

Original Article**DETERMINATION OF TOTAL CARBOHYDRATES, FLAVONOIDS, ORGANIC ACIDS, MACRO-AND MICROELEMENTS IN WOLFBERRY (*LYCIUM BARBARUM L.*) FRUIT CULTIVATED IN ALBANIA**BLEONA TUPA^{1*}, ELENA V. USPENSKAYA¹, OLGA G. POTANINA², NIKOLAY N. BOYKO³, RAUDAS A. H. NASSER³¹Department of Pharmaceutical and Toxicological Chemistry, Peoples Friendship University of Russia (RUDN University), 6 Miklukho-Maklaya St, Moscow, 117198, Russian Federation, ²Moscow State University Named After M. V. Lomonosov, Leninskie Gory, 1, Moscow, 119991, Russian Federation, ³Department of Pharmaceutical Chemistry and Pharmacognosy of the Center for Collective Use (Research and Education Center) "Center for Quality Control of Drugs", Peoples' Friendship University of Russia (RUDN University), 6 Miklukho-Maklaya St, Moscow, 117198, Russian Federation
Email: tupableona@gmail.com*Received: 09 Feb 2022, Revised and Accepted: 02 Apr 2022***ABSTRACT**

Objective: The purpose of the study is to determine the biological activity components content of wolfberry (*Lycium Barbarum L.*) fruit originating from Albania. To unify the requirements for quality control of medicinal plant raw materials, it is advisable to study the qualitative composition and quantitative content of the components of this plant that determine the complexity of biological action-anti-inflammatory, neuroprotective, anti-cancer, vision-improving, and reproduction-enhancing effects.

Methods: *Lycium barbarum L.* fruits were analyzed for the content of total carbohydrates and fructans by spectrophotometry method. The organic acids, one flavonoid, and one scopoletin were identified and quantified by the HPLC method. Macro-and microelements were analyzed by ICP-OES.

Results: The results of the spectrophotometric analysis showed that total carbohydrate content and fructans content lie in the range of 21.763%-70.384% and 19.90-20.25%. Rutin, the main flavonoid compound in *Lycium barbarum L.* fruits, and scopoletin, a coumarin compound, contents lie respectively in the range of 2.10-5.48 mg/g and 0.48-0.76 mg/g. Potassium (K) is the predominant element in fruits, the content of which was 6740.75 µg/g.

Conclusion: *Lycium barbarum L.* fruit is a rich source of important biologically active substances. Further, the resulting data are going to be used to establish a monograph for *Lycium barbarum L.* fruits.

Keywords: *Lycium barbarum L.*, Extracts, Polysaccharides, Fructans, Flavonoids, Organic acids, Mineral elements

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INTRODUCTION

Scientific interest in *Lycium barbarum L.* was manifested in the middle of the 20th century and was associated with numerous data testifying to the biological effects of this plant and instrumental capabilities of isolation and analysis of biologically active substances [1]. *Lycium barbarum L.* fruits (wolfberry, goji berry, Duke of Argyll's tea plant, Tibetan barberry) included in the *Solanaceae* family are a traditional food, therapeutic/medicinal plants of South-East Asia which have also become increasingly popular in Europe and North America [2]. The increasing therapeutic use of the plant in therapy has urged scientists to investigate its biological potency [3]. All varieties of *Lycium* belong to the taxonomic rank: division-flowering, class-dicotyledonous, family-Solanaceae, tribe-Lycieae, genus-*Lycium*, including more than 90 species. In China, are found seven of these species. According to [4] in the world are registered only 35 *Lycium* species for use in dietetics and medicine: in America and Africa are found 85% of *Lycium* species, of which only 26% are used. Of fourteen species in Eurasia-only nine species are used. and *L. Chinense*-in China. *Lycium barbarum L.* is a perennial deciduous branching shrub 1-2.5 m tall with long. The fruit is a red berry, 8-18 mm long. The berries are harvested ripe, dried to the state of a wrinkled pericarp, and exposed to direct sunlight until the exocarp becomes dry, solid, and the pulp is soft [5]. For a long time, fruits, leaves, bark, roots, and young shoots of *Lycium* have been used as components of food and biologically active substances. *Lycium barbarum L.* fruits contain polysaccharides, carotenoids, polyphenols, flavonoids, alkaloids, amides, peptides, anthraquinones, coumarins, lignanoids, terpenoids, steroids, organic acids, glycolipids and a stable analog of ascorbic acid named 2-O-β-d-glucopyranosyl-l-ascorbic acid (AA-2βG). The ability of polyphenols and flavonoids are well known and may prevent skin cancer development due to the reduction in oxidative stress and alteration of cytokine activity. This cell-protective effect is highly desirable in the treatment of Alzheimer's disease [6].

The biological activity of *Lycium* is mainly due to *Lycium barbarum* polysaccharides (LBP), which have demonstrated immunoregulatory and neuroprotective properties, antioxidant, anti-aging, antidiabetic, and anticancer effects [6]. The LBP is a novel prebiotic candidate for *Bifidobacterium* and *Lactobacillus* [7]. Prebiotic inulin-type fructans (polysaccharides such as β-2,1-linked fructans), present in LB fruits, can modify gut microbiota, increasing peristalsis, maintaining the balance of the microflora of the body, and improving diabetes [8]. In *Lycium barbarum*, fruits have been identified several flavonoids like the aglycones myricetin, quercetin, kaempferol, rutin, gentistic acid, and quercetin. Flavonoids from *Lycium barbarum L.* protect the blood cells and mitochondria against oxidative damages, according to [9]. *Lycium barbarum L.* (LB) also contains organic acids, such as citric, malic, fumaric, and shikimic acid. Organic acids can maintain the quality and nutritive value of fruits. They are used as antioxidants and preservatives. Also, several minerals of importance have been found in LB. [10]. For optimal functioning of the body, is necessary an adequate daily amount of minerals. The quality standards for *Lycium* plant (Fruit, Root Bark) are presented in the *Pharmacopoeia* monographs of China, Korea, Japan, USA, Great Britain and, as a rule, include sections: description, identification, purity, assay, containers, and storage [11].

A comparative analysis of the requirements for *Lycium barbarum L.* demonstrates a tendency to harmonize approaches to quality control of medicinal plant raw materials. However, there is significant variability in the qualitative and quantitative composition of biologically active components in *Lycium* plant raw materials of various origins [12]. Thus, the purpose of this research is to develop methods for the quantitative determination of *Lycium barbarum L.* biologically active components-total carbohydrates, flavonoids, organic acids, and mineral elements in dried *Lycium barbarum L.* fruits cultivated in Albania for unifying *Pharmacopoeia* requirements.

MATERIALS AND METHODS

Plant materials

Lycium Barbarum L. dried fruits were purchased on a farm in Western Albania, National park Divjake-Karavasta (N 48° 13'23.2" E 25° 11'42.0), during the summer season in 2020. The plant raw

material was identified by the most important external key features: the form of life, the type of fruit, and by microscopy identification (table 1) according to American Herbal Pharmacopoeia (fig. 1). *Lycium Barbarum* L. dried fruits were placed into separate plastic bags and vacuum sealed for storage and further use.

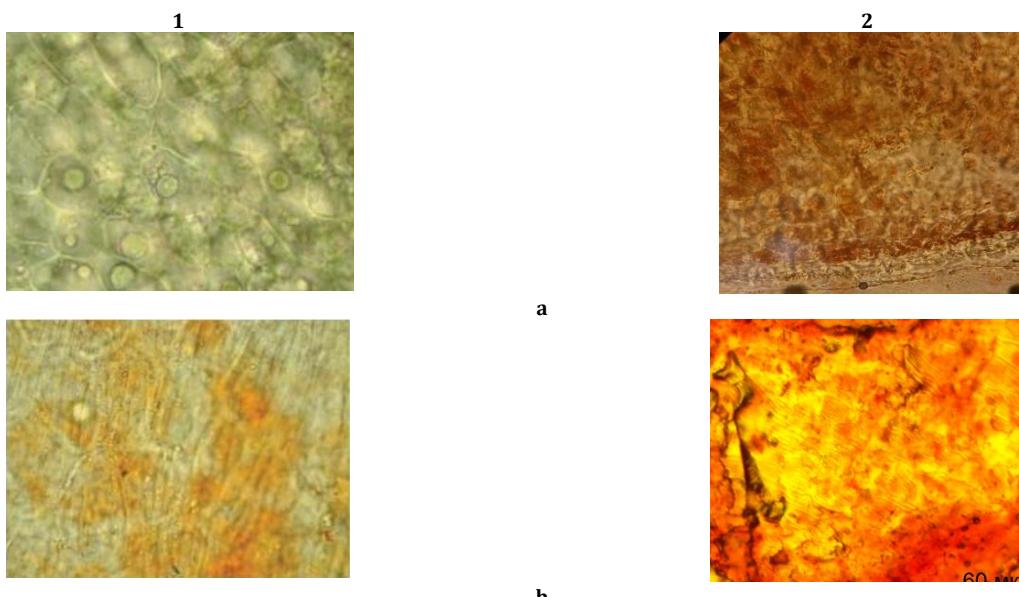


Fig. 1: Light microscopy identification of *Lycium barbarum* fruits from 1) American Herbal Pharmacopoeia and 2) Microscopy *Lycium barbarum* sample. a-endosperm of thin-walled cells with oil droplets; b-exocarp showing prominent parallel cuticular striations

Chemicals and standards

Standards of polysaccharides D-fructose, D-glucose were of analytical grade (>95 % purity), standards of organic acids tartaric acid, malic acid, citric acid, ascorbic acid were of analytical grade (>99 % purity), rutin ≥94%, and scopoletin ≥99% were obtained from Sigma-Aldrich Chemical Co. (St. Louis, MO, USA). Nitric acid (Trace Metal Grade), Hydrogen peroxide, H₂SO₄ ACS reagent 95-98%, lead acetate ACS reagent, ≥99%, sodium phosphate ACS reagent ≥99.0%, resorcinol Reagent Plus 99%, hydrochloric acid ACS reagent 37%, were from Sigma-Aldrich Chemical Co.

Instruments used

Bathwater LOIP LB-162, laboratory-scale Sartorius LP 1200S, Liquid Chromatograph Pro Star Varian, Cary 100 UV-Vis Spectrophotometer, Pharmaceutical refrigerator Sanyo MPR-414F, temperature-controlled microwave (MW oven/system) (Milestone Ethos), Agilent 1200 HPLC with DAD, ICP-OES (Varian 720-ES, axial view of plasma).

Experimental

Determination of the total carbohydrate (TC) content of the extracts

The *Lycium* plant raw materials extracts were prepared as described in [13] for *Tussilaginis farfarae* folia and in a previous article [14]. The extract was obtained from 0.500 g of raw materials (d = 7 mm) in a mixture of H₂O: H₂SO₄conc. (5.7:1) using by water bath with a reverse refrigerator (30 min) was filtered into a flask V = 100 ml with purified water. A blank determination was also carried out. The determination of TC was carried out by a spectrophotometric method with a phenol/sulfuric previously color reaction [15]. A mixture of 5% phenol solution and H₂SO₄conc. (1:5) was added to test tubes containing solutions of the standard sample (SS, glucose standard 4·10⁻³% in water), blank solution (BS), and extract (test solution) in volumes of V=1 ml, followed by a dilution of H₂SO₄ conc. to 10 ml. The absorbance measurements were performed after 30 min, at a wavelength of 490 nm, using a Cary UV-Visible

Spectrophotometer Model 100 (USA) [16-18]. The TC content (X, %) was calculated by using the formula [14]:

$$X = \frac{A \cdot a_0 \cdot P \cdot V}{A_0 \cdot a \cdot (100 - W)} \cdot 100\% \quad \dots \quad (1)$$

Where, A is TC absorbance; A₀ is SS absorbance; a is raw materials sampling weights, g; a₀ is SS sampling weights, g; V is dilution volume for TC; W is moisture content of raw materials, %; P is the SS content, %.

β-2,1-Linked fructans of *Lycium barbarum* polysaccharide (LBP)

The hydrolytic cleavage of fructans into fructose

Extracts were obtained from a sample of 2.000 g of crushed raw materials (d=1 mm) in a boiling bath in two stages, including extraction of 60 ml of water (60 min) and 30 ml of water (30 min). The obtained extracts were combined in a flask V=100 ml, followed by the addition of 2.0 ml of 10% (w/v) of lead acetate aqueous solution and 4.0 ml of 5% (w/v) of sodium phosphate aqueous solution. The solution was filtered, discarded the first 10 ml, and diluted with water 20 times (solution A); SS-5.6·10⁻⁴% fructose solution; the reference sample (RS) is purified water.

The absorbance measurements of products formed after the cleavage of inulin with resorcinol in an acidic medium have performed at λ_{max}=483 nm: 5.0 ml of 0.1% alcohol resorcinol solution and 10.0 ml of 30% hydrochloric acid were added to 5 ml of solution A, followed by heating at 80 °C (20 min) and dilution to 25 ml HCl. A blank determination was also carried out [19-21]. The fructans content (X, %) was calculated by the formula (1).

Flavonoids and coumarins

Quantitative analysis of flavonoids and coumarins was carried out by reverse-phase high-performance liquid chromatography (RP HPLC) with a diode-array detector (DAD). The sample preparation technique consisted in crushing the *Lycium Barbarum* L. fruits

followed by double extraction with ethanol of different concentrations: 30, 50, 60, 70, 80, 96 % for 45 min and 15 min. The resulting extract was filtered into a 100 ml flask, brought to the mark with 70% ethanol [22]. Ethanol solutions of rutin (0.05, 0.005, 0.0005 %) and scopoletin (0.01, 0.001, 0.0001 %) were used as standard samples (SS). Quantitative analysis of flavonoids in the extracts was carried out by RP HPLC method using a Liquid Agilent 1200 chromatograph DAD, with an automatic dispenser, vacuum degasser, gradient pump, thermostat, and spectrophotometric diode-matrix detector of the same series manufactured in the USA. Chromatographic column: Nucleosil of Macherey-Nagel GmbH and Co. KG, Germany (100-3 C18, length-100 mm, inner diameter-4.6 mm, particle size-2.7 microns); temperature of the chromatographic column: +35.00 °C. HPLC analysis was carried out under the following conditions: mobile phase (a)-aqueous solution of formic acid (1%); (b)-ethyl alcohol 96% with the following gradient elution: 90% A and 10% B initially, decreased to 80% A in 10 min, decreased to 70% A in 20 min, decreased to 50% A in 30 min, decreased to 10% A in 40 min then maintained until 60 min. Flow rate 1 ml/min and sample volume 1 µl.

Organic acids

The determination of tartaric, malic, ascorbic, and citric acids was carried out by high-performance liquid chromatography (Shimadzu Prominence HPLC) with Restek HPLC column (l=300 mm, d=4.6 mm); A mixed solution of 0.1 mol/l H₃PO₄ and 0.1 mol/l KH₂PO₄-H₂O was used as the pretreatment mobile phase the mobile, pH 2.2, using a with a flow rate of 1 ml/min isocratic elution procedure; λ =210 nm. Method: the prepared homogenate from 50 g of raw materials in 500 ml of water was centrifuged for 10 min at 4000 g. The supernatant fluid was passed through a 0.45 microns filter [23].

Vitamin C

The content of vitamin C (ascorbic acid) was determined by redox titration of 0.001 mol/l with a titrant-solution of 2, 6-dichlorophenolindophenolate sodium [24, 25] by using the formula:

$$X = \frac{V \cdot 0.000088 \cdot K \cdot 300 \cdot 100 \cdot 100}{m \cdot 1 \cdot (100 - W)} \quad \dots \dots (2)$$

Where 1 ml of 0.001 M solution of titrant is equivalent to 0.0088 mg of ascorbic acid; V is the volume of titrant, ml; m-the weight of raw materials, g; W is moisture content of raw materials, %; K-the correction factor.

Macro-and microelements in *Lycium* plant raw materials

The composition of elements in the sample's raw materials was determined using inductively coupled plasma-optical emission spectrometry ICP-OES (Varian 720-ES, axial view of plasma).

(ICP-OES, Thermo Fischer Scientific, Bremen, Germany). The *Lycium Barbarum* L. fruits were mineralized by treatment with concentrated solutions of HNO₃ and H₂O₂ at 170 °C until digestion was complete [26, 27]. The content of any elements (which are analyzed by ICP-OES methods) X (ppm) was calculated by the formula:

$$X = \frac{C \cdot 50}{m} \quad \dots \dots (3)$$

Where C is the TC element concentration, µg/ml; 50 is dilution volume; m is the weight of raw materials, g.

Statistical analysis

Results are expressed as mean±SD. The statistical analysis was carried out using a one-way ANOVA analysis. The p-value of 0.05 or less was considered significant for all experiments.

RESULTS AND DISCUSSION

Determination of the total carbohydrate content of the extracts

Since carbohydrates are a crucial source of energy for the body, to avoid muscle breakdown, ketosis, and dehydration, it is necessary to take from 50 to 100 g of carbohydrates per day [28]. Several studies have shown the high concentration of carbohydrates in the fruits of *Lycium barbarum* [29-31]. In this paper, a quantitative assessment of the total carbohydrate content in the *Lycium barbarum* L. fruits was carried out using a simple, quick, sensible, reproducible method of phenol-sulfuric acid (PSA), developed by DuBois *et al.* [15]. The method used was based on the reaction of phenols with the oxidation products of monosaccharides. During the reaction with concentrated sulfuric acid, saccharides dehydrate and form furfural derivatives [32], which react with phenol and form colored complexes that absorb light in the visible range at $\lambda_{\text{max}}=490$ nm. After adding phenol and sulfuric acid, the color of the solutions was bright yellow and permanent and had a definite absorption peak [33-36]. To establish the completeness of the extraction of carbohydrates from the fruits of *Lycium barbarum*, the effect of the ratio of raw materials and extractant, the multiplicity of extraction, and the optimal extraction time were studied. The total carbohydrate content in *Lycium barbarum* dried fruits, in terms of glucose and absolutely dry raw materials, was determined in extracts, and the results are presented in table 1.

Table 1: Extraction conditions for the extraction of polysaccharides from the *Lycium barbarum* L. fruits

Unchangeable parameter	Variable parameter	The content of TC, % (n=3)
Degree of grinding, d, mm		
The ratio of raw materials: extractant 1:200	7	34.846±0.312
	5	31.998±0.421
	4	28.821±0.954
	3	21.763±0.847
The ratio of the mass of the raw material to the volume of the extractant		
The degree of grinding	1:25	25.547±0.672
7 mm	1:50	24.996±0.354
	1:125	27.434±0.165
Extraction time (min)		
Raw material ratio: extractant 1:125, grinding 7 mm	15	29.133±0.454
	30	33.479±0.193
	90	33.361±0.431
	120	31.814±0.268
Multiplicity of extraction		
The ratio of raw materials: extractant 1:125, the degree of grinding 7MM	1	30.364±0.354
	2	56.721±0.477
	3	70.384±0.489

The content of TC was performed in triplicate and the results are the mean of three values±standard deviation

After multiple extractions, it was concluded that the optimal conditions for the extraction of polysaccharides from the *Lycium barbarum* L. fruits are as follows: the degree of grinding of the

sample is d=7 mm, the ratio of solvents is 1:125, the best extraction time is 90 min. An increase in the extraction multiplicity contributes to an augmentation in the yield of reducing mono-saccharides from

the polysaccharide complex of fruits. An increase in the multiplicity of extraction contributes to an increase in the yield of reducing monosaccharides from the polysaccharide complex of fruits. It took

3 extractions to maximize the extraction of polysaccharides. Fig. 2 shows the absorption spectra of the standard (glucose) solution and the tested (extracts) samples at $\lambda=487\pm 3$ nm

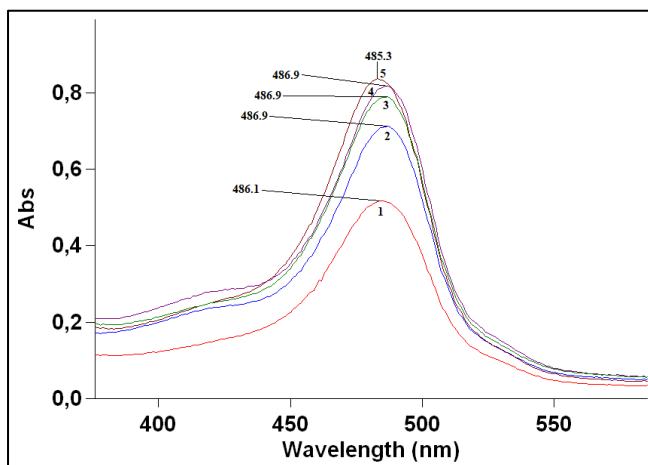


Fig. 2: Absorption spectra of glucose (SS) and extracts by acid hydrolysis reaction. 1--standard, 2-extract (extraction time $t_E=15$ min, 3- $t_E=120$ min, 4- $t_E=30$ min, 5- $t_E=90$ min ($n=3$, $p<0.05$)

The graphic demonstrates the maximum absorbance of glucose standard and extracts of *Lycium barbarum* L. in different extraction times. The maximum absorption of the standard sample of glucose was recorded in the wavelength range of 487 ± 3 nm. The maximum absorption in the spectra obtained after the interaction of extracts with phenol was in the range of 482-489 nm (similar maxima were given by standard samples). The content of the sum of polysaccharides and free sugars in terms of glucose in absolutely dry raw materials as a percentage (X, %) was calculated by the formula (1) and ranged from $21.763\pm 0.847\%$ to $70.384\pm 0.489\%$. The highest carbohydrate content was $70.384\pm 0.489\%$, in optimal conditions for extraction. The TC content was calculated by using the formula (1), which is $X=70.384\pm 0.489\%$.

The sum of fructans and fructose

Fructans are polymers of fructose with β -linked fructose units, which are generally regarded as dietary fiber, macromolecules that resist digestion by human endogenous enzymes because the majority passes through the stomach and small intestine mainly intact. There are three main fructans: inulins, levans (or phleins), and graminans [37]. Inulin is a general term covering all β (2->1) linear fructans, classified as 'indigestible' vegetable carbohydrates, which are dietary fibers that improve bowel movement [38]. Classical methods of quantitative determination determine the total

concentration of fructan based on the equivalents of fructose released after mild acid hydrolysis or enzymatic cleavage. The formation of a deep cherry-red complex containing dehydrated ketose with two resorcinol equivalents first was reported by Selivanov (1887) for the detection of fructose or sucrose (the Selivanov test) or the Resorcinol assay. The reaction is usually carried out under slightly acidic conditions, while the resulting complex demonstrates the highest absorption at 480-540 nm. The spectrophotometric method based on the analysis of resorcinol turned out to be easy to carry out, less expensive, and more accurate. Under the action of hydrochloric acid, one inulin molecule is split into 34-35 fructose molecules and one glucose molecule. Under these conditions, only fructose interacts with resorcinol. Thus, there is a direct relationship between the concentration of inulin and fructose formed from hydrolysis [39, 40]. The resorcinol assay was used for the quantitative determination of fructans, in terms of fructose, in the *Lycium barbarum* L. fruits. The maximum absorption of water extraction from the fruits complexation was at a wavelength of 485 nm. The absorption of water extract was compared with the standard of fructose. The maximum absorption of the standard of fructose with resorcinol coincides with the absorbance of the plant extract. Thus, it can be concluded that the wavelength of 485 nm can be used for the spectrophotometric determination of fructans in *Lycium barbarum* L. fruits in terms of fructose (fig. 3).

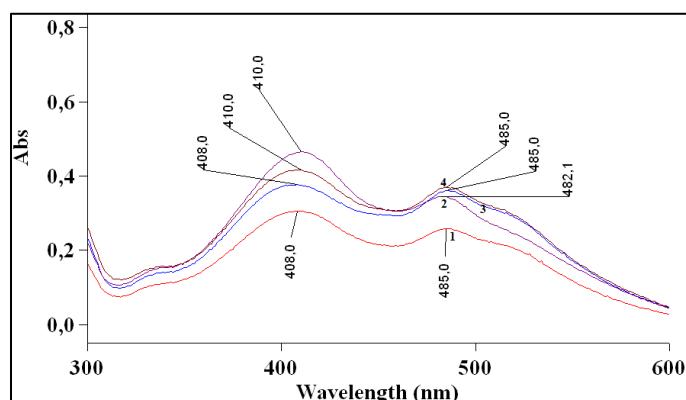


Fig. 3: Absorption spectra of reaction products of standard samples of fructose and extracts from *Lycium barbarum* L. fruits obtained with resorcinol, 1-Standard solution; 2,3,4-Extract solutions

The content of the fructans in terms of fructose in the *Lycium barbarum* L. fruits harvested in Albania, according to our data, lies in the range of 19.90-20.25%. For example, in dried chicory root, the inulin amount equals 15-20% or 35.7-47.6 g/100g in raw plant material. Thus, *Lycium barbarum* L. fruits can be considered a concentrated source of inulin [41]. Chicory has been exploited at industrial levels for inulin-type fructan extraction [42]. The quantity of inulin in *Lycium barbarum* L. is approximately the same as in chicory [43]. In this regard, Albanian *Lycium barbarum* L. fruits can be a promising source of fructans and new medicinal products. The obtained results of quantitative determination of inulin can serve as

convincing arguments for expanding the range of raw materials used for the industrial production of inulin.

Flavonoids and coumarins

Flavonoid and coumarin compounds have been identified and quantified in *Lycium barbarum* L. fruits by high-performance liquid chromatography (RP HPLC-DAD) with diode-array [44]. The identification by RP HPLC-DAD is based on the comparison of the UV absorption spectrum and a time (RT) with corresponding standard compounds: rutin of the flavonoid group and scopoletin of the coumarin group (fig. 4).

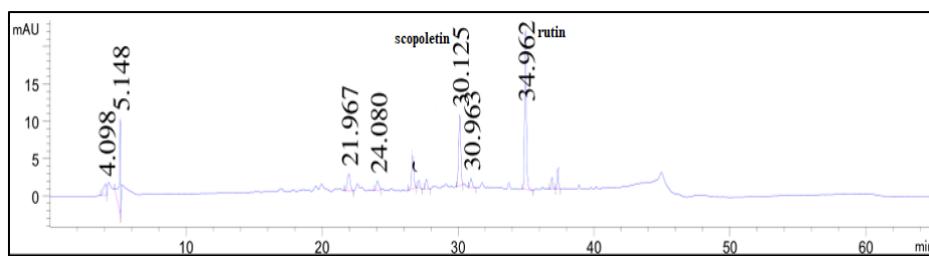


Fig. 4: Chromatogram of *Lycium barbarum* extract (n=5, p<0.05)

Rutin was the main flavonoid compound in *Lycium barbarum* L. fruits with a value of 2.10-5.48 mg/g dry sample (in calculation on rutin). The total yield of rutin depends on the solvent concentration. The maximum amount of rutin extracted from *Lycium barbarum* L. fruits (5.48 ± 0.26 mg/g) is extracted with 80% ethanol, and the minimum yield is when using ethanol with a 96% concentration (2.10 ± 0.09 mg/g). With ethanol, the amount of extracted rutin is also high at 5.48 ± 0.26 mg/g. Then with an increase in ethanol concentration, the amount of extracted rutin decreases to 3 ± 0.15 mg/g, then increases again with ethanol 80%. The maximum amount of scopoletin extracted from *Lycium*

barbarum L. fruits (0.76 ± 0.03 mg/g) is extracted with 70% ethanol, and the minimum yield is when using ethanol with a 96% concentration (0.48 ± 0.02 mg/g). The total yield of scopoletin is almost constant despite the solvent concentrations. The results are shown in (fig. 5).

The qualitative determination of flavonoids and polysaccharides

The qualitative tests for flavonoids and polysaccharides were performed, and all the tests were given positive by formation colored products (table 2) [45].

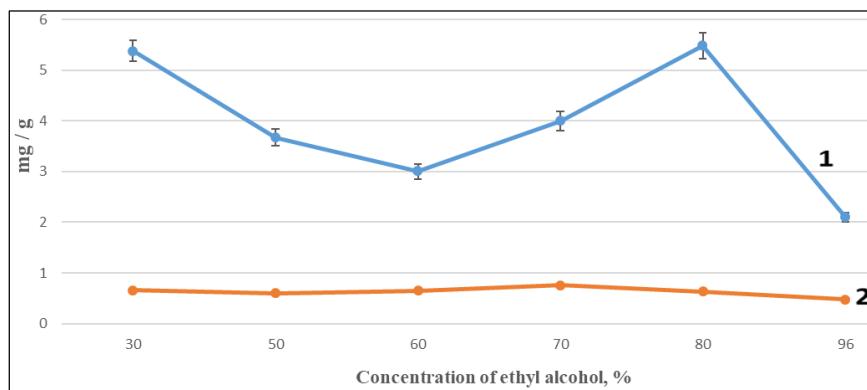


Fig. 5: The content of rutin (1) and scopoletin (2) in *Lycium barbarum* L. fruits (n=3)

Table 2: The results of flavonoids and polysaccharides qualitative determination

Compound	Test	Results	Conclusion
Polysaccharides	Molisch's test	A purple-colored product was formed.	Contains carbohydrates.
	Fehling's test	Appearance of a reddish-brown sediment.	Are present reducing sugars.
	Benedict's Test	The color changes to green when boiling.	The solution should contain sugar.
	Ferric(III) chloride 1%	Brown color	Flavonoids are present (3-OH group)
	Sodium hydroxide 10%	Yellow coloring	Flavones, flavonols, flavonones are present
	Lead Acetate 1%	Form yellow flakes in solution, precipitating	Flavones, chalcones and aurones are present
	Aluminum chloride 2%	Yellow coloring with bright green fluorescence in UV rays.	Flavonoids are present
Flavonoids	Ammonia solution	Yellow coloring, turning into orange or red when heated	Flavones, flavonones, flavonols and flavononols are present
	1 % vanillin solution	Red-crimson coloring	Catechins are present

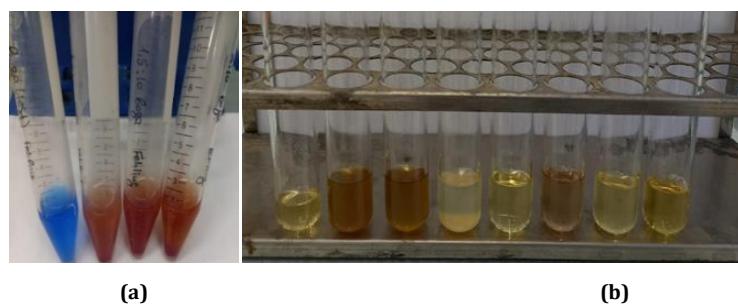


Fig. 5: Chromogenic reactions for (a) polysaccharides/Fehling test, (b) flavonoids

Organic acids

The observed peaks were identified by checking the retention time and the absorption spectra of each organic acid of both *Lycium barbarum* L. fruits and the standards at 210 nm. From the fruits

samples were separated and identified three organic acids: tartaric, malic, and citric acid. There were well-resolved and high peaks at a retention time of 4.35 min and 13.85 min. This peak could be organic acids, but the retention time did not match any of the standards, thus, identification was not feasible (fig. 6).

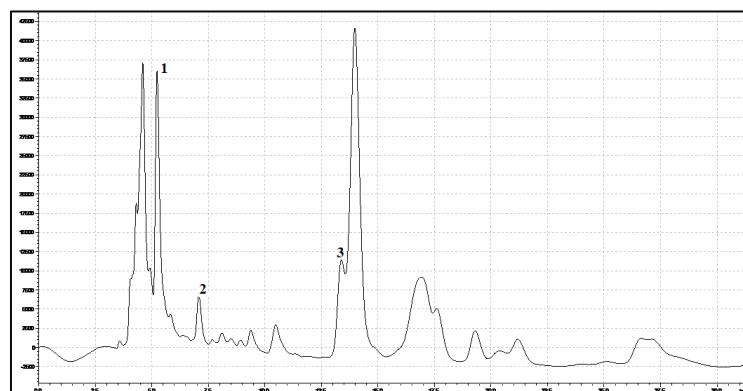


Fig. 6: Chromatogram of *Lycium barbarum* fruits extract at 210 nm. 1-tartaric acid, 2-malic acid, 3-citric acid (n=5, p<0.05)

Also, ascorbic acid was not detected by HPLC. So, this acid was determined using the titrimetric method with 2,6-dichlorophenolindophenol reagent as indicated in the Materials and Methods. The contribution to the daily vitamin C intake from a serving (100 g) of *Lycium barbarum* L. fruit is relatively high, around 244 mg/100g, while recommended dietary intake of vitamin C is 90 mg/day. The amount of ascorbic acid found in this fruit was higher than other data reported in the literature, such as by Donno D. et al. for

Italian *Lycium barbarum* L. fruit (48.94 mg/100g) and Nzeuwa I. BY et al. for chinese *Lycium barbarum* L. (33.15-113.86 mg/100 g) [46]. Tartaric acid (420 mg/100g) was the major organic acid, followed by malic acid (180 mg/100g) and citric acid (390 mg/100g).

Micro-and macroelements

The average concentrations of the macro and essential trace elements in *Lycium barbarum* L. fruits samples are shown in table 3.

Table 3: Macro-and micro-elements in *Lycium barbarum* L. fruits

Element	Nº	m, g	C, µg/ml	X, ppm	XCP, ppm	XCP, mg/100g
Ca	1	15.0001	0.76	5.07	4.82	0.48±0.053
	2	15.0000	0.70	4.67		
	3	15.0000	0.71	4.73		
Fe	1	15.0001	0.04	0.27	0.27	0.03±0.001
	2	15.0000	0.04	0.27		
	3	15.0000	0.04	0.27		
Mg	1	15.0001	1.62	10.83	10.73	1.07±0.021
	2	15.0000	1.60	10.67		
	3	15.0000	1.60	10.70		
Na	1	15.0001	209.211	1394.73	1400.73	140.1±12.928
	2	15.0000	210.610	1404.10		
	3	15.0000	210.511	1403.40		
K	1	15.0001	1014	6759.42	6740.75	674.07±4.099
	2	15.0000	1010	6734.73		
	3	15.0000	1009	6728.09		

The elemental composition was performed in triplicate and the results are the mean of three values±standard deviation

Potassium (K) is the predominant element (6740.75 µg/g), followed by sodium (Na, 1400.73 µg/g.). Regular consumption of this fruit will provide adequate amounts of the macro and essential trace

elements needed for humans. Fruit is a valuable source of minerals, particularly K and Na. Potassium and sodium are electrolytes with a significant role in regulating blood pressure by helping the body

maintain fluid and regulate the body's acid-base balance. [47]. Essential and nonessential element concentration is dependent on the soil characteristics, the physiology of the plant, fertilizers, insecticides, pesticides used in the plantations, and water. Each plant accumulates different levels of elements; some are easily absorbed, and some others are not [48].

CONCLUSION

The fruits of *Lycium barbarum L.* are a promising and rich source of important biologically active substances, the standards for which are in some of the world's pharmacopeias. It is noteworthy to study *Lycium* plants growing in different regions to unify the methods of testing the quality control of plant raw materials. The results obtained in this study are the basis for the unification of the structure of the pharmacopoeial monograph for medicinal plant materials and medicinal plant preparations based on *Lycium barbarum L.*

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AUTHORS CONTRIBUTIONS

POG developed the concept of the study. TB and NNB designed the study and guided and supervised the study. TB and UEV contributed to the interpretation of data and wrote the first draft. TB and RAN were associated with supervising, advising, and positioning the manuscript. All authors read and made corrections to the finalized manuscript before submission.

CONFLICT OF INTERESTS

The authors declare that there is no conflict of interest.

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