FORMULATION AND EVALUATION OF FAMOTIDINE MICRO BALLOONS WITH ENHANCED ANTI-ULCER ACTIVITY

RITESH KUMAR1, PAWAN KUMAR GAUTAM2, AMRISH CHANDRA3*

1Department of Pharmacy, IFTM University, Moradabad, 244102, Uttar Pradesh, India, 2Department of Pharmacy, S. N. Medical College, Agra, 282002, Uttar Pradesh, India, 3Amity Institute of Pharmacy, Amity University, Noida, 201313, Uttar Pradesh, India

Email: chandra.amrish@gmail.com

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ABSTRACT

Objective: The aim of the present study was to formulate and optimize famotidine loaded micro balloons for enhancing bioavailability, increasing gastric residence time of drug and to achieve sustained release in the stomach.

Methods: Microballoons were prepared using emulsion solvent diffusion method using HPMC K4M as the polymer. All the formulated microspheres were subjected to various evaluation parameters such as % drug entrapment, micromeritics properties, % buoyancy and in vitro drug release studies. The formulation was optimized using 3² full factorial design. Optimized formulation was subjective to in vivo floating (X-ray) and in vivo antiulcer studies.

Results: The microballoons were smooth and spherical in shape and were porous in nature due to hollow cavity. Sustained/controlled release of drug was observed for more than 12 h. based on the results of % drug entrapment, in vitro drug release and % buoyancy studies, formulation F6 was selected as optimized formulation. The release kinetics of optimized formulation followed Higuchi model and mechanism of release was non-Fickian diffusion. Examination of the X-ray radiographic images taken during the study indicated that the optimized formulation remained buoyant and uniformly distributed in the gastric contents for a long period. In ethanol-induced ulcer model, drug-loaded microballoons treated group showed significant ulcer protection index of 83.26% as compared to the marketed brand of famotidine 76.09% and untreated control group.

Conclusion: Famotidine-loaded floating micro balloons were successfully prepared and prove to be useful for the prolonged gastric residence of the drug, better bioavailability, patient compliance and anti-ulcer activity.

Keywords: HPMC, Famotidine, In vitro release, Factorial design, % drug entrapment, In vivo floating

INTRODUCTION

Different types of ulcers such as duodenal ulcers, gastric ulcers, Zollinger-Ellison syndrome and gastroesophageal reflux disease are major side effects of many drugs and H₂-receptor antagonists are used for their treatment. Famotidine is one such potent histamine H₂-receptor antagonist with a short biological half-life (2.5-4 h). Famotidine is readily but incompletely absorbed with site-specific absorption in the upper part of the gastrointestinal tract (GIT) with an oral bioavailability of 43-50% due to narrow absorption window in GIT [1-2]. Famotidine is also stable in acidic pH and has low conventional dose (20-100 mg daily in divided dose) so it is able to allow incorporation of polymer and other formulation excipients. Famotidine shows site-specific action in the treatment of different types of ulcers thus providing drug continuously to its absorption sites in a controlled manner, extending absorption phase and increasing the magnitude of drug effect [3].

Therefore the development of controlled/sustained release formulation in the form of multiple units floating drug delivery system (FDDS) such as microballoons (floating microspheres) would be an ideal approach for oral delivery of famotidine. Microballoons are gastro retentive drug delivery systems based on non-effervescent approach. Microballoons are spherical empty particles without core ideally having a size less than 200 micrometres. Microballoons float and can be retained in the stomach due to their lower bulk density than the gastric contents and remain buoyant in the stomach for a prolonged period of time without affecting the gastric emptying rate of other contents. Microballoons assist in improving the oral sustained delivery of drugs that have an absorption window in a particular region of the gastrointestinal tract.

Microballoons are considered as one of the most promising buoyant systems, as they possess the unique advantages of multiple unit systems as well as better floating properties, because of central hollow space inside the microsphere [4-6]. The aim of present study was to formulate and optimize gastro-retentive microballoons of famotidine for increasing gastric residence time of formulation thereby giving controlled release of famotidine in the gastric fluid for site-specific treatment of peptic ulcer, increasing the magnitude of drug effect, improving the oral bioavailability of the drug and patient compliance.

MATERIALS AND METHODS

Materials

Famotidine was obtained as gift sample from Sun Pharmaceuticals Pvt Ltd, Gujarat, India. Hydrochloric acid was purchased from qualigens fine chemicals, Mumbai, India. HPMC K4M was procured from Titan Biotech Ltd, India. All the ingredients used were of pharmaceutical grade. Solvents of reagent grade were used in all experiments.

Methods

Drug-excipient compatibility study

Drug-excipient compatibility study was carried out to check the compatibility of the drug with the polymer in preparation of microspheres of famotidine. The drug and polymer mixture of quantity 10 mg and 400 mg of KBr were acquired in a mortar and were triturated. A little volume of the triturated sample was seized and kept on to the sample holder and examined from 4000 cm⁻¹ to 400 cm⁻¹ in Shimadzu FTIR 8400 spectrophotometer. The spectra obtained were matched with that of the peaks obtained from the FTIR study of the pure drug sample and interpreted for the interaction of drug and polymers if any [7-9].

Preparation of famotidine loaded floating microballoons

Floating microballoons were prepared by the emulsion solvent diffusion method. Nine formulations of floating microballoons were prepared. Different ratios of famotidine and HPMC K4M were mixed in a mixture of ethanol and dichloromethane (DCM) in the ratio of 2:1.
The resulting suspension was added slowly into stirring to 0.50% w/v 400 ml solution of polyvinyl alcohol (PVA) at room temperature. The emulsion formed was stirred continuously for 2 h using a propeller-type agitator at different rpm (900, 1200 and 1500 rpm). The temperature was maintained at 40°C. The finely dispersed droplets of the polymer solution of the drug were solidified in an aqueous phase via the diffusion of the solvent, leaving the cavity of microspheres filled with water. Microspheres formed were filtered using a nylon cloth and washed repeatedly with distilled water. The collected floating microspheres were dried at room temperature and stored in desiccators [10].

Evaluation of famotidine loaded floating microspheres

Micrometrics properties of microspheres

Angle of repose

The angle of repose is the angle a pile forms with the ground. Angle of repose was determined using fixed funnel method. The height of the funnel was adjusted in such a way that the tip of the funnel just touches the heap of the blends. The accurately weighed blend is allowed to pass through the funnel freely onto the surface. The height and diameter of the powder cone were measured and angle of repose was calculated using the following equation.

\[ \theta = \tan^{-1}\left(\frac{h}{r}\right) \]

Where, \( h \) = height of pile, \( r \) = radius of pile, and \( \theta \) = angle of repose

Bulk density and tapped density

Bulk density is defined as the mass of powder divided by the bulk volume. Bulk density is determined by pouring a known quantity of microspheres into the measuring cylinder without compacting and read the unset apparent volume.

Bulk density is the ratio between a given mass of a microsphere and its bulk volume.

Bulk Density = Mass of powder/Bulk volume of the powder

Tapped density is determined by transferring a known quantity of microspheres (2 gm) into a measuring cylinder (10 ml) and is tapped mechanically till a constant volume is obtained. This volume is the bulk volume and it includes the true volume of the powder and the void space among the microspheres. Tapped density is calculated by using the formula.

\[ \text{Tapped density} = \frac{\text{Weight of powder}}{\text{Tapped volume of the powder}} \]

Carr’s index

Carr’s compressibility index CI (Carr, 1965) is defined as follows:

\[ CI = \frac{\rho_t - \rho_a}{\rho_a} = \frac{\rho_c - \rho_v}{\rho_v} \]

Where \( \rho_t \) and \( \rho_a \) = tapped and poured bulk density; and \( \rho_c \) and \( \rho_v \) = tapped and poured bulk volume respectively.

Hausner’s ratio

A similar index has been defined by Hausner [11].

Hausner’s ratio = Tapped density/Poured density

Determination of particle size

The particle size of microspheres was measured by using an optical microscope fitted with an eyepiece micrometre and a stage micrometre. The mean particle size was determined by measuring 200 particles with the help of a calibrated eyepiece micrometre [12].

Surface morphology

The surface characteristics and internal textures of the famotidine loaded floating microspheres were studied by scanning electron microscope. The samples were mounted on double-sided tape that has previously been secured on copper stubs and then analyzed at different magnifications. Photomicrographs were captured randomly using scanning electron microscope.

Drug entrapment efficiency

Estimation of drug entrapment efficiency (DEE) in floating microspheres was carried out by dissolving the weight amount (100 mg) of crushed microspheres in required quantity of 0.1N HCL in 1.20 and analyzed spectrophotometrically at a wavelength of 266 nm using the calibration curve. The polymer debris was removed by filtering through Whatman filter paper (No. 40). Each batch should be examined in a triplet manner. The entrapment efficiency of floating microspheres was calculated by dividing the actual drug content by the theoretical drug content of microspheres [13]. The % DEE of floating microspheres was calculated using this following formula:

\[ \text{DEE} \% = \left( \frac{\text{Actual drug content of microspheres}}{\text{Theoretical drug content of microspheres}} \right) \times 100 \]

In vitro buoyancy study

This study was carried out by USP type II dissolution test apparatus by spreading the microspheres on a simulated gastric fluid (900 ml of 0.1 N HCL, pH 1.20) containing 0.02% w/v Tween 20 as a surfactant. The media is stirred at 100 rpm at 37±0.5°C for 12 h. After a specific interval of time, both the fractions of microspheres (floating and settled microspheres) were collected and buoyancy of the floating microspheres was determined by using formula [14].

\[ \text{Buoyancy} \% = \frac{Q_f}{Q_f + Q_s} \times 100 \]

Where Q_f is the weight of floating microspheres and Q_s is the weight of settled microspheres respectively.

In vitro dissolution study

The in vitro dissolution studies were carried out using USP type II dissolution apparatus (Paddle Type). The study was carried out in 900 ml of 0.1N HCL (pH 1.20). The dissolution medium was kept in a thermostatically controlled water bath, maintained at 37±0.5°C. Microspheres containing drug equivalent to 40 mg were spread over the surface of 900 ml of dissolution media (0.1N HCL, pH 1.20). The paddle was rotated at 50 rpm. At predetermined time intervals i.e. 1, 2, 4, 6, 8, 10, 12 and 15 h 5 ml of sample was withdrawn from the dissolution apparatus and replaced with fresh media to maintain sink conditions. The drug concentration was analyzed by using UV spectrophotometer (Shimadzu 1800, Japan) at 266 nm [15-17].

Optimization and validation of experimental design

Optimization technique based on response surface methodology was utilized. It is generally used to determine the optimum combination of factors that yield the desired response and describes the response near the optimum. The runs or formulations, which are designed based on 3% factorial designs, are evaluated for the response variables. The response values are subjected to multiple regression analysis to find out the relationship between the factors used and the response values obtained. Independent variables or factors studied were a drug: polymer ratio \( (X_1) \) and stirring speed \( (X_2) \). The response values or dependent variables subjected for this analysis were % entrapment efficiency, % drug release at 12 h and % Buoyancy. The effect of formulation variables on the response values was statically evaluated by applying one-way ANOVA at 0.05 level using a commercially available software package Design Expert® 10.0.3 (Stat-Ease, USA). The design was evaluated by polynomial cubic model, which bears the form of the following equation.

\[ Y = b_0 + b_1X_1 + b_2X_1^2 + b_3X_2 + b_4X_1X_2 + b_5X_1^2 + b_6X_2^2 + b_7X_1^3 + b_8X_2^3 + b_9X_1^2X_2 + b_{10}X_1X_2^2 \]

Where \( Y \), is the dependent variable; \( b_0 \) is the arithmetic mean of the 9 runs; and \( b_1 \) and \( b_2 \) are the estimated coefficients for the independent factors \( X_1 \) and \( X_2 \) respectively. The main effects \( (X_1 \text{ and } X_2) \) represent the average result of changing one factor at a time from its low to high value. The interaction term \( (X_1X_2) \) shows how the response changes when 2 factors are simultaneously changed. The polynomial terms \( (X_1^2 \text{ and } X_2^2) \) are including investigating nonlinearity. The student t-test was conducted to examine the probability of each coefficient being equal to zero. All tests were performed at a 95% level of significance. In the final
We incorporated the results into the extra design checkpoint formulation. The values for the variables obtained after transformation were designated as F10. The extra design checkpoint formulation (F10) was obtained from transformation of each of two variables at three levels. In general, the formula for transformation is as follow [18].

\[
\text{X} = \frac{1}{2} \text{average of the three levels}
\]

The values for the variables obtained after transformation were incorporated into the extra design checkpoint formulation designated as F10.

The model, a comparison between the experimental and predicted values of the responses is also presented in terms of % Bias. % Bias was calculated by the following equation:

\[
\% \text{Bias} = \frac{\text{Predicted value} - \text{Experimental value}}{\text{Predicted value}} \times 100
\]

To demonstrate graphically the influence of each factor on responses and to indicate the optimum level of factors, the contour and response surface plots were generated using Design-Expert 10.0.3 software, p<0.05 was considered significant.

Drug release kinetics of optimized formulation

In vitro drug release data of optimized formulation (F6) was subjected to various mathematical models such as zero order, first order, Higuchi and Peppas kinetics models to understand the mechanism from the floating microballoons of famotidine. All curve fitting and plotting were performed using Microsoft excel software and regression coefficient (r^2) values were calculated. In this by comparing the regression coefficient (r^2) values obtained, the best fit model was selected [19-23].

In vivo floating behaviour study

Healthy rabbits weighing approximately 2 kg were used to assess in vivo floating behaviour. Ethical clearance for the handling of experimental animals was obtained from the institutional animal ethical committee (Registration No. 1321/PO/ReBi/S/10/PCPSEA) constituted for the purpose. The animals were housed in individual cages, and the experiments were carried out in a sanitized room. Rabbits fasted for 12 h and the first X-ray photograph was taken to ensure the absence of radio-opaque material in the stomach. Barium sulphate was incorporated into the microballoons to make the microballoons X-ray opaque. The amount of barium sulphate was kept 10 mg per g of microballoons to ensure visibility by X-ray and to enable the microballoons to float. Rabbits were made to swallow barium sulphate loaded microballoons with sufficient quantity of water. During the experiment, rabbits were not allowed to eat but the water was provided ad libitum. At predetermined time intervals (0, 6 and 12 h) the radiographs of the abdomen were taken using an X-ray machine [24-25].

Antiulcer activity

Albino Wistar rats of either sex weighing between [150-200 gms] were used to determine antiulcer activity. Ethical clearance was obtained from the institutional animal ethical committee (Registration No. 1321/PO/ReBi/S/10/PCPSEA) constituted for the purpose. Albino Wistar rats were housed in individual cages, and the experiments were carried out in a sanitized room. Albino Wistar rats were divided into four groups of six animals in each group. Group I served as control rats, administered 80% v/v ethanol (1 ml/200 gm, p. o); Group II administered optimized formulation (20 mg/kg, p. o); Group III administered optimized formulation modified with addition of 25 mg of gallic acid (20 mg/kg, p. o) to study the effect of gallic acid (natural antioxidant); Group IV administered famotidine marketed drug (20 mg/kg, p. o). The animals fasted for 24 h with free access of water. The animals were treated with formulations or control according to the various groups as designed. 1 hour later 80% v/v alcohol was administered p.o. to each animal. Animals were sacrificed after 1 hour of alcohol administration, stomachs were isolated, cut and opened along the greater curvature and pinned on a soft board. Ulcer index parameters were studied [26, 27].

Erosions formed on the glandular portion of the stomach were counted and each was given a severity rating on 1-3 scale, based on the diameter of the ulcer. Scoring of ulcers for normal stomach, red coloration, spot ulcer, hemorrhagic streak, ulcers and perforation were carried out as 0, 0.5, 1, 1.5, 2 and 3 respectively.

Mean ulcer score for each animal will be expressed as ulcer index. % Ulcer protection can be calculated by using this formula.

\[
\% \text{Ulcer protection} = \frac{\text{Mean ulcer in control} - \text{mean ulcer in test}}{\text{Mean ulcer in control}} \times 100
\]

Calculation of ulcer index can be carried out using this formula.

\[
\text{UI} = (\text{UN} + \text{US} + \text{UP}) \times 10^{-1}
\]

UI = Ulcer index
UN = Average of number of ulcer per animal
US = Average of severity score
UP = Percentage of an animal with ulcer

The Dunnett’s test was employed for statistical comparison. In all the cases, values of P<0.05 were considered significant. All values were presented as mean±SEM.

RESULTS AND DISCUSSION

Drug-excipient compatibility study

After performing FTIR of the famotidine, HPMC and microballoons, it was found that the peaks obtained in the formulation were in between the range of main principle peaks and were found to be very near to previously performed FTIR of pure drug famotidine. No major deviation in peaks was obtained in IR spectra, hence this indicates that drug was compatible with excipients without any significant interaction between famotidine and HPMC. The results of IR spectra suggest that selection of excipient for preparation of microballoons were suitable. Hence it cannot alter the therapeutic efficacy of famotidine, and it also supports to continue further research works.
Preparation of famotidine loaded floating microballoons

The floating famotidine loaded microballoons were prepared using emulsion solvent diffusion method as discussed in preparation of floating microballoons of famotidine. The polymer used was HPMC K4M and its composition was tabulated in table 1. Further, the stirring speed was varying from 900 to 1500 rpm. The finely dispersed droplets of the polymer solution of the drug were solidified in an aqueous phase via the diffusion of the solvent, leaving the cavity of microspheres filled with water. The benefits of preparation technique include low processing time, lack of exposure of drug to high temperature due to which stability of drug increased during the processing leading to high % entrapment efficiency of the drug in microballoons.

Table 1: Composition of floating microballoons using 3² full factorial experimental design

<table>
<thead>
<tr>
<th>Formulations code</th>
<th>Variable level in the coded form</th>
<th>X₁</th>
<th>X₂</th>
</tr>
</thead>
<tbody>
<tr>
<td>F1</td>
<td>-1</td>
<td>-1</td>
<td></td>
</tr>
<tr>
<td>F2</td>
<td>0</td>
<td>-1</td>
<td></td>
</tr>
<tr>
<td>F3</td>
<td>1</td>
<td>-1</td>
<td></td>
</tr>
<tr>
<td>F4</td>
<td>-1</td>
<td>0</td>
<td></td>
</tr>
<tr>
<td>F5</td>
<td>0</td>
<td>0</td>
<td></td>
</tr>
<tr>
<td>F6</td>
<td>1</td>
<td>0</td>
<td></td>
</tr>
<tr>
<td>F7</td>
<td>-1</td>
<td>1</td>
<td></td>
</tr>
<tr>
<td>F8</td>
<td>0</td>
<td>1</td>
<td></td>
</tr>
<tr>
<td>F9</td>
<td>-1</td>
<td>1</td>
<td></td>
</tr>
</tbody>
</table>

Concentration/value of independent variables

<table>
<thead>
<tr>
<th>Level</th>
<th>Concentration of HPMC K4M (mg)</th>
<th>Stirring Speed (rpm)</th>
</tr>
</thead>
<tbody>
<tr>
<td>-1</td>
<td>100</td>
<td>900</td>
</tr>
<tr>
<td>0</td>
<td>200</td>
<td>1200</td>
</tr>
<tr>
<td>1</td>
<td>300</td>
<td>1500</td>
</tr>
</tbody>
</table>

Response variables

<table>
<thead>
<tr>
<th>Y₁</th>
<th>% Entrapment Efficiency</th>
</tr>
</thead>
<tbody>
<tr>
<td>Y₂</td>
<td>% Drug Release in 12 h</td>
</tr>
<tr>
<td>Y₃</td>
<td>% Buoyancy</td>
</tr>
</tbody>
</table>

Micromeritic properties of microballoons

All the prepared batches of famotidine microballoons were evaluated for micromeritic properties such as bulk density, tapped density, carr’s compressibility index, hausner’s ratio and angle of repose as shown in table 2.

Angle of repose

The values for angle of repose of all prepared nine formulation batches were determined in the range of 22.14°±0.67 to 27.19°±0.62 showing that the different batches of microballoons were free-flowing with good flow property. In case of pure drug angle of repose was 39.4°.
±1.34 suggesting poor flow properties. All formulation batches possess smoother and regular surface of the particles.

### Bulk density

The bulk density values for all nine formulation batches were determined. The result of poured bulk density (g/cm³) ranged from 0.49±0.24 to 0.69±1.06 as shown in table 2. The bulk densities of different formulations were found to be much less than the density of the gastric fluid (1.004 g/cm³) and 0.1 N HCl, pH 1.20 (0.997 g/cm³). Low density of microballoons increased the porosity of microballoons. Being less in density, the microballoons were expected to float immediately with less or no lag time.

### Tapped density

The tapped density of microballoons of all nine formulation batches were found to be in the range of 0.55±0.58 to 0.76±0.45 g/cm³ as tabulated in table 2. Therefore, it was expected to be suitable for formulation of floating microballoons as they were having less density than 0.1 N HCl, pH 1.20.

### Compressibility index

Carr’s compressibility index was evaluated to determine compressibility characteristics of the microballoons. Carr’s compressibility index of floating microballoons of all nine formulations ranged from 7.93±0.0009 to 13.11±0.0003 % indicating excellent flow properties of microspheres with good compressibility while in case of pure drug carr’s compressibility index was 24.86±0.0012 % showed poor flow properties of the pure drug. Therefore, prepared floating microballoons showed improvement in flow properties as compared to pure drug and packability inside the capsules meant an ease of filling the microballoons inside the capsules.

### Hausner’s ratio

Hausner’s ratio for all nine formulations was in the range of 1.08 to 1.15 (<1.25) indicating good flow properties of microballoons while in case of a pure drug, Hausner ratio was 1.33. It suggested that Hausner ratio of all formulations of microballoons were less than pure drug and the result of the study showed enhancement of flow properties of the pure drug in case of floating microballoons.

### Determination of particle size

The mean particle size of all formulations was ranged from 109.43±1.18 µm to 160.58±0.73 µm as tabulated in table 2. Formulation batches F7 showed smallest and F3 showed the largest size of microballoons. Higher particle size was obtained when the proportion of HPMC K4M was increased in formulations. This was due to a significant increase in viscosity in a fixed volume of solvent, thus causing an increase in emulsion drop size and finally increases in size of particles. Increased speed of mixing (stirring) resulted in the formation of small-sized microballoons.

### Table 2: Micromeritics properties of different formulations of floating microspheres

<table>
<thead>
<tr>
<th>S. No.</th>
<th>Formulation code</th>
<th>Angle of repose (°)</th>
<th>Poured Density (g/cm³)</th>
<th>Tapped density (g/cm³)</th>
<th>Carr’s compressibility index (%)</th>
<th>Hausner’s ratio</th>
<th>Particle size (µm)</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>F1</td>
<td>22.14±0.67</td>
<td>0.49±0.24</td>
<td>0.55±0.50</td>
<td>10.90</td>
<td>1.12</td>
<td>151.25±0.25</td>
</tr>
<tr>
<td>2</td>
<td>F2</td>
<td>23.32±0.12</td>
<td>0.53±0.63</td>
<td>0.61±0.48</td>
<td>13.11</td>
<td>1.15</td>
<td>158.17±0.86</td>
</tr>
<tr>
<td>3</td>
<td>F3</td>
<td>25.28±0.61</td>
<td>0.61±0.72</td>
<td>0.67±0.35</td>
<td>8.95</td>
<td>1.09</td>
<td>160.58±0.73</td>
</tr>
<tr>
<td>4</td>
<td>F4</td>
<td>23.42±0.37</td>
<td>0.53±0.57</td>
<td>0.58±0.67</td>
<td>8.62</td>
<td>1.09</td>
<td>129.56±0.68</td>
</tr>
<tr>
<td>5</td>
<td>F5</td>
<td>23.84±1.74</td>
<td>0.59±0.39</td>
<td>0.65±0.53</td>
<td>9.23</td>
<td>1.10</td>
<td>135.28±1.15</td>
</tr>
<tr>
<td>6</td>
<td>F6</td>
<td>26.14±0.49</td>
<td>0.68±0.42</td>
<td>0.76±0.45</td>
<td>10.52</td>
<td>1.11</td>
<td>141.62±1.24</td>
</tr>
<tr>
<td>7</td>
<td>F7</td>
<td>23.68±0.88</td>
<td>0.58±0.82</td>
<td>0.63±0.50</td>
<td>7.93</td>
<td>1.08</td>
<td>109.43±1.18</td>
</tr>
<tr>
<td>8</td>
<td>F8</td>
<td>24.72±0.39</td>
<td>0.69±1.06</td>
<td>0.75±0.46</td>
<td>8.00</td>
<td>1.08</td>
<td>113.32±0.92</td>
</tr>
<tr>
<td>9</td>
<td>F9</td>
<td>27.19±0.62</td>
<td>0.62±0.76</td>
<td>0.68±0.73</td>
<td>8.82</td>
<td>1.09</td>
<td>120.14±0.78</td>
</tr>
</tbody>
</table>

* All values are expressed as mean±SD, n =3

### Surface morphology

The surface morphology of famotidine loaded floating microballoons was investigated by SEM. It was seen that microballoons were spherical in appearance and exhibit a range of sizes within each batch. The surface was observed to be smooth, dense and less porous, whereas the internal core was highly porous. The less porous outer surface and highly porous internal surface supported controlled release of drug from the microballoons and good buoyancy. The porous nature and cavity formed in the microballoons would dictate the floating behaviour of famotidine loaded floating microballoons.

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(a) (b) (c)

**Fig. 2:** SEM photographs of (a) range of floating microspheres (b) smooth texture of floating microspheres (c) porous internal surface of floating microspheres
Drug entrapment efficiency

The percent drug entrapment efficiency (% DEE) of all formulation batches were found to be in the range of 63.47±1.46% to 70.34±0.98% as shown in fig. 2. The % drug entrapment efficiency was more when the concentration of polymer (HPMC K4M) was increased in the drug: polymer ratio. Drug encapsulation efficiency was found to be increased due to increase in the concentration of gelling polymer leading to increase in crosslinking structure and viscosity of internal phase which leads to reduce migration of drug in aqueous phase. % DEE of formulations was found to decrease with an increase in stirring speed from 900 to 1500 rpm due to smaller size microspheres formed at a higher speed of rotation and leakage of drug from microspheres.

In vitro buoyancy study

In vitro buoyancy (%) for all the formulations batches, F1 to F9 were varied from 64.17±1.16% to 83.21±1.07%. It was observed that all the formulations were able to remain float in a continuous manner on the dissolution medium over a period of 12 h. The in vitro buoyancy of hollow microspheres of famotidine was due to the presence of pores, a hollow structure and low bulk density. It may also deduce that these microballoons can float in gastric fluid retarding the passage of the microballoons into the intestinal region and increasing their residing time in the stomach. The in vitro % buoyancy was found to increase with increasing concentration of polymer (HPMC K4M) while decreased with an increase in stirring speed.

In vitro dissolution study

In vitro drug release of famotidine from microballoons was performed in 0.1 N HCl, pH 1.20. The in vitro release profile showed initial burst release up to 1 h which may be due to the surface associated drug, followed by a sustained release phase as the entrapped drug slowly diffused into the dissolution medium. There was the sustained release of drug at a constant rate. The results of the study showed that on increasing the concentration of HPMC K4M it decreased drug release from microballoons. The high concentration of HPMC K4M decreases the drug release from microballoons due to more stiffness of microballoons and increased diffusional path length. The other reason may be that smaller microballoons were formed at lower concentrations of HPMC K4M and have a large surface area exposed to the dissolution medium resulting in faster drug release. Higher stirring speed also lead to increase in drug release due to the breakdown of microballoons into small pieces or rupturing the surface of microballoons. Formulations with low stirring speed showed sustained release of drug from microballoons.

Fig. 3: Comparative chart of % encapsulation efficiency and % buoyancy of different formulations

Fig. 4: In vitro drug release profile of floating microballoons for (a). F1 to F5 (b). F6 to F9
Optimization and validation of experimental design

The nine formulations of microspheres were prepared to study the effect of polymer concentration and speed of stirring speed. The effect of formulation variables on the response variables was statically evaluated by applying ANOVA at 0.05 level using a commercially available software package Design Expert® 10.0.3 (Stat-Ease, USA). Mathematical polynomial cubic equations were generated for all the three dependent variables or response parameters such as % entrapment efficiency, % drug release at 12 h and % Buoyancy. The mathematical models were tested for significance. The response values are subjected to multiple regression analysis to find out the relationship between the factors used and the response value obtained.

As there were insignificant terms, the model reduction was required. After reduction, reduce cubic model was used and thus the data points were better fitted with the model and all the response models were significant with the response parameters. It was found that quadratic model is best fitted to determine the effect of independent variables on response variables.

The quadratic equations generated for responses were given as:

% EE = 68.52 + 0.54X₁ - 1.08X₂ - 2.39X₁X₂ - 0.77X₁² - 1.72X₂²
% Drug release = 86.38 - 2.61X₁ + 3.15X₂
% Buoyancy = 73.41 + 7.31X₁ - 2.32X₂

Where X₁ and X₂ represent the coded values of the polymer concentration and stirring speed (rpm) respectively. The positive value of a factor in the above equations points out the enhancement of that response and vice versa.

To demonstrate graphically the influence of each factor on responses and to indicate the optimum level of factors, the contour and response surface plots were generated using Design-Expert 10.03 software.

Fig. 5: Response surface and contour plot of effect of X₁ and X₂ on (a). % entrapment efficiency (b). % drug release at 12 h (c) % buoyancy
By calculating actual polymer concentration (225 mg) and stirring speed (1275 rpm) from transformed proportions of each variable, the extra design checkpoint formulation (F10) was designed.

The extra design checkpoint batch was observed to have % entrapment efficiency, % drug release and % buoyancy of 68.33%, 84.21% and 78.66% respectively. The statistical insignificance of the observed values for extra design checkpoint formulation (F10) was evaluated with the predicted value using Student’s t-test. It was found to non-significant with 95% confidence interval. This statistical insignificance of the difference between the predicted and observed responses of extra design checkpoint formulation not only validate the design adopted for optimization but also confirmed the usefulness of a polynomial equation in predicting the % entrapment efficiency, % drug release and % buoyancy.

For the purpose of selecting the optimized formulation, the batch having maximum similarity when compared with the response variables of the extra design checkpoint batch and predicted values can be considered as the optimized batch.

Table 3 showed that the observed values of the prepared batch (F6) were very close to the predicted values, with low percentage bias, suggesting that the optimized formulation was trustworthy and rational.

<table>
<thead>
<tr>
<th>Responses variable</th>
<th>Predicted value</th>
<th>Observed value</th>
<th>Bias %</th>
</tr>
</thead>
<tbody>
<tr>
<td>% Entrapment Efficiency</td>
<td>67.0957</td>
<td>67.63</td>
<td>-0.796</td>
</tr>
<tr>
<td>% Drug Release</td>
<td>83.5238</td>
<td>83.52</td>
<td>0.0045</td>
</tr>
<tr>
<td>% Buoyancy</td>
<td>79.6226</td>
<td>80.14</td>
<td>-0.649</td>
</tr>
</tbody>
</table>

Drug release kinetics of optimized formulation

The best selected in vitro drug release data of optimized formulation were fitted to various kinetics models such as zero order, first order, Higuchi and Peppas models. The highest regression coefficient ($r^2$) values were obtained for Higuchi model (0.9979) followed by Korsmeyer-Peppas (0.9968), zero order (0.9364) and first order (0.9277) model using Microsoft excel software. It indicates diffusion to be the predominant mechanism of drug release from floating microballoons. Release mechanism was studied by Peppas Equation and value of slope or diffusion exponent ($n$) was determined. The value of diffusion exponent ($n$) was found to be 0.4956 that indicates non-Fickian or anomalous diffusion mechanism which leads to the conclusion that drug release mechanism indicates a combination of diffusion and spheres erosion. Zero order, first order, Higuchi and Korsmeyer-Peppas plots were shown in fig. 6 (a), (b), (c) and (d).

In vivo floating behavior study

The in vivo floating behavior of microballoons loaded with barium sulphate was investigated by radiographic images (X-ray photographs) of rabbit’s stomach at periodic time intervals. The radiographs obtained at 0, 6 and 12 h are shown in fig. which indicates a uniform distribution of formulation over the gastric fluid and in vivo duration of floating for more than 12 h.
Antiulcer activity

In control group of animals, oral administration of absolute ethanol produced characteristic lesions in the glandular portion of rat stomach which appeared as elongated bands of thick, black and dark red lesions. The in vivo evaluation showed that animals treated with the optimized formulation, an optimized formulation with gallic acid and administered marketed brand of famotidine showed significant protection index of 83.26%, 89.04% and 76.09% respectively in comparison to control. Gallic acid is a well-known natural antioxidant that is basically a secondary polyphenolic metabolite. Floating microballoons of an optimized formulation containing gallic acid showed high ulcer protection index of 89.04% as compared to 83.26% in the optimized formulation (F6). It may be due to the synergistic effect of famotidine with gallic acid. The combination of both may provide formulations with increased therapeutic effect. Macroscopical changes of ethanol-induced models are shown in fig. 8(a), (b), (c) and (d).

Table 4: Effect of various formulations on ethanol-induced gastric ulcer in rats

<table>
<thead>
<tr>
<th>Treatment</th>
<th>Ulcer index</th>
<th>% Ulcer Protection</th>
<th>pH of gastric juice</th>
</tr>
</thead>
<tbody>
<tr>
<td>Control</td>
<td>4.1±1.25</td>
<td>-</td>
<td>1.9±0.62</td>
</tr>
<tr>
<td>Optimized formulation</td>
<td>0.62±1.06*</td>
<td>83.26</td>
<td>4.3±0.39*</td>
</tr>
<tr>
<td>Formulation with gallic acid</td>
<td>0.49±1.35*</td>
<td>89.04</td>
<td>4.8±0.81*</td>
</tr>
<tr>
<td>Marketed brand of famotidine</td>
<td>0.98±10.83*</td>
<td>76.09</td>
<td>3.9±0.24*</td>
</tr>
</tbody>
</table>

Values expressed as mean±SEM; *P<0.05 when compared with control group.

Fig. 7: X-ray photograph of rabbit stomach: (a) without drug loaded microballoons (b) showing buoyancy of microballoons after 6 h (c) showing buoyancy of microballoons after 12 h

Fig. 8: (a) Ulcerated control stomach (b) Optimized formulation treated stomach (c) Optimized formulation with gallic acid treated stomach (d) Famotidine marketed drug treated stomach
CONCLUSION

Floating microballoons of famotidine with swellable hydrophilic polymers were successfully prepared using emulsion solvent diffusion method, by applying 3² factorial design. The prepared microballoons had a different size and the % entrapment efficiency of the drug by varying the formulation variables such as polymeric concentration and stirring rate. The prepared formulations were further evaluated and based on the results of in vitro drug release studies, % entrapment efficiency and % buoyancy, F6 was chosen as the best formulation.

Optimized formulation (F6) followed Higuchi kinetics and the release mechanism was non Fickian diffusion (n=0.4956). SEM study showed porous nature and cavity formed in the microballoons. In vivo radiographic images were helpful to locate the position of floating microballoons in the gastrointestinal tract and also confirmed that the prepared microballoons were able to retain in the stomach for a prolonged period of time and sustained release of famotidine. Thus, the prepared buoyant microballoons may prove to be potential candidates for a multiple-unit delivery system for intragastric drug delivery. In vivo anti-ulcer evaluation showed that animals treated with the optimized formulation, an optimized formulation with the gallic acid of famotidine showed significant ulcer protection of 83.26% and 89.04% respectively in comparison to control. Thus, such floating microballoons of famotidine prove to be useful for the prolonged gastric residence of the drug, better bioavailability and anti-ulcer activity.

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AUTHORS CONTRIBUTIONS

All the authors have contributed equally

CONFLICT OF INTERESTS

There are no conflicts of interest

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