzimidazole and its derivative.

**Keywords:** 2-Aminobenzimidazoles, Derivatives of 2-aminobenzimidazole, Pharmacological applications, Reactions.

© 2022 The Authors. Published by Innovare Academic Sciences Pvt Ltd. This is an open access article under the CC BY license (<http://creativecommons.org/licenses/by/4.0/>) DOI: <http://dx.doi.org/10.22159/ajpcr.2022v15i7.44899>. Journal homepage: <https://innovareacademics.in/journals/index.php/ajpcr>

### INTRODUCTION

2-Aminobenzimidazole may be a readily available chemical and may even be easily prepared within the laboratory. This review highlights different methods for the preparation of varied 2-aminobenzimidazoles. The utility of this organic synthon in preparing a good type of substituted benzimidazoles and benzimidazole heterocycles is systematically discussed with the applications of such compounds for the event of some new chemotherapeutic agents [1]. 2-aminobenzimidazole may be a member of the category of benzimidazoles, that's benzimidazole during which the hydrogen at position 2 is replaced by an amino group. 2-aminobenzimidazole (white shiny plates; m.p.220–222°C) is one in every of the longest known nitrogen heterocycles and has been recently recognized as a useful building block for the synthesis of a good kind of substituted benzimidazoles and benzimidazole heterocycles of educational, pharmaceutical, and industrial interest. The polyfunctionality resulting from the cyclic guanidine residue has made 2-aminobenzimidazole a flexible material in organic synthesis [2].

The 2-amino derivative is often found within the medicinal chemistry literature as a starting material [3]. A plethora of medicinal properties has been shown by derivatives of 2-substituted benzimidazole as a target molecule [4].

### SYNTHESIS OF 2-AMINOBENZIMIDAZOLE

#### Using cyanamide

2-aminobenzimidazole is a crucial parent substance, for example, pesticides. Although it is often obtained by ring closure of o-phenylenediamine with cyanogen bromide [5], this process is unsuitable for industrial work. We have been able to prepare in almost quantitative yield and in excellent purity by treating o-phenylenediamine with cyanamide, which is quickly available on an industrial scale [6] (Scheme 1).

#### Using mercuric oxide

The present invention consists a process for preparing 2-aminobenzimidazole comprising ring-closing an N-(o-aminophenyl)



Scheme 1: Synthesis of 2-aminobenzimidazole using cyanamide

thiourea with mercuric oxide to administer a 2-aminobenzimidazole [7] (Scheme 2).

### SYNTHESIS OF 2-SUBSTITUTED BENZIMIDAZOLES

#### Using thiourea and related derivatives

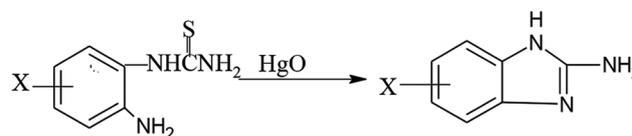
The reaction is occur through formation of carbodiimides as an intermediate further cyclization of 1-(2-amino-phenyl)thiourea results in the formation of 2-Alkyl-2-aryl and 2-acylamino benzimidazoles [8] (Scheme 3).

The cyclisation carried out with help of lead oxide and mercuric oxide in dry chloroform. Dimethyl sulfate and methyl iodide in ethanol mercury(II)chloride were proposed as a reagent [9-13]. The cyclization of (o-aminophenyl) thiourea was resulted in benzimidazole thione as a product, instead of synthesizing unsubstituted 2-aminobenzimidazole [14,15]. The formation of 2-alkoxycarbonylamino benzimidazole was prepared from reaction of N-[alkoxy(methylthio)methylene] and o-phenylenediamines with alkyl esters of carbalkoxythiocarbamic. The synthesise of carbamates was planned by Murray and Dains, in 1934, using o-phenylenediamine with 1,3-bis(alkoxycarbonyl)-S-methylisothiurea [16-22].

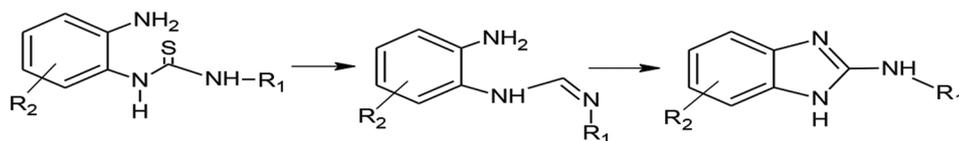
#### Using cyanogen halide, cyanamide and cyanoguanidines

The synthesis of several 2-aminobenzimidazole is synthesized using the Pierron process [23].

2-amino and 2-dimethylaminobenzimidazoles produce with reaction of dimethyl cyanamide and hydrochloride salts of o-phenylenediamine with cyanamide, in low yields [24-26]. The reaction of chlorocarbonic acid esters and acyl chlorides with cyanamide in the presence of pyridine or hydroxide triethylamine or with calcium cyanamide to obtain ethoxycarbonyl substituted cyanamide [27-31] (Scheme 4).

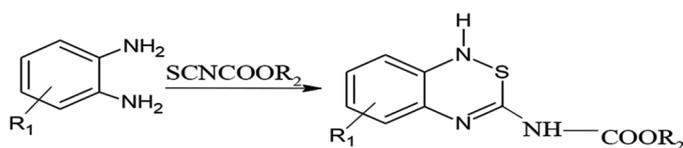
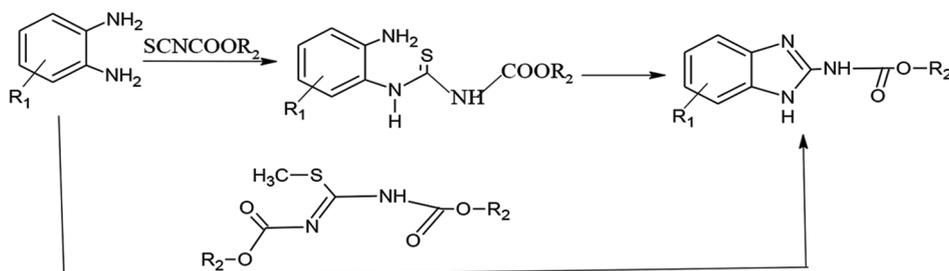


Scheme 2: Synthesis of 2-aminobenzimidazole using mercuric oxide



$R_1$  = alkyl, benzoyl, aryl, etc.

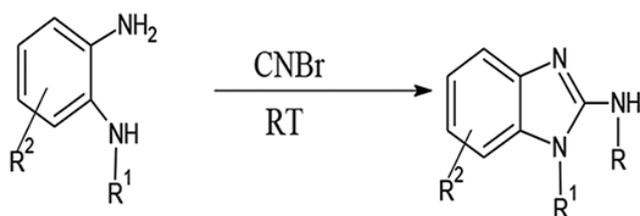
$R_2$  = H, alkyl, halogen, etc.



$R_1$  = alkyl, nitro

$R_2$  = methyl, ethyl

Scheme 3: Synthesis of 2-aminobenzimidazole using thiourea and similarly assembled compounds



$R$  = amide, alkyl, hydrogen, etc.

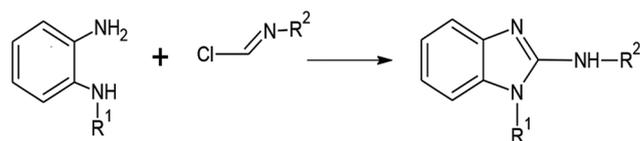
$R_1$  = hydrogen, alkyl, cycloalkyl, etc.

$R_2$  = halogen, hydrogen, nitro, alkyl, etc.

Scheme 4: Preparation of 2-aminobenzimidazole using cyanogens halide

#### Using N-substituted carbon imidoyl dichlorides

2-sodio-1-methyl benzimidazole and O-methylhydroxylamine were used as unsuccessful attempts to make the corresponding amine derivatives. The formation of 2-diethylamino-1-methylbenzimidazole derivative in low yield with the action of  $\text{BrN}(\text{C}_2\text{H}_5)_2$  on the 2-sodio byproduct. The synthesis of 2-aminobenzimidazole has synthesized by the help of N-substituted carbon imidoyl dichloride. The reaction between o-phenylenediamine and N-phenyl-o-phenylenediamine with N-aryl carbon imidoyl dichloride was resulted in 2-arylamino benzimidazoles and 1-phenyl-2-anilino benzimidazole. Exploitation of N-alkoxy carbonyl carbon imidoyl dichloride resulted in 2-alkoxycarbonylamino benzimidazoles in the presence of dioxane, chloroform and triethylamine [32,33] (Scheme 5).



$R^1$  = H, phenyl, etc.

$R^2$  = phenyl, alkyl, etc.

Scheme 5: Synthesis of 2-aminobenzimidazoles using N-substituted carbon imidoyl dichloride

Benzimidazole derivatives initiated by rearrangement with some quinoxaline derivatives by photochemically, the yield depends on the pH of the medium and the solvent [34] (Scheme 6).

#### Using urea derivatives with aryl substituted guanidines

Cyclisation of (N-Phenacyl amino) phenyl urea and phosphorus oxychloride by heating leads to 1-phenacyl-2-aminobenzimidazole and 2-benzylamino benzimidazole resulted using 1-(o-amino phenyl)-3-benzyl urea in toluene and p-toluene acid by reflux [35-37] (Scheme 7).

Cyclization of N-(o-Amino phenyl)-N-methyl-N',N'-disubstituted guanidines to aminobenzimidazole by treating with  $\text{CS}_2$ ,  $\text{CSCl}_2$ , and  $\text{HC}(\text{OC}_2\text{H}_5)_3$  [38]. De composition of 2-Acyloxyguanidines results in 2-dialkylamino benzimidazoles at room temperature [39]. An imino nitrene intermediate was obtained as result of the rearrangement of N-aryl-N-hydroxyamidines to benzimidazoles [40,41] (Scheme 8).

#### REACTIONS OF 2-AMINO BENZIMIDAZOLE

##### Oxidation reaction

2-acetyl aminobenzimidazole results from 2-aminobenzimidazole and acetic anhydride. Benzimidazoles are stable to oxidation (Scheme 9).

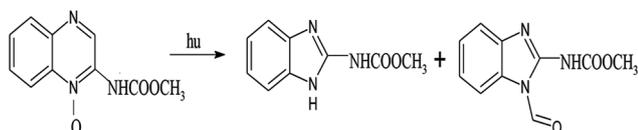
A little amount of imidazolidicarboxylic acid was obtained in the presence of strong oxidizing agent. A spread of benzimidazole carboxylic acids is prepared by the substituent group [42].

**Electrophilic reaction**

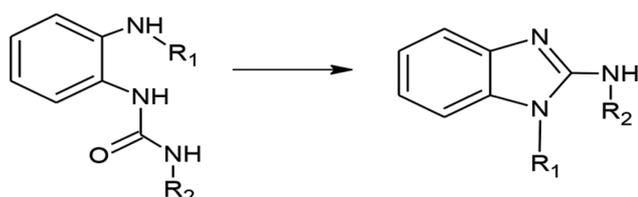
Nuclear substitutions of 2-aminobenzimidazoles using various electrophiles have yielded several 5,6-substituted benzimidazoles. Thus, 2-amino-5-chlorobenzimidazole is obtained by reaction of 2-aminobenzimidazole with oxide or hydrochloric acid or sodium hypochlorite. Within the latter case, four other benzimidazoles were also isolated from the reaction mixture [43]. Similarly, bromination of 1-alkyl-2-aminobenzimidazoles gave both 5-bromo- and 5,6-dibromobenzimidazoles (1) and (2). Nitration of 1-alkyl-2-aminobenzimidazoles and 1,3-diethyl-2-iminobenzimidazole yielded 5,6-dinitro-2-aminobenzimidazoles and therefore the corresponding 2-imino derivative, respectively. However, nitration of 1,3-dimethyl-2-iminobenzimidazole gave both 5-nitro- and 5,6-dinitro-1,3-dimethyl-2-iminobenzimidazoles [44] (Scheme 10).

**Reaction with nitriles**

2-aminobenzimidazole reacts with cyanates to yield 1-substituted-2-aminobenzimidazoles (3), which can be cyclized with an aldehyde or ketone in presence of an acid or a base to yield

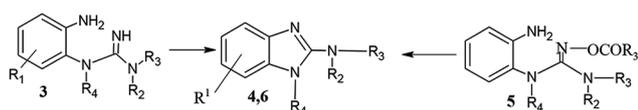


**Scheme 6: Synthesis of 2-aminobenzimidazole using N-substituted carbon imidoyl chlorides**



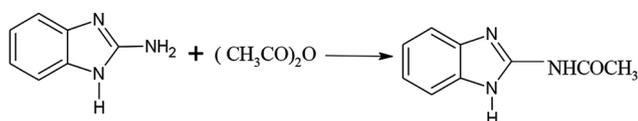
R<sub>1</sub> = H, alkyl, benzyl, etc.  
R<sub>2</sub> = H, alkyl, benzyl, etc.

**Scheme 7: Synthesis of 2-aminobenzimidazole using urea derivatives**



In 3,4 R<sub>1</sub>,R<sub>2</sub>,R<sub>3</sub>,R<sub>4</sub> = H, alkyl  
In 5,6 R<sub>1</sub>,R<sub>2</sub>,R<sub>3</sub>,R<sub>4</sub> = H, alkyl, benzyl, phenyl, phenyl amine

**Scheme 8: Synthesis of 2-aminobenzimidazole using Aryl substituted guanidines**



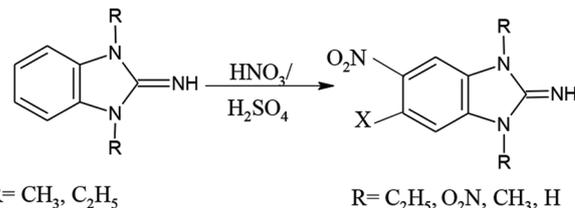
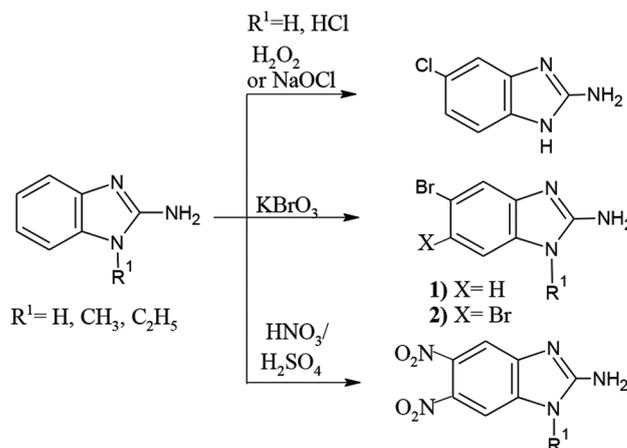
**Scheme 9: Oxidation reaction of 2-aminobenzimidazole**

the triazinobenzimidazoles(4) [45,46]. Similarly, treatment of 2-aminobenzimidazole with N-cyanofornimidates yields aminobenzimidazo[1,2-a]-1,3,5-triazines which can have the structures (5) or (6). The structure (5) was proved by its unambiguous synthesis from 2-guanylbenzimidazole and triethyl orthoformate [47] (Scheme 11).

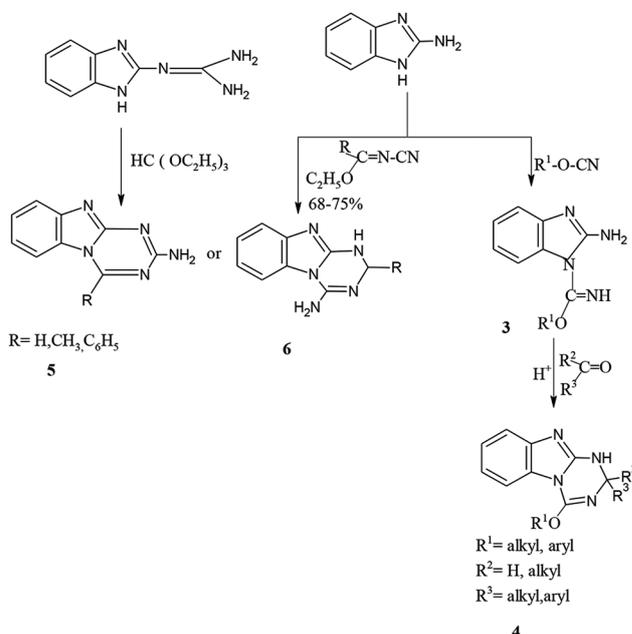
**Substitution reactions**

*Alkylation and arylation reactions*

Alkylations and arylations of 2-aminobenzimidazoles with alkyl or aryl halides create to moderate yields of 1-substituted-2-aminobenzimidazoles or 1,3-disubstituted-2-iminobenzimidazoles, depending on the reactivity of organic compound, which ends up



**Scheme 10: Electrophilic reaction of 2-aminobenzimidazole**



**Scheme 11: Reaction of 2-aminobenzimidazole with nitriles**

from their respective quaternary salts on action with a base. In an alternate method, 1-alkyl-2-aminobenzimidazole could also be treated with an organic compound to afford the corresponding 1,3-dialkyl-2-aminobenzimidazoles, which also involves quaternary salts as an intermediate product. Reaction of 2-aminobenzimidazole with alkyl iodides with hydrated oxide is to afford 1-alkyl-2-aminobenzimidazoles [48] (Scheme 12).

Another simple method to organize 1,3-dimethyl-2-methyliminobenzimidazole involves the methylation of 2-aminobenzimidazole with dimethyl sulfate. An identical reaction of 2-phenylaminobenzimidazole with dimethyl sulfate gives 1,3-dimethyl-2-phenyliminobenzimidazole [49] (Scheme 13).

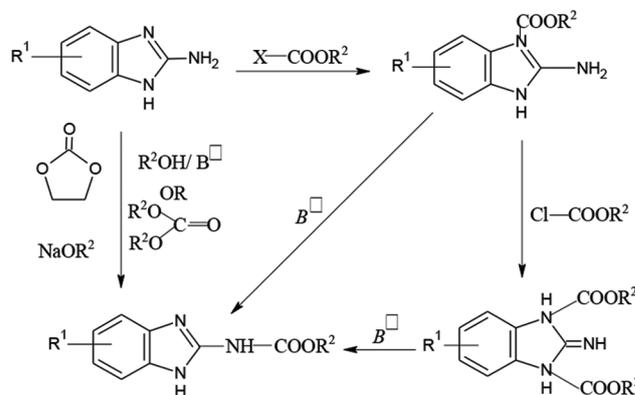
*Acylation and arylation reaction*

Acylation of 2-aminobenzimidazoles with an oversized form of alkyl carbonhalidates gives alkyl benzimidazole-2-carbamates in good yields. Better yields of alkyl benzimidazole-2-carbamates could also be obtained by treating 2-aminobenzimidazole, generated in place, with alkyl carbonochloridates [50]. The latter can also be prepared either by reacting 2-aminobenzimidazole with alkyl carbonates in presence of a base or with ethylene carbonate in presence of an alcohol and a base. The formation of alkyl benzimidazole-2-carbamate may proceed through the formation of the mono- or disubstituted benzimidazoles which when treated with a base, bring about to alkyl benzimidazole-2-carbamate [51] (Scheme 14).

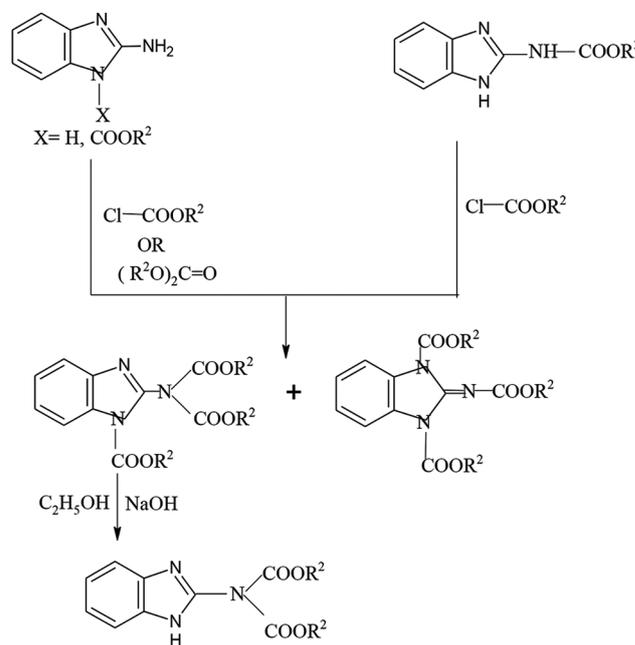
Using an appropriate quantity of alkyl carbonohalidates or the dialkyl carbonates within the above reaction, the tricarboxylates 1-ethoxycarbonyl-2-(diethylcarbonylamino)-benzimidazole and 2-ethoxycarbonylimino-1,3-diehoxycarbonyl-1,3-dihydrobenzimidazole were isolated from the reaction mixture; these compounds could also be also obtained by reaction of 1-ethoxycarbonyl-2-aminobenzimidazole with dialkyl carbonates. Better yields of

1-ethoxycarbonyl-2-(diethylcarbonylamino)-benzimidazole were obtained by treating 2-ethoxycarbonylamino benzimidazole with ethyl carbonochloridate. The merchandise 1-ethoxycarbonyl-2-(diethylcarbonylamino)-benzimidazole is hydrolyzed with alkaline ethanol to afford the corresponding 2-(N,N'-dialkoxy carbonylamino)-benzimidazole [52] (Scheme 15).

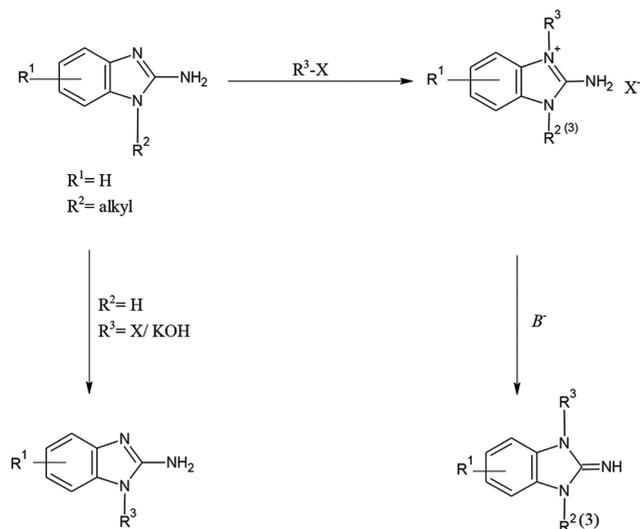
The opposite 2-N-acylaminobenzimidazoles could also be obtained by treating 2-aminobenzimidazole with acyl halides or acetic anhydride [53] (Scheme 16).



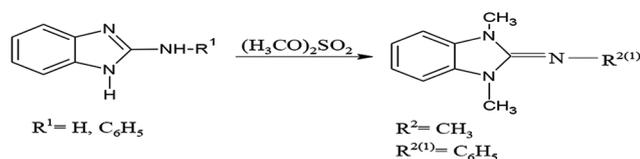
**Scheme 14: Acylation reaction of 2-aminobenzimidazole**



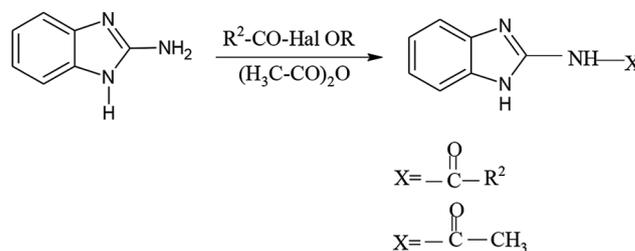
**Scheme 15: Acylation reaction of 2-substituted aminobenzimidazoles**



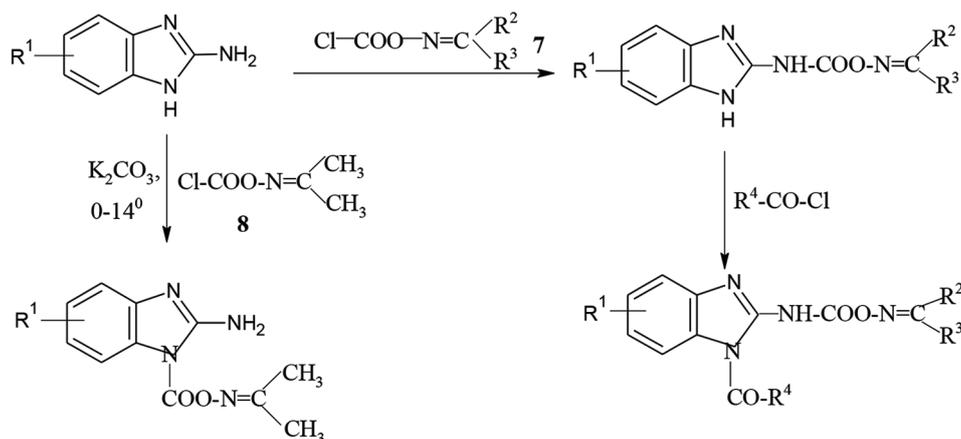
**Scheme 12: Alkylation and arylation reaction of 2-aminobenzimidazole**



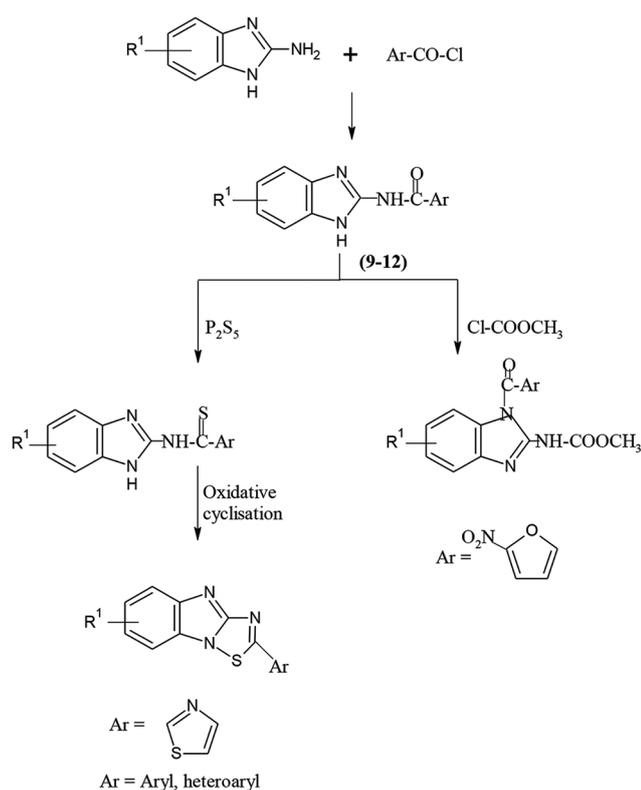
**Scheme 13: Alkylation and arylation reaction of 2-aminobenzimidazole**



**Scheme 16: Acylation reaction of 2-substituted 2-aminobenzimidazole**



Scheme 17: Acylation reaction of 2-aminobenzimidazole



Scheme 18: Aroylation reaction of 2-aminobenzimidazole

A number of benzimidazolyl carbamates are prepared as potential antiparasitic agents. Thus, reaction of 2-aminobenzimidazole with (7) gave the corresponding benzimidazole-2-carbamate, some of which were further acylated to afford the desired 1,2-disubstituted benzimidazoles. When the reaction of 2-aminobenzimidazole with (8) was allotted in acetone or dichloromethane within the presence of potassium carbonate at 0–140°C, a 74% yield of the 1-substituted-2-aminobenzimidazole was obtained [54] (Scheme 17).

The aroylation of 2-aminobenzimidazole yields both mono- and disubstituted benzimidazoles with a broad spectrum of biological activity. Thus, reaction of 2-aminobenzimidazole with aromatic acid chlorides yields the corresponding 2-arylamino benzimidazoles (9-12), which can be further acylated to allow 1,2-disubstituted benzimidazoles or thiolated to the corresponding thiones. The latter was cyclized within the presence of bromine to present the 2-substituted-

1,2,4-thiadiazolo[2,3-a]benzimidazoles. Several other thiadiazolo[2,3-a]benzimidazoles have also been prepared similarly [55] (Scheme 18).

Another method to arrange various 2-[N-(2-benzimidazo)-carbonyl]aminobenzimidazoles involves the nucleophilic ring opening of 6,13-dioxo-6H,13H-pyrazino[1,2-a:4,5-a]bisbenzimidazoles with 2-aminobenzimidazole [56] (Scheme 19).

Benzylation of 2-amino-1-methylbenzimidazole with benzoyl chloride in acetone at temperature gave the benzimidazolium chloride, which afforded 2-benzoylamino-1-methylbenzimidazole when heated with a base in chloroform. The merchandise 2-benzoylamino-1-methylbenzimidazole was obtained through an imine intermediate; the structure of an imine intermediate was established by converting it into 1-methylamino-2-methylbenzimidazole [57] (Scheme 20).

Further work on the mechanism of the aroylation of 2-aminobenzimidazoles involved reaction with various aroyl chlorides to make intermediate 1,3-diaroyl-2-iminobenzimidazoles which were readily transformed into 2-N-(aroylamino)-benzimidazoles within the presence of a base. The structure of 2-(N-aroylamino)-benzimidazoles was supported by the very fact that reaction of 2-(N-aroylamino)-benzimidazoles with benzyl chloride yielded 1-benzyl-2-(4-chloro-3-nitrobenzoyl)-aminobenzimidazole which was alternatively obtained by treating 1-benzyl-2-aminobenzimidazole with 4-chloro-3-nitrobenzoyl chloride [58] (Scheme 21).

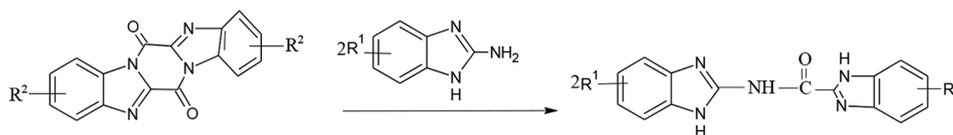
#### Sulphonylation reactions

Reaction of 2-aminobenzimidazoles with various sulphonyl chlorides yields, in general, 1-sulphonyl-2-aminobenzimidazoles, which, as an example when  $X=m-O_2$   $N---C_6H_4$ , afford, after the treatment with caustic soda, 2-aminobenzimidazole benzenesulfonate. Some 1-sulphonylamino-2-aminobenzimidazoles are acylated to yield the corresponding 1,2-disubstituted benzimidazoles possessing fungicidal activity. One such compound (13), obtained by treating 1-sulphonyl-2-aminobenzimidazole with methylmagnesium bromide, showed 100% *in vitro* inhibition of the expansion of polio virus at a amount of 1.5 µg/ml [59] (Scheme 22).

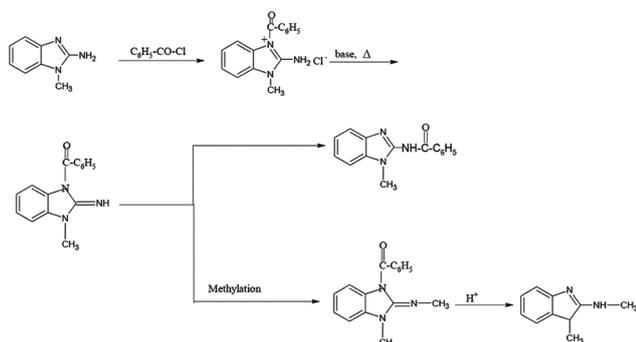
The preparation of 2-(4-acetylamino benzenesulphonyl)-aminobenzimidazole is achieved by treating 2-aminobenzimidazole with 4-acetaminobenzenesulphonyl chloride, while its deacetylated product (14) can be obtained by the reaction of o-phenylenediamine with p-aminobenzenesulphonylguanidine. The latter is obtained by reaction of p-aminobenzenesulfonamide with dicyanodiamide (Scheme 23).

#### Reaction with carbonyl compounds

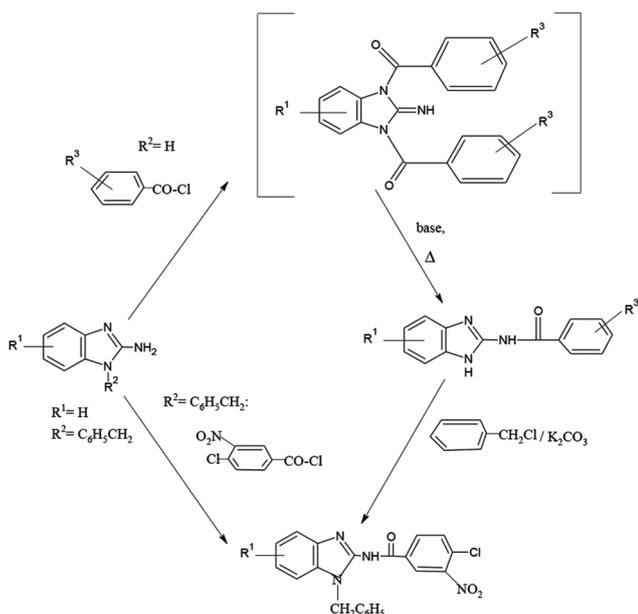
A number of aldehydes and ketones are known to react with 2-aminobenzimidazole to create the corresponding Schiff bases with a broad



Scheme 19: Aroylation reaction of 2-aminobenzimidazole



Scheme 20: Aroylation reaction of 2-aminobenzimidazole

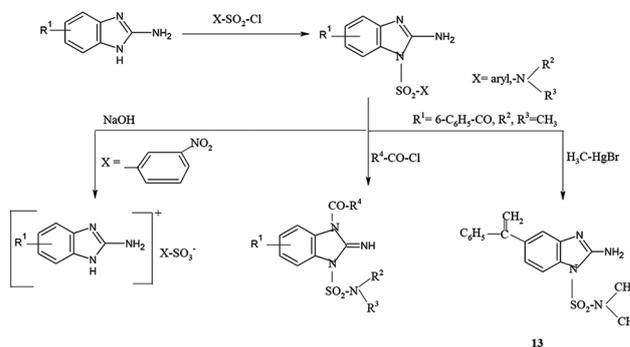


Scheme 21: Mechanism of aroylation reaction of 2-aminobenzimidazole

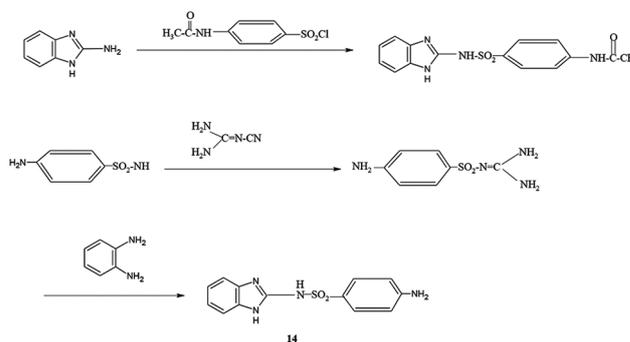
spectrum of biological activity. Thus, reaction of 2-aminobenzimidazole with 5-substituted furfural or 2-thiophenecarboxaldehyde yielded 2-[(5-substituted furfurylidene/thiophenylidene)-amino]benzimidazoles (azomethines) [60] (Scheme 24).

Treatment of various aliphatic anhydrides with 2-aminobenzimidazoles results in the formation of the monoamides which can be cyclized with anhydride to yield the corresponding imides (Scheme 25).

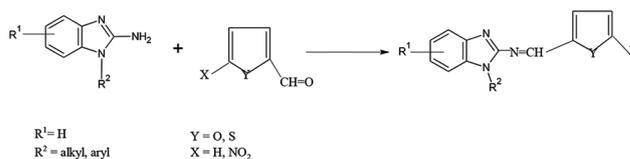
Condensation of 2-aminobenzimidazolyl-1-phenylimidate (15a) and 2-aminobenzimidazolyl-1-amidines (15b) with aromatic aldehydes or acids resulted within the formation of 2-aryl-4-substituted-s-triazino[1,2-a]benzimidazoles (17a) or (17b), respectively. The reaction of (15a) with an aromatic aldehyde first gives the azomethine (16) which cyclizes to create 2-aryl-4-phenoxy-1,2-dihydro-s-triazino[1,2-a]benzimidazole (18). Dehydrogenation of (18) with sulfur yields (17a) (Scheme 26).



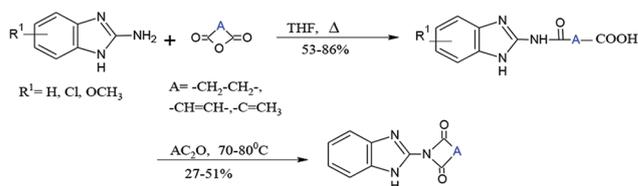
Scheme 22: Sulfonylation Reaction of 2-aminobenzimidazole



Scheme 23: Sulfonylation reaction of 2-aminobenzimidazole



Scheme 24: Reaction of 2-aminobenzimidazole with 2-substituted carboxyaldehyde

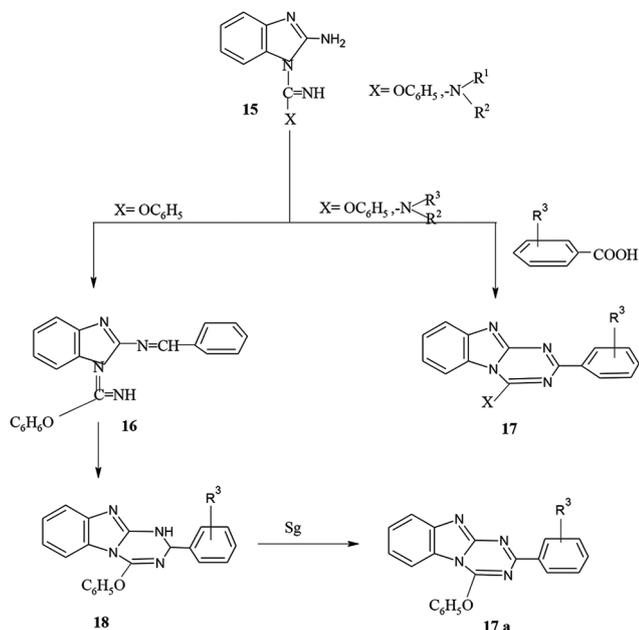


Scheme 25: Reaction of 2-aminobenzimidazole with aliphatic anhydrides

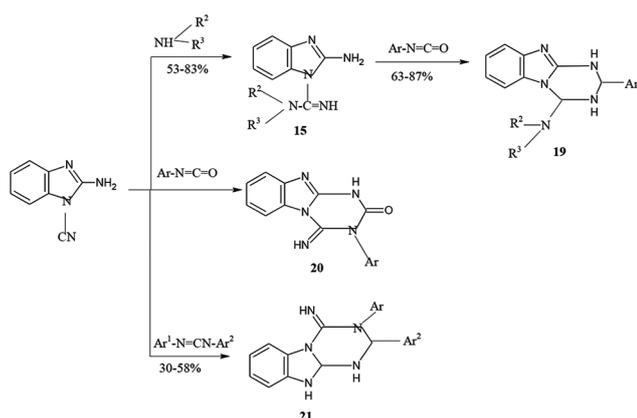
The preparation of (15a) is also accomplished by treating 2-aminobenzimidazole with phenyl cyanate while (15b) is obtained by reaction of 2-amino-1-cyanobenzimidazole with the required amines. Among these compounds, (15b) and 2-amino-1-cyanobenzimidazole react with aryl isocyanates to afford 2-aryl-4-amino-1,2-dihydro-s-triazino[1,2-a]benzimidazoles (19) and 2-oxo-3-aryl-4-imino-1,2,3,4-tetrahydro-s-triazino[1,2-a]benzimidazoles (20), respectively.

In addition, 2,3-diaryl-4-imino-1,2,3,4-tetrahydro-s-triazino[1,2-a]benzimidazole(21) could also be prepared by condensing 2-amino-1-cyanobenzimidazole with various azomethines [61] (Scheme 27).

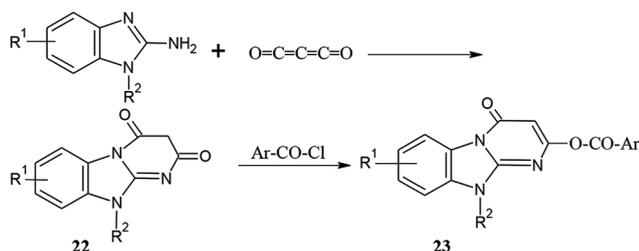
Carbon suboxide is thought to react with 2-aminobenzimidazole to afford 2,3-(dioxotetrahydropyrimidino)-benzimidazoles (22) in high yields, which can be aroylated to grant the corresponding O-aryl product (23) [62] (Scheme 28).



Scheme 26: Condensation reaction of 2-substituted aminobenzimidazole



Scheme 27: Reaction of 2-aminobenzimidazole with phenyl cyanate, aryl isocyanate, and azomethines



Scheme 28: Reaction of 2-aminobenzimidazole with carbon suboxide

Reaction of 2-aminobenzimidazole with 2-hydroxymethylene-3-androstanones (24 a-d) in absolute ethanol yielded androstano(or androst-4-eno or cholestano)[3,2-b]-(pyrimido[1,2-a]benzimidazoles) (25 a-d) in good yields [63] (Scheme 29).

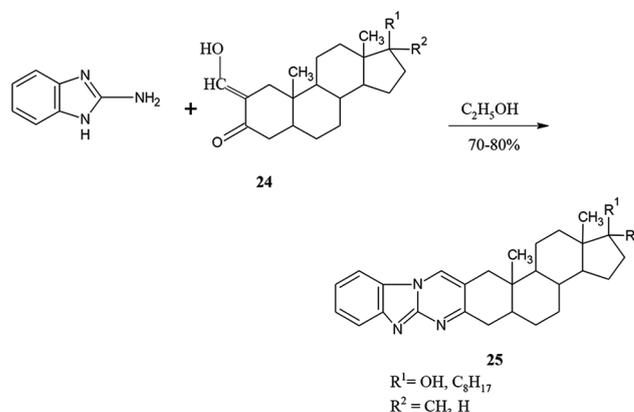
A series of 2,4-unsymmetrically dialkylated pyrimido[1,2-a]benzimidazoles (27) as possible antibacterial and antiarrhythmic agents has been prepared by treating 2-aminobenzimidazole with the  $\beta$ -diketones (26). Other heterocycles associated with (27) may additionally be prepared by treating 2-aminobenzimidazoles with  $\beta$ -dicarbonyl compounds [64] (Scheme 30).

Cyclocondensation of 2-aminobenzimidazole with diphenylcyclopropenone (28) gives diphenyl-2-oxo-1,2,3,4-tetrahydropyrimido[1,2-a]benzimidazole (29) [65] (Scheme 31).

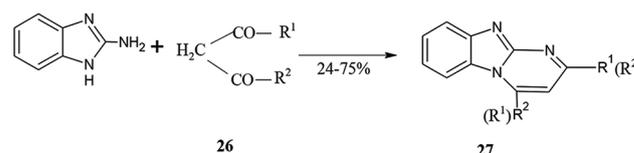
### Substitutions involving heterocyclization

Reaction of 2-aminobenzimidazole with ethoxycarbonylacetonitrile at 100°C resulted within the 2-N-acylation to make 2-(1-cyanoacetyl)-aminobenzimidazole. However, when the above reaction was disburbed at 140°C, it absolutely was impractical to isolate 2-(1-cyanoacetyl)-aminobenzimidazole; instead 4-imino-2-oxo-1,2,3,4-tetrahydropyrimido[1,2-a]benzimidazole was isolated from the reaction mixture. An analogous reaction occurred when 2-aminobenzimidazole was treated with ethyl acetoacetate, dimethyl acetylenedicarboxylate, or diethyl malonate resulting in the formation of the cyclocondensation products(30). The preparation of (30) can also be achieved by treating 2-aminobenzimidazole with a  $\beta$ -aminocrotonate (Scheme 32).

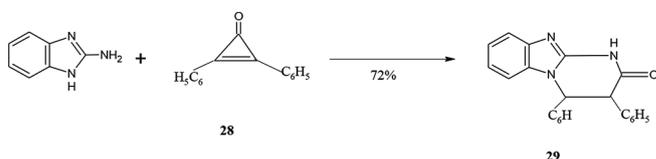
The above reaction has also been extended to synthesize a good form of heterocycles of a sort (31-33) by condensing ethyl



Scheme 29: Reaction of 2-aminobenzimidazole with 2-hydroxymethylene-3-androstanones



Scheme 30: Reaction of 2-aminobenzimidazole with  $\beta$ -diketones



Scheme 31: Cyclocondensation of 2-aminobenzimidazole with diphenylcyclopropenone

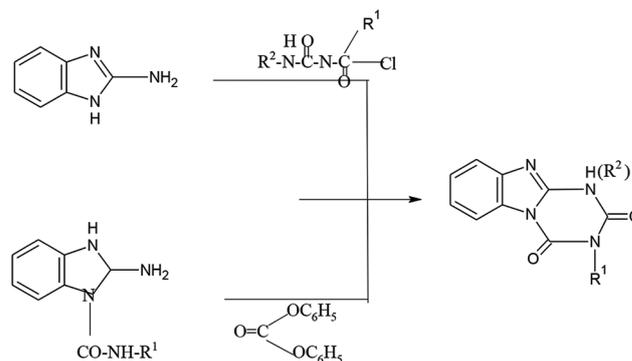
2-oxocyclohexanecarboxylate, ethyl 2-oxocyclopentanecarboxylate,  $\beta$ -aminocrotonitrile, or  $\beta$ -aminocinnaminonitrile, respectively, with 2-aminobenzimidazole [66] (Scheme 33).

A similar kind of cyclocondensation occurs when 2-aminobenzimidazole is treated with the  $\beta$ -aminoesters to convey the corresponding 2-amino-4-oxo-3,4-dihydropyrimido[1,2-a]benzimidazoles in 76-82% yields [67] (Scheme 34).

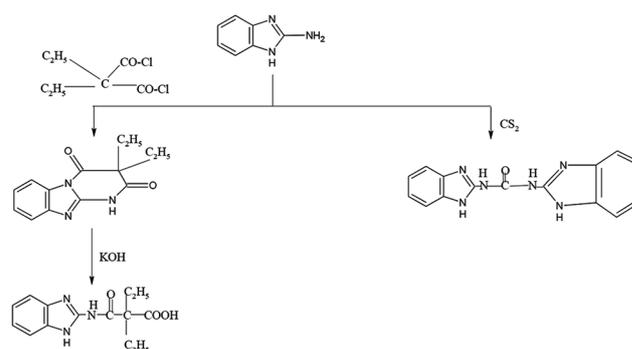
Treatment of the two-amino-1-aminocarbonylbenzimidazole with diphenyl carbonate affords the 2,4-dioxo-1,2,3,4-tetrahydro-s-triazino[1,2-a]benzimidazoles, which can even be prepared by condensing 2-aminobenzimidazole with the N-chlorocarbonyl urea (Scheme 35).

A number of activated acid chlorides react with 2-aminobenzimidazole to yield several benzimidazole-heterocycles of diverse biological interest. Thus, reaction of 2-aminobenzimidazole with diethylmalonic acid dichloride in anhydrous pyridine gave 3,3-diethyl-2,4-dioxo-1,2,3,4-tetrahydropyrimido[1,2-a]benzimidazole, which was converted into 3-[2-benzimidazolylaminocarbonyl]-2-ethyl-3-oxobutanoic acid by reaction with aqueous caustic potash. Dehydration of 3-[2-benzimidazolylaminocarbonyl]-2-ethyl-3-oxobutanoic acid with anhydride gives back 3,3-diethyl-2,4-dioxo-1,2,3,4-tetrahydropyrimido[1,2-a]benzimidazole. The actual fact that 2-aminobenzimidazole acts as a primary amine within the

above reaction and not as a phenyleneguanidine was evident by its reaction with chemical compound to yield N,N'-bis[2-benzimidazolyl] thiourea [68] (Scheme 36).

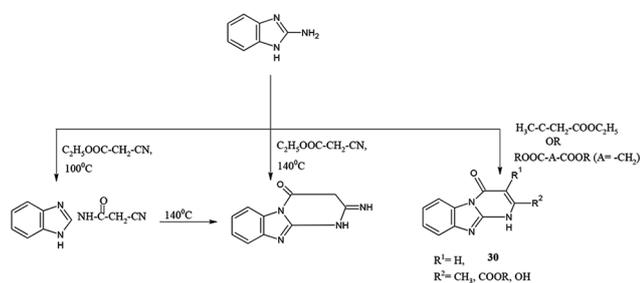


Scheme 35: Reaction of 2-aminobenzimidazole with diphenyl carbonate and N-chlorocarbonyl urea

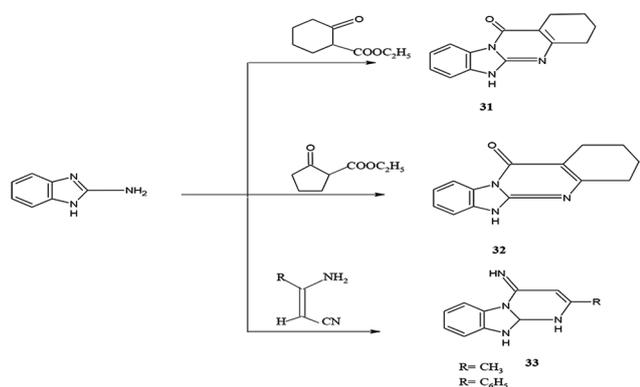


Scheme 36: Reaction of 2-aminobenzimidazole with diethylmalonic acid dichloride and carbon disulfide

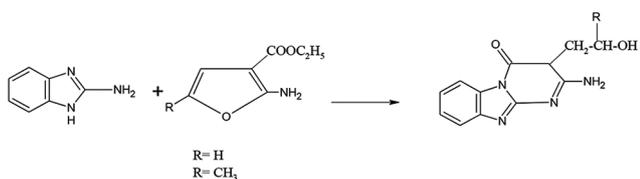
Scheme 32: Reaction of 2-aminobenzimidazole with ethoxycarbonylacetonitrile and  $\beta$ -aminocrotonate



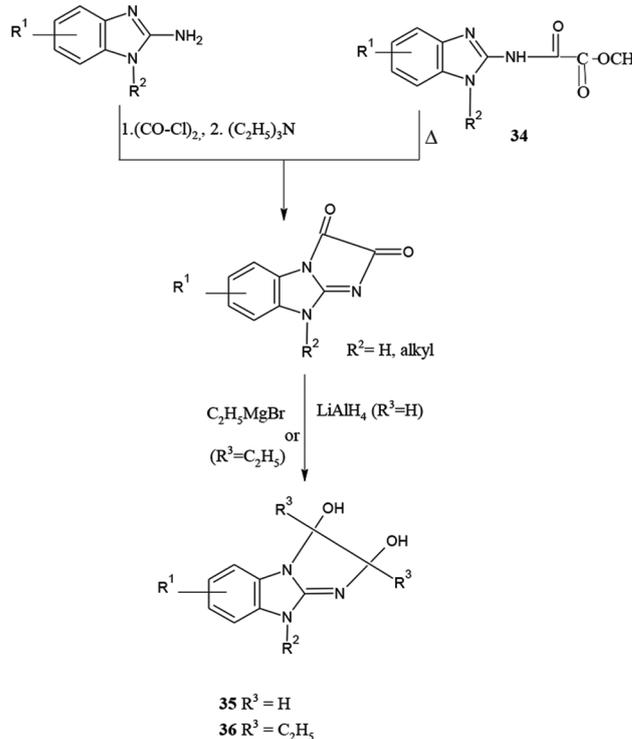
Scheme 32: Reaction of 2-aminobenzimidazole with ethoxycarbonylacetonitrile and  $\beta$ -aminocrotonate



Scheme 33: Condensation reaction of 2-aminobenzimidazole



Scheme 34: Cyclocondensation reaction of 2-aminobenzimidazole with  $\beta$ -aminoesters

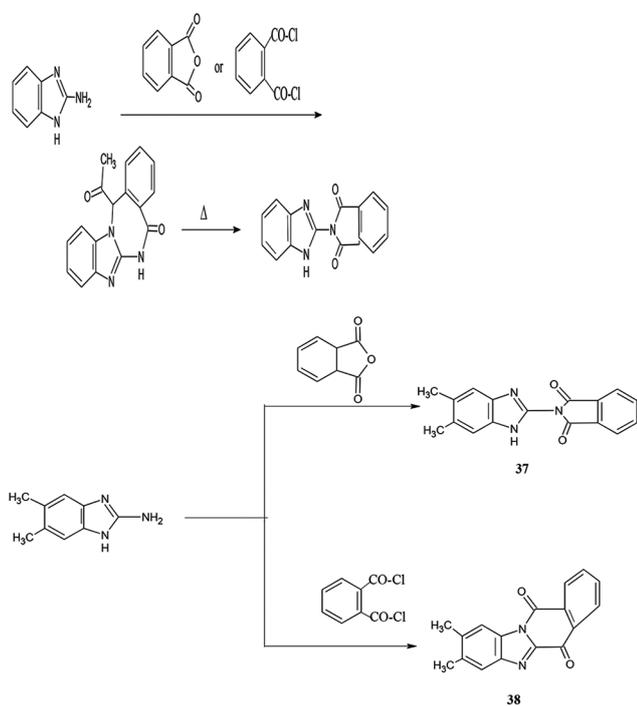


Scheme 37: Reaction of 2-aminobenzimidazole by the action of oxalyl chloride and triethylamine

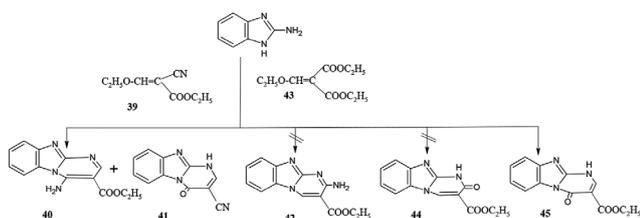
2,3-Dioxo-2,3-dihydro-9H-imidazo[1,2-a]benzimidazoles were prepared by the action of oxalyl chloride on the corresponding 2-aminobenzimidazole in absolute dioxane followed by thermal cyclization of the resulting intermediate in presence of triethylamine. Compound 2,3-dioxo-2,3-dihydro-9H-imidazo[1,2-a]benzimidazoles was also obtained in 20–70% yield by thermal cyclization of (34) [69] and further went to prepare (35,36) by action of lithium aluminum hydride and ethylmagnesium bromide, respectively (Scheme 37).

Katritzky and Yates have studied the phthaloylation reaction of 2-aminobenzimidazole and 2-amino-5,6-dimethylbenzimidazole with phthaloyl chloride and anhydride. Reaction of 2-aminobenzimidazole with the phthaloylating agents gave 6H-benzimidazol[1,2-b][2,4]benzodiazepin-7,12-dione which underwent thermal isomerization to afford 2-phthalamidobenzimidazole. However, reaction of 2-amino-5,6-dimethylbenzimidazole with phthaloyl chloride gave (38) while (37) was obtained by treating 2-amino-5,6-dimethylbenzimidazole with anhydride (Scheme 38).

Reaction of 2-aminobenzimidazole with diethyl ethoxy-methylenemalonate(43) and ethyl ethoxymethylene-cyanoacetate (39) failed to yield ethyl 2-oxo-1,2-dihydropyrimido[1,2-a]benzimidazole-3-carboxylate (44) and ethyl 2-aminopyrimido[1,2-a]benzimidazole-3-carboxylate(42); instead their isomeric esters(40) and (45) were obtained, respectively. Within the latter case, the nitrile (41) was also obtained [70] (Scheme 39).



Scheme 38: Phthaloylation reaction of 2-aminobenzimidazole



Scheme 39: Reaction of 2-aminobenzimidazole with diethyl ethoxy-methylenemalonate and ethyl ethoxymethylene-cyanoacetate

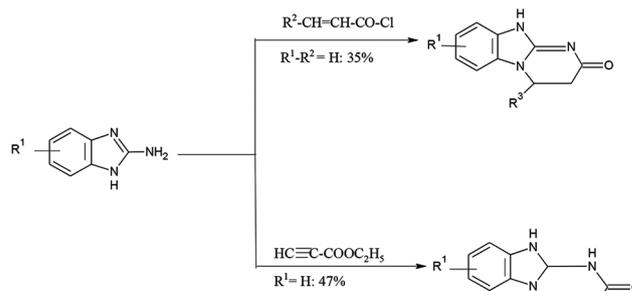
However, reaction of 2-aminobenzimidazole with  $\alpha,\beta$ -unsaturated acid chlorides or esters gave 2-oxo-3,4-dihydro-2H,10H-pyrimido[1,2-a]benzimidazoles. Similarly, reaction of 2-aminobenzimidazole with ethyl propynoate yielded 2-oxo-1,2-dihydropyrimido[1,2-a]benzimidazoles. Using 2-amino-1-methylbenzimidazoles, Trachenkoet [71] have prepared (46) by cyclization of (47) and (48), obtained by reaction of the previous with  $\alpha,\beta$ -unsaturated acid chlorides, by refluxing in ethanol (Scheme 40 and 41).

Reaction of 2-aminobenzimidazole with phenacyl bromide in acetone gives both mono- and diphenacyl derivatives (49) and (50) in 41 and 56% yields, respectively. They need been characterized by cyclizing them to corresponding imidazobenzimidazoles (51) and (52), respectively, in boiling hydrochloric acid. Compounds of the kind (51) have also been obtained by reaction of 1-alkyl-2-aminobenzimidazoles with anilides of chloroacetic acid followed by cyclization of the resulting product with phosphoryl chloride [72] (Scheme 42).

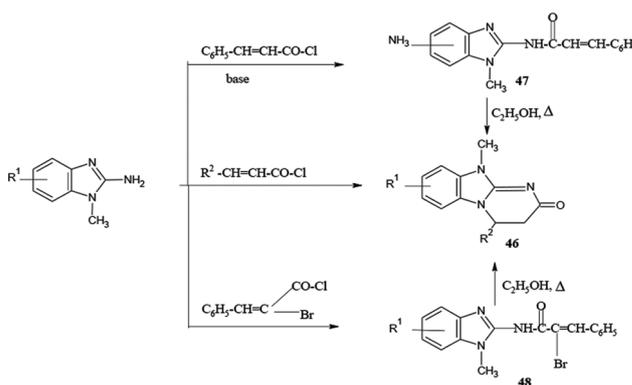
Alkylations and acylations [73] of 2-amiobenzimidazoles using the alkyl and acyl halides, possessing another reactive center, have yielded a series of benzimidazole heterocycles [53-56] (Scheme 43).

PHARMACOLOGICAL ACTIVITY

Benzimidazole a flexible heterocycle is possessing good spectrum of biological activity because of the incorporation of an imidazole nucleus which is a biologically accepted pharmacophore within the benzimidazole molecule. Benzimidazoles is having antiparasitic, anticancer, antiprotozoal, antihelminthic, anti-HIV, antiepileptic, anti-inflammatory, antiherpetic, anti-allergic, anti-histaminic, vasodilative, a narcotic analgesic, medication, antifungal, antiulcer, and pharmacological activities; one except the same activities from 2-aminobenzimidazole.



Scheme 40: Reaction of 2-aminobenzimidazole with  $\alpha,\beta$ -unsaturated carboxylic acid chlorides and esters



Scheme 41: Reaction of the 2-aminobenzimidazole with  $\alpha,\beta$ -unsaturated carboxylic acid chlorides



prepare yellow to orange dyes with bright and fast shades [80].

## CONCLUSION AND OUTLOOK

The present survey has clearly demonstrated that 2-aminobenzimidazole also successfully want to synthesize a good kind of benzimidazole heterocycles of educational and pharmaceutical interest. Moreover, in general, the required compounds are also obtained during a single step with high yields. Thus, the look of a molecule supported that the 2-aminobenzimidazole nucleus is also expected to guide to compounds with pharmacodynamic properties.

## REFERENCES

- Rastogi R, Sharma S. 2-Aminobenzimidazoles in organic syntheses. *Synthesis*. 1983;1983:861-82. doi:10.1055/s-1983-30546.
- 2-Amino Benzimidazole. Available from: <https://pubchem.ncbi.nlm.nih.gov/compound/2-Aminobenzimidazole>. [Last accessed on 2021 Jun 10].
- Stoermer M, Egan S, Forsyth CM, Moloney G. (Z)-N'-(1H-Benzo [d] imidazol-2-Yl)-Arylimidamide adducts of 2-aminobenzimidazole and aromatic nitriles: Structural and spectroscopic proof of regiochemical and stereochemical outcomes. *Eur J Org Chem* 2019;3:???
- Azam M, Khan AA, Al-Resayes SI, Islam MS, Saxena AK, Dwivedi S, *et al.* Synthesis and characterization of 2-substituted benzimidazoles and their evaluation as anticancer agent. *Spectrochim Acta A Mol Biomol Spectrosc* 2015;142:286-91. doi:10.1016/j.saa.2015.01.106. PMID 25706598.
- Pierron P. ??? *Ann Chim Phys* 1968;15:189.
- Offenlegungsschrift D 2214600, Mar 25; 1972.
- Stedman RJ, Inventor; SmithKline Corp, Assignee. 2-Aminobenzimidazole and A Process for Preparing 2-Aminobenzimidazoles. Vol. 3. United States Patent; 1969. p. 948.
- Kiffer D, Levy R. A New synthetic method for 2-alkyl-2-aryl- and 2-Acylaminobenzimidazoles. *CR Hebd Séances Acad Sci* 1968;267:1730.
- Omar AM, Amme O. Novel synthesis of 2-substituted aminobenzimidazoles from thiourea derivatives. The cyclodesulphurization of thio-compounds. *Pharmazie* 1972;27:798-9.
- Omar AM, Ragab MS, Farghaly AM, Barghash AM. The cyclodesulphurization of thio-compounds. Part 12: A new simple method for the synthesis of N-alpha-substituted benzimidazoles from thiourea derivatives. *Pharmazie* 1976;31:348-50. PMID 959294.
- Omar AM. The Cyclodesulfurization of thio-compounds; VII. A new facile synthesis of  $\alpha$ -substituted benzimidazoles. *Synthesis* 1974;1974:41-2. doi:10.1055/s-1974-23234.
- Omar AM, El-Dine SH, Hazzaa A. ??? *Pharmazie* 1973;28:682.
- Kiffer D. Cyclisation reaction of carbodiimides-study of conditions for synthesis of benzimidazoles. *Bull Soc Chim Fr* 1970;6:2377.
- Stedman RJ. US Patent No. 3,455,948; 1969.
- Griffin TS, Woods TS, Klayman DL. Thioureas in the synthesis of heterocycles. *Adv Heterocycl Chem* 1975;18:99-158. doi:10.1016/S0065-2725(08)60129-4.
- Murray JA, Dains FB. The action of amines on the esters of carboxy substituted ureas, thioureas and guanidines. III. *J Am Chem Soc* 1934;56:144-6. doi:10.1021/ja01316a044
- Olin JF, Dains FB. The action of the halogen hydrins and of ethylene oxide on the thioureas. *J Am Chem Soc* 1930;52:3322-7. doi:10.1021/ja01371a047.
- Klopping HL. Chem Abstract. United States Patent, 2,933,502.
- Klopping HL. Chemical Abstract. United States Patent 2,933,504.
- Averkin EA, Beard CC, Dvorak CA, Edwards JA, Fried JH, Kilian JG, *et al.* Methyl 5 (6). *J Med Chem* 1975;18:1164-6. doi:10.1021/jm00245a029, PMID 1177265.
- McKellar QA, Scott EW. The benzimidazole anthelmintic agents-a review. *J Vet Pharmacol Ther* 1990;13:223-47. doi:10.1111/j.1365-2885.1990.tb00773.x. PMID 2231864.
- Di HCl, HCl Di. Benzimidazoles- Table 1.89. Bisbenzimidazoles Compound Mp (C) Ref. Benzimidazoles and Cogeneric Tricyclic Compounds (1); 2009. p. 244.
- Alaqeel SI. Synthetic approaches to benzimidazoles from o-phenylenediamine: A literature review. *J Saudi Chem Soc* 2017;21:229-37. doi:10.1016/j.jscs.2016.08.001
- Adcock B, Lawson A, Miles DH. 2-Amino-2-imidazolines and 2-amino-2-oxazolines. *J Chem Soc* 1961;1014:5120-7.
- Weiss S, Michaud H, Prielzel H, Krommer H. Neue, einfache synthese von 2-aminobenzimidazol. *Angew Chem* 1973;85:866-7. doi:10.1002/ange.19730851910.
- Sawatari K, Mukai T, Suenobu K, Kamenozono S, Ika T. Chem Abstract Japanese Patent No. 76.16669; 1976. p. 63069.
- Simonov AM, Anisimova VA. Synthesis and transformation of 2-aminobenzimidazoles (review). *Chem Heterocycl Compd* 1979;15:705-23. doi:10.1007/BF00473548.
- Adams CD, Schlatter R. Chem [Abstract], 7.0057837. Republic of South Africa Patent No. 6706.589; 1968. p. 57837.
- Adams CD, Schlatter R. Chem Abstr. 7.0012629; Republic of South Africa Patent No. 6906.326; 1970.
- Craig PN, Hoover JR. InChem Abstract. British Patent No. 1111957. 1966. p. 1968-9.
- Harsanyi K, Toth G, Simay A, Gonczi C, Takacs K. Hungarian Patent No. 5800.
- Murphy DB, Picard JP. A Study of tautomerism in the 5-aminotetrazoles I. *J Org Chem* 1954;19:1807-14. doi:10.1021/jo01376a015
- Sheverdina NI, Kocheshkov KA. Synthesis of primary amines by the reaction of-methylhydroxylamine with organomagnesium and organolithium compounds. *Zh Obshch Khim* 1938;8:1825-30.
- Watkins DA. Benzimidazole pesticides: Analysis and transformations. *Pestic Sci* 1976;7:184-92. doi:10.1002/ps.2780070213.
- Depost G, sillion B, salle R. Formation of 2-benzamidobenzimidazole, Study on its Thermal-stability. *Comptes Rendus Hebdomadaires Seances Académie Des Sci Serie c*; 1972. p. 697.
- Ogura H, Kikuchi K. Heterocyclic compounds. XI. 1,3-Dipolar cycloaddition of benzimidazolium ylide with acetylenic compounds. *J Org Chem* 1972;37:2679-82.
- Ogura H, Takayanagi H, Yamazaki Y, Yonezawa S, Takagi H, Kobayashi S, *et al.* Heterocyclic compounds. 10. Synthesis of some imidazo [1, 2-a] benzimidazoles with potent analgetic activities. *J Med Chem* 1972;15:923-6.
- Lugosi P, Béla ÁG, Hornyák G. Synthesis and ring closure reactions of some N-(o-aminophenyl)-N-methyl-N', N''-Disubstituted-guanidines and of N-(2-amino-4-methoxyphenyl)-N-methyl-N'-phenylthiourea. *Period Polytech Chem Eng* 1975;19:307-16.
- Held P, Gross M. 2-Dialkylamino-benzimidazole aus O-acylierten N-Phenyl-2-hydroxy-3, 3-dialkylguanidinen. *Z Chem* 1973;13:292-3. doi:10.1002/zfch.19730130809.
- Kirmse W. Reaktionen mit carbenen und iminen als zwischenstufen zur photochemischen und thermischen umwandlung organischer stickstoff-verbindungen. *Angew Chem* 1959;71:537-41. doi:10.1002/ange.19590711702.
- Partridge MW, Turner HA. 430. Cyclic amidines. Part VII. Preparation of benzimidazoles from N'-aryl-N-hydroxyamidines. *J Chem Soc* 1958;430:2086-92. doi:10.1039/JR9580002086.
- Kalidhar U, Kaur A. An overview on some benzimidazole and sulfonamide derivatives with anti-microbial activity. *J Pharm Biol Chem Sci* 2011;2:1116-35.
- Roche Y, Vigier J, Guyod JL, Boucherle A. *Bull Trav Soc Pharm (Lyon)* 1979;23, 27 CA; 1981. p. 42082.
- Yutilov YM. Materially 4-oj[Chetvertov] Nauchn. [Konf. Aspirantov]. Rostov-on-Don: Rostovsk. Univ.] Sb. 1962, 110; CA; 1964. p. 9262.
- Martin D, Graubaus H, Esters C. Triazino-benzimidazoles from 2-amino-benzimidazole, aryl cyanates and carbonyl compounds. *J Prakt Chem* 1979;321:315-9.
- Martin D, Graubaus H. German Patent (DDRP). C.A. 1980, 92; 1979. p. 136499.
- Lalezari I, Nabahi S. Reaction of N-cyanoformimidates with some heterocyclic compounds. A new synthesis of 5-azaadenine and related compounds. *J Heterocycl Chem* 1980;17:1121-3. doi:10.1002/jhet.5570170559.
- Kikugawa Y. A facile N-alkylation of imidazoles and benzimidazoles. *Synthesis*. 1981;1981:124-5. doi:10.1055/s-1981-29356.
- Murphy DB, Musco J. Methylation of 2-aminobenzimidazole. *J Org Chem* 1971;36:3469-70. doi:10.1021/jo00821a047.
- Van Gelder JL, Raeymaekers AH, Roevens LF. German Patent. 2029637. 1971, CA; 1971. p. 100047.
- Klopping HL. US CA Patent 2933504; 1960. p. 55.
- Mayer KH, Lauer DH. *Synthesis* 1975:673.
- Craig PN, Hoover JR. CA British Patent, No. 1111957; 1968. p. 96721.
- Daun W, Scheinpflug H, Hamburger B. German Patent. 2140863, CA; 1973. p.124589.
- Beard CC, US. CA Patent 3976654; 1966. p. 1977.
- Rastogi R, Sharma S, Iyer RN. Synthesis of benzimidazole-2-carboxamides as potential anthelmintic agents. *Indian J Chem* 1979;18:464-7.

57. Khristich BI, Simonov AM, Suvorova GM. *Khim. Geterotsikl Soedin* 1293; CA; 1973. p. 146456.
58. Rastogi R, Sharma S, Iyer RN. *???*. *Eur J Med Chem* 1979;14:489.
59. Paget CJ, Chamberlin JW, Wikel JH. CA Belgian Patent 851630; 1978. p. 170148.
60. Kuzmierkiewicz W, Tyczynska B. Schiff-bases of 2-aminobenzimidazole. *Acta Pol Pharm* 1980;37:39-42.
61. Augustin M, Kuppe KR. Cyclisierungsreaktionen von 2-aminobenzimidazolen zu s-triazino [1, 2-a] benzimidazolen. *Tetrahedron* 1974;30:3533-8. doi:10.1016/S0040-4020(01)97537-4.
62. Dashkevich LB, Korbelainen ES. *Khim. Geterotsikl Soedin*, No. 37836 CA; 1967. p. 37836.
63. Bajwa JS, Sykes PJ. New steroidal heterocycles: Androstano [3, 2-b](pyrimido [1, 2-a] benzimidazoles). *J Chem Soc Perkin Trans* 1980;1:1859-61.
64. Basaglia L, Mariani B. *???*. *Ann Chim (Rome)* 1963;53:755.
65. Eicher T, Franke G. Zur reaktion von diphenylcyclopropenon mit guanidinen. *Liebigs Ann Chem* 1981;1981:1337-53. doi:10.1002/jlac.198119810802.
66. Antaki H, Petrow V. 119. New syntheses of heterocyclic compounds. The condensation of ethyl  $\beta$ -aminocrotonate with some cyclic amidines-Part XII. *J Chem Soc* 1951:551-5.
67. Elnagdi MH, Wamhoff H. A novel synthesis of azolopyrimidines. *Chem Lett* 1981;10:419-22. doi:10.1246/cl.1981.419.
68. Crippa GB, Perroncito G, Sacchetti G. *Gazz Chim Ital*; CA; 1935. p. 4007.
69. Simonov AM, Koshchienko YV, Suvorova GM, Tertov BA, Malysheva EN. *Khim. Geterotsikl Soedin*; 1976. p. 72521.
70. Chow AW, Jakas DR, Trotter BP, Hall NM, Hoover JR. Reaction of 2-aminobenzimidazole with bifunctional carboxylic acid derivatives. Formation of pyrimido [1, 2-a] benzimidazolones. *J Heterocycl Chem* 1973;10:71-5. doi:10.1002/jhet.5570100116.
71. Trachenko PV, Simonov AM, Popov II. *Khim. Geterotsikl Soedin*; 1978. p. 190727.
72. Simonov AM, Borisova TA. U.S.S.R. Patent, 478007. CA; 1976. p. 17339.
73. Schulze W, Letsch G. Potential nitrogen lost transport forms. Bis-(beta-chlorethyl)-carbonyl derivatives of benzimidazolones, benzimidazolthiones and benzimidazolnimides. *Pharmazie* 1973;28:367-71.
74. Krichovskaya AA, Mendzneritskaya LG, Butko VV. *Oksibioticheskie Anoksibioticheskie protsessy Eksp. Klin Patol* 1975;115:87321.
75. Dransch G, Hoerlein G. German Patent. (Dos) 2250469. CA; 1974. p. 13512.
76. Tunçbilek M, Kiper T, Altanlar N. Synthesis and *in vitro* antimicrobial activity of some novel substituted benzimidazole derivatives having potent activity against MRSA. *Eur J Med Chem* 2009;44:1024-33. doi:10.1016/j.ejmech.2008.06.026, PMID 18718694.
77. Simonov AM, Po O, Azolov K. (Rostov-On Don State University, U.S.S.R.); 1965.
78. Olbe LE, Carlsson E, Lindberg P. A proton-pump inhibitor expedition: The case histories of omeprazole and esomeprazole. *Nat Rev Drug Discov* 2003;2:132-9. doi:10.1038/nrd1010, PMID 12563304.
79. Bywater WG, McGinty DA, Jenesel ND. Antithyroid studies: II. The goitrogenic activity of some imidazoles and benzimidazoles. *J Pharmacol* 1945;85:14-22.
80. Vonder CJ, Frey C. Swiss Patent 613465. CA; 1980. p. 43279.