

ISOLATION AND DETERMINATION CARPAINE ALKALOID IN PAPAYA (*CARICA PAPAYA* L.) LEAF EXTRACT BY THIN-LAYER CHROMATOGRAPHY SCANNER

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ABSTRACT

Objective: This research was conducted to isolate the alkaloid carpaine by chromatography method and to determine it quantitatively by Thin Layer Chromatography Scanner.

Methods: Dried leaves were macerated with ethanol 70% and fractionated with dichloromethane. Isolation of carpaine alkaloid from the dichloromethane fraction was carried out by column chromatography and preparative thin-layer chromatography according to the R_f value in Thin Layer Chromatography (TLC) after exposure by Dragendorff reagent.

Results: The content of carpaine alkaloid was 7.5 mg with R_f 0.58 and dichloromethane: methanol (9.2:0.8) as eluent. Validation showed the linearity (R^2) 0.9988, the limit of detection (LOD) was 0.05 ppm, the Limit of Quantification (LOQ) was 0.19 ppm, the recovery from 98.93-102.43%, and the % coefficient of variation was 0.16%.

Conclusion: Carpaine alkaloid in papaya leaf extract was 10.52%.

Keywords: *Carica papaya*, Papaya leaf, Carpaine alkaloid, Validation method, TLC scanner

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INTRODUCTION

Papaya (*Carica papaya* L.) comes from the family *Caricaceae*. This plant is cultivated in almost all subtropical and tropical countries such as South America, India, Philippines and Indonesia. Almost all parts of the papaya plant have medicinal properties [1]. Papaya leaves extract have properties as hematopoietic, antimalarial, antifungal, anti-HIV-1, hypoglycemic, anti-inflammatory, antimicrobial, anti-tumor, immunomodulatory, hepatoprotector, and anti-thrombocytopenic activity [2-10]. In addition, papaya leaves can significantly increase platelet counts and increase the number of red blood cells [11] and showed analgesic activity in mice compared to aspirin [12]. Papaya leaves contain alkaloids carpaine, dehydrocarpaine I and II, pseudocarpaine, choline, carposide, flavonols, benzylglucocinnolate, papain, tannins, vitamin C, and E [13]. The major component of papaya leaves is carpaine alkaloid. This compound reported can reduce blood pressure, antitumor and antiplasmodial activity [14]. In relation to lowering blood pressure, it was reported that carpaine showed a direct effect on the myocardium, where this effect was thought to be related to the macrocyclic dilactone structure, a possible cation chelating structure [15]. The isolated carpaine is able to maintain mouse platelets up to $555 \pm 85.17 \times 10^9/L$ which has been induced by busulfan for 20 d with a dose of 200 mg/kg papaya leaf extract containing 2 mg/kg carpaine alkaloid [16].

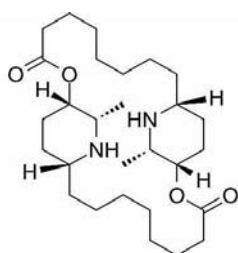


Fig. 1: Carpaine structure [17]

Carpaine alkaloid found with molecular docking has the potential to be antiviral dengue because they have interactions with Thr413 amino acid residues against RNA-dependent RNA polymerization in dengue viruses [18].

The research of Wang *et al.* (2015) states that the levels of carpaine alkaloid in papaya leaves are 0.93 g/kg by High-Performance Liquid Chromatography (HPLC) method [19]. Chromatography is a simple method that can be used to identify secondary metabolites in plants, including alkaloids [20]. Densitometry method with TLC scanner has been used at research of alkaloid compounds in *Chelidonium majus* L. plants which are capable of detecting many samples at the same time [21]. The densitometry method is simpler than HPLC in determining the level of identity compounds in an extract. This method has the advantages like being more effective in preparation and cost-efficient, even though it has a less sensitive sensitivity compared to HPLC. Determination of levels in extracts with TLC scanner is needed validation to determine a method that can use as the determination requirements. Therefore, this study aimed to isolate and determine carpaine alkaloids from papaya leaf extract *Carica papaya* L. using the TLC densitometry method.

MATERIALS AND METHODS

Plant material

Papaya (*Carica papaya* L.) leaves taken from botanical garden Manoko Lembang, Bandung, West Java, Indonesia. The plant identification was determined in the Taxonomy Laboratory, Department of Biology, Faculty of Mathematics and Natural Sciences, Universitas Padjadjaran, Indonesia.

Chemical materials

All solvents and chemicals used to prepare Dragendorff, Mayer, and Liebermann-Burchard reagents had an analytical grade purchased from Merck, including silica gel GF₂₅₄.

Isolation of carpaine alkaloid

Two kilograms of dried leaves were macerated with 70% ethanol at room temperature and protected from light. This extraction was repeated three times until most of the chemical components were extracted. All extracts were evaporated at 50 °C and 60 rpm with a rotary evaporator. The concentrated extract was dissolved in 500 ml of 2% hydrochloride acid (pH 2.0-3.0), then extracted with *n*-hexane to remove fatty acids and other non-polar compounds. The acid fraction was adjusted to pH 8.0-9.0 with ammonia solution and extracted with dichloromethane. The dichloromethane fraction was evaporated with a rotary evaporator at 50 °C. The dichloromethane fraction was dried and then placed in the put into column chromatography and eluted gradient with a solvent mixture of *n*-hexane: ethyl acetate (5: 5) to (0:10) and continued by dichloromethane: methanol (0:10) to (9:1). Each sub-fractions were monitored by TLC using dichloromethane: methanol (9:1) as mobile phase; then the eluted plates were exposed to Dragendorff reagent. The sub-fractions contained carpaine alkaloid were purified by preparative-TLC.

Validation method [22]

Stock preparation

Five milligrams of carpaine alkaloid dissolved with acetone 0.5 ml in Eppendorff tube then diluted to 5 various concentrations (312.5, 625, 1250, 2500, and 5000 ppm).

Linearity, LOD, and LOQ

The standard solution was diluted showed on the TLC plate, then eluted 10 cm long and measured in the area at the maximum wavelength.

Accuracy

Each standard concentration showed on the TLC plate then eluted 10 cm long and measured in the area at the maximum wavelength and repeat twice.

Precision

Each standard concentration showed on the TLC plate then eluted 10 cm long, measured in the area at the maximum wavelength, and repeated twice.

Carpaine analysis in papaya leaf extract

Papaya leaf extract was weighed 3 mg and dissolved with 1 ml acetone, repeated twice. Each sample solution was placed in the TLC plate, then measured by a TLC scanner.

RESULTS

This plant was taxonomically determined at Plants Taxonomy Laboratorium, Faculty of Mathematics and Natural Sciences, Universitas Padjadjaran as *Carica papaya* L. of the family Caricaceae, collection-number 476/HB/01/2017.

Two kilograms of dried papaya leaves were macerated with 70% ethanol and the yield was 18.75% or 375 g. A total of 300 g of concentrated extract were fractionated with *n*-hexane and dichloromethane. Then, each fraction was obtained at 11.65 g (3.89%) and 13.67 g (4.56%), respectively. Both of these fractions were determined by TLC and a dichloromethane fraction was showed the presence of carpaine alkaloid. The dichloromethane (13.67 g) fraction was dried, then put into column chromatography and eluted gradient with a solvent mixture of *n*-hexane: ethyl acetate (5: 5) to (0:10), then continued by dichloromethane: methanol (0:10) to (9:1). Each sub-fractions were monitored by TLC using dichloromethane: methanol (9:1) as mobile phase and the eluted plates were exposed to Dragendorff reagent. The sub-fractions contained carpaine alkaloid was CpII₉, which continued for further separation by column chromatography, then 8 sub-fractions were obtained. Sub-fractions 1-5 were detected of carpaine alkaloid (fig. 2).

The sub-fractions 1 to 5 were combined, then performed by preparative-TLC. Carpaine alkaloid was detected at R_f value of 0.58 with dichloromethane: methanol (98:2) as mobile phase. Isolated carpaine was observed in the form of a yellow powder with a weight of 7.5 mg. Calibration curves are made in a concentration range of 312.5-5000 ppm produced a linear regression equation $y = 0.3266x + 54.468$ with a correlation coefficient of 0.9988.

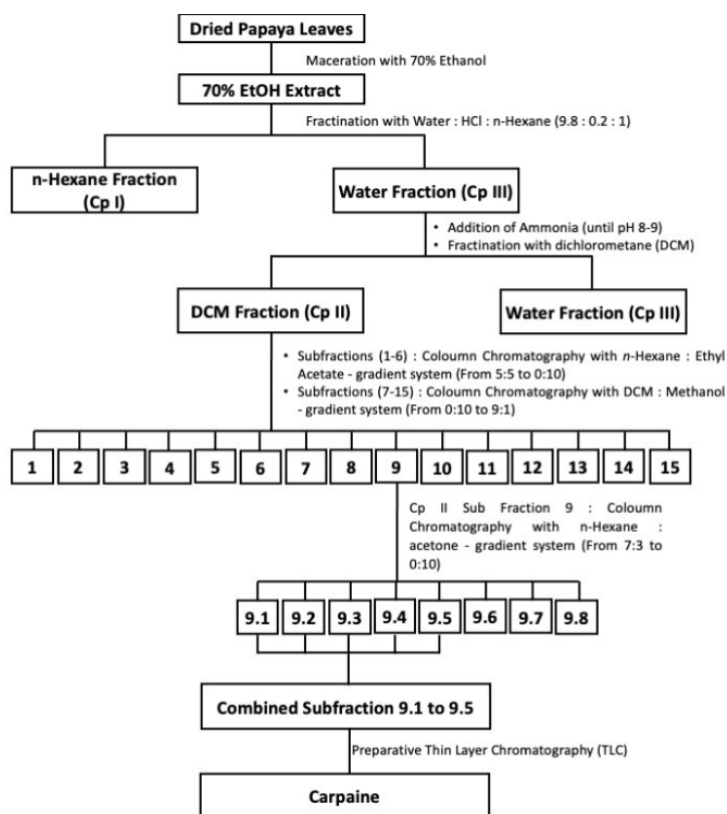


Fig. 2: Isolation process

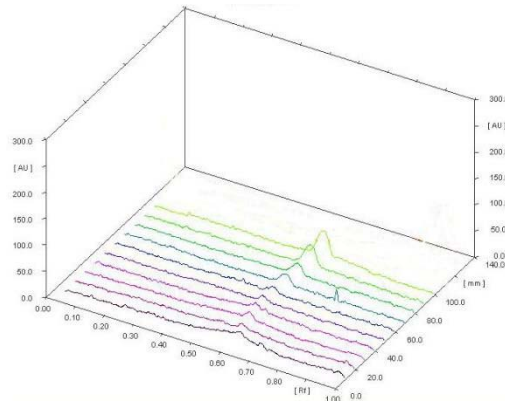


Fig. 3: Densitogram of linearity

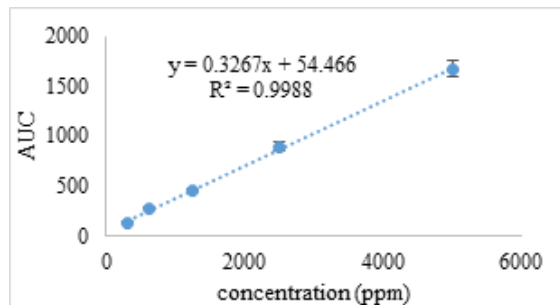


Fig. 4: Calibration curve of carpaine alkaloid

Table 1: Value of LOD and LOQ of carpaine alkaloid

Concentration (ppm)	AUC (Y)	Peak area (Yi)	(Y-Yi)	(Y-Yi) ²	LOD (ppm)	LOQ (ppm)
312.5	130.87	156.53	-25.66	658.46	0.05	0.19
625	276.89	258.59	18.297	334.78		
1250	456.66	462.72	-6.058	36.69		
2500	896.45	870.97	25.482	649.33		
5000	1675.90	1687.50	-11.58	134.05		
		Total		1813.32		

Table 2: Value of accuracy of carpaine alkaloid

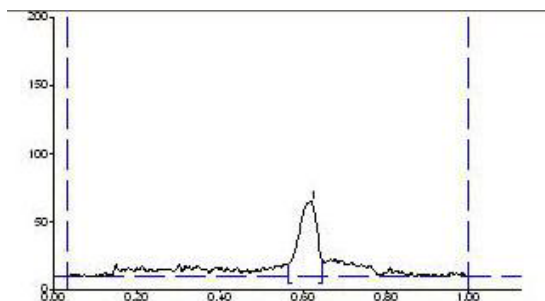
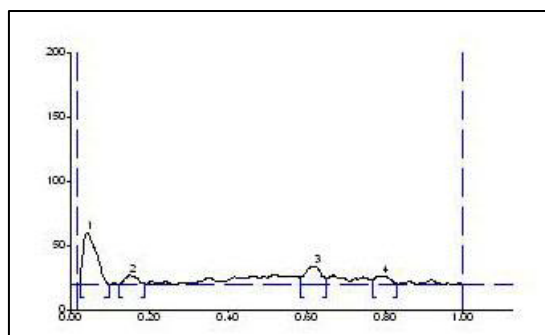
Concentration (ppm)	Peak area (AUC)	Recovery (ppm)	% Recovery	Average of % recovery
312.5	168.98	350.62	98.15	98.93
	144.99	277.16	99.70	
625	276.89	681.02	100.20	99.95
	238.66	563.97	99.70	
1250	467.21	1263.80	101.10	101.01
	466.45	1261.40	100.91	
2500	897.09	2579.80	103.19	102.43
	884.56	2541.60	101.66	
5000	1667.90	4940.00	98.80	99.11
	1678.00	4971.10	99.42	

Table 3: Value of precision of carpaine alkaloid

Concentration (mg/ml)	Peak area (AUC)
1.25	465.78
1.25	466.98
1.25	467.11
1.25	468.76
1.25	467.01
1.25	468.90
Total	7586.40
Average	2167.60
SD	0.0017
%KV	0.16

Table 4: Determination of carpaine alkaloid in papaya leaf extract

Concentration (ppm)	Peak area (AUC) sample	Assay (ppm)	Assay (%)
3000	159.65	322.05	10.73
	155.45	309.19	10.30
Average		315.62	10.52

Fig. 5: Densitogram of carpaine alkaloid R_f 0.62Fig. 6: Densitogram of extract sample with R_f 0.62

DISCUSSION

Plant identification was conducted to distinguish one species from all others are needed for diagnosis. The distinction between species helps characterize of genetic diversity in germplasm collections of endangered plant species [23]. The identification results showed that the plant used in the study was correct, namely *Carica papaya* L.

Fig. 2 showed the isolation of carpaine alkaloid was carried out in several steps. Carpaine isolation was started by maceration using ethanol to extract all secondary metabolites. Maceration is chosen because of preserves the secondary metabolites and the high yield. The maceration principle is the slow transfer of solutes to the solvent until equilibrium [24]. Ethanol was chosen as the extraction solvent because although methanol is a better solvent to dissolve secondary metabolites, methanol is more toxic than ethanol [25]. In addition, carpaine alkaloid is soluble in ethanol and chloroform [26].

After the extract was concentrated, it was then fractionated with *n*-hexane and 2% hydrochloric acid, so that the alkaloids turned into water-soluble alkaloid salts [27]. Ammonia was added to the hydrochloric acid fraction containing carpaine alkaloid, so that carpaine alkaloid will transform to alkaline form and be dissolved in organic solvents, such as dichloromethane [27]. Dichloromethane was chosen to be easily evaporated to obtain crude carpaine alkaloid. Crude carpaine alkaloid was sub-fractionated using column chromatography so that it was separated from other compounds. Crude carpaine alkaloid from column chromatography was separated again by preparative-TLC, then purified by 2-dimensional TLC to obtain purified carpaine alkaloid as shown from a single spot.

Validation of analytical methods is an act of assessing certain parameters based on laboratory tests, to prove that these parameters meet the requirements for their use and to confirm that the analytical

methods are suitable for their intended use [22]. Fig. 3 showed the responses of the TLC scanner to different concentrations of standard carpaine alkaloid as densitogram of linearity. Fig. 4 showed a coefficient correlation of 0.9988, which met the requirement [22]. (LOD) validation method is the smallest number of analytes in the sample that can provide a significant response, while (LOQ) means the smallest number of analytes in the sample that can be quantified or meet the criteria of precision and accuracy [22]. LOD and LOQ validation method was obtained based on the results of the calibration curve of the carpaine alkaloid. The LOD value is 0.05 ppm and the LOQ is 0.19 ppm (table 1).

Accuracy is the closeness of a value from the measurement results to the actual value in the analyte. Accuracy results are determined by the value of %recovery [22]. The area of the result of the calibration curve is then substituted on the equation for the calibration curve. The results of the accuracy test of carpaine alkaloid isolates indicate that the average of recoveries requirements, which were in the range of 98-102% (table 2).

Precision is the percent value of the Coefficient of Variation (%CV) which is calculated from the AUC value and the results of the concentration obtained. [12] Precision tested on standard solutions with a concentration of 1250 ppm in six times replication at a wavelength of 680 nm and a percent coefficient of variation was obtained or %CV was 0.16 (table 3).

The densitogram of standard carpaine alkaloid was given R_f of 0.62 (fig. 5). So, to determine the concentration of carpaine alkaloid in the papaya leaf extract, the focus was on finding the R_f which was the same as the standard, which is 0.62 (fig. 6). Carpaine alkaloid concentration in 3000 ppm papaya leaf extract was 315.62 ppm, equivalent to 10.52% (table 4).

Carpaine alkaloid concentration was lower than Vien's study (2017) [28]. That study obtained 63% of carpaine alkaloid in the total papaya extract that showed concentration of carpaine alkaloid in this study was lower because of the geographical and climate differences in Indonesia and Vietnam, make the nutrients in the soil would be vary causing differences content of secondary metabolites.

CONCLUSION

Carpaine alkaloid was isolated by column chromatography-TLC method and obtained 7.5 mg of carpaine isolate at R_f 0.58 with the mobile phase dichloromethane: methanol (9.2:0.8). The maximum wavelength was 680 nm. Carpaine alkaloid in extract was obtained 10.52% with validation of linearity regression (R²) of 0.9988, (LOD) 0.05 ppm, (LOQ) 0.19 ppm, accuracy value 98.93-102.43%, and the precision value 0.16%.

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Nil

AUTHORS CONTRIBUTIONS

All the authors contributed equally.

CONFLICT OF INTERESTS

The authors declare no conflict of interest.

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