**Original Article** 

# SYNTHESIS, CHARACTERISATION AND DNA PHOTOCLEAVAGE ACTIVITY OF NEW 2-(THIOXO/OXO) QUINOLINE-4,6-DIMETHYL PYRIMIDINYL HYDRAZONES

# AJAY SHARMA<sup>a</sup>, RAJSHREE KHARE<sup>a\*</sup>, VINOD KUMAR<sup>a\*</sup>, VIKAS BENIWAL<sup>b</sup>

<sup>a</sup>Department of Chemistry, Maharishi Markandeshwar University, Mullana, Ambala 133207, Haryana, India, <sup>b</sup>Department of Biotechnology, Maharishi Markandeshwar University, Mullana, Ambala 133207, Haryana, India. Email: rajshreekhare@gmail.com

# Received: 24 Jun 2014 Revised and Accepted: 14 Aug 2014

# ABSTRACT

**Objective:** The main objective of present work is to synthesize, characterize and evaluate DNA photocleavage activity of hydrazones containing quinoline and pyrimidine rings.

**Methods:** The syntheses of new 2-(Thioxo/Oxo)quinoline-4,6-dimethyl pyrimidinyl hydrazones has been achieved by the reaction of 2-(Thioxo/Oxo)quinoline-3-carbaldehydes and 2-hydrazino-4, 6-dimethylpyrimidine. The structure of synthesized compounds is established on basis of data obtained from the spectroscopic techniques such as <sup>1</sup>H NMR, <sup>13</sup>C NMR, FT-IR and mass. The synthesized compounds were evaluated for their DNA photocleavage activity at 40  $\mu$ g/ $\mu$ l concentration by agarose gel electrophoresis method.

Results: The synthesized compounds 5e showed complete cleavage of DNA while 5a, 5b and 6e showed significant cleavage potential.

**Conclusion:** A series of novel hydrazones bearing quinoline and pyrimidine moiety has been synthesized and well characterized on the basis of spectroscopic data and further evaluated for their DNA photocleavage activity. It has been observed that compounds having bromo and thio group displayed good activity.

Keywords: Qunioline-3-carbaldehyde, 2-Hydrazinopyrimidine, Hydrazone, DNA photocleavage activity.

# INTRODUCTION

Biological potential of heterocyclic nitrogeneous compounds like quinolines[1] and pyrimidines[2] are well recognized by the synthetic 2-Chloroquinoline-3chemists and biologists. carbaldehyde is a versatile building block which is used in the synthesis of heterocyclic compounds and some of its derivatives showed a wide spectrum of pharmacological activities[3]. In addition, pyrimidine and its derivatives are another important class of biologically active compounds which shows numerous pharmacological activities[4-6] such as anti-microbial, anticonvulsant, analgesic, anti-inflammatory, anti-platelet, antitubercular, anti-HIV, DNA cleaving agents[7] etc.

Hydrazones are also important class of compounds with a wide range of applications[8-13] and biological importance[8, 14, 15]. The presence of an azomethine -NHN=CH- group makes each compounds as interesting intermediate for synthesis of various heterocyclic compounds. It has been reported that the incorporation of biologically active moieties into another new biologically important heterocyclic compounds with different functionality in same molecule possessed good activity profile[16-18]. In view of these facts and in continuation of our research work[19,20], some novel hydrazones in combination with quinoline and pyrimidine nucleus have been synthesized and evaluated of their DNA nicking activity in the present study.

# Experimental

All the chemicals and solvents were purchased from common commercial suppliers (Hi-media, Loba, S. D. Fine Chemicals and Rankem). Double distilled water was used and melting points were determined using Digital melting point apparatus (paraffin bath) and are uncorrected. Thin layer chromatography was performed on silica gel G for TLC (Rankem) and spots were visualized by iodine vapours or by irradiation with UV light (254 nm). Infra red spectra were recorded on Perkin Elmer RZX FTIR spectrophotometer using KBr discs. The <sup>1</sup>H-NMR and <sup>13</sup>C-NMR spectra were scanned at 400 and 100 MHz, respectively on Bruker spectrophotometer instrument using TMS as an internal reference standard in DMSO-*d*<sub>6</sub>. Coupling

constants (J) are given in Hz. The mass spectra was recorded on Q-ToF Micro Waters LC-MS spectrometer. The starting compounds **1ae**, **2a-e**, **3a-e** and **4** were prepared by the reported methods[21-23].

#### General procedure for synthesis of 2-Thioxoquinoline-4,6dimethylpyrimidinyl hydrazones (6a-e)

3-Formylqinoline-2-thiones (0.01 mol) was dissolved in DMF (10 ml) and added 2-hydrazino-4,6-dimethylpyrimidine (0.01 mol). The mixture was stirred and heated occasionally for 20- 30 min. The progress of the reaction was monitored by TLC till the completion of the reaction. The reaction mixture was poured into ice cold water and stirred for 10-15 min, filtered, precipitated product was washed with cold water and recrystallized from ethanol.

# 1-[(2-Thioxo-1,2-dihydroquinoline-3-yl)methylene]-2-(4,6-dimethylpyrimidin-2-yl) hydrazine (5a)

Orange red; **Yield**: 86%; **M. p.**: 250-252°C; **IR (KBr)** cm<sup>-1</sup>: 3305(N-H str.), 3007 (Ar. C-H str.), 1585,1543 (C=N str.), 1488, 1452 (Ar. C=C str.), 1130 (C=S str.); <sup>1</sup>H **NMR (DMSO-d<sub>6</sub>)**,  $\delta_H$ : 2.39 (s, 6H, 4", 6"-CH<sub>3</sub>), 6.77 (s, 1H, 5"-H), 7.36-7.40 (m, 1H, 6'- H), 7.62-7.67 (m, 2H, 5'-H, 7'-H), 7.92 (d, 1H, 8'-H, J = 8 Hz), 8.55 (s, 1H, N=CH), 8.97(s, 1H, 4'-H), 12.03 (s, 1H, N-NH), 13.92 (s,1H, 1'-NH); <sup>13</sup>C **NMR (DMSO-d<sub>6</sub>)**  $\delta_C$ : 23.4 (4", 6"-CH<sub>3</sub>), 111.9 (C-5"), 122.3 (C-6'), 124.3 (C-5'), 125.4 (C-4a'), 126.1 (C-7'), 127.1 (C-8'), 130.6 (C-4'), 132.3 (C-3'), 133.9 (C-methylene), 137.5 (C-8a'), 159.1 (C-4'',6''), 167.3 (C-2''), 178.2 (C-2'); **MS (ES')** *m*/z Observed [M+1]': 310.0, calcd: 309.1.

#### 1-[(6-Methoxy-2-thioxo-1,2-dihydroquinolin-3-yl)methylene]-2-(4,6-dimethylpyrimidin-2-yl)hydrazine (5b)

Orange red; **Yield**: 88%; **M. p.**: 259-261 °C; **IR (KBr) cm<sup>-1</sup>**: 3308 (N-H *str.*), 3007 (Ar. C-H *str.*), 1592,1548 (C=N *str.*), 1496, 1457 (Ar. C=C *str.*), 1130 (C=S *str.*); <sup>1</sup>**H NMR (DMSO-** $d_6$ **)**,  $\delta_{H}$ : 2.33 (s, 6H, 4", 6"-CH<sub>3</sub>), 3.84 (s, 3H, OCH<sub>3</sub>), 6.63 (s, 1H, 5"-H), 7.26 (dd, 1H, 7'-H, *J* = 8 Hz, *J*= 2 Hz), 7.45 (d, 1H, 5'-H, *J* = 1 Hz), 7.59 (d, 1H, 8'-H, *J* = 9.2 Hz), 8.32 (s, 1H, N=CH), 8.89 (s, 1H, 4'-H), 11.40 (s, 1H, N-NH), 13.81 (s, 1H, 1'-NH); <sup>13</sup>C NMR (DMSO- $d_6$ **)**  $\delta_C$ : 23.4 (4", 6"-CH<sub>3</sub>), 55.5 (6'-OCH<sub>3</sub>), 108.4 (C-5'), 111.9 (C-5''), 117.4 (C-7'), 121.75 (C-4a'), 123.4 (C-8'), 129.4 (C-4'), 133.3 (C-3'), 134.1 (C-methylene), 139.3 (C-8a'),

155.9 (C-6'), 159.5 (C-4'', 6''), 167.3 (C-2''), 177.6 (C-2'); **MS (ES+)** *m/z* Observed [M+1]\*: 340.0, calcd.: 339.1.

#### 1-[(8-Methyl-2-thioxo-1,2-dihydroquinolin)-3-yl)methylene]-2-(4,6-dimethylpyrimidin-2-yl) hydrazine (5c)

Dark Orange; **Yield**: 82%; **M. p.**: 256-258°C; **IR (KBr) cm**-1:3313 (N-H str.), 3021 (Ar. C-H str.), 1588, 1543 (C=N str.), 1486, 1439 (Ar. C=C str.), 1133 (C=S str.); <sup>1</sup>H NMR (DMSO- $d_6$ ),  $\delta_{H}$ : 2.34 (s, 6H, 4", 6"-CH<sub>3</sub>), 2.60 (s, 3H, 8'-CH<sub>3</sub>), 6.63 (s, 1H, 5"-H), 7.24-7.28 (m, 1H, 6'-H), 7.45 (d, 1H, 7'-H, *J* = 7.3 Hz), 7.75 (d, 1H, 5'-H, *J* = 7.6 Hz), 8.31 (s, 1H, N=CH), 8.89 (s, 1H, 4'-H), 11.40 (s, 1H, N-NH), 12.33 (s, 1H, 1'-NH); <sup>13</sup>C NMR (DMSO- $d_6$ )  $\delta_c$ : 17.1 (8'-CH<sub>3</sub>), 2.34 (4", 6"-CH<sub>3</sub>), 111.1 (C-5"), 122.5 (C-5'), 124.1 (C-4a'), 124.3 (C-6'), 126.8 (C-7'), 130.4 (C-4'), 132.6 (C-3'), 137.3 (C-2"), 180.7 (C-2'); **MS (ES')** *m/z* Observed [M+1]\*: 324.0, calcd: 323.1.

# 1-[(6-Methyl-2-thioxo-1,2-dihydroquinolin-3-yl)methylene]-2-(4,6-dimethylpyrimidin-2-yl) hydrazine (5d)

Orange; Yield: 80%; M. p.: 255-257 °C; IR (KBr) cm<sup>-1</sup>: 3307 (N-H str.), 3017 (Ar. C-H str.), 1583, 1543 (C=N str.), 1493, 1448 (Ar. C=C str.), 1139 (C=S str.); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>),  $\delta_{H}$ : 2.33 (s, 6H, 4", 6"-CH<sub>3</sub>), 2.42 (s, 3H, 6'-CH<sub>3</sub>), 6.63 (s, 1H, 5"-H), 7.41 (d, 1H, 7'-H, *J*= 9.42), 7.61 (s, 1H, 5'-H), 7.63 (d, 1H, 8'-H, *J*= 9.20), 8.65 (s, 1H, N=CH), 8.83 (s, 1H, 4'-H), 11.78 (s, 1H, N-NH), 12. 48 (s, 1H, 1'-NH):; <sup>13</sup>C NMR (DMSO-d<sub>6</sub>)  $\delta_{c}$ : 21.7 (6'-CH<sub>3</sub>), 23.4 (4", 6"-CH<sub>3</sub>), 112.2 (C-5"), 124.7 (C-4a'), 125.9 (C-8'), 127.7 (C-7'), 128.5 (C-4'), 132.4 (C-3'), 134.9 (C-6'), 135.8 (C-methylene), 143.5 (C-8a'), 159.1 (C-4",6"), 167.8 (C-2"), 178.2 (C-2'); MS (ES\*) *m/z* Observed [M+1]\*: 324.0, calcd: 323.1.

#### 1-[(6-Bromo-2-thioxo-1,2-dihydroquinolin)-3-yl)methylene]-2-(4,6-dimethylpyrimidin-2-yl) hydrazine (5e)

Dark red; Yield: 79 %; M. p.: 270-272°C; IR (KBr) cm<sup>-1</sup>: 3317 (N-H str.), 3019 (Ar. C-H str.), 1594, 1552 (C=N str.), 1498, 1459 (Ar. C=C str.),1137 (C=S str.); <sup>1</sup>H NMR (DMSO- $d_6$ ),  $\delta_H$ : 2.39 (s, 6H, 4", 6"-CH<sub>3</sub>), 6.78 (s, 1H, 5"-H), 7.65-7.68 (m, 2H, 5'-H, 8'-H), 8.42-8.45 (m, 1H, 7'-H), 8.52 (s, 1H, N=CH), 8.73 (s, 1H, 4'-H), 12.10 (s, 1H, N-NH), 13.78 (s, 1H, 1'-NH); <sup>13</sup>C NMR (DMSO- $d_6$ )  $\delta_C$ : 23.4 (4", 6"-CH<sub>3</sub>), 111.2 (C-5"), 120.2 (C-7"), 126.1 (C-5"), 127.6 (C-8"), 128.9 (C-4a'), 129.7 (C-4'), 132.1 (C-3'), 133.2 (C-methylene), 134.1 (C-6'), 143.4 (C-8a'), 159.8 (C-4", 6"), 167.9 (C-2"), 180.1 (C-2"); MS (ES<sup>+</sup>) m/z Observed [M+2]<sup>+</sup>: 389.0, calcd: 387.02.

#### General procedure for synthesis of 2-Oxoquinoline-4,6dimethylpyrimidinyl hydrazones (6a-e)

2-Oxo-3-formylqinolines (0.01 mol) was dissolved in DMF (10 ml) and added 2-Hydrazino-4, 6-dimethyl pyrimidine (0.01 mol). The mixture was stirred and heated occasionally for 20- 30 min. The progress of reaction was monitored by TLC till the completion of the reaction. The reaction mixture was poured into ice cold water, stirred for 10-15 min, filtered, precipitated product was washed with cold water and recrystallized from ethanol.

# 1-[(2-Oxo-1,2-dihydroquinolin-3-yl)methylene]-2-(4,6dimethylpyrimidin-2-yl) hydrazine (6a)

Yellow; Yield: 86%; M. p.: >275°C; IR (KBr) cm<sup>-1</sup>: 3218 (N-H str.), 1651 (C=0 str.), 1556 (C=N str.); <sup>1</sup>H NMR (DMSO- $d_6$ ),  $\delta_{H}$ : 2.32 (s, 6H, 4", 6"-CH<sub>3</sub>), 6.64 (s, 1H, 5"-H), 7.17-7.21 (m, 1H, 6'- H), 7.31 (d, 1H, 5'-H, J = 8 Hz), 7.47-7.51 (m, 1H, 7'-H), 7.81 (dd, 1H, 8'-H, J = 8Hz, J = 0.8 Hz, J = 0.72 Hz), 8.35(s, 1H, N=CH), 8.37(s, 1H, 4'-H), 11.37 (s, 1H, N-NH), 12.0 (s,1H, 1'-NH); <sup>13</sup>C NMR (DMSO- $d_6$ )  $\delta_{c}$ : 23.4 (4", 6"-CH<sub>3</sub>), 111.4 (C-5"), 124.3 (C-6'), 126.9 (C-5'), 127.2 (C-4a'), 127.7 (C-7'), 128.1 (C-8'), 129.6 (C-4'), 133.3 (C-3'), 134.6 (C-methylene), 145.3 (C-8a'), 159.1 (C-4",6"), 160.7 (C-2'), 167.7 (C-2"); MS (ES<sup>+</sup>) m/z Observed [M+1]<sup>+</sup>: 294.1, calcd: 293.1.

#### 1-[(6-Methoxy-2-oxo-1,2-dihydroquinolin-3-yl)methylene]-2-(4,6-dimethylpyrimidin-2-yl)hydrazine (6b)

Yellow; Yield: 87%; M. p.: >275 °C; IR (KBr) cm<sup>-1</sup>: 3270 (N-H str.), 1690 (C=0 str.), 1596 (C=N str.); <sup>1</sup>H NMR (DMSO- $d_6$ ),  $\delta_H$ : 2.33 (s, 6H, 4", 6"-CH<sub>3</sub>), 3.80 (s, 3H, OCH<sub>3</sub>), 6.60 (s, 1H, 5"-H), 7.11 (d, 1H, 7'- H, J = 8 Hz), 7.24-7.29 (m, 2H, 5'-H), 8'-H), 8.33 (s, 1H, N=CH), 8.36 (s, 1H, 4'-H), 11.29 (s, 1H, N-NH), 11.85 (s, 1H, 1'-NH); <sup>13</sup>**C** NMR (DMSO-d<sub>6</sub>)  $\delta_c$ : 23.4 (4", 6"-CH<sub>3</sub>), 55.3 (6'-OCH<sub>3</sub>), 109.3 (C-5'), 111.7 (C-5''), 116.2 (C-7'), 119.6 (C-4a'), 119.8 (C-8'), 126.5 (C-4'), 132.4 (C-3'), 132.07 (C-methylene), 136.2 (C-8a'), 154.3 (C-6'), 159.5 (C-4'', 6''), 160.6 (C-2'), 167.2 (C-2''); MS (ES<sup>+</sup>) m/z Observed [M+1]<sup>+</sup>: 324, calcd.: 323.1.

# 1-[(8-Methyl-2-oxo-1,2-dihydroquinolin-3-yl)methylene]-2-(4,6-dimethylpyrimidin-2-yl) hydrazine (6c)

Yellow; Yield: 83%; M. p.: >275 °C; IR (KBr) cm<sup>-1</sup>: 3268 (N-H str.), 1660 (C=0 str.), 1555 (C=N str.); <sup>1</sup>H NMR (DMSO- $d_6$ ),  $\delta_{H}$ : 2.31 (s, 6H, 4", 6"-CH<sub>3</sub>), 2.44 (s, 3H, 8'-CH<sub>3</sub>), 6.63 (s, 1H, 5"-H), 7.09-7.13 (m, 1H, 6'-H), 7.34 (d, 1H, 7'-H, *J* = 8 Hz), 7.65 (d, 1H, 5'-H, *J* = 8 Hz), 8.33 (s, 1H, N=CH), 8.36 (s, 1H, 4'-H), 11.13 (s, 1H, N-NH), 11.31 (s, 1H, 1'-NH); <sup>13</sup>C NMR (DMSO- $d_6$ )  $\delta_c$ : 17.1 (8'-CH<sub>3</sub>), 23.4 (4", 6"-CH<sub>3</sub>), 111.9 (C-5"), 119.3 (C-5"), 122.03 (C-4a'), 123.3 (C-6'), 125.8 (C-7'), 126.6 (C-4'), 131.7 (C-3'), 133.4 (C-8'), 135.9 (C-methylene), 136.8 (C-8a'), 159.5 (C-4", 6"), 161.5 (C-2'), 167.3 (C-2"). MS (ES<sup>+</sup>) *m/z* Observed [M+1]<sup>+</sup>: 308.1, calcd.: 307.1.

# 1-[(6-Methyl-2-oxo-1,2-dihydroquinolin-3-yl)methylene]-2-(4,6-dimethylpyrimidin-2-yl) hydrazine (6d)

Yellow; Yield: 80%; M. p.: >275°C; IR (KBr) cm<sup>-1</sup>: 3295 (N-H str.), 1670 (C=0 str.), 1575 (C=N str.); <sup>1</sup>H NMR (DMSO- $d_6$ ),  $\delta_H$ : 2.37 (s, 6H, 4", 6"-CH<sub>3</sub>), 2.79 (s, 3H, 6'-CH<sub>3</sub>), 6.63 (s, 1H, 5"-H), 7.21 (d, 1H, 7'-H, *J*= 9.52), 7.63 (d, 1H, 8'-H, *J*= 9.18), 8.26 (s, 1H, N=CH), 8.29 (s, 1H, 4'-H), 11.13 (s, 1H, N-NH), 11.36 (s, 1H, 1'-NH); <sup>13</sup>C NMR (DMSO- $d_6$ )  $\delta_c$ : 21.1 (6'-CH<sub>3</sub>), 23.4 (4", 6"-CH<sub>3</sub>), 111.5 (C-5"), 124.2 (C-5'), 125.5 (C-4a'), 126.3 (C-8'), 127.2 (C-7'), 129.9 (C-4'), 133.8 (C-3'), 134.8 (C-6'), 135.1 (C-methylene), 143.6 (C-8a'), 159.4 (C-4", 6"), 160.2 (C-2'),167.5 (C-2"); MS (ES<sup>+</sup>) *m/z* Observed [M+1]<sup>+</sup>: 308.1, calcd.: 307.1.

# 1-[(6-Bromo-2-oxo-1,2-dihydroquinolin-3-yl)methylene]-2-(4,6-dimethylpyrimidin-2-yl) hydrazine (6e)

Yellow green; Yield: 78%; M. p.: >275°C; IR (KBr) cm<sup>-1</sup>: 3305 (N-H str.), 1680 (C=0 str.), 1565 (C=N str.); <sup>1</sup>H NMR (DMSO- $d_6$ ),  $\delta_H$ : 2.34 (s, 6H, 4", 6"-CH<sub>3</sub>), 6.66 (s, 1H, 5"-H), 7.46-7.49 (m, 2H, 5'-H, 8'-H), 8.23-8.26 (m, 1H, 7'-H) 8.36 (s, 1H, N=CH), 8.39 (s, 1H, 4'-H), 11.33 (s, 1H, N-NH), 12.02 (s, 1H, 1'-NH); <sup>13</sup>C NMR (DMSO- $d_6$ )  $\delta_c$ : 23.8 (4", 6"-CH<sub>3</sub>), 110.3 (C-5"), 122.5 (C-7'), 125.1 (C-5'), 126.6 (C-8'), 128.6 (C-4a'), 129.2 (C-4'), 132.9 (C-3'), 133.7 (C-methylene), 134.9 (C-6'), 144.4 (C-8a'), 159.2 (C-4", 6"), 161.8 (C-2'), 167.6 (C-2''); MS (ES<sup>+</sup>) m/z Observed [M+2]<sup>+</sup>: 373.04, calcd.: 371.04.

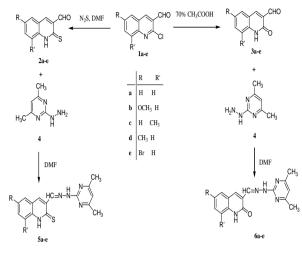
# DNA Photocleavage activity

The photocleavage of plasmid DNA was determined by agarose gel electrophoresis. The experiments were performed in a volume of 10  $\mu$ l containing the plasmid DNA in TE (*Tris* 10 mM, EDTA 0.01 mM, pH 8.0) buffer in presence of 40  $\mu$ g of the synthesized compounds. The samples were taken in polyethylene microcentrifuge tubes, which were then irradiated for 30 min at room temperature in trans-illuminator (8000mW/cm) at 360 nm. Further, the samples were incubated at 37°C for one hour. 6X loading dye containing 0.25% bromophenol blue and 30% glycerol (8  $\mu$ l) was mixed with an irradiated sample. The analysis of samples was carried out on 0.8% agarose horizontal slab gel in *Tris*-Acetate EDTA buffer (40 mM Tris, 20 mM acetic acid, 1 mM EDTA, pH: 8.0). Untreated plasmid DNA was maintained as a control in each run of gel electrophoresis which was carried out at 5V/cm for 2.0 h. Gel was stained with ethidium bromide (1  $\mu$ g/mL) and photographed under UV light [20].

# **RESULTS AND DISCUSSION**

Quinoline-3-carbaldehydes and 2-Hydrazino-4,6-dimethyl pyrimidine were used as starting materials for synthesis hydrazones (Scheme 1). First, 2-thioxoquinoline-3-carbaldehydes **2a-e** were treated with 2-hydrazino-4,6-dimethylpyrimidine **4** in DMF to give 2-thioxoquinolinepyrimidinyl hydrazones **5a-e**. In the similar way, 2-Oxoquinolinepyrimidinyl hydrazones **6a-e** were also prepared from 2-oxoquinoline-3-carbaldehydes **3** and 2-hydrazino-4,6-dimethylpyrimidine **4** in DMF.

The formation of compounds **5** and **6** have been confirmed due to appearance of the characteristics bands in a range 3320-3210 cm<sup>-1</sup> due to –NH stretching of hydrazone group in IR spectra and two characteristics singlets at  $\delta$  8.20-8.75 and  $\delta$  11.1-12.1 due to N=CH and N-NH proton in <sup>1</sup>H NMR spectra, respectively. The results are also supported by mass spectra.



Scheme 1: Synthetic scheme for 2-(Thioxo/ Oxo)quinoline-4,6dimethyl pyrimidinyl hydrazones (5 and 6)

The heterocyclic compounds containing conjugated C=N bond system have ability to cleave DNA photochemically due to the generation of photoexcited (n-  $\pi^*$ ) state which would have radical character [24]. In a recent communication, compounds having quinoline or pyrimidine pharmacophore have been reported for DNA cleavage photochemically[7, 25]. The compound with C=S group showed significant DNA photocleavage activity.

The bromo group containing compound of this category showed complete cleavage of DNA. The ability of synthesized compounds to interact with plasmid DNA and induce cleavage is shown in the electrophoretogram (**Figure 1**). All solutions were prepared in DMSO. The compounds **5e** showed complete cleavage while **5a**, **5b** and **6e** showed significant nicking in DNA.

# CONCLUSION

In this work, we have reported the synthesis of new hydrazones having quinoline and pyrimidine motifs which were characterized by IR, <sup>1</sup>H NMR, <sup>13</sup>C NMR, mass spectral data and elemental analysis. The synthesized compounds were screened for their DNA photocleavage activity at 40  $\mu$ g/ $\mu$ l concentration. The compounds having bromo and thio group showed good DNA photocleavage activity.

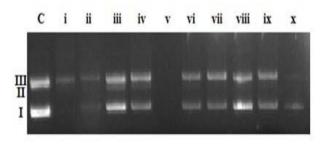


Fig. 1: Gel electrophoretogram of compounds (5a-e) and (6a-e)

C: DNA (control); Lane-(i): DNA + 5a; Lane-(ii): DNA+ 5b; Lane-(iii): DNA+ 5c; Lane-(iv): DNA+ 5d; Lane-(v): DNA+ 5e; Lane-(vi): DNA+ 6a; Lane-(vii): DNA+ 6b; Lane-(viii): DNA+ 6c; Lane-(ix): DNA+ 6d; Lane-(x): DNA+ 6e.

## REFERENCES

- Katritzky AR, Rees CW. Comprehensive Heterocyclic Chemistry vol.,2. Oxford NY: Pergamon Press; 1997.
- Brown DJ. Pyrimidines and their Benzo Derivatives, In: Katritzky AR, Rees CW, editors. Comprehensive Heterocyclic Chemistry vol. 3, Oxford: Pergamon Press; 1997. p. 55-155.
- Abdel-Wahab BF, Khidre RE, Farahat AA, El-Ahl AAS. 2-Chloroquinoline-3-carbaldehydes: synthesis, reactions and applications. J Arkivoc 2012;(i):211-76.
- Selvam TP, James CR, Dniandev PV, Valzita SK. A mini review of pyrimidine and fused pyrimidine marketed drugs. J Res Pharm 2012;2:1-9.
- Dinakaran VS, Bomma B, Srinivasan KK. Fused pyrimidines: The heterocycle of diverse biological and pharmacological significance. J Der Pharma Chemica 2012;4:255-65.
- Jain KS, Chitre TS, Miniyar PB, Kathiravan MK, Bendre VS, Veer VS, Shahane SR, Shishoo CJ. Biological and medicinal significance of pyrimidines. J Curr Sci 2006;90:793-803.
- Keri RS, Hosamani KM, Shingalapur RV, Hugar MH. Analgesic, anti-pyretic and DNA cleavage studies of novel pyrimidine derivatives of coumarin moiety. Eur J Med Chem 2010;45:2597-605.
- Rollas S, Kucukguzel SG. Biological activities of hydrazone derivatives. J Molecules 2007;12:1910-39.
- 9. Singh RB. Hydrazones as analytical reagents: A review. J Talanta 1982;29:77-84.
- Katyal M, Dutt Y. Analytical applications of hydrazones. J Talanta 1975;22:151-66.
- Fouda AA, Al-Sarawy AA, Radwan MS. Some aromatic hydrazone derivatives as inhibitors for the corrosion of C-steel in phosphoric acid solution. J Ann Chim 2006;96:85-96.
- Negm NA, Morsy SMI, Said MM. Corrosion inhibition of some novel hydrazone derivatives. J Surfactants Deterg 2005;8:95-8.
- Sherif ESM, Ahmed AH. Synthesizing new hydrazone derivatives and studying their effects on the inhibition of copper corrosion in sodium chloride solutions. J Synth React Inorg, Met-Org Nano-Met Chem 2010;40:365-72.
- Zelenin KN, Khorseeva LA, Alekseev VV. Physiologically active complexes of hydrazones (review). Pharm Chem J 1992;26:395-405.
- Hajipour AR, Baltork IM, Bigdeli M. A convenient and mild procedure for the synthesis of hydrazones and semicarbazones from aldehydes or ketones under solvent-free conditions. J Chem Res Synop 1999:570-1.
- 16. Siddiqui SM, Salahuddin A, Azam A. Synthesis, characterization and antiamoebic activity of some hydrazone and azole derivatives bearing pyridyl moiety as a promising heterocyclic scaffold. Eur J Med Chem 2012;49:411-6.
- El-Sayed MAA, Abdel-Aziz NI, Abdel-Aziz AAM, El-Azab AS, Asiri YA, El-Tahir KEH. Design, synthesis, and biological evaluation of substituted hydrazone and pyrazole derivatives as selective COX-2 inhibitors: Molecular docking study. J Bioorg Med Chem 2011;19:3416-24.
- Ozkay Y, Tunal Y, Karaca H, Iskdag I. Antimicrobial activity and a SAR study of some novel benzimidazole derivatives bearing hydrazone moiety. Eur J Med Chem 2010;45:3293-8.
- Kumar V, Gupta GK, Kaur K, Singh R. Fluorophenylhydrazones as potential COX-2 inhibitors: A novel, efficient, one pot solid phase synthesis, docking study and pharmacological evaluation. J Med Chem Res 2013;22:5890-900.
- Sharma A, Khare R, Kumar V, Gupta GK, Beniwal V. 1-(Subsituted)-4, 4, 6-trimethyl-3, 4-dihydropyrimidine-2(1*H*)thione: Green synthesis, antibacterial activity and DNA photocleavage activity. Int J Pharm Pharm Sci 2014;6:171-5.
- 21. Srivastava A, Singh RM. Vilsmeier-Hack reagent: A facile synthesis of 2-chloro-3-formylquinolines from *N*-arylacetamides and transformation into different functionalities. Ind J Chem 2005;44B: 1868-75.
- 22. Kumar V, Aggarwal R, Tyagi P, Singh SP. Synthesis and antibacterial activity of some new 1-heteraryl-5-amino-4-phenyl-3-trifluoromethylpyrazoles. Eur J Med Chem 2005;40:922-7.
- 23. Aggarwal R, Kumar V, Tyagi P, Singh SP. Synthesis and antibacterial study of some new 1-heteroaryl-5-amino-

3*H*/methyl-4-phenylpyrazole. Bioorg Med Chem 2006;14:1785-91.

24. Toshima K, Takano R, Maeda Y, Suzuki M, Asai A, Matsumura S. 2-Phenylquinoline-Carbohydrate hybrids: molecular design, chemical synthesis, and evaluation of a new family of lightactivatable DNA-cleaving agents. J Angew Chem Int Ed 1999;38:3733-35.

 Bindu PJ, Mahadevan KM, Satyanarayan ND, Naik TRR. Synthesis and DNA cleavage studies of novel quinoline oxime esters. J Bioorg Med Chem Lett 2012;22:898-900.