

Original Article

ANTIBACTERIAL ACTIVITY OF FLAVONOID FROM KEPEL (*STELECHOCARPUS BURAHOL*) LEAVES AGAINST *STAPHYLOCOCCUS EPIDERMIDIS*

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Received: 11 Apr 2017 Revised and Accepted: 31 Aug 2017

ABSTRACT

Objective: The objective of this research was to separate flavonoids of kepel (*Stelechocarpus burahol*) leaves from methanol extract as antibacterial agent against *Staphylococcus epidermidis* that have a role in body odor.

Methods: The methanol extracts of kepel was fractionated in methanol: water (7:3), *n*-hexane and chloroform consecutively. Methanol: water (7:3) extract with the highest flavonoids content, fractionated by silica gel column chromatography (isocratic elution, with *n*-buthanol: methanol: acetic acid (1:8:1) as eluent) to produce 7 fractions. All fractions were tested for antibacterial activity with a microdilution method. The most active fractions was determined using UV-VIS spectrophotometer (Shimadzu, Japan) and FTIR (Bruker, Germany).

Results: Fraction V was the most active fraction with minimum inhibitory concentration (MIC) 0.06 mg/ml and minimum bactericidal concentration (MBC) of 0.50 mg/ml. Fraction V was further separated by preparative thin layer chromatography (TLC) and gave three fractions. Fraction V3 was the most active fraction with MIC 1.00 mg/ml and MBC of 2.00 mg/ml. Identification of fraction V3 based on assessments on ultraviolet-visible and infrared spectrum showed the maximum wavelength at 327 nm. These results indicate a transition $\pi \rightarrow \pi^*$ and $n \rightarrow \pi^*$ generated from the chromophore conjugated C = C and C = O. Based on the allegations of the functional group obtained, then alleged that in Fraction V3 containing flavones compounds.

Conclusion: These results suggest that flavonoid of *S. burahol* leaves extracts is potential as antibacterial agents against *S. epidermidis* and therefore justifies their usage in traditional medicine for the treatment of body odor.

Keywords: Fractionation, Flavonoid, *Stelechocarpus burahol*, Antibacterial, *Staphylococcus epidermidis*

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DOI: <http://dx.doi.org/10.22159/ijpps.2017v9i10.19071>

INTRODUCTION

Axillary body odor is individually specific and potentially a rich source of information about its products. Odor individuality partly results from genetic individuality, but the influence of ecological factors such as eating habits are another main source of odor variability [1]. The generation of malodour on various sites of the human body is caused by the microbial biotransformation of odourless natural secretions into volatile odorous molecules. On the skin surface, distinctive odours emanate, in particular, from the underarm (axilla), where a large and permanent population of microorganisms thrives on secretions from the eccrine, apocrine and sebaceous glands [2]. The sweat issued by someone very involved in the onset of body odor. Infection of the apocrine glands that produce sweat by bacteria, can play a role in the decay process. Bacteria induced the body odor are *Staphylococcus epidermidis*, *Corynebacterium acne*, *Pseudomonas aeruginosa*, and *Streptococcus pyogenes* [3]. Several herbals have been used as deodorant to treat axillary body odor.

Kepel (*Stelechocarpus burahol* (Blume) Hook. f. and Thomson) is one of fruit tree originally found in Indonesia, included in the family Annonaceae. Kepel is a plant that has been used traditionally as a fragrance especially among the palaces in Indonesia. Part of the plant used for medicinal are leaves, bark and fruit [4]. Consumed the fruit can reduce the smell of sweat, breath and urine [4]. It was reported that fruit pulp had potential as a natural deodorant since it had adsorbent function and probiotic activation by increase the growth of Bifidobacteria so the population of odorant producing bacteria will be reduced [5]. The fruit pulp of kepel also reported contains high antioxidant [6], and contains alkaloids, flavonoids, polyphenols, triterpenoids, saponins and quinones as well have the effect of anti-implantation [7]. Kepel's leaf contains terpenoids and flavonoids [7]. Leaves extracts of kepel contains flavonoids include

auron, flavanones and flavanols that can be used for antibacterial [9]. Ethanol and *n*-hexane extracts of kepel's leaf has potential for lowering uric acid levels in mice [10, 11]. Phenylalanine ammonia lyase (PAL) enzyme activity and anthocyanins content in young leaves were relatively higher than the mature and the medium leaves, but mature kepel leaves have the highest flavonoid content and total chlorophyll than others. Increased rainfall will decrease the content of flavonoid in kepel leaves, but increased anthocyanin, PAL activity and total chlorophyll. Therefore kepel mature leaves can be used as raw material for medicine [12].

Utilization of antibiotics will usually make the bacteria become resistant and permanent nature multiplies within its host. Therefore, it is necessary to investigate the antibacterial active compounds in kepel leaves extracts that capable to inhibit *S. epidermidis* growth. This study aimed to identify the flavonoid from kepel leaves extract that has potential as an antibacterial against *S. epidermidis*. *S. epidermidis* is one of the suspected bacteria that cause body odor. Furthermore, *S. epidermidis* are generally resistant to penicillin, amoxicillin and methicillin [13], so it's very important to investigate alternative compounds that can inhibit the growth of this bacteria.

MATERIALS AND METHODS

Collection of plant material

Stelechocarpus burahol leaves were collected in April 2011 from Cilacap, Central of Java, Indonesia. The plant identification and authentication certificate issued by the Herbarium LIPI. Voucher specimen (No. voucher specimen: BMK0199092016) is deposited at Tropical Biopharmaca Research Center, Bogor Agricultural University, Indonesia. The leaves taken from plant of kepel 3 y old, and mixed among the young leaves and old leaves. The materials were washed, wet sortation, dried at 50 °C in oven dryer and grinded into powder.

Chemicals and reagents

All chemicals were purchased from Sigma-Aldrich Co., Inc. (St. Louis, MO, USA). All solvents used were of HPLC grade, obtained from E-Merck Ltd. (Darmstadt, Germany). The chromatography plates were TLC aluminium plated precoated with silica gel 60 PF₂₅₄ (20 x 10 cm, 0.2 mm thick) obtained from E-Merck Ltd. (Darmstadt, Germany). The test organism used in this study was *Staphylococcus epidermidis* ATCC 12228. Bacterial culture agar and broth were purchased from E-Merck Ltd. (Darmstadt, Germany). Tetracycline and trichloro-carbon (TCC) were purchased from Pharmaceutical Company (Jakarta, Indonesia).

Extract preparation and fractionation

Extraction was performed according to the methods of Sukadana with a little modification [14]. The dried leaves powder of kepel was macerated with methanol (1 g dried leaf powder: 10 ml methanol, w/v) for 24 h at room temperature. Solvent was evaporated by rotary evaporator.

The methanol extract of kepel leaves was suspended in methanol: water (7: 3) mixture and then partitioned with 25 ml *n*-hexane. The layer of extract suspended in *n*-hexane was evaporated, while the methanol: water partitioned again with 25 ml chloroform in order to get the methanol: water extracts and chloroform extracts. Each of the extract was evaporated to remove the solvent and analyzed with flavonoid phytochemical test [15]. Methanol: water (7:3) extract with the highest flavonoids content, fractioned by silica gel column chromatography (isocratic elution, with *n*-butanol: methanol: acetic acid (1:8:1) as eluent) to produce 7 fractions. Fractions that showed similar *R_f* value and TLC profile were combined. All fractions were tested for antibacterial activity with a microdilution method. The fraction that has the highest antibacterial activity further separated using TLC preparative to obtain the most active fractions.

Antibacterial assay

Antibacterial assay was performed by following procedure of Batubara et al. [16]. The test organism used in this study was *Staphylococcus epidermidis* ATCC 12228. The medium used in this study was trypticase soy broth (TSB). Sterilized medium (100 µl), sample [40 µl, serial concentration, diluted in dimethylsulfoxide (DMSO) 20 %] or positive control (40 µl), and inoculum (5 µl) were added to each well of a 96-well plate. The inoculum was prepared at the concentration of 10⁻² CFU/ml. *S. epidermidis* was incubated in the medium for 48 h at 37 °C. The extract concentration at which there was no visually detectable bacterial growth was described as the minimum inhibitory concentration (MIC). Next, 10 µl of each medium with no visually detectable bacterial growth was inoculated in 100 µl of fresh medium for 48 h at 37 °C. The concentration at which there was no bacterial growth after the second inoculation

was described as the minimum bactericidal concentration (MBC). The positive controls used were tetracycline and TCC. The antibacterial assay was conducted a minimum of three times, each at different times.

Determination active compounds

Determination of compounds contained in the most active fractions was performed using UV-VIS spectrophotometer (Shimadzu, Japan) and FTIR (Bruker, Germany).

RESULTS AND DISCUSSION

Extraction and investigation of flavonoid content in leaf extracts

Extraction was made with methanol as a solvent, referring to the polar nature of methanol in extracting the flavonoid compound. Generally, flavonoids are soluble in polar solvents such as ethanol, methanol, *n*-butanol, acetone, dimethyl sulfoxide, and water. The presence of sugar bound to the flavonoid cause flavonoids more soluble in water [17]. Maceration process is particularly advantageous in the isolation of natural compounds because in the immersion will occur break down cell walls and membranes cell. This happens due to the pressure difference between the inside and outside of the cell, so that the secondary metabolites present in the cytoplasm will be dissolved in a solvent. The yield of leaf extracts was 15.15 % (w/w).

The methanol extract of the leaves was separated by a liquid-liquid extraction. This leads to mass transfer from the originating solvent to the solvent extraction. Polar components will be distributed on a methanol: water, semipolar components will be distributed in chloroform, and nonpolar components will be distributed on *n*-hexane. The components that distributed in *n*-hexane and chloroform are fat, terpenes, chlorophyll, and xantofil [17]. The yield of *n*-hexane, chloroform and methanol: water extract obtained from the leaf extracts were 31.98, 11.81, and 27.82 % (w/w) respectively. The yield of *n*-hexane extract higher compared to the other showed in the leaf of kepel contain nonpolar components. Based on qualitative flavonoid test, methanol: water extracts have the high flavonoid content. *n*-Hexane extract has no flavonoids content (table 1). Flavonoids are hydroxylated phenolic substances known to be synthesized by plants in response to microbial infection and they have been found to be antimicrobial substances against a wide array of microorganisms *in vitro*. Several flavonoids, including flavones such as apigenin, flavonols such as galangin, flavonol glycosides, isoflavones, flavanones, and chalcones have been shown to possess potent antibacterial activity [18]. Flavonoids have the ability to complex with proteins and bacterial cells mainly with nucleophilic amino acids. This complex often leads to inactivation of the protein and loss of its functionality [19].

Table 1: Flavonoid present in leaves extracts of *S. burahol*

Extracts	Results
<i>n</i> -hexane	-
Chloroform	+
Methanol: Water (7:3)	+++

+indicating positive test and -indicating negative test

Antibacterial activity

The methanol: water extracts have the high flavonoid content, so this extract fractioned by silica gel column chromatography (isocratic elution, with *n*-butanol: methanol: acetic acid (1:8:1) as eluent). Fractions that eluted be precious on column chromatography separation showed antibacterial activity (table 2). Fraction V is the most active fraction with MIC of 0.06 mg/ml and MBC of 0.50 mg/ml. The MIC and MBC value of fraction V is equal to tetracycline that means the antibacterial activity of Fraction V and tetracyclines are same. When compared with the TCC, the Fraction V had higher activity than TCC. Tetracycline and TCC are a group of broad-spectrum anti-infective used commercially in soaps, shampoos, deodorants and other household products.

Fraction V as the most active fraction is a mixture of 3 spot, thus further separated using preparative TLC with *n*-butanol: acetic acid: water (4:1:5) to separate the flavonoid [17]. Based on the separation obtained three fractions. All fractions subjected to antibacterial assay. The results showed the fraction V3 was the most active fraction compared with other fractions (table 3). The MIC and MBC value of Fraction V3 was higher than tetracycline and TCC, which means the fraction V3 has a lower activity than tetracycline and TCC. Activities Fraction V higher than the Fraction V3, it is suspected that there were several compounds mixed in Fraction V are potential as antibacterial. The complex mixtures of Fraction V composed mainly of flavonoid caused a synergistic effect as antibacterial against *S. epidermidis*. MIC is the lowest concentration of an antimicrobial inhibit the visible growth of a microorganism.

MICs are important in diagnostic laboratories to confirm resistance of microorganisms to an antimicrobial agent and also to monitor the activity of new antimicrobial agents [20]. Aliyannis *et al.* [21], classifying extracts of plant material on the basis of MIC is as follows:

-strong inhibition: MIC<500 µg/ml; -Moderate inhibition: MIC from 500 µg/ml to 1500 µg/ml; -Weak inhibition: MIC>1500 µg/ml. Therefore, antibacterial activity of methanol: water (7:3) extract and the fractionation results included in strong inhibition classification.

Table 2: Antibacterial activity of fractionation I-VII from silica gel column chromatography

Fraction name	MIC (mg/ml)	MBC (mg/ml)
I	2.00	>2.00
II	0.12	2.00
III	0.25	1.00
IV	0.25	2.00
V	0.06	0.50
VI	>2.00	>2.00
VII	>2.00	>2.00
Tetracycline*	0.06	0.50
TCC*	0.12	1.00

*positive control

Table 3: Antibacterial activity of TLC preparative fraction

Fraction name	MIC (mg/ml)	MBC (mg/ml)
FractionV1	2.00	>2.00
Fraction V2	2.00	>2.00
Fraction V3	1.00	2.00
Tetracycline *	0.06	0.50
TCC*	0.50	1.00

*positive control

Bioactive compounds

Fraction V3 (the most active fraction) were obtained from the preparative TLC analyzed using UV-VIS spectrometer. Imaging was done with the wavelength changes of 2 nm. The results of the UV-Vis analysis showed Fraction V3 has a maximum wavelength at 327 nm. UV-VIS spectrum of Fraction V3 is shown in fig. 1. These results

indicate a transition $\pi \rightarrow \pi^*$ and $n \rightarrow \pi^*$ generated from the chromophore conjugated C = C and C = O. Compounds that have the transition $\pi \rightarrow \pi^*$ and $n \rightarrow \pi^*$ will absorb light in the UV region at a wavelength of 200-400 nm [22]. Absorption at 200-400 nm indicated the presence of chromophore group, which is one of the characteristics of flavones [23]. According to Markham [17], the maximum wavelength range 310-350 nm is the flavonoid class of flavone.

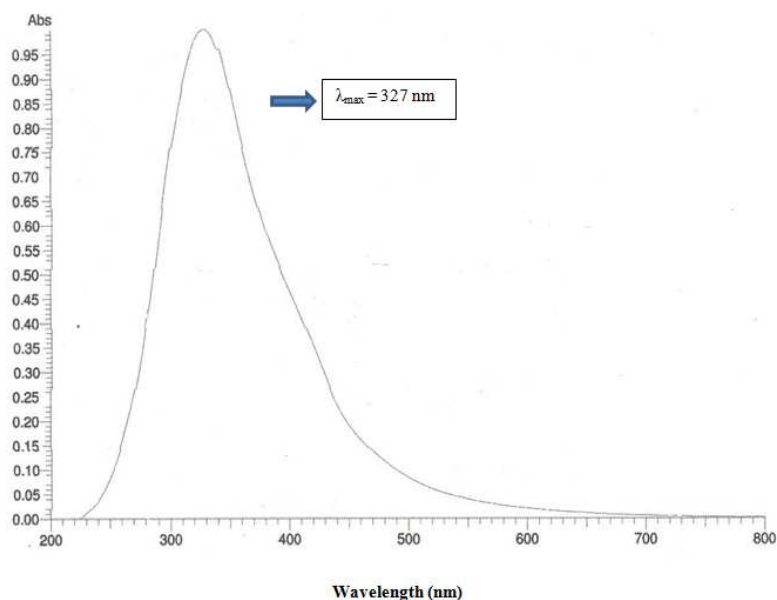


Fig. 1: UV-VIS spectrum of fraction V3

FTIR spectrum of Fraction V3 can be seen in table 4 and fig. 2. Based on the FTIR spectrum of Fraction V3 contain some functional groups such as-OH (3406.00 cm^{-1}) which is supported also by the emergence of regional absorption at wavenumber 1109.55 cm^{-1} to C-

O bond. Aliphatic stretching C-H appeared at 2924.11 cm^{-1} and is reinforced by the presence of bending regional absorption at wave numbers 1398.31 cm^{-1} . The functional groups of the C = C aromatic demonstrated by the presence of regional absorption at wave

number 1628.59 cm^{-1} and there is a stretching $\text{C}=\text{O}$ at 1735.83 cm^{-1} region. Based on the allegations of the functional groups obtained, then alleged that in Fraction V3 containing flavone compounds (fig. 3). Hazra *et al.* [24] reported pentahydroxy flavones from the seeds of *Mimusops elengi* Linn showed strong inhibitory activity against

Gram positive and Gram negative bacteria. The flavonoid class of flavone from kepel leaves also reported by Sunarni *et al.* [25] as DPPH scavenger. They were identified flavon with hydroxyl group on C-3, C-7, C-3', C-4' and methyl on C-5 and showed activity with an EC_{50} value of 6.43 $\mu\text{g}/\text{ml}$.

Table 4: FTIR characteristic bands of fraction V3

Wave numbers (cm^{-1})	References*	Functional groups
3406.00	3200-3450	stretch O-H
2924.11	2850-3000	stretch C-H aliphatic
1628.59	1500-1675	stretch $\text{C}=\text{C}$ aromatic
1735.83	1650-1900	stretch $\text{C}=\text{O}$
1398.31	1300-1475	bend C-H aliphatic
1109.55	1000-1300	stretch C-O

*Sources: Creswell *et al.* [22] and Field *et al.* [26]

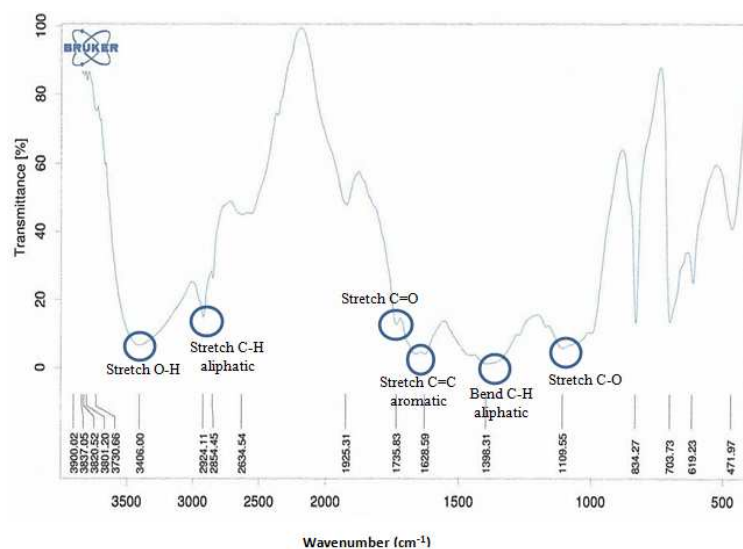


Fig. 2: FTIR spectrum of fraction V3

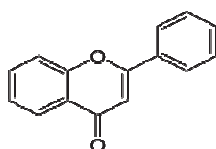


Fig. 3: Structure of flavone compounds

CONCLUSION

The fractionation of methanol: water (7:3) extract from kepel leaves with isocratic elution (*n*-buthanol: methanol: acetic acid 1:8:1) using column chromatography resulted 7 fractions. V3 fraction is the most active with MIC 1.00 mg/ml and MBC 2.00 mg/ml. The identification results using UV-VIS and FTIR alleged that the active compounds as antibacterial is flavone group.

ACKNOWLEDGEMENT

The authors would like to acknowledge the financial support for the project received from Tropical Biopharmaca Research Center, Bogor Agricultural University, Indonesia.

AUTHOR CONTRIBUTION

Susi Indariani, Aisyah Hidayat, Latifah K Darusman and Irmanida Batubara contributed equally in collection of plant material, experiment design, performing the experiment and data compilation. All of the authors wrote, read and approved the final manuscript.

CONFLICTS OF INTERESTS

All authors have none to declare

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- How to cite this article**
- Susi Indariani, Aisyah Hidayat, Latifah K Darusman, Irmanida Batubara. Antibacterial activity of flavonoid from kepel (*Stelechocarpus burahol*) leaves against *Staphylococcus epidermidis*. Int J Pharm Pharm Sci 2017;9(10):292-296.