# DEVELOPMENT AND VALIDATION OF RP-HPLC METHOD FOR SIMULTANEOUS ESTIMATION OF PIPERAQUINE PHOSPHATE AND DIHYDROARTEMISININ IN BULK 

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#### Abstract

High Performance Liquid Chromatography (HPLC) methods are described for determination of drugs as a single or in combination in bulk or pharmaceutical formulation. The objective of the present study was to develop and validate novel, accurate, sensitive, precise, rapid and isocratic reverse Phase HPLC (RP-HPLC) method for the simultaneous determination of Piperaquine phosphate and Dihydroartemisinin in bulk because no method is available for simultaneous estimation of these drugs. The separation was achieved on GRACESMART RP-18 column ( $250 \mathrm{~mm} \times 4.6 \mathrm{~mm}$, $5 \mu \mathrm{~m}$ ) with mobile phase consisting of 10 mM Ammonium acetate ( pH 4.6 , adjusted with Acetic acid): Methanol ( $15: 85 \% \mathrm{v} / \mathrm{v}$ ) at a flow rate of 1.2 $\mathrm{ml} / \mathrm{min}$. UV detection at 220 nm . PQP and DHA obeyed linearity in the concentration range of $5-25 \mu \mathrm{~g} / \mathrm{ml}(\mathrm{r} 2=0.9993)$ and $5-25 \mu \mathrm{~g} / \mathrm{ml}(\mathrm{r} 2=$ 0.9987 ) respectively. The asymmetric factors were found to be 1.17 for PQP and 1.2 for DHA. The developed method was validated as per ICH guidelines fulfill all the acceptance criteria and can be use for routine analysis.


Keywords: Piperaquine phosphate, Dihydroartemisinin, Simultaneous, RP-HPLC.

## INTRODUCTION

Piperaquine Phosphate (PQP) i.e. 4,4-(1,3-Propanediyldi-4,1Piperazinediyl) bis (7- chloroquinoline) phosphate, molecular mass is $633.51 \mathrm{~g} / \mathrm{mole}$ is a bisquinoline anti-malarial drug and shows good activity against chloroquine-resistant Plasmodium strains. Evidence suggesting the inhibition of the heme-digestion pathway in the parasite food vacuole is most convincing. Piperaquine's bulky bisquinoline structure may be important for activity against chloroquine resistant strains and may act by inhibition of the transporters that efflux chloroquine from the parasite food vacuole.

Dihydroartemisinin (DHA) is (4S,5R,8S,9R,10S,12R,13R)-1,5,9Trimethyl 11,14,15,16- tetraoxatetracyclo [10.3.1.0[4,13].0[8,13]] hexadecan-10-ol. DHA is Dihydroartemisinin (DHA) is a semisynthetic compound obtained by reduction of artemisinin extracted from the leaves of Artemisia annua L followed by purification and blending. DHA is highly active against asexual forms of the four species of Plasmodium that infect humans with very rapid reductions in parasitaemia but relatively short plasma halflives. There is also activity against the sexual forms (gametocytes) and therefore it is having some potential to reduce transmission rates.[1,2] Few methods have been reported for the estimation of these drugs as single drug or in combination with other drugs Piperaquine Phosphate (PQP) has been reported to be estimated by HPLC[3,4], LC-MS with impurities[5], and LC-UV for stability studies. While for DHA estimation was carried out with HPLC[7-10], core enhanced Accucore column[11], LC-tandem Mass[12,13], spectrophotometry [14]. So far no analytical method was available for simultaneous estimation of PQP and DHA.

## MATERIALS AND METHOD

## Materials

HPLC grade Methanol (Fischer scientific), Ammonium acetate and HPLC grade Acetic acid were used. Deionized HPLC grade water was used to prepare mobile phase and diluents solutions. Both the drugs; PQP and DHA were obtained from Shreya Life Sciences Pvt. Ltd., Aurangabad, M.S., India.

## Equipments

HPLC analysis was performed on a Dionex HPLC system with P680 Pump, automated sample injector, a GRACESMART RP-18 column ( 250 mm X 4.6 mm i.d., with particle size $5 \mu \mathrm{~m}$ ) and a programmable
variable wavelength UV-visible detector. Data were collected and processed using "Chromeleon version 6.0" software.

Preparation of Ammonium Acetate Buffer pH 4.6
Accurately weighed 0.3854 g of Ammonium acetate was dissolved in deionized HPLC grade water to make 10 mM buffer strength. Then pH of buffer adjusted to 4.6 using HPLC grade Acetic acid. [15]

Preparation of Mobile Phase
The two components of mobile phase; Methanol (HPLC grade) and Buffer (prepared previously) were separately filtered through a $0.45 \mu \mathrm{~m}$ nylon membrane filter. They were mixed respectively in the ratio of $85: 15 \% \mathrm{v} / \mathrm{v}$ and sonicated for 15 min . [15]

Preparation of Standard Laboratory Mixture Solution
Accurately weighed quantity of 10 mg of PQP and 10 mg of DHA were transferred to 100 ml volumetric flask, dissolved in mobile phase and volume was made up to mark with same solvent. From stock solution suitable aliquot was transferred to 10 ml volumetric flask and diluted to mark with the mobile phase, to obtain the concentration of $10 \mu \mathrm{~g} / \mathrm{ml}$ of PQP and $10 \mu \mathrm{~g} / \mathrm{ml}$ of DHA. A volume of $10 \mu \mathrm{l}$ of solution was injected. All measurements were repeated three times for each concentration.

## RESULT AND DISCUSSION

Under the stated chromatographic conditions, the retention time of drugs was 2.02 min for PQP and 2.46 min for DHA. A model Chromatogram for pure laboratory mixture is shown in figure 1.


Fig. 1: Chromatogram for Pure Drug Mixture

Method Validation: [16-22]
The proposed method was validated as per ICH parameters and guidelines.

## Linearity

Appropriate aliquots of the standard stock solutions of PQP and DHA were pipette out and transferred to a series of 10 ml volumetric flasks respectively.

The volume was made up to the mark with mobile phase to obtain working standard solutions of each drug separately of concentrations $5,10,15,20$, and $25 \mu \mathrm{~g} / \mathrm{ml}$. From these solutions, 10 $\mu \mathrm{l}$ injections of each concentration of the drug were injected into the HPLC system three times separately and chromatogram was obtained under the conditions as finalized by developed method. The peak areas were recorded. The standard calibration curves for PQP and DHA were plotted separately as peak area Vs the respective concentration.

Table 1: Linearity study of DHA at 220 nm for HPLC

| S. No. | Concentration <br> $(\boldsymbol{\mu g} / \mathbf{m l})$ | Peak Area <br> Mean $\pm$ S.D. $[\mathbf{n}=\mathbf{3 ]}$ | \%RSD |
| :--- | :--- | :--- | :--- |
| 1 | 5 | $3.39 \pm 0.0832$ | 2.450 |
| 2 | 10 | $6.25 \pm 0.0757$ | 1.211 |
| 3 | 15 | $9.403 \pm 0.0404$ | 0.429 |
| 4 | 20 | $12.38 \pm 0.0808$ | 0.652 |
| 5 | 25 | $16.17 \pm 0.195$ | 0.195 |
| Average \% RSD |  |  | $\mathbf{0 . 5 3 6}$ |

Table 2: Linearity study of PQP at 220 nm for HPLC

| S. No. | Concentration <br> $(\boldsymbol{\mu g} / \mathbf{m l})$ | Peak Area <br> Mean $\pm$ S.D. $[\mathrm{n}=\mathbf{3 ]}$ | \%RSD |
| :--- | :--- | :--- | :--- |
| 1 | 5 | $4.586 \pm 0.0611$ | 1.333 |
| 2 | 10 | $9.26 \pm 0.0754$ | 0.814 |
| 3 | 15 | $14.103 \pm 0.0665$ | 0.471 |
| 4 | 20 | $20.416 \pm 0.104$ | 0.509 |
| 5 | 25 | $26.066 \pm 0.0709$ | 0.261 |
| Average \% RSD |  |  | $\mathbf{0 . 6 7 7}$ |



Fig. 2: Calibration Curve of Standard DHA


Fig. 3: Calibration Curve of Standard PQP

## Precision

It is expressed as $\pm$ S.D. of series of measurements. Precision was carried out by two methods according to ICH guidelines:

1) Repeatability studies
2) Intermediate precision - variation in days and analyst.

## Repeatability

It is measured by multiple injections of a homogenous sample of 10 $\mu \mathrm{g} / \mathrm{ml}$ of PQP and $10 \mu \mathrm{~g} / \mathrm{ml}$ of DHA that indicates the performance of the HPLC instrument under chromatographic conditions. Results are shown in Table 4.

Table 3: Results of recovery studies of PQP and DHA for HPLC

| Pre-analysed <br> Sample Solution <br> $(\boldsymbol{\mu g} / \mathbf{m l})$ | \% Level <br> of Addition | Excess drug <br> Added $(\boldsymbol{\mu g} / \mathbf{m l})$ <br> $(\mathrm{n}=\mathbf{3})$ | Amount <br> Recovered <br> $(\boldsymbol{\mu g} / \mathbf{m l})$ | Average <br> \%Recovery | Average <br> \% RSD |
| :--- | :--- | :--- | :--- | :--- | :--- |
| PQP (10) | 80 | 8 | 17.96 | 0.139 |  |
|  | 100 | 10 | 19.96 | 99.81 | 0.051 |
| DHA (10) | 120 | 8 | 22.04 | 100.80 |  |
|  | 80 | 10 | 17.98 | 99.88 | 0.090 |
|  | 100 | 12 | 20.03 | 100.15 | 0.037 |

Table 4: Results of Repeatability Studies of PQP and DHA for HPLC
$\left.\begin{array}{llll}\hline \text { Analyte } & \begin{array}{l}\text { Amount taken } \\ (\mu \mathrm{g} / \mathrm{ml})\end{array} & \begin{array}{l}\text { Amount Found } \\ (\mu \mathrm{g} / \mathrm{ml}) \pm \text { S. } .\end{array} & \text { \%RSD } \\ {[\mathrm{n}=6]}\end{array}\right)$

Table 5: Results of Intra-Day and Inter-Day Precision of PQP and DHA for HPLC

| Analyte Conc. ( $\mu \mathrm{g} / \mathrm{ml}$ ) |  | Intra-Day Amount Found ( $\mu \mathrm{g} / \mathrm{ml}$ ) |  | Inter-Day Amount Found ( $\mu \mathrm{g} / \mathrm{ml}$ ) |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
|  |  | $\begin{aligned} & \text { Mean } \pm \text { S.D. } \\ & {[\mathrm{n}=3]} \end{aligned}$ | \%RSD | $\begin{aligned} & \text { Mean } \pm \text { S.D. } \\ & {[n=3]} \end{aligned}$ | \%RSD |
| PQP | 5 | $4.93 \pm 0.014$ | 0.248 | $4.97 \pm 0.0251$ | 0.505 |
|  | 10 | $10.01 \pm 0.035$ | 0.349 | $9.96 \pm 0.0585$ | 0.587 |
|  | 15 | $14.81 \pm 0.025$ | 0.169 | $14.84 \pm 0.036$ | 0.242 |
|  |  | Avg.\%RSD | 0.255 | Avg. \%RSD | 0.444 |
| DHA | 5 | $4.97 \pm 0.0305$ | 0.613 | $5.92 \pm 0.0246$ | 0.446 |
|  | 10 | $9.877 \pm 0.025$ | 0.254 | $9.84 \pm 0.0254$ | 0.258 |
|  | 15 | $14.99 \pm 0.058$ | 0.387 | $14.95 \pm 0.031$ | 0.210 |
|  |  | Avg.\%RSD | 0.418 | Avg. \%RSD | 0.304 |

Table 6: Results of precision after variation in Analyst

| Analyte | Amount taken <br> $(\mu \mathrm{g} / \mathrm{ml})$ | Analyst I <br> Avg Amount <br> Found <br> $(\mu \mathrm{g} / \mathrm{ml})$ | \%RSD | Analyst II <br> Avg Amount <br> Found <br> $(\mu \mathrm{g} / \mathrm{ml})$ |
| :--- | :--- | :--- | :--- | :--- | :--- |
| PQP | 9.963 |  | 9.95 |  |
| DHA | 10 | 9.89 | 0.208 | 0.361 |

## Intermediate precision

1) Variation in days: Intra-day and Inter-day Precision

In the intra-day studies, 3 replicates of 3 different concentrations ( 5 , $10,15 \mu \mathrm{~g} / \mathrm{ml}$ ) of PQP and ( $5,10,15 \mu \mathrm{~g} / \mathrm{ml}$ ) of DHA were analyzed in a day and percentage RSD was calculated.

For the inter day variation studies, 3 replicates of different concentrations were analyzed on 3 consecutive days and percentage RSD were calculated. Results are shown in Table 5.

## 2) Variation in Analyst

## Specificity and Selectivity

The analytes should have no interference from other extraneous components and be well resolved from them. Specificity is a procedure to detect quantitatively the analyte in presence of component that may be expected to be present in the sample matrix, while selectivity is the procedure to detect qualitatively the analyte in presence of components that may be expected to be present in the sample matrix.

The method is quite selective. There was no other interfering peak around the retention time of PQP and DHA; also the base line did not show any significant noise.

## Sensitivity (LOD and LOQ)

Sensitivity of the proposed method was estimated in terms of Limit of Detection (LOD) and Limit of Quantitation (LOQ). LOD was found to be $1.15 \mu \mathrm{~g} / \mathrm{ml}$ and $0.329 \mu \mathrm{~g} / \mathrm{ml}$ for PQP and DHA respectively.

LOQ was found to be $1.39 \mu \mathrm{~g} / \mathrm{ml}$ and $0.997 \mu \mathrm{~g} / \mathrm{ml}$ for PQP and DHA respectively.

## Robustness

Robustness of the proposed method was assessed by making deliberate changes in flow rate, and proportion of mobile phase which was performed by injecting sample solution containing 10 $\mu \mathrm{g} / \mathrm{ml}$ of PQP and $10 \mu \mathrm{~g} / \mathrm{ml}$ of DHA.

And chromatographic resolution between two drugs was evaluated. No significant change was observed during the study. Results are shown in Table7.

Table 7: Results of Robustness of PQP and DHA

| Parameters | $\begin{gathered} \text { PQP } \\ (10 \mu \mathrm{~g} / \mathrm{ml}) \end{gathered}$ |  |  | \%RSD |  | $\begin{gathered} \text { DHA } \\ (10 \mu \mathrm{~g} / \mathrm{ml}) \end{gathered}$ | \%RSD |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | RT |  | Peak Area |  | RT |  | Peak AREA |  |
| Mobile phase Composition (v/v) (Ammo. Acetate Buffer pH 4.6:MeOH) |  |  |  |  |  |  |  |  |
| 20:80 | 2.05 |  | 9.73 |  | 2.47 |  | 4.74 |  |
| 15:85 | 2.02 |  | 9.77 | 1.22 | 2.46 |  | 4.90 | 1.13 |
| 10:90 | 2.09 |  | 9.83 |  | 2.44 |  | 4.68 |  |
| Flow Rate Variation |  |  |  |  |  |  |  |  |
| 1.1 ml/min | 2.15 |  | 9.79 |  | 2.56 |  | 4.75 |  |
| $1.2 \mathrm{ml} / \mathrm{min}$ | 2.02 |  | 9.84 | 0.457 | 2.46 |  | 4.79 | 0.613 |
| $1.3 \mathrm{ml} / \mathrm{min}$ | 1.94 |  | 9.88 |  | 2.35 |  | 4.73 |  |

Table 8: Summery of Validation Parameter for RP-HPLC Estimation of PQP and DHA

| Parameters | PQP | DHA |
| :--- | :--- | :--- |
| Range | $5-25 \mu \mathrm{~g} / \mathrm{ml}$ | $5-25 \mu \mathrm{~g} / \mathrm{ml}$ |
| Linearity <br> Regression equation $(\mathrm{y}=\mathrm{mx}+\mathrm{C})$ and <br> correlation coefficient <br> LOD | $\mathrm{y}=1.133 \mathrm{x}-1.459$ | $\mathrm{y}=0.690 \mathrm{x}-0.923$ |
| LOQ | $\mathrm{R}^{2}=0.995$ | $\mathrm{R}^{2}=0.999$ |
| Recovery (\% RSD) | 1.15 | 0.329 |
| Precision (\% RSD) | 3.49 | 0.997 |
| Repeatability | 0.173 | 0.093 |
| Intra-Day (n=3) |  |  |
| Inter-Day (n=3) | 0.168 | 0.547 |
| Ruggedness (\% RSD) | 0.418 | 0.255 |
| $\bullet \quad 0.304$ | 0.444 |  |
| Analyst I (n=3) |  |  |
| Analyst II (n=3) | 0.288 | 0.208 |
| Sensitivity | 0.374 | 0.361 |
| Robustness | Sensitive | Sensitive |

## CONCLUSION

The research work concluded that the Antimalerial drugs (PQP and DHA) have been simultaneously analyzed with modern tools and techniques like HPLC. It can be further concluded that the RP-HPLC method being developed is reproducible, sensitive, precise, specific, and accurate. The mobile phase is simple to prepare and the sample recovery in formulation was in good agreement with their original claim. Statistical analysis proves that the developed method can be used for routine analysis of Piperaquine phosphate and Dihydroartemisinin in bulk.
Though the method shows good results there is need to study more about the chemical and thermal stability of Dihydroartemisinin as it have created critical problems during experimental work. Still there is wide scope for further research in this topic such as method development for the pharmaceutical formulation of PQP and DHA.

## CONFLICT OF INTERESTS

## Declared None

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## REFERENCES

1. WWW. chemspider.com
2. WWW. chemicalize.org
3. Hunga TY, Timothy MED, Kenneth FI. Measurement of Piperaquine in plasma by liquid chromatography with ultraviolet absorbance detection. J of chrom B 2003;791:93-101.
4. Vemula VR, Sharma PK, Singhvi I. HPLC method development and validation for simultaneous estimation of Arterolane maleate and Piperaquine phosphate in bulk and tablet dosage form. Int J of Res in Pharm and Chem 2013;3:3-6.
5. Dongre VG, Karmusea PP, Ghugare PD, Gupta M, Nerurkar B, Shaha C, et al. Characterization and Quantitative determination of impurities in Piperaquine phosphate by HPLC and LC/MS/MS. J of Pharm and Biomed Analysis 2007;43:186-95.
6. Reddy GS, Reddy SP, Reddy SL. Development and validation of a stability indicating liquid chromatographic method for simultaneous estimation of Arterolane maleate and Piperaquine phosphate in combined dosage form. Oriental J of Chem 2013;29:1371-80.
7. Muhia DK, Mberu EK, Watkins WM. Differential extraction of Artemether and its metabolite Dihydroartemisinin and determination by high performance liquid chromatography. J of Chrom B 1994;660:196-9.
8. Modi H, Patel P, Baria H, Patel S, Kabani N. Development and validation of RP-HPLC methods for determination of Dihydroartemisinin and Piperaquine phosphate in bulk and their pharmaceutical dosage form. J Inventi Rapid Pharm Analysis and Quality Assurance 2012;2:9-13.
9. USP medicines compendium (https:// mc.usp.org). Dihydro artemisinin:performance-based monograph. J Chromatogr.
10. Thomas CG, Ward SA, Edwards G. Selective determination of Artemether and its major metabolite Dihydroartemisinin in plasma by high performance liquid chromatography with UV detection. J of Chroma 1992;583:131-6.
11. Gartland J, Cheshire UK, Accucore H. Thermo Fisher Scientific, Runcorn, Analysis of Artesunate and Dihydroartemisinin using a core enhanced technology column, Application note, anccscetartdihy Thermo Fisher Scientific Inc. J Chromatogr 2011; 1011.
12. Hanpithakponga W, Kamanikoma B, Dondorpa AM, Singhasivanona BP, Whitea NJ, Daya NPJ, et al. A liquid chromatographic tandem mass spectrometric method for determination of Artesunate and its metabolite Dihydroartemisinin in human plasma. J of Chrom B 2008;876:61-8.
13. Terish JD, Kumar S, Ramesh N, Sasi JSL. Development and validation of high-performance liquid chromatography-tandem mass spectrometric method for simultaneous determination of Artemether and its metabolite Dihydroartemisinin in human plasma:a pharmacokinetic study application. Eur J of Experimental Bio 2011;1 (3):169-79.
14. Sreevidya TV, Narayana B. Spectrophotometric determination of Artemisinin and Dihydroartemisinin. Indian J of Chemical Technology Jan 2008;15:59-62.
15. Snyder LR, Kirkland JJ. Practical HPLC method development. 2nd ed. John wiley:sons;1997.
16. Ewing GW. Instrumental methods of Chemical Analysis. 5th ed. Mcgraw Hill international editions:Chemistry series;1985.
17. Meyer VR. Practical high performance liquid chromatography. 4th ed. John wiley and sons;2004.
18. Sethi PD. High performance liquid chromatographyquantitative analysis of pharmaceutical formulation. 1st ed. CBS publishers and distributors;2001.
19. Lindsay S. Analytical Chemistry by open learning HPLC. 2nd ed. Wiley India pvt. Ltd.
20. ICH guidelines, validation of analytical procedures:text and methodology, q2 (r1) Nov. J Chromatogr 2005:1-13.
21. ICH guidelines, text on validation of analytical procedures, q 2 a Oct. J Chromatogr 1994.
22. ICH guidelines, validation of analytical procedure:methodology, q2b Nov. J Chromatogr 1996.
