A STABILITY-INDICATING AND VALIDATED REVERSE-PHASE HIGH-PERFORMANCE LIQUID CHROMATOGRAPHIC METHOD FOR THE SIMULTANEOUS ESTIMATION OF PHENYLEPHRINE AND FEXOFENADINE IN BULK AND TABLET DOSAGE FORMS

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INTRODUCTION

Phenylephrine (ACB) is chemically named as 3-[(1R)-1-hydroxy-2-(methyl amino)ethyl] phenol (Fig. 1). It is used as a sympathomimetic amine that acts predominantly on α-adrenergic receptors. It is mainly used to treat nasal congestion, but may also be useful in treating hypotension and shock, hypotension during spinal anesthesia, prolongation of spinal anesthesia, paroxysmal supraventricular tachycardia, symptomatic relief of external or internal hemorrhoids, and to increase blood pressure as an aid in the diagnosis of heart murmurs [1-3].

Fexofenadine is chemically named as 2-(4-(1-hydroxy-4-[4(hydroxydiphenylmethyl) piperidin-1-yl] butyl)phenyl)-2-methylpropanoic acid (Fig. 2). Fexofenadine, like other second- and third-generation antihistamines, does not readily pass through the blood-brain barrier, and so causes less drowsiness than first-generation histamine receptor antagonists.

Methods: Chromatography was carried out on a Kromasil C-18 column (4.6 mm×250 mm, 5 μm particle size) with a isocratic mobile phase composed of ortho phosphoric acid buffer: acetonitrile, (75:25%) v/v) at a flow rate of 1 mL/minutes. The column temperature was maintained at 30°C and the detection was carried out using a photodiode array detector at 210 nm.

Results: The retention times for phenylephrine and fexofenadine were 2.156 minutes and 3.359 minutes, respectively. The percentage recoveries of phenylephrine and fexofenadine were 100.63% and 99.84%, respectively. The relative standard deviation for assay of tablets was found to be <2%. The detection and quantification limits were found to be 0.10 and 0.31 µg/mL for phenylephrine and 0.01 and 0.03 µg/mL for fexofenadine, respectively.

Conclusion: Thus, the method was fast, accurate, precise, and sensitive; hence, it can be employed for routine quality control of tablets containing both drugs in quality control laboratories and pharmaceutical industries.

METHODS

Chromato phyic systems: An HPLC system consisting of a pump (Waters 600E), a column oven (Waters 510), a photo diode array detector (Waters 474), and an integrator (Waters 746) was employed in this method. The Empower software was used for LC peak integration along with data acquisition and data processing. The column used for separation of analytes was Kromasil C18 (250 mm×4.6 mm, 5 μm). Mobile phase consisting of orthophosphoric acid buffer:acetonitrile in the ratio of 75:25% v/v at a flow rate of 1.0 mL/minutes. It was filtered through 0.45 μm nylon filter and sonicated for 5 minutes in ultrasonic bath. Samples were analyzed at 210 nm at an injection volume of 10 μL.

Preparation of solutions

Preparation of phosphate buffer pH 4.8
One milliliter of orthophosphoric acid was taken in a 1000 mL volumetric flask, add about 900 mL of milli-Q water, degas to sonicate, and finally make up the volume with water.
Phenylephrine stock preparation (100 µg/mL)
Accurately weighed and transferred 2.5 mg of phenylephrine into 25 mL clean dry volumetric flask, add 17 mL of diluent (water:methanol - 50:50), then sonicate for 10 minutes, and make up the volume with diluent.

Fexofenadine stock preparation (1200 µg/mL)
Accurately weighed 30 mg of fexofenadine and transferred into 25 mL clean dry volumetric flask, add 17 mL of diluent (water:methanol - 50:50), then sonicate for 10 minutes, and make up the final volume with diluent.

Standard preparations
Phenylephrine standard preparation (10 µg/mL)
From the above phenylephrine stock solution, 1 mL was pipetted out into 10 mL clean dry volumetric flask and make up the final volume with diluent.

Fexofenadine standard preparation (120 µg/mL)
From the above fexofenadine stock solution, 1 mL was pipetted out into a 10 mL clean dry volumetric flask and make up the final volume with diluent.

Sample preparation
One tablet weighed and transferred into 25 mL volumetric flask, 10 mL diluent added and sonicated for 5 minutes. Further the volume made up with diluent and filtered. Further diluted 1 mL of filtrate to 10 mL with diluent and mixed.

Label claim
Phenylephrine 2.5 mg and fexofenadine 30 mg.

Chromatographic condition
Samples were analyzed at 210 nm at an injection volume of 10 µl and separation was carried out using Kromasil C18 (250 mm×4.6 mm, 5 µm) column and orthophosphoric acid buffer:acetonitrile in the ratio of 75:25% v/v as mobile phase at a flow rate 1.0 mL/minutes. The proposed method was optimized to give a sharp peak with minimum tailing for phenylephrine and fexofenadine (Fig. 3). The optimized conditions are given in Table 1.

Method development
To establish and validate an efficient method for analysis of these drugs in pharmaceutical formulations, preliminary tests were performed. Different chromatographic conditions were employed for the analysis of the phenylephrine and fexofenadine in both bulk and pharmaceutical dosage forms. The pure drugs of phenylephrine and fexofenadine were injected into the HPLC system and run using standard organic solvents commonly used for HPLC studies. Water and methanol were tested separately and commonly used for HPLC studies.

In focus to develop good symmetrical peak, water was replaced by phosphate buffer and it was observed that phosphate and acetonitrile gave satisfactory results. This mobile phase system was tried with different proportions and with different flow rates. Finally, the optimal condition of the mobile phase was chose as phosphate buffer:acetonitrile in the ratio of 75:25% v/v. This composition of the mobile phase resolved the two drugs very well. All measurements were carried at ambient temperature of the column. To optimize the flow rate, various flow rates were used. The optimal flow rate was 1 mL/minute for the presented work.

Method validation
The validation of the method was carried out as per ICH guidelines [13]. The parameters assessed were specificity, linearity, precision, accuracy, stability, limit of detection (LOD), and limit of quantitation (LOQ).

Specificity
Specificity is the ability of the analytical method to measure the analyte response in the presence of interferences including degradation products and related substances.
The accuracy was determined by calculating the percentage recoveries of phenylephrine and fexofenadine. It was carried out by adding known amounts of each analyte corresponding to three concentration levels (50%, 100%, and 150%) of the labeled claim to the excipients. At each level, six determinations were performed and the accuracy results were expressed as percent analyte recovered by the proposed method.

**Precision**

Precision of an analytical method is usually expressed as the standard deviation. The repeatability studies were carried out by estimating the response of phenylephrine and fexofenadine six times. The intraday and interday precision studies (intermediate precision) were carried out by estimating the corresponding responses three times on the same day and on three different days for three different concentrations and the results are reported in terms of relative standard deviation (RSD).

**Linearity**

The purpose of the test for linearity is to demonstrate that the entire analytical system (including detector and data acquisition) exhibits a linear response and is directly proportional over the relevant concentration range for the target concentration of the analyte. The linear regression data for the calibration plot are indicative of a good linear relationship between peak area and concentration over a wide range. The correlation coefficient was indicative of high significance.

**Robustness**

Robustness of the method was investigated under a variety of conditions including changes of composition of buffer in the mobile phase, flow rate, and temperature. This deliberate change in the method has no effect on the peak tailing, peak area, and theoretical plates, and finally the method was found to be robust.

**LOD and LOQ**

The LOD can be defined as the smallest level of analyte that gives a measurable response and LOQ was determined as the lowest amount of analyte that was reproducibly quantified. These two parameters were calculated using the formula based on the standard deviation of the response and the slope. LOD and LOQ were calculated using equations:

- \[ \text{LOD} = 3.3 \times \frac{s}{S} \]
- \[ \text{LOQ} = 10 \times \frac{s}{S}. \]

Where \( s \) = Standard deviation, \( S \) = Slope of the calibration curve.

**Assay of phenylephrine and fexofenadine in tablet**

Assay of marketed product was carried out using the developed method. Sample solutions were prepared and injected into RP-HPLC system. The sample solution was scanned at 210 nm. The % drug estimated was found to be 99.86 for phenylephrine and 99.77 for fexofenadine. The chromatogram showed two single peaks of phenylephrine and fexofenadine was observed with retention times of 2.137 and 3.43 minutes (Fig. 4).

**Forced degradation studies**

Stress studies are performed according to ICH guidelines under conditions of hydrolysis (acidic and alkaline), photolysis, oxidation, and thermal studies.

**Oxidation**

To 1 mL of stock solution of phenylephrine and fexofenadine, 1 mL of 20% hydrogen peroxide (\( \text{H}_2\text{O}_2 \)) was added separately. The solutions were kept for 30 minutes at 60°C. For HPLC study, the resultant solution was diluted to obtain 850 and 5 µg/mL solutions and 10 µl solutions were injected into the system and the chromatograms were recorded to assess the stability of sample.

**Acid degradation studies**

To 1 mL of stock solution phenylephrine and fexofenadine, 1 mL of 2 N hydrochloric acid was added and refluxed for 30 minutes at 60°C. The resultant solution was diluted to obtain 850 and 5 µg/mL solutions and 10 µl solutions were injected into the system and the chromatograms were recorded to assess the stability of sample.

**Alkali degradation studies**

To 1 mL of stock solution phenylephrine and fexofenadine, 1 mL of 2 N sodium hydroxide was added and refluxed for 30 minutes at 60°C. The resultant solution was diluted to obtain 850 and 5 µg/mL solutions and 10 µl solutions were injected into the system and the chromatograms were recorded to assess the stability of sample.

**Dry heat degradation studies**

The standard drug solution was placed in oven at 105°C for 6 hrs to study dry heat degradation. For HPLC study, the resultant solution was diluted to 850 and 5 µg/mL solutions and 10 µl solutions were injected into the system and the chromatograms were recorded to assess the stability of the sample.

**Table 1: Optimized chromatographic conditions**

<table>
<thead>
<tr>
<th>Parameter</th>
<th>Condition</th>
</tr>
</thead>
<tbody>
<tr>
<td>Mobile phase</td>
<td>Orthophosphoric acid buffer:acetonitrile (75:25% v/v) pH adjusted to 4.8</td>
</tr>
<tr>
<td>Column</td>
<td>Kromasil 250 mm×4.6 mm, 5 µ</td>
</tr>
<tr>
<td>Wavelength</td>
<td>210 nm</td>
</tr>
<tr>
<td>Flow rate</td>
<td>1.0 mL/minutes</td>
</tr>
<tr>
<td>Injection</td>
<td>10 µl</td>
</tr>
<tr>
<td>volume</td>
<td></td>
</tr>
<tr>
<td>Runtime</td>
<td>6 minutes</td>
</tr>
<tr>
<td>Diluent</td>
<td>Water:acetonitrile (50:50)</td>
</tr>
</tbody>
</table>

Fig. 4: A typical chromatogram of phenylephrine and fexofenadine in tablet dosage form.
Photo stability studies

The photochemical stability of the drug was also studied by exposing the 850 and 5 µg/mL solutions to UV Light by keeping the beaker in UV chamber for 7 days or 200 watt hrs/m² in photo stability chamber. For HPLC study, the resultant solution was diluted to obtain 850 and 5 µg/mL solutions and 10 µl solutions were injected into the system and the chromatograms were recorded to assess the stability of sample.

RESULTS AND DISCUSSIONS

To establish and validate an efficient method for analysis of these drugs in pharmaceutical formulations, preliminary tests were performed. Different chromatographic conditions were employed for the analysis of the phenylephrine and fexofenadine in both bulk and pharmaceutical dosage forms.

Table 2: Results of forced degradation studies

<table>
<thead>
<tr>
<th>Serial number</th>
<th>Injection</th>
<th>Phenylephrine %</th>
<th>Fexofenadine %</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td></td>
<td>Assay Degradation</td>
<td>Assay Degradation</td>
</tr>
<tr>
<td>1</td>
<td>Acid</td>
<td>96.37 3.63</td>
<td>95.81 4.19</td>
</tr>
<tr>
<td>2</td>
<td>Base</td>
<td>96.88 3.12</td>
<td>96.22 3.78</td>
</tr>
<tr>
<td>3</td>
<td>Peroxide</td>
<td>98.90 1.10</td>
<td>97.91 2.09</td>
</tr>
<tr>
<td>4</td>
<td>Thermal</td>
<td>99.60 0.40</td>
<td>99.16 0.94</td>
</tr>
<tr>
<td>5</td>
<td>UV degradation</td>
<td>99.41 0.59</td>
<td>99.42 0.58</td>
</tr>
<tr>
<td>6</td>
<td>Neutral degradation</td>
<td>99.66 0.34</td>
<td>99.93 0.07</td>
</tr>
</tbody>
</table>

Table 3: Precision method of proposed RP-HPLC method

<table>
<thead>
<tr>
<th>Drug</th>
<th>Mean area</th>
<th>RSD %</th>
</tr>
</thead>
<tbody>
<tr>
<td>Phenylephrine</td>
<td>486451</td>
<td>0.4</td>
</tr>
<tr>
<td>Fexofenadine</td>
<td>2974450</td>
<td>0.6</td>
</tr>
</tbody>
</table>

Table 4: Percentage recovery data for phenylephrine and fexofenadine

<table>
<thead>
<tr>
<th>Drug</th>
<th>Spiked level %</th>
<th>Recovery %</th>
<th>RSD %</th>
</tr>
</thead>
<tbody>
<tr>
<td>Phenylephrine</td>
<td>50</td>
<td>99.96</td>
<td>0.7</td>
</tr>
<tr>
<td></td>
<td>100</td>
<td>100.36</td>
<td>1.2</td>
</tr>
<tr>
<td></td>
<td>150</td>
<td>101.17</td>
<td>0.7</td>
</tr>
<tr>
<td>Fexofenadine</td>
<td>50</td>
<td>100.06</td>
<td>0.7</td>
</tr>
<tr>
<td></td>
<td>100</td>
<td>100.34</td>
<td>1.1</td>
</tr>
<tr>
<td></td>
<td>150</td>
<td>99.77</td>
<td>0.6</td>
</tr>
</tbody>
</table>

Linearity was established by analyzing different concentrations of phenylephrine and fexofenadine respectively. The calibration curve was plotted with the area obtained versus concentration of both compounds.
phenylephrine and fexofenadine (Figs. 5 and 6). In the present study, six concentrations were chosen ranging between 2.5 and 15 µg/mL of phenylephrine and 30-180 µg/mL of fexofenadine.

The regression equation and correlation coefficient for phenylephrine and fexofenadine were found to be $y=47494x+2190$. And $R^2=0.9990$, respectively, and results are given in Table 5.

Robustness of the method is the ability of the method to remain unaffected by small deliberate changes in parameters such as flow rate, mobile phase composition and column temperature. To study the effect of flow rate of the mobile phase, it was changed to 0.1 units from 1.0 to 0.9 mL and 1.1 mL. The effect of column temperature was also checked by changing the temperature to ±5°C. This deliberate change in the above parameters has no significant effect on chromatographic behavior of the samples and results are given in Table 6.

LOD and LOQ of phenylephrine and fexofenadine were evaluated based on RSD of the response and slope of the calibration curve. The detection limits were found to be 0.01 and 0.08 µg/mL for phenylephrine and fexofenadine, respectively. The quantitation limits were found to be 0.08 and 0.24 µg/mL for phenylephrine and fexofenadine, respectively. The results are given in Table 7.

CONCLUSION

A new stability-indicating RP-HPLC method has been developed for the estimation of phenylephrine and fexofenadine in bulk and pharmaceutical dosage forms. The forced degradation studies were carried out in accordance with ICH guidelines and the results revealed suitability of the method to study stability of phenylephrine and fexofenadine under various degradation conditions such as acid, base, oxidative, thermal, UV, and photo lytic degradations. Finally, it was concluded that the method is simple, sensitive, and has the ability to separate the drug from degradation products and excipients found in the dosage form.

REFERENCES