FORMULATION OF ORALLY DISINTEGRATING TABLETS OF CINNARIZINE BY USING DIRECT COMPRESSION METHOD

RASHA KHALID DHAIHR1, MYASAR AL-KOTAIFI2
1-2Department of Pharmaceutics, College of Pharmacy, University of Mosul, Mosul, Iraq

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INTRODUCTION

Over the last decade, the development of orally disintegrating tablets (ODTs) has received increased interest among pharmaceutical industries and researchers [1]. Such dosage form is not only preferable for elderly and children, but also for adults consumers who like to have their medication obtainable at any time [2].

Orodispersible tablets are defined according to European Pharmacopoeia as “uncoated tablets intended to be placed in the mouth where they disperse rapidly before being swallowed” [3]. ODTs are also known as rapid dissolve, fast melting, quick disintegrating, fast dispersing, and/or rapid melt tablets [4]. After being disintegrated in saliva, ODTs produce a solution or suspension that can be swallowed easily by the patient [5]. It means that this dosage form mixes the advantages of both liquid and solid formulations. Also, the opportunity for pregastric absorption of the drug may increases as the drug passes through the mouth, pharynx, and esophagus toward the stomach. In such case bioavailability of the drug is significantly improved when compared with that of the conventional tablets [6].

Direct compression can be regarded as the easiest, most appropriate, and simplest way to obtain a rapidly disintegrating tablet with adequate structural integrity. In the direct compression method, the most important factor to be taken in concern is the selection of suitable excipients with good disintegration properties and compatibility. Several factors should be considered in selecting the super disintegrants because these agents do not affect the rate of disintegration only, but they also affect friability, hardness and how the tablet feels in the mouth. The disintegration and dissolution properties of the direct compressible ODTs are based on the single or combined action of disintegrants and water-soluble excipients. However, the disintegrants have the major role. Choosing of a suitable disintegrant type and in optimal concentration is essential for ensuring high disintegration rate [7].

Cinnarizine is histamine H1-receptor blocker and voltage-gated calcium channel blocker used for the treatment of vertigo, Meniere's disease, motion sickness, nausea, vomiting, and other symptoms of vestibular origin [8]. The drug is tasteless and water-insoluble [9]; therefore it can be considered as a model drug to be formulated as ODTs. Pharmaceutically, it belongs to the biopharmaceutical classification system (BCS) class II, and drug dissolution is the limiting step for its absorption [10].

Preparation of cinnarizine ODTs with enhanced disintegration and dissolution rate is beneficial because diseases like motion sickness, nausea and vomiting require immediate release of the drug from the dosage form. In addition, cinnarizine ODTs may improve patient compliance and reduce the side effect of the drug. The aim of the present work was to formulate cinnarizine orally disintegrating tablets by direct compression method, to ensure that high percentage of the drug can be dissolved in the stomach within a short period of time which increases the opportunity for high bioavailability of the drug.

MATERIALS AND METHODS

Materials

Cinnarizine powder was purchased from Apollo Healthcare Resources, Singapore. Lactose, Super Tab® 40LL was provided by DFE pharma, Germany. Microcrystalline cellulose PH 102 was taken from JRS pharma, US. Pearlitol, mannitol (400) DC was purchased from Roquette Pharma, French. Kyron T-314 was obtained from COREL PHARMA CHEM, India. Crospovidone (CP), sodium starch glycolate (SSG), sodium saccharin, citric acid anhydrous, talc, and magnesium stearate were supplied from Awamedica Drug Industries, Iraq.

Preparation of cinnarizine ODTs by direct compression method using different types of super disintegrants

Twelve different formula of orally disintegrating tablets of cinnarizine (25 mg) were prepared by direct compression method using three...
different types of disintegrants as demonstrated in table 1. Sodium saccharine and citric acid were added to mask any unacceptable taste while magnesium stearate and talc were added as lubricant and glidant respectively. All the ingredients were passed through a sieve (mesh # 60) and blended thoroughly for 10 min, except talc and magnesium stearate which were added later and blended for further 2-3 min. The powder blend was compressed in tablets using 8 mm normal flat punches to get tablets of 180 mg weight using multi-rotary tablet compression machine (Mini press II MT tablet compression, Rimek, India).

Table 1: Composition of cinnarizine (25 mg) ODTs with a total weight of 180 mg

<table>
<thead>
<tr>
<th>Ingredients</th>
<th>F1 % (w/w)</th>
<th>F2 % (w/w)</th>
<th>F3 % (w/w)</th>
<th>F4 % (w/w)</th>
<th>F5 % (w/w)</th>
<th>F6 % (w/w)</th>
<th>F7 % (w/w)</th>
<th>F8 % (w/w)</th>
<th>F9 % (w/w)</th>
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<th>F9b % (w/w)</th>
<th>F9c % (w/w)</th>
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<td>13.9</td>
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<td>Lactose, Super Tab® 401L</td>
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<td>Kyron-T314</td>
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<td>-</td>
<td>-</td>
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<td>-</td>
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<td>6</td>
<td>7</td>
<td>8</td>
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<tr>
<td>Total weight % (w/w) 180 mg</td>
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<td>100</td>
<td>100</td>
<td>100</td>
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</tr>
</tbody>
</table>

*W/w: Percentage by weight

Flowability study of the prepared powder mixtures

Measuring of the angle of repose

The angle of repose is a characteristic related to inter-particle friction or resistance to movement between particles [11]. It was determined using fixed height funnel method by measuring the height of the cone and the base formed by the powder falling through a funnel. It is calculated by applying the following equation:

\[
\tan \theta^o = \frac{h}{r}
\]

Where \( \theta^o \) = Angle of repose; \( h \) = height of powder cone; \( r \) = radius of powder cone.

Compressibility index (Carr’s Index) and Hausner’s ratio

*Compressibility is the measure obtained from the shrinkage of the apparent volume for a defined mass of powder after the rearrangement of the grains/particles during the application of tapping*[12]. Compressibility index (Carr’s Index) and Hausner’s ratio were determined by gently loading a certain amount of the powder through a funnel into 100 ml cylinder, and tapped until the volume stopped changing. Carr’s index and Hausner’s ratio were calculated according to the equation below:

\[
\text{Carr’s index} = \frac{V^t - V^f}{V^t} \times 100, \text{ Hausner’s ratio} = \frac{V^f}{V^t}
\]

Where \( V^t \) = untapped apparent volume in ml, \( V^f \) = tapped volume of blend in ml.

Evaluation of the prepared cinnarizine tablets

Hardness (Crushing strength)

Hardness test was done to determine the resistance of a tablet to the mechanical shocks during handling, shipping, storage, and transportation [13]. It was measured by tablet breaking force tester-USP (Electrolab, India) and expressed in Newton (N).

Friability (Resistance to abrasion)

Friability test can be done to evaluate the ability of the tablets to withstand abrasion in packing, handling and transporting. Twenty tablets are randomly taken, dedusted, weighed accurately and transferred into friability tester-1 (EF-2, Electrolab, India). After 100 rotations, the tablets were dedusted and weighed again, the friability percentage was calculated for each formula by the following equation [14]:

\[
\text{Friability } \% = \frac{W_{\text{initial}}-W_{\text{final}}}{W_{\text{initial}}} \times 100
\]

Uniformity of weight

Twenty tablets were selected at random, and the average weight was calculated. The variation of the weight of the individual tablet from the average weight was calculated. In this work, the percentage limits allowed of weight variation according to the average weight of the tablets is 7.5% [3].

Disintegration test

Disintegration time for cinnarizine ODTs was determined by disintegration tester-3 USP, (Electrolab, India), using 900 ml of distilled water, at 37 °C±0.5 [3]. The time needed for complete disintegration of the tablet with no palpable mass remaining in the apparatus was measured in seconds [15]. The disintegration time was also measured for the marketed cinnarizine conventional tablet (Stugeron® 25 mg tablet) as a reference.

Wetting time

The wetting time was measured using a simple procedure in which five circular tissue papers of 10 cm in diameter were placed in a 10 cm circular petri dish. 10 ml solution of a water-soluble dye (0.02% tartrazine dye) was added. A tablet was carefully placed on the surface of the tissue paper, the time spent for water to reach the upper surface of the tablet represents the wetting time [16].

Water absorption ratio

To measure the water absorption ratio, the same procedure as that of wetting time was followed. The initial weight of the tablet before placing it in the petri dish was recorded (W_a), the wetted tablet was taken and reweighed (W_b). water absorption ratio (R) was determined according to the following equation [16]:

\[
R = \frac{W_a-W_b}{W_b} \times 100
\]

Where \( R \) = water absorption ratio; \( W_a \) = weight of the tablet after absorption; \( W_b \) = weight of the tablet before absorption.

Dissolution test (In vitro drug release)

Dissolution test for the prepared ODTs of cinnarizine was performed using a dissolution test apparatus II (Pharma test, PT-IT77, Germany) with a paddle stirrer. A volume of 900 ml of 0.1 N HCl (pH 1.2) was used as a dissolution medium at a speed of 50 rpm and a temperature of 37 °C±0.5. A sample of 5 ml of the medium was withdrawn at the predetermined time interval and replaced by the same volume of fresh drug-free dissolution medium. The samples were filtered through a 0.45 µm filter, diluted and analyzed at 254 nm by UV spectrophotometer (Specord 40, analytic Jena, Germany).
the drug concentration was calculated from the calibration curve [15]. In addition, the dissolution test was conducted for the conventional cinnarizine tablet (Stugeron® 25 mg tablet) as a reference using the same procedure and conditions as that for the prepared ODTs of cinnarizine.

Drug content (assay)

From each formula of the prepared cinnarizine ODTs, 10 tablets were randomly selected and powdered. An equivalent amount of 25 mg of cinnarizine was taken and dissolved in 0.1 N HCl filtered and analyzed for drug content at λmax (254) by UV -visible spectrophotometer [17].

Fourier transforms infrared (FTIR) studies

Fourier transforms infrared (FTIR) spectra of cinnarizine alone, and mixtures of 1:0.57 of cinnarizine with Kyron, and 1:5.3 of cinnarizine with lactose (which represented the actual ratio of the constituent of mixtures of 1:0.57 of cinnarizine with Kyron, and 1:5.3 of cinnarizine with lactose) were recorded to detect any possible interaction between cinnarizine and other excipients. IR-spectrophotometer (Thermo Mattson IR 300, USA) was used with potassium bromide disc; the scanning range was 500-4000 cm⁻¹ [18].

Stability study

The selected ODTs formula (formula F9c) was subjected to accelerated stability study for three months at 40 °C in a stability chamber (Newtronic, India) with 75%±5 relative humidity (RH). In addition stability study at 25 °C with 60±5 relative humidity (RH) for three months was conducted where samples were withdrawn every month and evaluated [19].

Statistical analysis

The software Excel 2013 using t-test and ANOVA was used for statistical analysis, the difference was considered to be statistically significant when P<0.05 and non-significant when P>0.05. Mean, and standard deviation was used to express the values in the data.

RESULTS AND DISCUSSION

Flowability study of the prepared powder mixtures

The results of the flowability study showed that those formulae that were prepared using mannitol as a diluent (F1, F2, and F3) had excellent flow properties. The angle of repose was in the range of (24.98±0.36 to 28.31±1.52), Carr’s index was less than 10%, and Hausner’s ratio was less than 1.1. This was due to the free-flowing nature of the granulated form of mannitol [Perlitol1500 DG] [20].

The formulae that were prepared with MCC diluent (F4, F5, and F6) showed flow properties ranged from good to fair. The angle of repose was in the range of (32.0±2.65 to 44.4±1.70), Carr’s index was in the range of (10.69±0.88% to 16.39±0.27%), and Hausner’s ratio was in the range of (11.1±0.015 to 11.95±0.005) because MCC is made up of crystalline powder that is porous in nature with a low bulk density which hinders its flowability [21]. The formulae (F7, F8, and F9) that were prepared with the diluent lactose (super Tab@40Ll) showed good to excellent flow properties, the angle of repose was in the range of (24.03±2.11 to 34.4±0.43), Carr’s index was in the range of (6.37±0.94% to 14.49±0.05%) and Hausner’s ratio was in the range of (1.06±0.007 to 1.16±0.021) due to the presence of lactitol monohydrate in its composition which enhanced its flow properties [22].

Evaluation of cinnarizine orally disintegrating tablets

The hardness of the tablets from all of the formulae was within an acceptable range (21.97±1.76 N to 31.55±4.40 N). The highest hardness value was observed for the formulae F4, F5, and F6 due to the strong inter-particle force between MCC particles [7]. In addition to the acceptable friability of less than 1%. Tablets thickness was in the range of (2.88±0.03 mm to 2.97±0.02 mm), except for the formulae F4, F5, and F6, in which higher thickness range was reported (3.46±0.02 mm to 3.59±0.07 mm), due to the low bulk density of MCC that was used as a diluent for them [21]. The weight variation was acceptable and within the pharmacopoeial limits, and the drug content was found to be good and not less than 90% or more than 110% which is the required limit [5].

Disintegration time

The disintegration time was in the range of (43.3±1.53s to 28.0±7.1s) as shown in table 2. The formulae F5 and F6 that were prepared with the diluent MCC showed the shortest disintegration time which may be due to the self-disintegration property of MCC (it disintegrates by capillary and swelling mechanism) which was complimentary with the disintegration properties of the super disintegrants that were used [23]. Lactose showed shorter disintegration time than mannitol which may be attributed to the reported spontaneous disintegration property of lactose [24].

Coming to the effect of the super disintegrant on disintegration time, CP revealed the shortest disintegration time among the other super disintegrants in the formulae that were prepared with mannitol (F1, F2, and F3), (ANOVA, *P<0.05) and lactose (F7, F8, and F9), (ANOVA, **P<0.01). The order of the disintegration time in such formulae was found to be CP>Kyron T-314>SSG. CP with its porous structure quickly absorbs water into the tablet by capillary action followed by secondary swelling which results in rapid disintegration of the tablet [25]. Kyron disintegrates evenly by wicking mechanism mainly, also by swelling on hydration [26]. SSG disintegrates by swelling, but unlike CP and Kyron T-314, it gets at a high level which may result in the formation of a thick viscous barrier that retards disintegration [27]. These results are in agreement with the results obtained by Swatantra et al. [2014], in a comparative study on the effect of various super disintegrants in the formulation of ibuprofen fast dissolving tablets [28].

Conversely, the formulae that were prepared with the diluent MCC; CP revealed the longest disintegration time in the formula F4 compared with that of F5 and F6. It may be due to the lower water absorption ratio of the formula F4 with this super disintegrant when compared with that of SSG and Kyron T-314 (F5 and F6) as shown in table 2.

Wetting time

Wetting time can be regarded as one of the vital parameters to predict how much of the fluid is enough for breaking the tablet when placed upon the tongue (fig. 1). All of the tested formulae were wetted within an acceptable time of less than one minute as illustrated in table 2, which indicated their ability to absorb water and swell. The results were in the range of 2.72±0.75s to 43.6±6.4s. The shortest wetting time was with the formulae that were prepared with MCC (F4, F5, and F6) with the wetting time of 10.03±3.01s, 4.62±0.68s, and 2.72±0.75s respectively, this could be due to its porous structure [21].

The type of the super disintegrant had a significant effect on wetting time in the formulae F4, F5, and F6 that were prepared with the diluent MCC. Like the disintegration time, the formula F4 that was prepared with the super disintegrant CP showed the longest wetting time (10.03±3.01s) compared with those prepared with SSG and Kyron-T314 (4.62±0.68s and 2.72±0.75s) respectively (ANOVA, *P<0.05).

Water absorption ratio

The highest water absorption ratio was observed for the formulae that were prepared with MCC, followed by those prepared with lactose then mannitol, as illustrated in table 2. The effect of the super disintegrant on water absorption ratio appeared when MCC was used as a diluent. The super disintegrant SSG significantly increased the water absorption ratio from 115.27±5.68% and 182.49±13.75% when CP and Kyron T-314 were used as super disintegrants to 232.39±8.62% which may lead to less patient compliance due to the long stay on the tongue (ANOVA, **P<0.01). These results are in agreement with the results obtained by Lakshmi et al. [2013] who made a comparative evaluation of natural and synthetic superdisintegrants in the formulation of fast dissolving tablets [29]. Moreover, when the concentration of Kyron T-314 increased from 5% in the formula F9 to 8% in formula F9c, the water absorption ratio increased from 58.52±0.51% to 81.45±7.04% (ANOVA, *P<0.05), while the disintegration and wetting time were not significantly affected.
Dissolution test

The ODTs of cinnarizine were subjected to the dissolution test using 0.1N HCl (pH=1.2) as a dissolution medium. The release data obtained through the dissolution study is shown in table 2. The release profiles in fig. 2, fig. 3, and fig. 4 demonstrated that among the different types of diluents, lactose revealed the highest percentage of the drug released that reached up to 61.46±4.60% in the formula F9 after 2 min. The reason may be related to the high water solubility of the type of lactose that was used in this work (SuperTab® 40L) when compared with the others. It was composed of anhydrous lactose and lactitol monohydrate, the dissolution rate of anhydrous β-lactose is 10 times more than that of α-lactose monohydrate [30], while MCC is practically insoluble in water [31].

In fact, the type of the super disintegrants had no significant effect on the percentage of the drug released in the formulae that were prepared with the diluent mannitol (F1, F2, and F3) (fig. 5). While the super disintegrant Kyron T-314 significantly increased the percentage of the drug released in the formula F6 that was prepared with the diluent MCC, (51.24±8.34% after 2 min) when compared with the other formulae with this type of diluent (F4 and F5), (fig. 6), (ANOVA, **P<0.01). Similarly, in the formulae that were prepared with the diluent lactose, the super disintegrant Kyron T-314 revealed the highest percentage of the drug released (61.46±4.60% after 2 min in the formula F9) among the formulae that were prepared with lactose (ANOVA, **P<0.01) and all other formulae (F1-F9), (fig. 7). Kyron T-314 increased the porosity of cinnarizine tablets due to which the absorption of water takes place at a higher rate that results in breaking of the tablets and therefore faster disintegration/dissolution [32]. These results are in agreement with the results obtained by Gattu and Lakshmi (2011), who made a comparative evaluation of natural and synthetic super-disintegrants with newer superdisintegrant Kyron-T-314 [33].

Table 2: Evaluation of cinnarizine ODTs

<table>
<thead>
<tr>
<th>Formula type</th>
<th>Disintegration time* (s)</th>
<th>Wetting time** (s)</th>
<th>Water Absorption ratio after 10 s (%)</th>
<th>Drug content*** (%)</th>
<th>Drug released after 2 min (%)</th>
<th>K1 (min−1)</th>
</tr>
</thead>
<tbody>
<tr>
<td>F1</td>
<td>8.33±0.57</td>
<td>21.23±7.1</td>
<td>31.81±2.04</td>
<td>92.43±0.38</td>
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<td>F2</td>
<td>28.02±7.1</td>
<td>43.64±6.41</td>
<td>15.95±0.169</td>
<td>90.42±0.30</td>
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<td>F3</td>
<td>16.10±3.60</td>
<td>34.67±7.70</td>
<td>20.74±1.41</td>
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<td>45.77±5.86</td>
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<td>F4</td>
<td>12.33±1.53</td>
<td>10.03±3.01</td>
<td>115.27±5.68</td>
<td>90.06±0.02</td>
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<td>F5</td>
<td>5.66±0.57</td>
<td>4.62±0.68</td>
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<td>F6</td>
<td>4.33±1.53</td>
<td>2.72±0.75</td>
<td>182.49±13.75</td>
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<td>F7</td>
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<td>F8</td>
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<td>F9a</td>
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<td>F9b</td>
<td>11.2±2</td>
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<td>F9c</td>
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<td>17.75±1.67</td>
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<td>93.15±0.51</td>
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<tr>
<td>Stugeron® 25 mg tablet</td>
<td>3124±32</td>
<td>-</td>
<td>-</td>
<td>92.29±1.40</td>
<td>8.75±2.15</td>
<td>-</td>
</tr>
</tbody>
</table>

K1: First order release constant, *Average of six reading±SD, **Average of three reading±SD

By increasing the percentage of Kyron-T314 from 5% in the formula, F9 to 8% in the formula F9c the percentage of the drug released after 2 min increased from 61.46±4.60 to 74.55±3.0 (fig. 8). (ANOVA, **P<0.01). These results were much higher than that of the commercially available cinnarizine tablets (Stugeron® 25 mg tablet), in which only 8.75±2.15% of the drug was released within 2 min. Results of the release kinetic models showed that all of the formulae followed the first order kinetics. The k1 value increased in all of the formulae with increasing of the percentage of the drug released (table 2). From the above results, formula F9c can be selected as the most appropriate formula that showed rapid disintegration (11.66±2.25 s) when compared with that of the marketed one (312±4.32 s), (t-test, **P<0.01), with the highest dissolution profile among the other formulae.

![Fig. 2: Percentage release of cinnarizine from the prepared ODTs with different types of diluents in the presence of the super disintegrant (crospovidone). n=6; Data are expressed as mean±SD](https://example.com/fig2.jpg)
The FTIR studies showed no detected chemical interaction between cinnarizine, Kyron T-314, and lactose. Fig. 9 shows that there were no noticed shift in peaks or extra peaks appeared when comparing IR of cinnarizine alone (fig. 9, a), lactose alone (fig. 9, b), Kyron T-314 alone (fig. 9, c) with the IR chart of the physical mixture of cinnarizine and lactose (fig. 9, d), cinnarizine and Kyron T-314 (fig. 9, e).
Stability study

The results of the stability study of the optimized formula (F9c) showed that there were no significant changes in hardness, friability, disintegration, drug release and drug content of the examined tablets through the stability study (table 3). The only exception is the noticed increase in weight that was observed for the tablets stored at 40 °C, 75%±5 RH which might be due to the hygroscopic nature of the anhydrous lactose that presents in the tested formula [34].

![FTIR spectra](image)

**Fig. 9:** FTIR spectra of cinnarizine alone (a), lactose alone (b), Kyron T-314 alone (c), cinnarizine and lactose (d), cinnarizine and Kyron T-314 (e)

**Table 3:** Stability study data of ODTs of cinnarizine

<table>
<thead>
<tr>
<th>Days</th>
<th>Stability conditions</th>
<th>Drug content* (%)</th>
<th>Disintegration time* (s)</th>
</tr>
</thead>
<tbody>
<tr>
<td>At time zero</td>
<td>-</td>
<td>93.15±0.51</td>
<td>11.66±2.25</td>
</tr>
<tr>
<td>30</td>
<td>25°C and 60±5 RH</td>
<td>93.23±0.06</td>
<td>11.03±1.34</td>
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<td></td>
<td>40°C and 75±5 RH</td>
<td>94.22±0.70</td>
<td>11.39±2.25</td>
</tr>
<tr>
<td>60</td>
<td>25°C and 60±5 RH</td>
<td>96.8±0.92</td>
<td>10.60±2.26</td>
</tr>
<tr>
<td></td>
<td>40°C and 75±5 RH</td>
<td>96.01±1.23</td>
<td>10.86±2.321</td>
</tr>
<tr>
<td>90</td>
<td>25°C and 60±5 RH</td>
<td>93.76±1.56</td>
<td>11.16±0.75</td>
</tr>
<tr>
<td></td>
<td>40°C and 75±5 RH</td>
<td>93.59±0.268</td>
<td>10.83±1.16</td>
</tr>
</tbody>
</table>

*Average of six reading±SD, **Average of three reading±SD
CONCLUSION
Owing to that cinnarizine is a class II drug according to the biopharmaceutical classification system (BCS), so any improvement in the dissolution will lead to enhancement of the bioavailability. It was successfully formulated as ODTs by using direct compression method. Among the various diluents used, lactose was selected. In addition, different super disintegrants were used, the super disintegrant Kyron T-314 was the most successful one when used with the diluent lactose. It played an important role in decreasing the disintegration time and enhancing the dissolution of the drug. The selected formula showed acceptable characteristics, rapid disintegration and dissolution, and good stability. Accordingly, the selected formula (F9c) represents a promising formula for preparing ODTs by using the simple, direct compression strategy.

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AUTHORS CONTRIBUTIONS
All the authors have contributed equally

CONFLICTS OF INTERESTS
Declared none

REFERENCES


