

## MI-SPE, M-SPE AND M-SPD RECENT APPLICATION ON SOLID PHASE EXTRACTION FOR COMPOUND EXTRACTION OF COMPLEX MATRICES

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### ABSTRACT

Solid Phase Extraction is a method used for extraction, purification of compounds contained in complex matrices and purification of samples in the fields of industry, pharmacy, and toxicology analysis. This review represents the recent advances of solid phase extraction from different mode of separation which are the most commonly used in recent days such as molecular imprinting solid phase extraction (MISPE), magnetic solid phase extraction (MSPE) and matrix solid phase dispersion (MSPD). In this review, the applications of MISPE in the analysis of agrochemicals such as herbicides, fungicides and insecticides are summarized. This review also states preparation of various magnetic composites (MSPE) based on graphene and its application as adsorbents in pre concentrating natural mixes, organic macromolecules. Besides this, this review summarizes the application of MSPD in biological, food, and environmental samples, including both organic and inorganic analyte.

**Keywords:** Solid phase extraction, Recent advances, Molecular imprinting solid phase extraction, Magnetic solid phase extraction, Matrix solid phase dispersion

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### INTRODUCTION

A method utilized for extraction, sanitization of compounds contained in complex matrices and purification of samples in the fields of industry, pharmacy, and toxicology analysis is called strong stage extraction. It is the most generally utilized analytical extraction strategies and can facilitate the analysis of a compound from biological materials such as blood, urine, water, etc. which contain many matrices. The problem that often arises is when there is a sample whose composition is unknown but is very complex and contains so many chemical components in the form of liquid and the presence of solid particles that expand inside. Therefore, we need a technique that can analyze the specific compounds contained in a sample [1].

Solid Phase Extraction (SPE) is a relatively new methodology but SPE is fast developing as the main tool for sample pre-treatment or for clean-up of dirty samples, for example, samples that have high matrix content such as salts, proteins, polymers, resins, etc. Filtration, homogenization, precipitation, chemical reactions, solvent exchange, concentration, matrix removal, solubilization are components that can be used singly or in combination to obtain samples with forms that are compatible with the analysis tools needed [1].

SPE has the advantage when compared to liquid-liquid extraction that is by using SPE the extraction process becomes more perfect, the separation of analyte from the matrix becomes more efficient, reducing the organic solvents used. SPE is an efficient separation process so that high recovery (>99%) is easier to achieve when compared to liquid-liquid extraction. In liquid-liquid extraction, several extractions are still needed to obtain high recovery, whereas with SPE only one-step is needed. Moreover, SPE has advantages and has been widely used because it can use various stationary phases, fast and simple procedures and high throughput. The main advantage of SPE is its easy use, fast time and generally only requires a small volume of extraction solvents [2].

To improve selectivity in sample analysis, the current SPE method are combined with other methods such as chromatography (GC-MS) UV-Vis Spectrophotometer, HPLC, Gas and Mass Chromatography. According to Barnes *et al.*, the combination of chromatography and SPE can be used more simply and effectively in purification, sample analysis. The advantages when compared to other methods such as HPLC, NMR, and MS are low costs, the ability and sample analysis are quite good [2].

### Methodology

Our review is based on the literature obtained from the PubMed and Scopus databases using the keywords "molecular imprinted solid phase extraction", "matrix solid phase dispersion" and "magnetic solid phase extraction". The specific keywords related to application of molecular imprinted solid phase extraction, matrix solid phase dispersion and magnetic solid phase extraction used were "molecular imprinted solid phase extraction in complex matrices," "magnetic solid phase dispersion in food," "molecular imprinted solid phase extraction in biological fluid," and "matrix solid phase dispersion in pharmaceuticals. The following exclusion criteria were applied: reviews, opinions or commentaries, non-English language, and unrelated topics such as other type of solid phase extraction beside molecular imprinted solid phase extraction, magnetic solid phase extraction and matrix solid phase dispersion.

### Different mode of separation

The following are the frequently utilized method of detachment in solid phase extractions in current days dependent on the articles, for example, molecular imprinted solid phase extraction (MI-SPE), Magnetic SPE (MSPE) and matrix solid phase dispersion. In the new advancements of test readiness procedures, molecularly imprinted polymers (MIPs) have picked up enthusiasm as a novel sort of sorbent with alluring properties. Such polymers may move toward becoming supplement to antibodies to be utilized in pesticide assurance [3, 4]. Magnetic solid phase extraction has been utilized as pretreatment procedure for the examination of a few mixes due to its points of interest when it is contrasted and great strategies. Magnetic solid phase extraction limits the utilization of additional steps, for example, precipitation, centrifugation, and filtration that diminishes the control of the example [5-77].

Uses of MSPD have secured natural, nourishment, and ecological examples, including both organic and inorganic analyte. Matrix solid phase dispersion (MSPD) has turned out to be a proficient example arrangement method for solid, semi-solid, and sticky examples [8].

### Molecular imprinting solid phase extraction

The molecular printing technique enables synthetic polymers to be prepared for a template molecule with specific binding sites. This can be achieved if the target acts as a molecular template during the polymerization process. Monomers with functional groups are

arranged by non-covalent or covalent interactions around the template. After copolymerization with cross-linking reagents, the functional groups are fixed in the polymer matrices in defined positions. After removal of the template by chemical cleavage or solvent extraction from the polymer networks, complementary binding cavities with size and shape are produced (fig. 1). Since MIPs contain binding sites "tailor-made," they have a high degree of selectivity and affinity for the template. These materials can be used in various fields, such as chemical analysis and detection [9, 10], separation and purification [11], drug delivery [12] and catalysis [13].

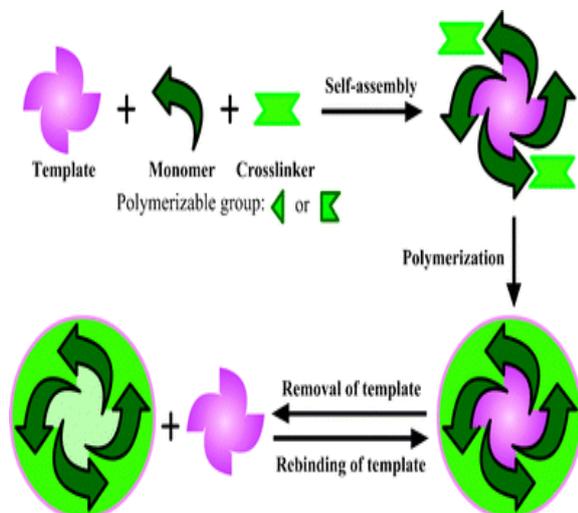


Fig. 1: Process of molecular imprinting [14]

Increasing the selectivity of SPE was done a great deal using the molecular imprinting polymer (MIP) technique, which combined this technique with the so-called molecular imprinting solid phase extraction (MI-SPE). Lately, molecularly imprinted polymers (MIPs) have made great influences in planned recognition capacity, stability relative simplicity and low readiness expenses, and potential application to a wide assortment of target molecules [15]. The utilization of MIP on SPE has a high advantage because it can selectively extract and eliminate the sample matrix in the analyte. MI-SPE has designs that follow natural mechanisms, such as specific antibody and biological receptor bonding mechanisms. MI-SPE is a versatile technique that enables biological and chemical molecules to be attracted [16].

#### a) MISPE application in agrochemicals

It has grown quickly in the region of agrochemical investigation recently. Nevertheless, just a single survey concerning the utilization of MISPE in agrochemical identification has been printed [17]. Latest analytical strategies require a high level of affectability to discover and measure agrochemicals. Some MIPs are combined through semi-covalent impressions. Silica embedded with MIP is additionally cutting-edge as a sorbent for SPE [18].

#### b) MISPE application in herbicides

The extraction and affirmation of triazine herbicides from various examples have gotten significant consideration in molecular imprinted solid phase extraction. The separation of triazines from vegetables and soils are made through precipitation polymerization for the preparation of MIPs in utilizing propazine [19]. MIPs were additionally arranged utilizing cyanazine as a format [20]. Their investigation demonstrated that a blend of photosynthetic-based biosensors and a pre-fixation strategy utilizing MIP permitted the location of photosynthesis-hindering herbicides in water at the dimension required by the European community [21], arranged for the utilization of new MIP-covered strong stage micro extraction

(SPME) strands and connected them to check triazines in convoluted examples. The outcomes demonstrated that the prometryne MIP-

covered strands had an extraordinary selectivity and liking to the seven triazines and that the affectability of the triazine assurance was clearly enhanced by the MIP-covered SPME-HPLC strategy [22], which incorporated a propazine engraved polymer utilizing acetonitrile or toluene as porogen. The chemical compound prepared to be broken down was the best introduction. At last, the MISPE procedure was utilized to clean water, soil and maize test separates and micellar electro kinetic chromatography (MEKC) was likewise used to decide the triazines [23]. Djozan *et al.*, in order to extricate triazine rice, onion and tap water, solid SPME fiber was created dependent on MIP. GC and GC-mass spectrometry (MS) inspected this herbicide.

Additionally, molecular imprinted solid phase extraction was utilized to separate phenyl urea and antidiabetic medicate herbicides[24], to plan two non-indistinguishable MIPs utilizing linuron or isoproturon as formats for precipitation polymerization. Detachment of phenyl urea herbicides in carrot, potato, maize and pea test are removed using MISPE strategy by utilizing HPLC-bright (UV) at 244 nm [25]. Cacho *et al.*, by utilizing fenuron or isoproturon as templates and acetonitrile or toluene as porogens, incorporated four distinctive MIPs dependent on methyl acrylic acid (MAA). The outcomes demonstrated that fenuron polymers and polymer were incredibly specific utilizing toluene as a porogen. MIPs were utilized in groundwater and soil tests as another material to separate phenyl urea [26]. Tamayo *et al.* blended and bundled MIPs in HPLC segments of stainless steel (125x 4, 6 mm).

Finally, the ideal engraved segment was utilized specifically for the LC-UV screening of phenyl urea herbicides from vegetable example extricates at a low fixation level in under 10 min [27]. Tang *et al.* utilized BSM as a layout particle to set up a precipitation polymerization MIP. The MISPE strategy has been enhanced for sulfonylurea extraction in soybean tests and a high recuperation has been accomplished. MIP recuperations for nicosulfuron (NS), metsulfuron-methyl (MSM), BSM and tribenuron-methyl (TBM) were likewise contrasted with C18 SPE cartridges; the outcomes demonstrated that MIP recuperation esteems were 29.34% for NS, 24.37% for MSM, 46.03% for BSM and 5.01% for TBM [28]. She *et al.* incorporated MIP for precipitation polymerization sulfonylurea herbicides utilizing chlosulfuron as the template particle. The MIPs have been packed into cartridges. The LC-Tandem Mass Spectrometry (MS-MS) dictated the examples after extraction. MISPE was likewise utilized for different kinds of herbicides. A procedure dependent on chemical measurements and quantum science was proposed by [29] for the structure and blend of engraved polymers of sham chloroacetamide. Metolachlor deschloro has been picked as the layout. MISPE indicated specific restricting properties for chloroacetamide herbicides contrasted with business SPE sections and the lattice impact was altogether reduced [30]. Baggiani *et al.* arranged polymers of bentazone utilizing non-covalent polymerization of atomic impressions. The MIPs were sectioned in HPLC. The sections indicated great recuperations (91-96%) and the convergence of 3.2-15.2 components.

#### c) MISPE application in pesticides

The utilization of MISPE for pesticide extraction and assurance has gotten a lot of consideration because of the wide circulation of pesticides in plants. Organophosphorus pesticides are generally utilized in agribusiness, as creepy insects and weeds can be controlled effectively to enhance the productivity and nature of the item. Regardless, their deposits may stay in the outcome of recuperation or capacity. The advancement of a touchy location system is critical to keep up these uncontrolled impacts on human prosperity and natural contamination.

Pereira *et al.* incorporated a MIP that could explicitly expel fenitrothion (FNT) from tomatoes. The clean-up and convergence of FNT before the HPLC examination was done with three separate MIP cartridges and a fortress measurement of 5 m g/g for three days. The MIP presented 59% medium extraction profitability. This MIP can be reused a few times after recuperation [31].

Another MIP course of action by utilizing a mix of trichlorphone and monocrotophos as a blended design. Another strategy for MISPE

joined with GC for the affirmation of organophosphates in vegetables was created utilizing this engraved polymer as the sorbent [32]. This was fused by a room temperature ionic liquid in the engraved polymer meddled with a mass polymerization technique utilizing dichlorvos as the configuration. The cartridge has been squeezed to pretreat vegetables with MIPs. This cartridge has been differentiated with the fragment of the C18 extraction [33]. The results from the chromatograms displayed that the engraved cartridge had a higher obsession affect and favored selectivity for dichlorvos over the C18 extraction area [34] Kang *et al.* organized a MIP by using diethyl(3-methylureido) (phenyl) methyl phosphonate (DEP) as a fake design by precipitation polymerization.

They found that explicit acknowledgment was basically dependent on the gathering of the investigator with the DEP-MIPs [35]. Xie *et al.* composed a precipitation polymerization MIP utilizing triazophos (TAP) as a design. In their work, a TAP-confirmation sensor for MIP-chemiluminescence (CL) was created utilizing TAP-graved particles as acknowledgment segments and the lumionl-H<sub>2</sub>O<sub>2</sub> CL reaction as the recognizable proof structure. The outcomes demonstrate that this procedure can be utilized for the online enhancement and area of residual organophosphate pesticides.

MISPE was additionally associated with different bug sprays, for example, pyrethroid and pirimicarb, for example [36]. Shi *et al.* utilized deltamethrin and cypermethrin as course of action particles (D-MIPs and C-MIPs) to orchestrate two sorts of MIPs. The outcomes demonstrated that the C-MIP-subordinate MISPE methodology indicated higher extraction recuperations and the proposed MISPE strategy joined with the GC-electron get identifier (ECD) was utilized viably to guarantee pyrethroid in ocean water. Differentiated and the customary Florisil SPE, their MISPE demonstrated more prominent productivity of separation and less utilization of common dissolvable [37]. Zhou *et al.* consolidated a sub-atomic stone landmark utilizing in situ polymerization utilizing pirimicarb as a format, MAA as a monomer, ethylene glycol dimethyl acrylate as a cross-linker and a toluene-dodecanol mix as a porogenic dissolvable. Another system for the sensitive, explicit affirmation of pirimicarb in tomato and pear has been created by solidifying littler scale extraction of MIP-polymer stone landmark with HPLC-photodiode display (PDA). They likewise analyzed the execution of MISPE and C18. The higher cleanup proficiency utilizing MISPE was appeared in the chromatogram and the MIP stone landmark could typically be reused without debilitating.

#### d) MISPE application in drugs

In recent decades, drugs classes such as antibiotics and beta blockers are widely used in the application of MISPE. Macrolide (MAL) is an antibiotics used in the water pollution. As a durable and highly specific SPE material, the molecularly impressed polymer (MIP) can overcome the aqueous matrix's influence to enrich ultra-trace amounts of target compounds. It is possible to use MISPE to separate and enrich numerous pharmaceuticals from environmental water, such as herbicides [38], phenolic compounds [39], pesticides [40] and hormones [41]. It is more practical to prepare an MIP that can selectively enrich multiple structural analogs due to the multiple antibiotics in water samples. MALs should be monitored in the long run as one group of the largest consumption of antibiotics in animal husbandry. A novel LC-MS/MS method was developed based on the MISPE to analyze the macrolide drugs in different water sources. The MISPE cartridge shows higher recognition capability towards macrolides compared to traditional SPE cartridges and can be recycled more than 20 times. The application of the method developed in real water samples has demonstrated its feasibility and practicality. Besides this, atenolol as a class of beta blockers is not only being used in the treatment of hypertension, cardiac arrhythmias, myocardial infarction and angina pectoris but also as doping for athletes due to its effectiveness in heart rate, anxiety and tremors [42]. MIP-SPE sorbents's ability to recognize analyte targets is influenced by the nature of the template, monomers and reactions in them [43]. Previous studies have been conducted using methacrylate acid (MAA) as a functional monomer to synthesize sorbent MIP-SPE atenolol [42]. In this study, MI-SPE atenolol sorbent was made from itaconic acid as a functional monomer with methanol as a porogen by bulk polymerization method so that after

its adsorption characteristics are identified it can be used as an alternative method of selective sample preparation of atenolol from the blood sample.

#### Magnetic solid phase extraction

A method that has been considered significantly in the examination of compounds in feeling tests are called magnetic solid phase extraction (MSPE). MSPE depends on the extraction of various compound from the example utilizing solids with attractive properties. This procedure can be envisioned as an attractive partition regularly used to isolate attractive stages from nonmagnetic stages [44]. The straightforwardness of this strategy affects the advancement and utilization of partition methods including the utilization of attractive fields.

This procedure comprises in the amalgamation of particles with attractive properties (for the most part magnetite Fe<sub>3</sub>O<sub>4</sub>) trailed by a covering of the attractive stage with various natural mixes. The changed solids are connected as adsorbents amid detachment, division, and pre grouping of the analyte [45-47]. The MSPE depends on the scattering of an attractive adsorbent in arrangement. The attractive adsorbents with the analyte adsorbed superficially can be disconnected and eluted with the expansion of fitting solvents. The methodology does not require extra advances, for example, centrifugation, precipitation, or filtration of the example.

MSPE method has been utilized for these perceived advantages, for example, effortlessness, productivity, cost, and high selectivity. This method utilizes the utilization of outer attractive field in the segregation of the attractive particles in the detachment and examination of various kinds of mixes with the fundamental target of discrete analyte of extensive volume without extra advances that could cause loss of analyte [48-50], since its disclosure has gotten significant consideration in the improvement of a few applications in the division of assorted analyte.

The blend by co precipitation is the strategy, which is increasingly connected to the combination of the attractive stage. Attractive solids can be covered with antibodies, explicit receptors, embodying of attractive particles with organic operators, polymers, and silica adjusted with various practical gatherings [51-53]. Extra impacts of their physical and compound attributes must be considered amid the plan of MSPE based philosophy.

In the nourishment territory MSPE has been connected to separation and specific pre centralization of proteins [54], metal particles [55-61] and natural mixes, for example, anti-infection agents, pesticides, colors, phenolic mixes, and advertisers developing that are available in a few food stuff tests [49, 62-57].

#### Detection of organic pollutants

##### Pesticides

In the assurance of agriculture harvests against weeds, pests and infections, pesticides are probably the most vital instrument. Anyhow, the application of pesticides is extensive to the point that pesticide buildups can be often times found in consumer goods such as plants or creature beginning, natural compartments such as ecology, air, bodies of water, and soil. Organophosphorus pesticides, organochlorine pesticides, pyrethroid pesticides, carbamate pesticides, neonicotinoid bug sprays, benzoyleurea bug sprays, triazole fungicides, imide fungicides and triazine herbicides are frequently used pesticides. The majority of pesticides have a place with ring structures based on carbon. In addition, G-based attractive composites have a comprehensive application as adsorbent to MSPE for the pesticide assurance buildups and it had been constantly created to acquire progressively viable, natural neighborly and specific investigative techniques. The most well-known adsorbent for their readiness techniques is MGC as it is faster, basic and more practical than those of different materials [68] are.

##### Drugs

Medication deposits investigation is made out of the examination of human and veterinary medications. Human and veterinary medications were done with buildup examination. Example of human

medications are anti-microbial, pain relieving, mitigating, antidepressants, and antipsychotics, which includes two perspectives. Firstly, the outlandish outflows of healing facilities. Secondly, industrial facilities, and the medications staying in the human bodies, metabolites, and fecal matter. Whereas, example of animal medications are sulfonamides, antibiotic medication, and quinolones. Centers around the medications stays in the creatures' bodies, metabolite, and fertilizer. Earth, plasma, blood, and pee was given the low focus dimensions of medications, sufficient example pretreatment strategies for the evaluation of medications is basic. In the medication buildup investigation, G-form attractive composites were broadly utilized. Adding on, the attractive G/GO composites have incorporated different distinctive functionalities to accomplish higher extraction efficiencies and reduced LODs. By utilizing porphyrin-functionalized nanocomposite Fe3O4-graphene oxide (TCPP/Fe3O4-GO) as an adsorbent for sulfonamide (SA) detection in ecological water tests, there is rapid progress and productive pre-focus approach. It has been shown that Fe3O4-GO, TCPP/Fe3O4-GO showed a higher adsorption limit through the gusto-stacking connections and electrostatic connections between the decidedly charged SAs and the adversely charged porphyrin Fe3O4-GO functionalized [69].

#### a) Detection of biological macromolecules

Peptides [71] and proteins [70], a biological macromolecules associated with the diagnosis of disease have always received a wide and current interest. Due to its unique advantages, the MSPE system has as of late been broadly utilized in the seclusion and preconcentration of organic macromolecules.

Phosphorylation is a standout amongst the most widely recognized structures post-translational types of protein alteration, and it can direct cell exercises, for example, cell proliferation, cell regulation, transformation, metabolism, differentiation, signal transduction and cell regulation. The direction of organic pathways in cancer cells is critical for protein phosphorylation. It is in this way fundamental to build up an examination for the identification of phosphoprotein. Lu and associates built up the MSPE-based Fe3O4/graphene/TiO2 strategy for the extraction and advancement of phosphopeptides from organic examples. Changing the TiO2 graphene with high particular adsorption is one of the advancement from the article [72].

Yin *et al.* stated that an easy method of preparation for the twofold sided mesoporous nanocomposites attractive graphene (mag graphene@mSiO2) and after that utilized the mag graphene@mSiO2 to selectively enrich and analyze endogenous peptides [73]. It has a large surface territory and an extensive pore volume. The hydrophobic inward pore dividers could productively adsorb target particles through hydrophobic-hydrophobic communication. The Fe3O4 particles on the two sides of the materials could then disentangling the advancement procedure, which was fundamental for the treatment of complex natural examples.

New hydrophilic polydopamine-covered attractive graphene nanocomposites (MG@PDA) have been incorporated and utilized for the exceptionally effective tryptic immobilization [74]. The MG@PDA nanocomposites also made the nanocomposites Able to immobilize high amounts of trypsin yet in addition great hydrophilicity of PDA, which enormously enhanced biocompatibility of nanocomposites. The work manufactured a solid establishment for the high-throughput proteome research. Presently, the Sun group built up a powerful partition strategy to distinguish low endogenous phosphorylated peptides [75].

b) In this investigation, adsorbents chose were size-exclusive magnetic graphene/mesoporous silica composites with titanium (IV)-removed pore walls (alluded to as Ti4empe-MGMS). The presentation of hydrophilic polydopamine abbreviated the engineered procedure of Ti45-0-MGMSs and the arranged mesoporous channels supported the particular catching of endogenous phosphopeptides while precluding vast proteins.

#### Detection of inorganic pollutants

Heavy metal pollution is one of the world's most important environmental problems today. Even at extremely low concentrations, it is highly toxic and can hoard in living things,

triggering a series of serious illnesses and disorders. The advancement of new adsorbents with a phenomenal adsorption limit is a standout amongst the most essential answers for the issue. G-based MSPE is becoming more and more popular in heavy metal isolation and enrichments such as Cd (II) [76], Zn (II) [77], As (III) and As (V) [78], Hg (II) [79], and Cr (VI) [80]. Cation connections, electrostatic collaborations, hydrogen bonds and dative bonds in the G-based attractive composite can consume metal particles. The pH of test arrangements affects the adsorbent's adsorption proficiency because the zeta capability of the attractive G/GO composite changes with the pH test. Zhou and partners demonstrated that an aqueous self-get together methodology could be utilized to combine the diminished graphene oxide/Fe3O4 composites and to quickly remove Cr (VI) from the composites [81].

It also showed rapid Cr (VI) evacuation with a high adsorption limit in corrosive and nonpartisan arrangements, while the expulsion productivity in soluble media diminished forcefully. This wonder was ascribed to the adjustment in control type and thickness of the adsorbent just as the current type of Cr(VI) with the example pH. Adsorption was also non-selective even if both Fe3O4 and RGO were able to adsorb Cr (VI). The composite Triethylenetetramine-magnetite reduced graphene oxide (TETMRGO) has been synthesized and used for Cu(II) removal to improve selectivity [82]. The TET-MRGO composite demonstrated a high selectivity and adsorption limit (209.1 mg g<sup>-1</sup>) towards Cu (II) particles to the benefit of the high liking between nitrogen molecules and Cu (II) ions [83].

Zero valent iron is a strong adsorbent with a specific limit of adsorption. By altering with zero valent iron, it is conceivable to upgrade the adsorption limit of attractive G. An attractive G, for instance nanoscale zero valent iron/G (0FG) composite, was along these lines arranged, portrayed and used to remove Co(II) from aqueous solution [84]. Zero valent iron showed a high adsorption capacity (134.27 mg g<sup>-1</sup>) towards Co (II) ions at 30 °C. Graphene nanocomposites (CoFe2O4-G and NiFe2O4-G) together with magnetic cobalt and nickel ferrites (CoFe2O4 and NiFe2O4) were blended through a solvothermal procedure and used to expel lead (Pb (II)) and cadmium (Cd (II)) particles from aqueous arrangement. [85].

The outcomes demonstrated that material sorts affected the adsorption limit. Qi *et al.* arranged three sorts of magnetite/decreased graphene oxide (MRGO) nanocomposites to find out if the preparation methods affect the capacity of adsorption examples are such as, solvothermal, hydrothermal and co-precipitation methods [86]. The results showed difference in surface functional groups, crystal structure, particle sizes, surface morphology and surface charge. These nanocomposites showed different performance in the adsorption of drugs, colors also, metal particles. The MRGO arranged by the co-precipitation steps demonstrated an extraordinary capacity to adsorb non-positive particles. Notwithstanding, those integrated by the solvothermal steps gave the others the best extraction limit and reusability and offered a decent prospect of attractive strong stage extraction. Hence, utilizing the correct readiness technique before application is profoundly prescribed to achieve the best extraction execution. In this way, it is profoundly suggested to use the right preparation method before application to achieve the best extraction performance.

#### c) MSPE applications of graphitic carbon

G-C3N4 is the most current 2D carbon material that is presently accepting extraordinary consideration in many research regions. From the explanatory outlook, due to its similitude to GN, great similarity with watery stage and expectable extraordinary fondness for a wide scope of mixes make g-C3N4 an exceptionally encouraging sorbent [87]. The simplicity of planning and minimal effort of the forerunners are without a doubt extra focal points appeared by this carbon material, that without a doubt can be procured by thermally impelled mass development or pyrolysis of melamine or urea [87-90, 91-92] likewise with no requirement for added substances such natural solvents [90]. It is clear that this foundation represents the extraordinary probability of g-C3N4 as sorbent for condition benevolent SPE [87].

The primary experiment for pre-fixation intentions are quite recent. These give an account of the use of g-C3N4 as SPME and SPE sorbents

[89], [90] and unquestionably are making ready for application in MSPE. Surely, concerning GN, the g-C<sub>3</sub>N<sub>4</sub> sheets can be advantageously beautified with attractive particles, along these lines anticipating re-conglomeration and empowering quick detachment from the arrangement [91]. The work by [89] examined g-C<sub>3</sub>N<sub>4</sub>-SPME for extraction of test blends of different structure and furthest point, for example amphetamine, nerolidol, ametryn, deltamethrin, dodecane, and acryl-amide. It was uncovered that g-C<sub>3</sub>N<sub>4</sub> was doing greater as far as extraction capacity plus strength contrasted with business filaments because of its extraordinary properties, predominantly free structure and multi type communication, combining hydrophobic impact, p-p stacking, hydrogen holding and dipole-dipole. Now g-C<sub>3</sub>N<sub>4</sub> has been tried likewise in the in-clump SPE mode to decide benzoylurea pesticides in juices [90]. The things were arranged by buildup of urea that yielded a nano-vessel shape like the human veins, prepared to do great extraction productivity and reusability.

Wang et al. [91] arranged an attractive g-C<sub>3</sub>N<sub>4</sub> nanocomposite for pre-grouping of phthalate esters from water, explicitly di-n-butyl phthalate, butyl phthalate, dihexyl phthalate, and di-(2-ethyl hexyl) phthalate. The attractive things arranged by precipitation technique in nearness of melamine-determined g-C<sub>3</sub>N<sub>4</sub>, showed a sheet shaped with Fe<sub>3</sub>O<sub>4</sub> nanoparticles on the g-C<sub>3</sub>N<sub>4</sub> base. Their hydrophobe and H bond held the analyte. Quantitative extraction (up to 99%) was gotten in ten minutes, also the sorbent demonstrated to keep up its execution for successive MSPEs.

A similar combination was as well revealed for the MSPE of phenolic acids (three, 4-dihydroxybenzoic acid, 3, 4-dihydroxycinnamic acid, (2E)-3-(4-hydroxy-3-methoxyphenyl) prop-2-enoic corrosive) is removed by watery plant [88]. It explained that the sorption limit was not just administered by hydrophobe, but rather likewise by p-p stacking, hydrogen holding and electrostatic collaborations. The analyte was measured separated from the example and reusability was exhibited for five revolutions, focusing on the probability of g-C<sub>3</sub>N<sub>4</sub> likewise in lattices more not able to be predicted than water, and in concurrence continuously [92] concentrating on MSPE of PAHs from eatable fat tests.

Nevertheless, of the blended mode conduct of g-C<sub>3</sub>N<sub>4</sub>, its selectivity can be tuned by synthetic change of the immaculate structure. For instance, carbon self-doped g-C<sub>3</sub>N<sub>4</sub> can be effectively acquired by using melamine pre-treated with ethanol as the forerunner and, in the meantime, oneself doping process yields a permeable object [93]. Development of delocalized p bonds that would expand the fondness for fragrant atoms by the substitution of spanning N particles with C iotas. This can be extremely supportive in connection with the science of the analyte to be pre-concentrated, as in fact was accounted for by [91] for MSPE of five BFR (polybrominated phenyl ethers). The last appealing composite contained ca. 30 wt.% of doped g-C<sub>3</sub>N<sub>4</sub> interfaced with Fe<sub>3</sub>O<sub>4</sub> particles (surface zone 100.7 m<sup>2</sup> g<sup>-1</sup>) that for this circumstance accept similarly an important activity in favoring the dispersibility of the carbon-material in the liquid game plan (C-doped g-C<sub>3</sub>N<sub>4</sub> is is point of fact more hydrophobic than unadulterated g-C<sub>3</sub>N<sub>4</sub>). Mix and extraction are schematically defined in fig. 2.

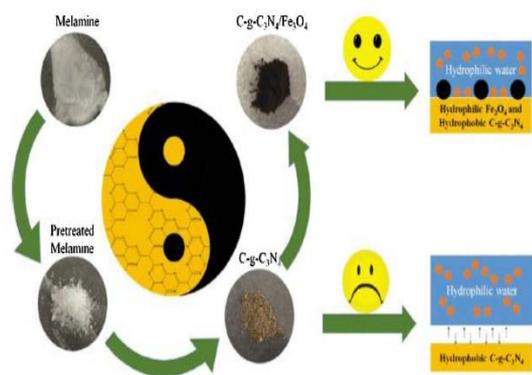


Fig. 2: Synthesis scheme of Fe<sub>3</sub>O<sub>4</sub>/C-g-C<sub>3</sub>N<sub>4</sub> and extraction mechanism proposed for brominated [91]

In divergence to the pure material [88], C~doped g-C<sub>3</sub>N<sub>4</sub> showed an extended sorption limit as per mixes hydrophobicity, featuring the job of selectivity tuning instigated by C-doping. The material was attempted in environmental waters, for example, tap and lake waters surrendering recuperation to 100%, and was reused for adsorption/desorption cycles with unaltered extraction capacity.

From this, obviously novel performing materials basing their sorption properties to g-C<sub>3</sub>N<sub>4</sub> are at present under scrutiny for various example arrangement systems (SPME, SPE, and MSPE). The starter reports show that the high profitability of this new carbon material could impressively augment the utilization of MSPE, in watery as well as in organic grids. Therefore, countless reports are normal sooner rather than later.

Graphene oxide (GO), the oxidized graphene derivative, is accessible by natural graphite powder oxidation [94]. On its nanosheet surface, GO contains abundant reactive oxygen functional groups in the form of epoxy, hydroxyl, and carboxyl groups, making it extremely hydrophilic and highly dispersive [95, 96]. This makes it difficult for graphene oxide to separate from aqueous solution directly. Preparing magnetic GO composite is a superior choice to overcome these problems, which can ensure convenient magnetic separation after adsorption.

The application based on magnetic GO in the preparation of samples to adsorb trace analytes has also received increased attention over the past few years [97, 98-100]. In this study, we synthesized magnetic Fe<sub>3</sub>O<sub>4</sub> graphene oxide (Fe<sub>3</sub>O<sub>4</sub>/GO) as an adsorbent for the extraction, pre-concentration and determination of three commonly used chemicals (bisphenol A, naproxen, and triclosan) frequently detected in wastewater. Removal of toxic heavy metals from wastewater achieved primarily through multiple processes such as cloud point extraction, precipitation/co-precipitation, extraction of liquid and solid-phase extraction (SPE) [101].

There is also a study of (MSPE-DLLME) which has been developed for the strong desire/formal decision about something of biphenyl and biphenyl oxide in water samples. In this dispersive liquid-liquid microextraction (DLLME) technique, associate applicable mixture of extraction and disperser solvents is employed. This technique has attracted lots of attention because of its blessings, like short extraction time, low consumption of organic solvent, and ease. In this study, the surface of Fe<sub>3</sub>O<sub>4</sub> nanoparticles was coated with polypyrrole as a conductive polymer. Using magnetic solid-phase extraction (MSPE), the analytes were extracted from the water samples followed by dispersive liquid-liquid microextraction. The effect of major factors were studied, including sorbent quantity, extraction time, type of elution solvent, volume of elution solvent, type of extraction solvent, volume of extraction solvent, and desorption time [102-104].

**Matrix solid phase dispersion (MSPD)**

Matrix solid phase dispersion (MSPD), first presented by [105] gives an elective way to deal with decrease dissolvable use and examination time for planning strong, semi-strong, and gooey examples [106]. In a run of the mill MSPD methodology, tests are mixed with sorbent to get homogeneous blend. The subsequent blend is moved and pressed into an extraction segment. At that point dissolvable is gone through the section to complete washing and elution venture for the extraction and seclusion of analyte from the framework. An extra co-sorbent could be stacked at the base of the section to additionally tidy up the eluent for some situations largely; the last concentrate can be breaking down by chromatography based systematic strategies. The traditional dissolvable extraction strategy, MSPD kills ventures of rehashed centrifugation and additionally filtration, and techniques of re-extraction during comparison.

Solid phase extraction (SPE) is different, in which isolated dissolvable extraction strategy is required to make strong examples reasonable for stacking into a SPE section, MSPD kills the dissolvable extraction step. The utilization of dissolvable and accordingly the required control time for the arrangement would significantly downsize it. The patterns and improvements of MSPD [106, 107, 108-113] has serious audits identifying it. In this research, we center consideration on the most recent improvements in MSPD sorbent,

scaling down of MSPD, on-line/in-line systems, and the adjustment of unique MSPD technique. Literary works amid 2015 and June 2018 are researched to keep away from the cover with later great audits [112, 113].

Shortly, MSPD extraction (MSPDE) was performed by mixing ox-like tissues with a proper measure of lipophilic strong stage pressing material (C18) in a glass mortar with a glass pestle, to create a semi-dry, evidently homogeneous, simple to deal with material [105] in the Barker's fundamental work. At that point, the got powder was stuffed into a typical syringe barrel SPE section, from which target mixes were segregated with an appropriate elution dissolvable. Truth be told, the mixing of test with a rough strong help permitted the mechanical interruption of the grid structure, as affirmed by checking electron microscopy perceptions [108]. The fundamental paces of the procedure is effectively depicted profoundly somewhere else [105, 107, 108, 114] are accounted for in fig. 3.

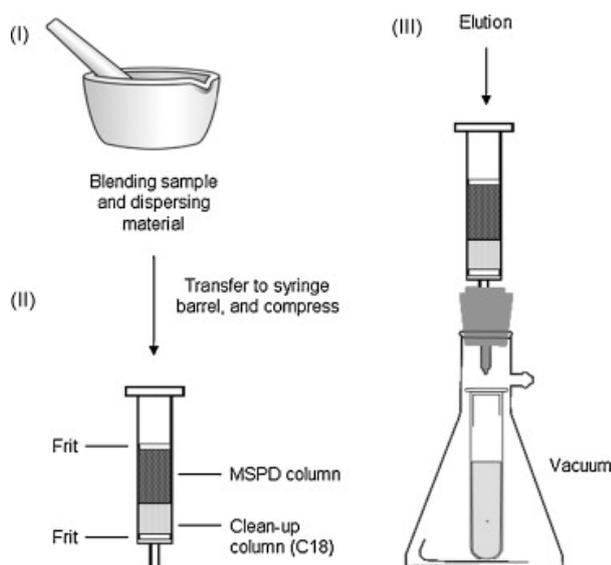


Fig. 3: Procedure of matrix solid phase dispersion

#### Recent advances in MSPD sorbent

Molecularly imprinted polymers (MIPs), the manufactured sorbents that show particular official of target sub-atomic, have been broadly used for the extraction of explicit mixes [115, 116]. Sorbent requires to be mixed with test to get homogeneous blend in MSPD. Engraved polymers can be integrated utilizing different sorbents as bearer to enhance mechanical quality of MIPs materials. For instance, MIPs were set up on carbon nanotubes (CNTs) for the MSPD readiness of malachite green in amphibian items [117]. Silica gel [118], silica nanoparticles [119], and mesoporous silica [120] equally have been accounted for as the bearer of MIPs to improve the selectivity of MSPD sorbents. In addition, Wang *et al.* explained the combination of blended layout MIPs for the extraction of multi-class veterinary medications [121]. This impressive MIPs sorbent was used for the concurrent MSPD extraction of 20 sedates in meat, including 8 fluoroquinolones, 8 sulfonamides and 4 antibiotic medications.

Graphene is one of the carbon-based nanomaterials, which demonstrates extraordinary guarantee in test readiness [122, 123]. Graphene gives substantial surface region and Nano sheet morphology for enhancing adsorption limit. Likewise, the delocalized  $\pi$  electron framework in graphene could make it shape solid  $\pi$ -stacking collaboration with mixes containing fragrant rings. These properties make graphene a decent possibility for the adsorption of benzenoid mixes. Sun *et al.* discovered that a graphene-embodied silica sorbent for the investigation of flavonoids in the leaves of *Murraya paniculata* (L.) Jack [124]. Immobilized on the surface of silica gel kept away from the collection and kept up the huge surface territory and  $\pi$ -electron rich structure graphene

amid the mechanical mixing. Contrasted and five sorbents (graphene, silica gel, C18, diatomaceous earth, and unbiased alumina), graphene-typified silica demonstrated a better extraction efficiency for the objective flavonoid mixes.

Tan *et al.* planned Phenyltrichlorosilane for the extraction of polycyclic aromatic hydrocarbons (PAHs) in soils. [125]. this material adventures the high liking between magnesium oxide and PAHs to upgrade the maintenance of target atoms. Joining the microspheres with phenyltrichlorosilane reduced the focused adsorption of chlorine-contained obstructions, which are broadly exist in soil tests. Using hexane and DCM as flushing and eluting dissolvable, separately, seven PAHs were effectively decided in HPLC-FLD with breaking points of identification (LODs) of 0.02–0.12  $\mu\text{g}/\text{kg}$ .

Wang *et al.* mentioned that the usage of polyethyleneimine (PEI)-changed attapulgite material as MSPD sorbent was considered for the guarantee of cadmium in fish items [126]. Presenting of PEI, which is a cationic polymer with high partiality to cadmium particle, brought about the high recovery of target particle in complex networks. High grouping of  $\text{HNO}_3$  (half, v/v) was compulsory to discharge the cadmium. Controlled by nuclear ingestion spectrometry (AAS), the LOD of cadmium in fish test was observed to be 2.5  $\mu\text{g}/\text{kg}$ . Moreover, sorbents, for example, mussel shell [127, 128], atomic strainer [129, 130], microcrystalline cellulose [131], and metal-natural system materials [132, 133] additionally have been considered.

#### CONCLUSION

In conclusion, SPE is a well-established technique and, its benefits over alternative sample preparation techniques, has been widely used for the analysis of diverse very different categories of compounds during a style of matrices. Moreover, SPE is essentially used for determination of natural pollutants in binary compound

Ecological examples because of an outsized sample volume may be utilized. In Addition, SPE is also used extensively in medical specialty and pharmaceutical analysis and within the analysis of food and beverages. The introduction of latest solid phases and additional selective activity modes, the development of latest experimental configurations to adapt SPE to specific things and therefore the improvement of automatic devices can lead to their application in numerous fields of qualitative analysis.

#### AUTHORS CONTRIBUTIONS

All the author have contributed equally

#### CONFLICT OF INTERESTS

Declared none

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