Objective: This study aimed to characterize physicochemical and chemical characteristics of Chaba maple (Hibiscus acetosella) homemade jam (CHJ) and determine its antioxidation ability.

Methods: The physicochemical and chemical characteristics of CHJ were investigated. The color, viscosity, and pH were observed as physicochemical data while chemical properties were obtained from sugar content and total polyphenol content (TPC), determined using high-performance liquid chromatography refractometer and Folin–Ciochaltu assay, respectively. The antioxidant activities of CHJ were identified using 1,1-diphenyl-2-picrylhydrazyl (DPPH) radical scavenging, ferric reducing antioxidant power (FRAP), and nitric oxide (NO) radical scavenging ability methods.

Results: The color and viscosity of CHJ were purple-red and 34,483.33±152.75 cP, respectively. The pH was at 3.78. The total sugar was not detected in CHJ. The TPC of CHJ showed the highest (47.18±1.80 mg gallic acid equivalent [GAE]/g of jam) followed by Streamline (SL) (23.66±0.32 mg GAE/g of jam), Doikham (DK) (21.99±0.50 mg GAE/g of jam), and Best food (BF) (9.75±0.38 mg GAE/g of jam), respectively. Antioxidant activities of CHJ with %1,1-diphenyl-2-picrylhydrazyl (DPPH) radical scavenging of 100.00±1.39% and FRAP value of 1690.70±8.26 uM. Both of activities exhibited the highest activity and significantly different when compared with other jams. The %NO scavenging activity of CHJ and SL was 72.43±1.93% and 73.82±1.66%, respectively, which higher than DK and BF.

Conclusion: This study shows good in both physicochemical and chemical characteristics of CHJ. The CHJ presents the highest TPC as well as antioxidant activities. Thus, a homemade jam of Chaba maple may be considered as a good source of antioxidants and functional foods.

Keywords: Chaba maple (Hibiscus acetosella), Homemade jam, Antioxidation, Total polyphenol content.

INTRODUCTION

Jam is one of the most popular products in the market made from several fruits such as apricot, pineapple, grape, strawberry, blueberry, cranberries, and blackcurrant for preserving the food. There are many parts of plants which have been used to prepare a jam and the different part of plant can provide a varied bioactive compounds, phenolics, flavonoids, and anthocyanins [1]. Typically, jam is composed of at least 40% of fruit mixed with sugar and gelling agent, using by thermal processing [2]. Low sugar and low-calorie fruit jams have increased in the market for protecting obesity and diabetes [3]. Therefore, sweeteners as sorbitol, xylitol, steviol glycosides, and erythritol have been replaced.

Chaba maple (Hibiscus acetosella) is the plant found in Pathum Thani, Sanburi, Nonthaburi, and Nakhon Nayok, Thailand. Chaba maple is in Malvaceae family and well known in African msem allow, false roselle, maroon mallow, cranberry hibiscus, or red-leaved hibiscus [4]. This plant is a perennial subshrub and ornamental plant as well as fresh food. All parts of this plant are purple-red color. In previous studied, flowers and leaves of Hibiscus species and Chaba maple showed biological activities including antioxidant, antityrosinase, and antibacterial activities [5-9]. All of these activities were obtained from phenolic compounds such as anthocyanins (i.e., cyanidin, delphinidin, and malvidin), flavonols (quercetin, kaempferol, and myricetin) in flowers, and caffeoyl-hydroxycitric acid and neochlorogenic acid in leaves. These compounds show a wide range of antioxidant activities which prevent to degenerate of neuronal disorders, cardiovascular disease, cancer, and diabetes.

In this study, the attractive purple-red color of Chaba maple parts was made to jam. To ensure the quality as antioxidation of Chaba maple jam, the total polyphenol content (TPC) and antioxidant activities of Chaba maple homemade jam (CHJ) were evaluated. Moreover, these results were compared with three commercial products.

METHODS

Plant material

The fresh flowers and leaves of Chaba maple were collected from Pathum Thani, Thailand, on September 2018. The petals and leaves were used for CHJ preparation.

Chemicals

Folin–Ciochaltu’s phenol reagent, gallic acid, L-ascorbic acid, aluminum chloride, DPPH, and sodium nitroprusside (SNP) were purchased from Sigma Chemical Co. (St. Louis, MO, USA). Griess Reagent Kit was obtained from Promega Corporation (Promega, State, USA). All of chemicals in the study were analytical grade. The available jams in the market were purchased from Tesco Lotus supermarket (Pathum Thani, Thailand) including, Best food (BF), Doikham (DK), and Streamline (SL) jam.

CHJ preparation

The jam formula was 44.44% of petals, 35.56% of leaf juice (fresh leaveswater, 1:4), 17.78% of erythritol, 1.78% of lemon juice, and 0.44% of salt. First, the leaves were boiled for 15 min and filtered. The filtrated solution was then used and heated to 70–80°C after that the petals were added and allowed to boil for 10 min. The lemon juice,
Characterization of CHJ

Color

The color of CHJ was evaluated by organoleptic and instrument. For the instrument, the measurement based on CIE \( \text{L}^*\text{a}^*\text{b}^* \) system of the color parameter using spectrophotometer (ColorQuest XE, HunterLab, USA). The \( \text{L}^* \) (lightness), \( \text{a}^* \) (greenness [−] to redness [+]), and \( \text{b}^* \) (blueness [−] to yellowness [+]) were measured.

Viscosity

The viscosity of CHJ was measured using a Brookfield Viscometer (DV2T, Brookfield Engineering Laboratories, USA) at 25°C. Viscometer was adjusted to zero and the spindle LV2 was set in the instrument.

pH measurement

The sample of CHJ was blended with deionized water (1:9 jam:water, w/w) for 1 min using vortex mixer (Wisemix®, Korea) and then was filtrated before measuring. The pH of filtered solution was determined using a pH meter (SP-2100, Suntex, Taiwan).

Total sugar

The total sugars were analyzed using high-performance liquid chromatography (HPLC) with refractive index detector. One gram of CHJ was dissolved in 25 ml acetonitrile:water (50:50 v:v). This solution was then centrifuged for 10 minutes at 8000 rpm. The supernatant was filtered using 0.45 nylon filter. Separation was carried out on an amino-bonded column with a mobile phase of acetonitrile:water:triethylamine (75:25:0.2) and sugar content were determined by Refractive Index Detector against the standard solution (0.1, 0.2, 0.5, 0.8 and 1.0 ppm of fructose, glucose, sucrose, maltose, and lactose). The column and the refractive index detector were maintained at 30°C. The injection volume was 10 μl and flow rate was 1.5 ml/min.

Sample preparation for phytochemical analysis

The CHJ and commercial jams, i.e., BF, DK, and SL were extracted with water solvent and assisted by ultrasonic method. The 250 mg of jams were mixed with 5 ml distilled water and immersed in a temperature controlled ultrasonic bath (Elmasonic Easy 60 H, Germany) at 30°C for 5 min. Then, the extracts were centrifuged (Universal 320 R, Germany) at 9000 rpm for 10 min. The supernatants were collected to obtain the extract samples and kept at −20°C before analysis.

Determination of TPC

The amount of TPC of CHJ, BF, DK, and SL was determined by Folin–Ciochette assay which modified from Kamtekar et al. [10]. Briefly, 10 μl of 10 mg/ml extract samples were filled into 96-well microplate and then 100 μl of 1:10 diluted Folin–Ciochette reagent was added. After incubation at room temperature for 7 min, 80 μl of 7% w/v of sodium carbonate was added. After storing in dark room temperature for 2 h, the mixture was recorded under microplate reader (GloMax-Multi Detection System, USA) at 750 nm. TPC was estimated from a calibration curve of standard gallic acid solution. All extracts were measured in 3 times and the results were expressed as milligram gallic acid equivalent (GAEE)/g of jam.

Determination of antioxidant activities

DPPH radical scavenging assay

Free radical scavenging activity of jams was measured by 1,1-diphenyl-2-picrylhydrazyl (DPPH) according to a modified method of Lee et al. In brief, 75 μl of jam solutions was mixed with 150 μl of 0.2 mM DPPH the jam solutions. The reaction of mixtures was carried out at room temperature for 30 min. After incubation, the mixtures were measured at 517 nm using microplate reader (GloMax-Multi Detection System, USA). L-ascorbic acid was used as a positive control and water was used as a blank. The experiment was done in triplicate. The antioxidant capacities of jams were expressed as % DPPH radical scavenging.

\[
\% \text{DPPH radical scavenging} = \left( \frac{A_{517 \text{sample}} - A_{517 \text{blank}}}{A_{517 \text{blank}}} \right) \times 100
\]

Where, \( A_{517 \text{blank}} \) is absorbance of negative control (water) and \( A_{517 \text{sample}} \) is absorbance of jam solution.

Ferric reducing antioxidant power (FRAP) assay

FRAP assay was measured according to Abolhasani et al. [11] method with some modifications, based on the reduction of Fe\(^{3+}\)-TPTZ to a blue-colored Fe\(^{2+}\)-TPTZ. First, the FRAP reagent was prepared by mixing of 10 ml of 0.3 M sodium acetate buffer solution, 1.0 ml of 10 mM TPTZ, and 1.0 ml of 20 mM FeCl\(_3\), and then incubated at 37°C for 4 min. The 150 μl of FRAP reagent was mixed with 20 μl of each jam solution (50 mg/ml) and incubated at 37°C for 30 min. The absorbance of the reaction mixture was measured at 560 nm using a microplate reader (GloMax-Multi Detection System, USA). The calibration curve was prepared using standard 1 mM FeSO\(_4\) solution. The FRAP value of the sample was expressed as μM.

Nitric oxide (NO) radical scavenging assay

NO donor was generated from SNP interacts with oxygen. This assay is modified from Sasaki and Kalaizichiyen [12]. Exactly, 10 mM of SNP in a phosphate-buffered solution (pH 7.4) was incubated with 1 ml of jam solution at 25°C for 3 h. A 100 μl of the testing solution was withdrawn to react with a Griess Reagent Kit, whereby the solution was reacted with 20 μl sulfuricamide for 10 min and then 20 μl \( N \)-(1-naphthyl) ethylenediamine dihydrochloride for another 10 min. The reaction mixture absorbance was measured at 560 nm and the NO concentrations were determined as the nitrite (NO\(^2\)) concentrations from the standard curve of a standard nitrite solution. Distill water and L-ascorbic acid were used as the negative and positive controls, respectively. Percentage inhibition of the nitrite ions generated is observed as NO scavenging capacity of jam solution following below equation.

\[
\% \text{NO radical scavenging} = \left( \frac{A_{560 \text{blank}} - A_{560 \text{sample}}}{A_{560 \text{blank}}} \right) \times 100
\]

Where, \( A_{560 \text{blank}} \) is absorbance of negative control (water) and \( A_{560 \text{sample}} \) is absorbance of jam solution.

Statistical analysis

Data are reported as the mean ± standard deviation of triplicated experiments. TPC and antioxidant activities were analyzed by one-way analysis of variance followed by Tukey’s (post hoc) using the SPSS 22.0 software. Significance was accepted at \( p<0.05 \).

RESULTS AND DISCUSSION

Characteristics of CHJ

The characteristic of CHJ is shown in Fig. 1. For physical characteristics, the color of jam is purple-red, similar to their petal and leave’s color. For the color analysis, CHJ showed dark purple-red by visual observation and showed low \( \text{L}^* \) value (1.90±0.01) that represented dark color while \( \text{a}^* \) showed high value (13.36±0.08) with redness (dark red color) and \( \text{b}^* \) showed quite blue color (Table 1). The pH of CHJ was at 3.78. The CHJ

![Fig 1: The color of Chaba maple homemade jam by organoleptic observation (a), the petals (b), and leaves (c) of Chaba maple.](image-url)
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<table>
<thead>
<tr>
<th>Test sample</th>
<th>TPC (mg GAE/g of jam)</th>
<th>Antioxidant activities</th>
<th>FRAP values (µM)</th>
<th>% NO radical scavenging</th>
</tr>
</thead>
<tbody>
<tr>
<td>BF</td>
<td>9.7±0.38</td>
<td>70.4±2.79</td>
<td>540.58±5.05</td>
<td>17.58±1.72</td>
</tr>
<tr>
<td>DK</td>
<td>21.99±0.50</td>
<td>97.0±0.39</td>
<td>1169.3±20.55</td>
<td>63.89±1.72</td>
</tr>
<tr>
<td>SL</td>
<td>23.66±35.52</td>
<td>95.6±1.82</td>
<td>1179.0±26.52</td>
<td>73.82±1.66</td>
</tr>
<tr>
<td>CHJ</td>
<td>47.18±1.80</td>
<td>100.0±1.39</td>
<td>1690.7±8.26</td>
<td>72.4±1.93</td>
</tr>
<tr>
<td>1 mM L-ascorbic acid</td>
<td>-</td>
<td>1706.3±0.002</td>
<td>-</td>
<td>-</td>
</tr>
<tr>
<td>0.1 mg/mL L-ascorbic acid</td>
<td>-</td>
<td>90.0±0.09</td>
<td>-</td>
<td>-</td>
</tr>
<tr>
<td>0.0025 mg/mL L-ascorbic acid</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>67.7±0.79</td>
</tr>
</tbody>
</table>

*a Test sample: BF: Best food, DK: Doikham, SL: Streamline, CHJ: Chaba maple homemade jam.
*b Values are expressed as means±SD (n=3).
*c p<0.05 is significantly different when compared with CHJ.

Table 1: Determination of physicochemical and chemical characteristics of Chaba maple homemade jam

<table>
<thead>
<tr>
<th>L*</th>
<th>a*</th>
<th>b*</th>
<th>pH</th>
<th>Viscosity (cP)</th>
<th>Total sugar (g/100 g)</th>
</tr>
</thead>
<tbody>
<tr>
<td>1.9±0.01</td>
<td>13.36±0.08</td>
<td>3.27±0.02</td>
<td>3.78</td>
<td>34.48±0.33</td>
<td>1179±152.75</td>
</tr>
<tr>
<td>63.89±1.27</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>Not detected</td>
</tr>
</tbody>
</table>

*Values are expressed as means±SD (n=3).

showed high viscosity (34, 483±152.75 cP) due to high quantity of petals. In the formulation, erythritol was chosen to use as sweetening which tastes like sugar, contains almost zero calories. Then, the result of total sugar by HPLC analysis represented a sugar including fructose, glucose, sucrose, maltose, and lactose. Therefore, CHJ may be beneficial for patients who want to control sugar or sugar control people.

TPC

It is known that edible plants contain huge phytochemicals that could be useful for disease protecting and lowering the risk of various cancers as well as prevention of free radical inducing properties in inflammation-related disorders and cardiovascular diseases [13]. Several plants contain phenolic and flavonoid compounds. A literature review, phenolic compounds in Chaba maple were caffeic acid, gallic acid, galacteinate, coumaric acid, 3,4-dihydroxybenzoic acid, caffeoyl-hydroxyxicrtic acid, chlorogenic acid, querectin-3-galactoside in leaves, and cyanidin and myrcetin in flowers [5,6,14,7]. The TPC of CHJ was determined compared to three commercial products, BF, DK, and SL. All of them contain the fruits, for example, mulberry, strawberry, and cranberry that contain phenolic compounds such as phenolic acids, flavonoids-flavonols, anthocyanins, tannins, and ascorbic acid [15].

Results are summarized in Table 2. The TPC of CHJ showed the highest value for 47.18±1.80 mg GAE/g of jam while BF showed the lowest. DK and SL showed TPC value 21.99 and 23.66 mg GAE/g of jam, respectively, which had no significantly different. As the lowest of TPC, BF is 23.66±0.32 mg GAE/g of jam. Comparing the different jam, TPC represented the quantity genuine plant and in product. The CHJ contains 44.44% Chaba maple petals and 35.56% Chaba maple leaves juice while DK, SL, and BF contain 72% mulberry, 50% blackcurrant, and 10% mixed berries (6% strawberry and 4% blueberry), respectively.

From our studied, total phenolic content, total flavonoid content, and anthocyanin were found in the leaves and flowers of Chaba maple. Therefore, CHJ showed good source of total phenolic compounds.

Antioxidant activities

Phenolic compounds are primary antioxidant that acts as free radical scavenger [16]. They can donate hydrogen atom scavenging free radicals and reducing power and then produce antioxidant radicals which are stable compounds. Antioxidant activities of CHJ and commercial jams were evaluated by the radical scavenging activity using DPPH and NO assay and the capacity of reducing power of jams using FRAP assay. For DPPH assay, CHJ presented the highest percentage radical scavenging properties with 100.00±1.39 followed by DK, SL, and BF, respectively (Table 2). As for NO radical scavenging activity, CHJ showed reduction of NO generation from SNP by 72.4±1.93% as well as SL (73.8±1.66%) and followed by DK and BF respectively. When compared to other jams, CHJ exhibited highest reducing power ability that caused the reduction of Fe3+/ferricyanide complex to the ferrous form. From previous studied, the leaves and flowers of Chaba maple in various extraction displayed total phenolic and flavonoids contents and antioxidant activities as well. Vilela et al. [18] revealed that phenolic compounds in Chaba maple leaves could provide protective effects against genotoxicity and mutagenicity induced by alkylating agents which important role in protection against DNA damage in mice [14]. The methanolic extract of Chaba maple leaves, especially 2-0-trans-cafeoyl-hydroxyxicrtic acid and major phenolic acid, showed antioxidant and anti-inflammatory activities by inhibiting intracellular reactive oxygen species (ROS) production by HL60 cells and extracellular ROS production by neutrophil polymorphonucleocytes and on the activity of pro-inflammatory enzyme myeloperoxidase [7]. The polyphenols and flavonoids in young Chaba maple leaves extract showed inhibit ROS produced by activated neutrophils [5]. From the previous research supported that the TPCs were well correlation between phenolic compounds and their antioxidant activities. It is noted that polyphenol compounds of jams were contributing to radical scavenging activity. Therefore, this study presents antioxidant activity of a homemade jam through difference mechanisms that may lower the risk of health disorders such as cancer, heart disease, and diabetes.

CONCLUSION

CHJ shows a good physical and chemical properties. This jam provides a good source of antioxidant similar with TPC which a choice in health-care food and functional food. To fulfill the jam product development, the forthcoming research in nutrition analysis, microbiological activity, and safety need to be evaluated.

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REFERENCES


